# Influence of Organic Acids on Nutrients Availability from Natural Alternative Fertilizers

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FIELD experiment was carried out during the summer A season with sesame and the winter season with wheat crops in a sandy soil at the Ismailia Agric. Res. Station. The work aimed to study the role of organic acids in nutrients availability and productivity of both indicated crops. Urea- formaldehyde, rock phosphate, feldspar and acidulation with different organic acids ((humic, fulvic, tartaric, salicylic and oxalic) and mineral acid sulphuric) at two concentrations 10 and 15% (C1) and (C2) were used. Results revealed that acidulation had positive effect on chemical properties as well as availability of soil N, P and K nutrients compared to the control treatments; soil pH decreased especially with C2 which relatively raised soil EC. Also, there was a significant increase in the availability of N, P and K in soil; again, the superiority was with the second concentration. There was a significant response to acidulation compared to the control treatment when assessing yield of each grains or seeds and straw of both wheat and sesame crops, especially when using the second concentration (C2). An increase in the total content of NPK elements with acidulation treatments compared to the control for both wheat and sesame crops was observed. The results also showed high response to salicylic acid compared to other acids while the oxalic acid being inferior. Finally, urea- form, rock phosphate and feldspar (alternative fertilizer) under acidulation with different acids could be a promising approach to enhance release of available N, P and K and to improve soil chemical properties which will be reflected on the yield along with total content of nutrients.

Keywords: Acidulation, Urea-form, Rock phosphate, Feldspar, Macronutrients availability, Wheat and sesame yield

It is worth mentioning that fertilization became expensive; therefore it is necessary to consider the use of some natural rocks such as feldspar, rock phosphate especially with acid treatment for easing the release of elements contained in these materials.

Mineral fertilizers, particularly those of mineral-N, are important for plant nutrition (Hartman, 1988). The efficiency of N fertilizer can be increased through the use of slow release N forms, which potentially reduce nitrogen leaching losses, particularly in sandy soils extending N availability over the growing season and improving the efficiency of

plant recovery (Wang and Alva, 1996). Also, Hassan *et al.* (2010) showed that ureaformaldehyde (UF) treatments, either as full or half dose, had positive effects on vegetative growth of plants. The authors added that UF treatments increased the available forms of N, P and K in the sandy soil for the four grown crops seedling in comparison with the traditional urea. Furthermore, Seddik *et al.* (2011) reported that application of UF as slow release fertilizer at a rate of (75%N) in presence of compost improved chemical properties and increased yield components of carrot as well as its total content of macronutrients and is economically feasible.

Phosphorus is a key nutrient element for the growth and development of crops (Wang *et al.*, 2011). Phosphorus is provided mainly by soil, however, it can easily be immobilized in soil, and the seasonal application rate of phosphate fertilizer for crops is only 10-25% (Liu *et al.*, 2012).

Application of rock phosphates in arid and semi-arid regions is not common, because of its low availability giving the fact that most soils are alkaline under drought stress, relatively high in pH, and low in organic matter. A number of studies have examined the solution of rock phosphate (RP) in soils and its subsequent effect on soil reactivity such as soil pH, particle size of the RP, along with concentrations of Ca and P in the soil solution (He *et al.*, 2005). On the other hand, calcareous soils are frequently characterized by its low bioavailability of plant nutrients due to high base status with pH variable between 7.5 - 8.5 and the presence of carbonate minerals (Marschner, 1995). The efficiency of P fertilizers in these soils is generally very low because P applied to the soil reacts with Ca forming minerals such as dicalcium phosphate dehydrate and tricalcium phosphate .

Concerning K, its main source for plants growing under natural conditions comes from the weathering of K minerals and organic K- sources such as compost and plant residues. Potassium from feldspar mineral could be solubilized and transformed into available form when incorporated with organic materials such as compost or vinasse (Seddik, 2011).

The increases in K uptake reduce the concentration of K in the soil solution and root surfaces thus induce the release of K held on the external surfaces of soil particles and also the gradual release of K from the interlayer. Therefore, the use of natural minerals, especially feldspar and compost combined with organic residues should be considered as natural alternatives and beneficial cheap sources of K fertilization for sustainable agriculture in sandy soils (Singh *et al.*, 2002).

In many studies, humic and fulvic acids preparations were reported to increase the uptake of mineral elements to promote the root length (Canellas *et al.*, 2002) and to increase the fresh and dry weights of crop plants (Chen *et al.*, 2004 a, b). Due to the positive effect of humic substances on the visible growth of plants, these chemicals have been widely used by the growers instead of other substances. This, however, has led to growers using higher amounts of commercial substances.

Humic acid is considered a suitable application for improving the vegetative growth, yield production and the antioxidant activity of the plants (Yasser *et al.*, 2011). Humic *Egypt. J. Soil Sci.* **56**. No.3 (2016)

acids may improve urea N use efficiency as well as reducing environmental pollution by increasing soil exchangeable  $NH_4^+$  and available  $NO_3^-$  while retaining more N in the soil (Yusuff *et al.*, 2009). Also, humic substances can improve water holding capacity for better drought resistance and reduction in water usage (Russo and Berlyn, 1990). In addition, soil structure is influenced by humic and fulvic acids participating in numerous bridging mechanisms, including water, Al, and Ca bridging (Tan, 2003). Stimulation of root growth is generally more apparent than stimulation of shoot growth (Nardi *et al.*, 1996).

Moreover, humic acid applications can improve the stability of soil structure. Physical and chemical properties of soil such as soil organic carbon, total nitrogen, modulus of rupture and aggregate stability were improved by HA amendment. According to reported results, HA (K-Humate) has potential to be used as an effective management tool for sustainability of the soil environment (Gümü and Seker, 2015)

It is worth mentioning that, citric acid, tartaric acid and oxalic acid of different concentrations application into soil, increase the content of available P. When the concentration of acids were reduced to 1 mmol/L, both the total and maximum release amount of soil P significantly declined, the order being citric acid < malic acid < oxalic acid and acetic acid (Lv *et al.*, 2015). Also, the outhouse added that, the parabolic diffusion equation showed that organic acids can improve the migration rate of phosphorus in the soil. Salicylic acid (SA) application has been reported to enhance plant tolerance to many abiotic stress SA alters some plants physiological functions. These functions may work as a key role in enhancing plants tolerance to salt stress (Srivastava and Dwivedi, 2000).

Therefore, the objective of this study is to evaluate efficiency of using natural sources (urea formaldehyde, rock phosphate and feldspar) as alternative fertilizers through acidulation with organic acids on availability of macronutrient and yield productivity of sesame and wheat.

### **Materials and Methods**

A field experiment was carried out at Ismailia Agric. Res. Station, ARC during the summer season with sesame (*Sesamumindicum* L. var., Giza 32) and the winter sprinkler irrigation system to evaluate the effectiveness of organic acids (tartaric, salicylic, oxalic, humic and fulvic) and (sulphuric acid) application on N, P and K release from urea formaldehyde, rock phosphate and feldspar. Soil chemical properties and nutrient availability as well as nutritional status of sesame and wheat crops were also evaluated. The institute farm is located at 30° 35′41.9″ N Latitude and 32° 16′ 45.8″ E longitude. The soil under study was analyzed according to methods described by Cottenie *et al.* (1982) as shown in Table 1. Relative chemical properties of organic acids are described in Tables 2 and 3. The experiment was designed in a randomized complete block design with three field replications.

WAFAA M.A. SEDEEK et al.

The control (1) treatment received mineral fertilizers (15 % P<sub>2</sub>O<sub>5</sub>) at a rate of 200 Kg fed.<sup>-1</sup> basically before sowing; potassium was added in the form potassium sulfate (48 % K<sub>2</sub>O) at 50 Kg fed.<sup>-1</sup>, nitrogen was added in the form ammonium sulfate (20 % N) rates of 450 and 100 Kg fed.<sup>-1</sup> as recommended for wheat and sesame production. The control (2) treatments received nitrogen fertilizer as urea- form (46 % N) at rate of 200 and 100 Kg fed<sup>-1</sup> as slow release nitrogen fertilizer, phosphorus as rock phosphate (15 % P2O<sub>5</sub>) at rate of 200 Kg fed.<sup>-1</sup> and feldspar (10% K<sub>2</sub>O) at rate of 480 Kg fed<sup>-1</sup>. Then it was mixed with soil surface two weeks before sesame and wheat cultivation. Soil surface acidulation was performed using either 10 (C1) or 15% (C2) concentrations of the studied acids. Solution volume used in acidulation was 400 L /feddan (before sesame and wheat cultivation and then added two doses within season). At maturity, sesame and wheat were harvested after 120 day to evaluate yield components (seeds or grains and straw) and nutrient status. Plant samples were oven dried at 70 °C until constant dry weight, then ground and digested using H<sub>2</sub>SO<sub>4</sub> and H<sub>2</sub>O<sub>2</sub> mixture described by Page et al. (1982). Soil chemical properties of soil along with analyses for natural minerals were evaluated according to Cottenie et al. (1982). Obtained results were subjected to statistical analysis according to Snedecor and Cochran (1982), and the treatments were compared by using the least significant difference (L.S.D) at 0.05 level of probability.

	TABLE 1. Physical and	l chemical pr	operties of the	experimental soil
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Soil characteristics	Value
Particle size distribution %	
Coarse Sand	50.4
Fine Sand	40.4
Silt	3.20
Clay	6.00
Texture class	Sandy
Chemical properties	
CaCO <sub>3</sub> %	1.40
pH(Suspension 1: 2.5)	7.92
EC dS/m (saturated past extract)	0.37
Organic matter %	0.40
Soluble cations and anions (meq L <sup>-1</sup> )	
Ca <sup>++</sup>	0.95
Mg <sup>++</sup>	0.89
Na <sup>+</sup>	1.51
K <sup>+</sup>	0.45
$CO_3$	-
HCO <sub>3</sub> -	1.42
Cl-	1.02
SO <sub>4</sub> -	1.36
Available nutrients (mg kg <sup>-1</sup> )	1
N	66.0
Р	12.0
K	45.6

Egypt. J. Soil Sci. 56, No.3 (2016)

506

Characteristics	Humic acid	Fulvic acid
pН	5.56	1.23
EC (dSm <sup>-1</sup> )	6.52	4.61
O.C %	9.50	4.20
	Total macronutrients %	
Ν	1.29	0.42
Р	0.25	0.15
K	2.00	2.00

ТАВ	LE 2.	Chemical	properties	of appl	ied hum	ic and f	ulvic acids
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Treatments	Conc.	рН	EC	
Humio coid	C1	1.90	2.80	
Humic acid	C2	1.83	3.45	
Eluvia agid	C1	2.00	1.60	
	C2	1.80	2.80	
Tartaric acid	C1	1.84	10.4	
	C2	1.58	10.3	
Salicylic acid	C1	1.84	11.8	
	C2	2.09	19.5	
Oxalic acid	C1	1.80	78.5	
	C2	1.82	98.2	
Sul-husis said	C1	1.20	18.2	
Sulphulle actu	C2	1.81	30.0	

 TABLE 3. pH and EC of applied acids for soil

### WAFAA M.A. SEDEEK et al.

#### **Results and Discussions**

### Effect of acidulation with different acids on soil characteristics Soil pH

Results in Table 4 revealed that application of acidulated to soil treated with urea- form, rock – phosphate and feldspar, decreased soil pH values compared with the control treatments without organic acids. Favorable effects were more encountered with the C2 (15%) of all organic or mineral acids; this trend was true for both seasons although the pH values in the second season (wheat crop) increased than the first season probably due to different seasons and related moisture and temperature. The effect of acidity resulting from applied acids were explained by Welch *et al.* (2002) who found that organic acids contribute to the lowering of solution pH as they dissociate in a pH dependent equilibrium, into their respective anion(s) and proton(s). Organic acids buffer solution pH and should continue to dissociate as protons are consumed by the dissolution reaction. Similarly, microorganisms often export organic acids as anions (Netik *et al.*, 1997).

#### Electric conductivity (EC)

Data in Table 4 indicated that, application of acidulated urea- form, rock – phosphate and feldspar caused increases in the EC values as compared to the control treatments, in both seasons. The observed increase in soil EC might be due to the indirect effect of lowering soil pH, which consequently leads to dissolving nutrients in the soil solution and thus resulting in higher EC. This is similar with the data of Seddik and Osman (2015) who found that, humic and fulvic acids contain more nutrients, different elements and function groups with high molecular weight and carbon contents. Also, organic acids play an important role in improving bioavailability of soil nutrients which cause an increase in the EC of the soil along with mineral acids generally increased EC values as compared to control treatment. This may be due to the presence of hydrogen ion which lowered the soil pH reflected on increasing the rate dissolution of feldspar and increasing the electric conductivity due to improving nutrients availability.

#### Soil available N, P and K

Data in Table 5 show an increase in available N, P and K, at both seasons, under treatments (urea form, rock phosphate and feldspar) as a result of acidification with organic or mineral acids compared to the control treatments (without acidification). This may be due to the effect of organic acid on mineral weathering rates through at least three mechanisms: changing the dissolution *Egypt. J. Soil Sci.* **56**. No.3 (2016)

rate through decreasing solution pH or forming complexes with cations at the mineral surface; and affecting the saturation state of the solution with respect to mineral and affecting the speciation in solution of ions (Song and Huang, 1988).

		Sesame	Wheat	Sesame	Wheat
Treatments	Conc.	pH (suspension 1:2.5)		EC	dSm <sup>-1</sup>
con 1		7.62	7.80	0.57	0.52
con 2		7.75 a	7.90 a	0.61 f	0.50 g
	C1	7.43	7.64 b c d	0.95 b c	0.81 d
Humic	C2	7.31 d	7.52 c d	1.10 a	1.01 a
	Mean	7.37 d	7.58 c d	1.02 a	0.91 a
Fulvic	C1	7.48 b c d	7.60 c d	0.82 c d e	0.78 d e
	C2	7.36 c d	7.47 d	1.00 a	0.93 a b c
	Mean	7.42 c d	7.54 d e	0.91 b	0.85 b
	C1	7.63 a b	7.71 a b c	0.73 e f	0.65 f
Tartaric	C2	7.50 b c d	7.62 b c d	1.00 a b	0.94 a b
	Mean	7.56 b	7.66 b c	0.86 b	0.79 c
	C1	7.51 b c d	7.66 b c d	0.87 b c d	0.85 c d
Salicylic	C2	7.40 b c d	7.52 e	0.94 b c	1.00 a
	Mean	7.45 b c d	7.59 e f	0.91 b	0.92 a
Oxalic	C1	7.60 a b c	7.80 a b	0.83 c d e	0.70 e f
	C2	7.51 b c d	7.71 b c	0.98 a b	0.86 b c d
	Mean	7.56 b c	7.76 b	0.91 b	0.78 c
	C1	7.36 c d	7.52 e	0.71 e f	0.66 e
Sulphuric	C2	7.33 d	7.48 d	0.80 d e	0.72 ef
	Mean	7.34 d	7.50 e	0.75 c	0.69 d

 TABLE 4. Effect of natural alternative fertilizes acidulation on pH and EC values of soil for sesame and wheat at two seasons

Control 1= fertilizer as recommended dose

Control 2= Urea -form + rock- P + K -feldspar

		Sesame season			Wheat season		
Treatments	Conc.	Available (mg Kg <sup>-1</sup> )			Available (mg Kg <sup>-1</sup> )		
		N	Р	K	N	Р	K
control 1*		110	18	50	105	19	62
control 2*		108 g	17 g	44 j	98 j	18 e	56 g
	C1	155 c	20 e f	84 d	150 d e	24 c	90 c d
Humic	C2	164 b	22 c d	100 a	163 b	33 a	104 b
	Mean	160 a	21 c	92 a	157 b	29 a	97 a
	C1	140 de	19 e f	80 e	148 e f	22 d	88 c d g
Fulvic	C2	152 c	20 e f	96 b	156 c	25 c	94 c
	Mean	146 b	19 c	88 b	152 c	24 b	91 b
	C1	120 f	21 d e	60 h	115 i	21 d	73 f
Tartaric	C2	134 e	28 b	75 f	127 h	22 d	87 c d e
	Mean	127 d	24 b	67 d	121 e	21 b	80 d
	C1	148 cd	23 c d	83 d	153 c d	25 c	92 c d
Salicylic	C2	180 a	30 a	95 b	172 a	31 b	111 a
	Mean	164 a	27 a	89 ab	162 a	28 a	102 a
Oxalic	C1	125 f	19 f	55 i	140 g	22 d	82 e
	C2	140 de	22 c	63 g	146 e f	24 c	88 c d e
	Mean	133 c	20 c	59 e	143 d	23 b	85 c d
	C1	141 de	19 f	80 e	137 d e	19 e	86 d e
Sulphuric	C2	155 c	22 cd	86 c	150 f g	22 d	90 c d
	Mean	148 b	21 c	83 c	144 d	21 bc	88 b c

TABLE 5. Effect of natural alternative fertilizers acidulation on availability of macronutrients in sandy soil

Control 1= fertilizer as recommended dose

Control 2= Urea -form + rock- P + K- feldspar

Obtained results also showed that available nutrients for the two seasons were greater with salicylic and humic acids treatment compared to other organic or mineral acids. Ali *et al.* (2015) found that, using organic acids such as humic and salicylic acids play important roles in nutrient soluble and attracting beneficial microorganisms. Also, organic acids improve calcium availability which buffers pH in soil. Oxalic acid treatment was inferior. This was explained by Chen *et al.* (2000) who found the various organic acids can effectively dissolve minerals and chelate metallic cations, the great effect of organic acid on dissolution of rocks and minerals may be attributed to the presence of hydrogen ions and the formation of cation-complexes. Similar results were reported by Lv *et al.* (2015). Finally, values of available N, P and K were superior at high rates of applied acids. Ugolini and Sletten (1991) found that the levels of organic acids were affected by a number of independent variables such as pH redox potential.

### Effect of acidulation on plant

Yield components

Data presented in Table 6 indicted that the yield values of seeds and straw for sesame and grains and straw for wheat crops at both studied seasons were generally, increased significantly by treatments with acidulated urea- form, rock- phosphate and feldspar (alternative fertilizer) compared to control treatment, the second rate of acids being superior. As expected, when compared to control with recommended dose of NPK treatments and control with including those of no acidulation of urea- form, rock- phosphate and feldspar the latter was inferior than advanced.

 TABLE 6. Effect of natural alternative fertilizer acidulation on yield components of sesame and wheat crops

	G	Sesame yi	eld Kg fed <sup>-1</sup>	Wheat yield Kg fed <sup>-1</sup>		
Treatments	Conc.	Seeds	Straw	Grain	Straw	
control 1*		260	1600	1765	2000	
control 2*		252 ј	1550 g	1700 j	1950 i	
	C1	300 j	1600 f	2600 f g h	2610 f g	
Humic	C2	335 e	1720 e	2700 d e f	2730 d e	
	Mean	317 d	1660 d	2650 c d	2670 c d	
	C1	320 f	1700 e	2665 e f g	2700 d e f	
Fulvic	C2	370 d	1820 d	2785 c d e	2800 d	
	Mean	345 c	1760 c	2725 с	2750 с	
Tartaric	C1	372 d	1830 d	2800 c d	2915 c	
	C2	390 c	1900 c	2970 b	3000 c	
	Mean	381 b	1865 b	2885 b	2958 b	
	C1	400 b	1950 b	2900 b c	3110 b	
Salicylic	C2	460 a	2100 a	3210 a	3300 a	
	Mean	430 a	2025 a	3055 a	3205 a	
Oxalic	C1	275 i	1600 f	2415 i	2480 h	
	C2	290 h	1700 e	2500 h i	2620 e f g	
	Mean	282 f	1650 d	2457 e	2550 e	
Sulphuric	C1	290 h	1600 f	2540 g h i	2570 g h	
	C2	320 f	1730 e	2680 d e f	2700 d e f	
	Mean	305 e	1665 d	2610 d	2635 d e	

Control 1= fertilizer as recommended dose

Control 2= Urea -form + rock- P + K- feldspar

### WAFAA M.A. SEDEEK et al.

The results in Fig. 1 showed the effect of acidulation, regardless the concentration, on total yield of sesame and wheat crops which had the same direction of seeds, grains and straw of two crops. The results also revealed that salicylic acid, again, was superior to other acids treatment as the order of acid effectiveness is as follows: salicylic > tartaric > fulvic > humic > sulphuric > oxalic acids. This was explained by Srivastava and Dwivedi (2000) who found the application of salicylic acid (SA) enhanced plant tolerance to many abiotic stresses; the SA favors some plants physiological functions which may work as a key role in enhancing plants growth.

Also, Yasser *et al.* (2011) mentioned that, humic acid is considered a suitable application for improving the vegetative growth, yield production and the antioxidant activity of plants. Also, Yusuff *et al.* (2009) suggested that, humic acid may improve urea nitrogen use efficiency as well as increasing soil exechangeable  $NH_4^+$  and available  $NO_3^-$  while retaining more N in soil.





Egypt. J. Soil Sci. 56, No.3 (2016)

512

### Status of N,P and K in grown plants

The effect of treatments on total contents of N, P and K of seeds and straw for seseame along with the grains and straw for wheat crops are shown in Fig 2. Results indicated that values of total N,P and K total content increased significantly with application of urea form, rock-phosphate and feldspar (alternative fertilizer) under the effect of acidulation compared to control treatments. of course, this may be due to the acidulation which improved the translocation of elements from roots to all parts of plant. Nardi *et al.* (1996) mentioned that the stimulation of root gorwth is generally more apparent than stimulation of shoot growth under conditions of organic acids application. Again, the response to salcivic acid was superuior than other acids; C2 (15%) were most effective.



Fig 2. Effect of natural alternative fertilizer acidulation on total content of seseme and wheat crops

Finally, the addation of urea form , rock – phosphate and feldspar (alternative fertilizer) mixed by other acids ethier organic or inorganic acids to sandy soil may improve the status and release of nutreients espcially P and K. Rengel and Domon (2008) found that the best performance of feldspar plus compost or organic acids could be attributed to differential exudation of organic compounds, which facilitate the release of non- exchangeable potassium (NEK) and considered to be a mechanism for differential K uptake efficiency. Also, according to the theory of phosphate solubilization mechanism (PSM), availability may be a resultant of the process of solubilization due to the production of organic acids which is accompanied by the acidification of the medium (Puente *et al.*, 2004).

### Conclusion

From the above-mentioned results, it could be concluded that the application of NPK from acidulated urea- form, rock phosphate or K- feldspar (alternative fertilizer) treated with acidulation especially the high rate of applied acids (15), may improve the soil chemical properties such as pH along with status of availability of NPK in sandy soil within release from these fertilizer. Also, results revealed that the effect of salicylic acid was more efficient than other organic acids. This has reflected on growth behavior of wheat and sesame crops including either total content of nutrients or total biological yields. Therefore, this work was a simulation in the availability of nutrients affected with exudate roots.

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Egypt. J. Soil Sci. 56, No.3 (2016)

514

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## تأثير الاحماض العضوية على تيسر العناصر الغذائية من الأسمدة الطبيعية البديلة

وفاء محد احمد صديق ، وجيدة زكريا حسن و هناء عطية زين العابدين معهد بحوث الأراضي والمياه والبيئة – مركز البحوث الزراعية – الجيزة حمص.

اجريت تجربة حقلية خلال الموسمين الصيفي و الشتوي على محصولى السمسم والقمح المنزعة فى تربة رملية بمحطة بحوث الأسماعلية – مركز البحوث الزراعية وذلك لدراسة دور الأحماض العضوية فى تيسر العناصر الغذائية وانتاجية محصول كل من السمسم والقمح. تم اسستخدام اليوريا فورمالدهيد ، الصخر الفوسفاتى و معدن الفلسبار المحمضة بالأحماض العضوية المختلفة (هيوميك، فالفيك، طرطريك ، سالسيلك، و اوكساليك،) وحمض معدنى (حمض الكبرتيك) بتركيزات 10 و15٪

اشارت النتائج الى استجابة موجبة لكل من الخواص الكيميائية وتيسر العناصر فى التربة نتيجة اضافة الاحماض مقارنة بمعاملة الكنترول ، انخفاض قيم pH الترية خصوصا عندالمستوى الثانى والارتفاع لقيم EC . كذلك اظهرت النتائج زيادة معنوية لكل من النيتروجين والبوتاسيوم و الفوسفور الميسر فى التربة خصوصا فى المستوى الثانى من تركيزات الاحماض.

اوضحت النتائج استجابة معنوية عند تقيم كل من الحبوب ،البذور والقش لمحصولى القمح والسمسم نتيجة اضافة المعاملات المتأثرة بالاحماض مقارنة بمعاملة الكنترول. اماالمحتوى الكلى للعناصر ( نيتروجين ، فوسفور والبوتاسيوم) فى النبات ، فقد اوضحت النتائج زيادة فى المحتوى الكلى للعناصر فى معاملات التحميض فى كل من السمسم والقمح مقارنة بالكنترول . اظهرت النتائج ايضا استجابة عالية لحمض السلسليك مقارنة بباقى الاحماض الاخرى بينما كان حمض الاوكسليك اقلهم. وفى النهاية قد تكون اضافة و اليوريا فورم و الفلسبار ، الصخر الفوسفاتى ( الاسمدة البديلة ) تحت تأثير الاحماض المختلفة طريقة مبشرة لزيادة انطلاق وتيسر العناصر الغذائية وكذلك تحسين الخواص الكميائية للتربة مما ينعكس على المحصول الكلى بمكوناتة بالاضافة الى المحتوى الكلى من العناصر الغذائية.