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Evaluation of Drainage Water Quality of El Hoks Drain at North Nile Delta, Kafr El-Sheikh Governorate, Egypt

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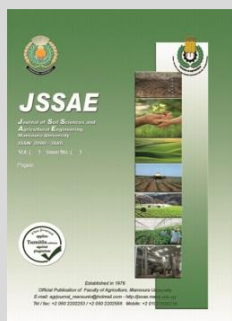


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ABSTRACT

There is a severe shortage in irrigation water supply in North Delta area, especially Kafr El-Sheikh Governorate, Egypt. The shortage in fresh water supply in farmlands located on the tail end of irrigation canals has to be replenished by the available drainage water. So, the current work focused on assessing water quality of El-Hoks Drain for irrigation with respect to its contents of ammonium- N ($\text{NH}_4\text{-N}$), nitrate-N ($\text{NO}_3\text{-N}$), salts, dissolved oxygen (DO), biological oxygen demand (BOD), chemical oxygen demand (COD), boron (B) and some heavy metals. Sixty water samples were collected from ten branch drains' discharge in El-Hoks Drain during winter (2017/018) and summer (2018) seasons. According to FAO (1985) and Egyptian standards (Law 48/1982), results showed that BOD, COD and $\text{NO}_3\text{-N}$ are the main pollutants, which BOD and COD values are classified between bad to slight and moderate, whereas, $\text{NH}_4\text{-N}$ and $\text{NO}_3\text{-N}$ values fall within the normal range for irrigation, but it is in the abnormal range based on the Egyptian standards. DO Values are in the normal range for irrigation according to Egyptian standards. Also, values of heavy metals Cu, Mn, Pb and Cd exceed the allowable limits for irrigation except Ni which falls within the normal range. B values in water samples fall between bad and good. Meanwhile, pH values ranged from 7.37 to 7.95, EC values ranged from 1.16 to 2.71 dSm^{-1} , while SAR values varied from 5.74 to 8.64 which are suitable for irrigation and their values were higher in winter than in summer.

Keywords: BOD, COD, heavy metals, nitrate and water quality.



INTRODUCTION

Water scarcity is one of the common problems in many parts of the world with increasing population and industrial growth (Motoshita *et al.*, 2018). Therefore, drainage water and other low-quality water such as wastewater have to be used for irrigation (Ali *et al.*, 2011). In Egypt, using low-quality water in the agricultural sector at the North Nile Delta is common, especially in Kafr El Sheikh Gov, which is located at the end of irrigation networks, where the farmers receive inadequate fresh water. They have to pump the drainage water directly to irrigate their fields. The local irrigation districts of Kafr El-Sheikh Governorate constructed a number of emergency feeders to feed the end of some irrigation canals which have water supply shortage from the nearby drains (Allam and Negm, 2013 and Gabr, 2018). The main drains in Kafr El Sheikh Gov., are West El-Burullus, Gharbia, El-Khashaah, Tirrah, No. 7, No. 8, No. 9, and El-Hoks, and Brinbal (EMI, 2012; Nassar and Gharib, 2014) and Abukila, (2015).

The long-time reuse of low-quality water such as drainage water and wastewater for irrigation could impair soil functions and cause environmental pollution due to salinization, increase of sodium ions relative to other cations, and accumulation of heavy metals (Kabata-Pendias and Mukherjee, 2007; Abegunrin *et al.*, 2016; Balkhair and Mohammad, 2016; Saliba *et al.*, 2018; Cao *et al.*, 2018; Ganjegunte *et al.*, 2018; Abd-Elwahed, 2018 and Barber *et al.*, 2019). There is a number of water quality guidelines related to irrigation according to FAO (Ayers and Westcot, 1985), but none of them is entirely satisfactory because of the wide variability in the environmental conditions, because they

concerned about the effect of water quality on soil and crops. Therefore, five categories are defined related to problems of water quality in irrigated agriculture: (a) salinity hazards (EC and total dissolved solids, TDS), (b) infiltration and permeability hazards (EC and SAR), (c) specific ion toxicity (SAR, boron and chloride), (d) trace element toxicity, and (e) miscellaneous impacts on sensitive crops (pH, NO_3 and CaCO_3). In the Egyptian standards (Law 48/1982), chemical water quality parameters (pH, TDS, dissolved oxygen, DO, biochemical oxygen demand, BOD, nitrates, $\text{NO}_3\text{-N}$, phosphate and heavy metals) are used to classify the suitability of agricultural drainage water (ADW) for irrigation. The expected water quality level may be different depending on the type of irrigation (Bouwer and Idelovitch, 1987).

There are several studies that have discussed the assessment of drainage water quality in Kafr El Sheikh Governorate. El-Shahawy and Ragab (2005) determined the concentration of some heavy metals in the El-Gharbia main drain and they found that the concentration of Pb, Cd, Ni and B were less than the permissible limits for irrigation. Taha *et al.* (2012) found that the water quality of El Gharbia, No.11 and No.10 drains in Kafr EL-Shiekh for irrigation purposes was classified as none and slight to moderate degree of impairment according to FAO (1985), but $\text{NO}_3\text{-N}$ concentrations in some locations on EL-Gharbia main drain during September and December were over the permissible limit. The quality evaluation of drainage water of Nashart Drain indicated that its quality doesn't meet the local standards for the direct use for irrigation based on the Egyptian standards (Allam and Negm 2013). The contamination factor of water in most locations on

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El-Hoks No. 7 and No. 8 drains indicated that their water quality had moderate suitability for irrigation (El-Amier *et al.*, 2015). El-Batrawy *et al.* (2018) showed that the average concentrations of Mn, Zn, Fe, Ni, Cu, Pb in Burullus Lake water at the front of Hoksia Drain were 2.58, 12.44, 14.94, 2.94, 0.78 and N.D. µg/L, respectively during winter and 0.10, 2.68, 13.96, 10.26, 1.04 and 8.10 µg/L, respectively during summer 2014. Jahin *et al.* (2020) showed that the water of some canals and drains in Kafr El-Sheikh Gov., North Nile Delta, Egypt was classified as low to good water quality. Therefore, this study aimed to assess water quality of El-Hoks Drain, North Nile Delta, Kafr El-Sheik Gov., Egypt for irrigation.

MATERIALS AND METHODS

Study area: Ten branch drains discharge in El-Hoks Drain at Kafr El-Sheik Governorate, North Nile Delta, Egypt were chosen (Fig.1) to assess the quality of their water for irrigation. The length of El-Hoks Drain is about 18.3 km, which received drainage water as well as sewage water from the adjacent villages.

Water sample collection: Sixty water samples were collected from the outlets of the ten branch drains (Table 1) during winter 2017/18 (November, January and March) and summer 2018 (May, July and September).

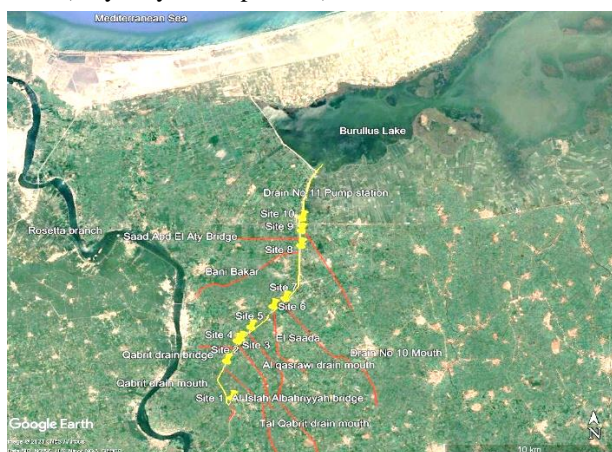


Fig. 1. Study area and sampling locations of drainage water.

Table 1. Locations of water sampling under the present study

Location	Name of the drain	Latitude	Longitude
Site 1	Al-Islah Albahriyyah bridge	31° 14' 46.40"N	30° 33' 4.23"E
Site 2	Qabrit drain mouth	31° 16' 4.92"N	30° 32' 51.76"E
Site 3	Qabrit drain bridge	31° 16' 47.07"N	30° 33' 17.39"E
Site 4	Al qasrawi drain mouth	31° 16' 52.94"N	30° 33' 22.29"E
Site 5	Tal Qabrit drain mouth	31° 17' 16.94"N	30° 33' 45.43"E
Site 6	El Saada	31° 18' 5.89"N	30° 34' 42.50"E
Site 7	No 10 Mouth	31° 18' 18.01"N	30° 35' 0.91"E
Site 8	Bani Bakar	31° 20' 10.32"N	30° 35' 37.72"E
Site 9	Saad Abd El Aty Bridge	31° 20' 40.22"N	30° 35' 38.22"E
Site 10	No 11 Pump station	31° 21' 9.93"N	30° 35' 39.44"E

Water samples analysis: The water samples were drawn at a depth of 0.5 meter below the water surface and stored in acid-washed high dens polypropylene vials (1 L). Water samples were stored in another set of 0.5 L polypropylene vials washed with 50% HNO₃ and double deionized water and acidified by 5 ml HNO₃ and transported in iceboxes to the lab for analysis within 24 h for trace element analysis.

The chemical analysis of water was carried out as follows:

- EC (dSm⁻¹), pH, soluble cations and anions (meq/L) were determined according to Jackson (1973).

- Sodium adsorption ratio (SAR) was calculated according to Richards (1954).
- NH₄-N was measured using Kjeldahl methods and NO₃-N concentration was determined calorimetrically using a spectrophotometer (Jackson, 1973).
- Boron was determined calorimetrically using a spectrophotometer with Azomethine-H method according to Bingham (1982).
- Heavy metals (Cu, Mn, Ni, Pb and Cd) were determined using atomic absorption spectrophotometry (GBC Σ Aventavir 1.3) as described in standard methods-302 A (APHA, 1985).
- Dissolved oxygen (DO), biological oxygen demand (BOD) and chemical oxygen demand (COD) were measured according to Ademoroti (1996).

Quality indices: Water quality classes according to USDA (1954) were; C₁, C₂, C₃ and C₄, representing low, medium, high and very high salinity, respectively and S₁, S₂, S₃ and S₄ representing low, medium, high and very high sodicity, respectively.

Sodium: Calcium activity ratio (SCAR) was calculated as follow:

$$SCAR = Na^2 \text{ meq/L} \div \sqrt{Ca} \text{ meq/L}$$

Residual sodium carbonate (RSC): was calculated as follow:

$$RSC = (CO_3^{2-} + HCO_3^-) - (Ca^{2+} + Mg^{2+})$$

Soluble sodium percentage (SSP) was calculated according the following:

$$SSP = \frac{Na}{\sum Cations} * 100$$

The permeability index PI was calculated according to Doneen (1964):

$$PI = \frac{Na + \sqrt{HCO_3}}{Na + Ca + Mg} \times 10$$

The other measurements were made by Irrigation Water Assessment Model (IWA-Mod), which is a computer model that can be used to assess the quality of irrigation water through some indicators such as salinity, SAR, Adj. SAR, SSP, RSC, SCAR, MAR and PI.

Statistical analysis: The data of water analysis for all variables were statistically analyzed for significant differences among seasons and locations using a one-way ANOVA. Also, the relationships between the studied variables with that for Nile River water were calculated using the Pearson correlation index.

RESULTS AND DISCUSSION

Water salinity (EC) and alkalinity (SAR):

The salinity and alkalinity of the drainage water might be affected by some factors such as land use, crop pattern, soil management, drain location, sampling location and drainage efficiency. Results in Table (2) indicate that the average values of EC in drainage water in both seasons ranged from 1.16 to 2.71 dSm⁻¹ and SAR varied from 5.74 to 8.64. The salinity and alkalinity values of water in the studied drains are mostly classified as slight to moderate according to the international guideline of FAO/RNEA (1993) with EC_w limits of ≤ 0.7 dS/m (slight) and 0.7-3.0 dS/m (moderate), while SAR limits are ≤ 3.0 (slight) and 3.0-9.0 (moderate). The increases in salinity and alkalinity in the drainage water of this area are mainly ascribed to the inflow of saline water due to sea water intrusion, so salinity and alkalinity values of drainage water increased towards the sea. Also, salinity and alkalinity of water of these drains were higher in winter (November, January and March) than in summer (May, July and September), may due to the

high amounts of irrigation water with rice crops which led to dilution of drainage waters. Data also revealed that salinity and alkalinity of water in drain No 11 pump station in January recorded the highest values compared to other drains. The EC values ($\leq 1 \text{ dSm}^{-1}$) were in the normal range for irrigation water according to FAO (1985), but they exceed the allowable limits of EC according to the Egyptian standards (Law 48/1982). Therefore, using water of these resources for irrigation required special management such as the selection of salt tolerant crops and effective drainage systems for salinity control.

Soil reaction (pH):

Water samples showed alkaline pH values (7.37-7.95) and its values were increased in July and September. The highest pH values were obtained in the outlet of Tal Qabrit Drain in July. The pH values fall within the normal range for irrigation according to the guidelines given by FAO (1985) and the Egyptian standards (Law 48/1982). These results may be due to the high influx of CO_3^{2-} and HCO_3^- to the agricultural drainage water (El-Gamal, 2017). CO_3^{2-} and HCO_3^- anions change the Ca^{2+} and Mg^{2+} to be insoluble salts leaving Na^+ as the predominant ion in water (Mandal *et al.*, 2019).

Table 2. Average pH, EC and SAR values in drainage water in different locations

Location	pH		EC (dSm^{-1})		SAR	
	Winter	Summer	Winter	Summer	Winter	Summer
	Al-Islah Albahriyyah bridge	7.38	7.76	1.77	1.64	6.96
Qabrit drain outlet	7.43	7.77	1.81	1.50	7.05	6.59
Qabrit drain bridge	7.42	7.74	2.22	1.51	7.80	6.61
Al qasrawi drain outlet	7.45	7.84	2.13	1.27	7.66	6.04
Tal Qabrit drain outlet	7.54	7.87	2.09	1.16	7.58	5.74
El Saada outlet	7.43	7.86	1.78	1.40	6.99	6.39
Drain No 10 outlet	7.57	7.86	2.34	1.53	8.01	6.66
Bani Bakar outlet	7.42	7.95	2.40	1.50	8.12	6.58
Saad Abd El Aty Bridge	7.37	7.83	2.18	1.47	7.74	6.48
Drain No 11 Pump station	7.45	7.89	2.71	1.60	8.64	6.77
C.V.(standard deviation)	0.12	0.10	0.33	0.19	0.58	0.45

Organic matter indicators (DO, BOD and COD):

Dissolved oxygen (DO) is one of the important parameters for water quality assessment. The values of DO in drainage water of the investigated drains ranged from 10 to 13 mgL^{-1} in both seasons (Table 3). Also, the data revealed that the highest DO value was recorded in the Tal Qabrit Drain outlet in September, while the lowest value was obtained in the El Saada Outlet in January. However, DO values fall within the normal range for irrigation based on the guidelines given by the Egyptian standards (Law 48/1982).

Biochemical oxygen demand (BOD) is also an important pollution indicator that reflects the load of organic waste in water. The average values of BOD in drainage water of the investigated drains ranged from 18.33 to 32.67 mgL^{-1} for both seasons, and the higher values were recorded at the Al Qasrawi Drain outlet in the winter season. The majority of BOD values recorded in the winter season were higher than that in the summer season. These results are similar to that observed by Safaa *et al.* (2012), who recorded that BOD varied from 7 to 120 mgL^{-1} in water of some drains in the Nile Delta.

Chemical oxygen demand (COD) concentration gives a reliable parameter for judging the extent of organic pollution in water. COD is the oxygen required for chemical oxidation

of organic matter. The results indicated that the average values of COD in drainage water of the studied drains ranged from 46 to 63 mgL^{-1} for both seasons, and its values in the winter season were higher than that in the summer season. The values of BOD and COD in water of the studied drains were classified as slight to moderate and bad according to the guidelines given by FAO (1985) and the Egyptian standards (Law 48/1982). The higher BOD and COD values reflect higher load of organic matter in water due to the increased level of pollution from the adjacent drains, which adversely affects the water quality. This indicates that a high percentage of organic matter in water samples might be resistant to microbial degradation (Badr *et al.*, 2013).

Table 3. Average of DO, BOD and COD in water of different locations

Location	DO		BOD		COD	
	Winter	Summer	Winter	Summer	Winter	Summer
	Al-Islah Albahriyyah bridge	11.33	11.67	31.67	23.33	53.67
Qabrit drain outlet	13.00	11.33	30.33	21.67	55.67	46.00
Qabrit drain bridge	11.67	11.33	29.33	26.00	49.00	50.33
Al qasrawi drain outlet	12.67	11.33	32.67	27.33	57.33	55.00
Tal Qabrit drain outlet	11.67	12.33	32.33	29.67	58.00	55.67
El Saada outlet	10.00	10.67	28.33	29.00	61.67	49.00
Drain No 10 outlet	12.00	11.00	28.67	30.33	58.67	50.00
Bani Bakar outlet	11.67	11.67	28.33	32.67	63.00	49.33
Saad Abd El Aty Bridge	11.67	11.67	26.33	26.33	58.67	48.33
Drain No 11 Pump station	12.00	11.00	26.00	18.33	51.67	46.67
C.V.(standard deviation)	0.80	0.47	2.34	4.35	4.33	3.35

Boron concentrations:

Data in Table (4) showed that the average values of boron concentrations in drainage water in drains under study for both seasons varied from 0.50 to 0.87 mgL^{-1} , and the highest values in both seasons were recorded in water of Drain No 11 Pump Station. The B concentration values were evaluated as good to bad according to guidelines given by FAO (1985) and the Egyptian standards (Law 48/1982), so the water can be used for irrigating the tolerant and semi-tolerant crops.

Nitrate and ammonium forms:

Nitrogen in drainage water is mainly in the form of nitrate ($\text{NO}_3\text{-N}$), but may also be in the form of ammonium ($\text{NH}_4\text{-N}$), although $\text{NH}_4\text{-N}$ is readily adsorbed by the colloids and rapidly oxidized into $\text{NO}_3\text{-N}$. Also, nitrate is the final stable form of oxidation/decaying of organic matter from domestic, industrial and agricultural sources. Data in Table (4) shows that the average values of $\text{NH}_4\text{-N}$ concentration in water of the studied drains for both seasons ranged from 1.23 to 3.60 mgL^{-1} , while the $\text{NO}_3\text{-N}$ concentration varied from 2.67 to 18.00 mgL^{-1} . These results are similar to Antar *et al.* (2012) who reported that the concentration of $\text{NH}_4\text{-N}$ is less than the $\text{NO}_3\text{-N}$ in drainage water. The concentration of $\text{NO}_3\text{-N}$ in drainage water increased towards the sea. Also, concentration of $\text{NO}_3\text{-N}$ was higher in the summer months than in the winter months and varied from one drain to another and from one month to another. This may be ascribed to several factors including soil properties, amount of irrigation water, temperature, drainage system and forms of applied fertilizers (Dinnes *et al.*, 2002). In general, the majority of $\text{NO}_3\text{-N}$ concentrations in drains exceed the maximum contaminant level for drinking water (10 mgL^{-1}) according

to U. S. Environmental Protection Agency (1991). Moreover, NH₄-N concentrations were higher in winter months than that in summer months, may be due to lower amounts of water discharged to the drains during winter. Also, NH₄-N is mostly produced by decomposition of organic matter and by hydrolysis of urea (El-Sheltawy *et al.*, 2007). Generally, NH₄-N values fall within the normal range for irrigation according to the evaluation given by FAO (1985), but it is in the abnormal range based on the Egyptian standards (Law 48/1982). On the contrary, NO₃-N values may not fall in the normal range for irrigation according to FAO (1985), but they fall within the normal range of the Egyptian Standards (Law 48/1982). Therefore, with using of drainage water of these drains with these concentrations of NH₄-N and NO₃-N, the application rates of N-fertilizers can be decreased.

Table 4. B, NH₄-N and NO₃-N concentrations in drainage water in different locations

Site	B		NH ₄ -N		NO ₃ -N	
	Winter	Summer	Winter	Summer	Winter	Summer
Al-Islah Albahriyah bridge	0.70	0.50	2.63	2.33	6.67	4.33
Qabrit drain outlet	0.73	0.61	2.93	2.60	4.00	2.67
Qabrit drain bridge	0.78	0.63	3.60	2.20	8.00	11.33
Al qasrawi drain outlet	0.75	0.67	2.57	2.87	7.33	14.00
Tal Qabrit drain outlet	0.79	0.68	3.30	2.83	9.33	15.33
El Saada outlet	0.79	0.61	3.27	2.97	10.33	17.00
Drain No 10 outlet	0.73	0.61	3.27	2.07	12.33	16.33
Bani Bakar outlet	0.73	0.66	3.20	1.70	11.00	16.67
Saad Abd El Aty Bridge	0.78	0.76	3.20	1.51	13.00	15.33
Drain No 11 Pump station	0.87	0.86	2.43	1.23	10.67	18.00
C.V (standard deviation)	0.05	0.10	0.38	0.60	2.77	5.40

Heavy Metals:

The concentrations of heavy metals (Ni, Mn, Cu, Cd and Pb) in water of the studied drains varied from one metal to another and from one drain to another (Fig 2). Heavy metals concentrations (mgL⁻¹) of water in all drains and months ranged from 0.009 to 0.17 for Ni, 0.12 to 0.52 for Mn, 0.20 to 0.41 for Cu, 0.03 to 0.09 for Cd and 0.036 to 0.092 for Pb. These results indicated that Ni concentration in all drains fall within the normal range for irrigation (0.2 mg/l) according to FAO (1985) and the Egyptian standards (Law 48/1982), while the concentrations of Mn, Cu, Cd and Pb exceed the allowable limits (0.2, 0.2, 0.01 and 0.05 mgL⁻¹, respectively). The results showed lower levels of Cu and Pb in drainage water in the summer months in different locations than in winter months might be due to the high amounts of irrigation water with rice fields which led to dilution of drainage waters. While, the high level of Mn in summer may due to the water current speedily in summer faster than in winter and the rate of evaporation was relatively high. Similar findings were obtained by Antar *et al.* (2012) and El-Batrawy *et al.* (2018). Also, the data showed that Cd concentrations in drainage water recorded approximately the same values in both summer and winter. Generally, drainage waters were highly contaminated by heavy metals in both winter and summer seasons at all sampling locations. Therefore, the use of these waters without good treatment lead to increases in the concentrations of heavy metals in soils and plants and consequently enter the food chain causing dangerous complications to man and other biota. These toxic metals may

cause kidney and liver failure, anemia and cancer in addition to chromosomal aberrations (El-Sanafawy *et al.*, 2010).

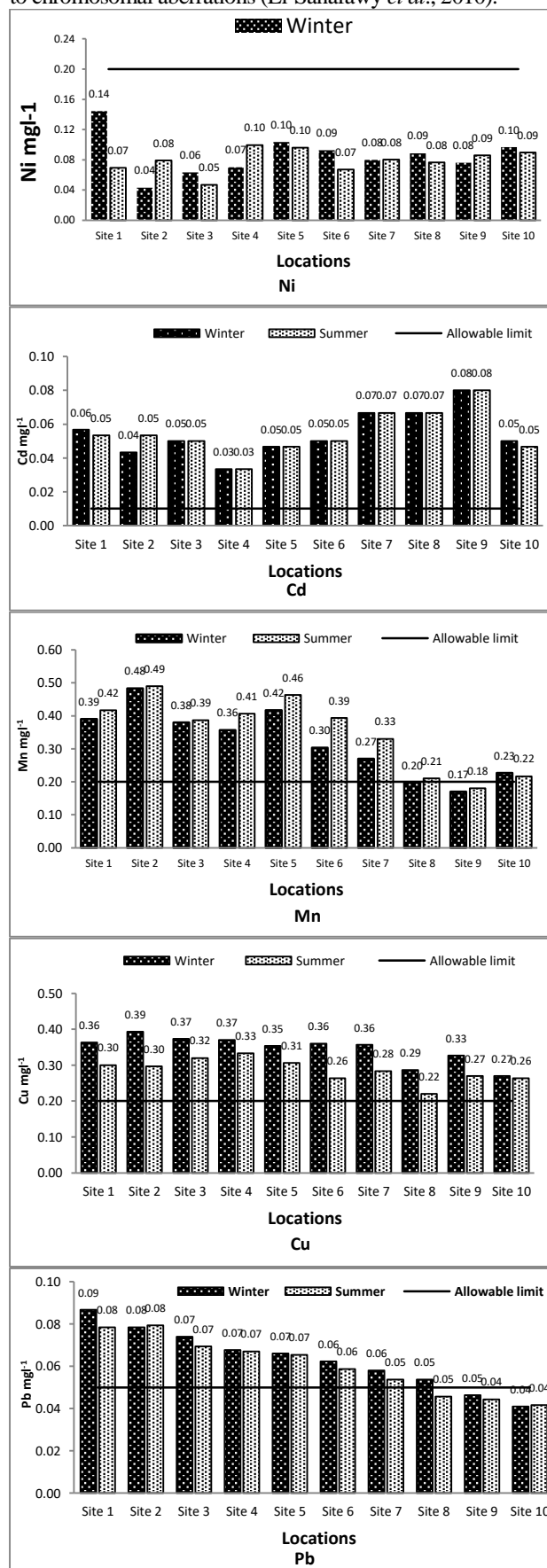


Fig. 2. heavy metals concentration in different sites along el hoksa drain

Water quality Assessment by IWA-Mod:

According to the diagram of USDL (1954), the studied water samples in all sites (Table 5 and Fig 3) are in class C₃S₂, except the sites No. 7, 8 and 9 and 10 in winter are in class C₄S₂. According to this classification, the water in the C₃S₂ class is high saline and medium alkalinity. Therefore, this water is considered slightly dangerous for irrigation purposes and can be used with adequate drainage, special management for salinity control and salt tolerant plants must be selected. On the other hand, C₄S₂ class indicates that the water is very high in saline with medium alkalinity. This water is not suitable for irrigation under ordinary conditions but may be used occasionally under very special circumstances, such as permeable soils, adequate drainage, high salt tolerant crops and excess irrigation water to provide considerable leaching of salts. Sodium ions in this water are present in appreciable high level and can cause hazardous effects in fine textured soils having high cation exchange capacity, especially under low leaching without gypsum application. So, this water can be used in course textured or organic soil with good permeability. The reductions in water infiltration can occur when irrigation water contains high sodium relative to the calcium and magnesium and when the residual sodium carbonate (RSC) was appreciably high in drainage water; they cause an appreciable sodicity hazard. The concept of RSC appears to relate better to sodicity problems in the field (FAO, 1988). The soil permeability is affected by long term use of irrigation water with high levels of sodium relative to calcium and magnesium. Doneen (1964) gave a criterion for assessing the suitability of drainage water for irrigation based on the permeability index classified as class 1 (>75%), class 2 (25-75 %) and class 3 (<25 %). Class 1 and 2 water is categorized as good for irrigation with a 75 % or more permeability index. Class 3 water is unsuitable with a 25 % permeability index as shown in Tables 5.

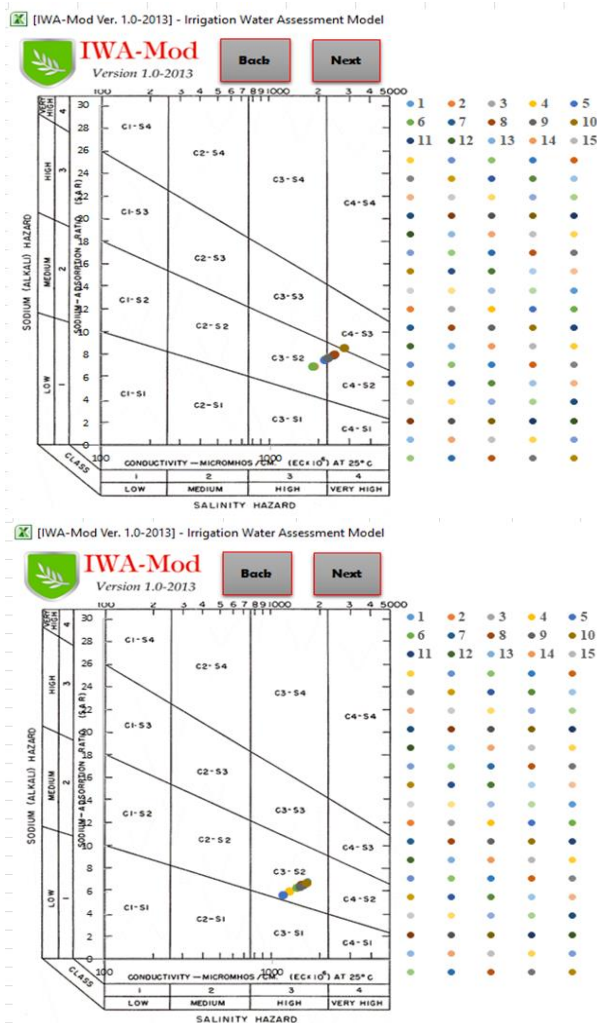


Fig. 3. Classification of drainage water in winter (a) and summer (b) seasons on USDA Diagram (1954)

Table 5. Assessment of drainage water quality

No.	SSP	SAR	Adj. SAR	RSC	SCAR	PI	MAR	Salinity hazard	Alkalinity hazard	USSL Index	USSL class
Winter months											
Site 1	65.33	6.97	11.59	-3.56	5.94	75.77	31.27	C3	S2	C3S2	Acceptable
Site 2	65.79	7.05	11.69	-3.75	6.02	75.48	31.26	C3	S2	C3S2	Acceptable
Site 3	65.83	7.80	13.33	-5.06	6.65	73.98	31.27	C3	S2	C3S2	Acceptable
Site 4	65.77	7.66	12.77	-4.92	6.53	74.04	31.23	C3	S2	C3S2	Acceptable
Site 5	65.91	7.58	12.58	-4.8	6.46	74.14	31.24	C3	S2	C3S2	Acceptable
Site 6	65.59	6.99	10.34	-4.34	5.96	74.09	31.23	C3	S2	C3S2	Acceptable
Site 7	65.77	8.00	13.46	-5.63	6.83	73.34	31.24	C4	S2	C4S2	Poor
Site 8	65.83	8.12	14.13	-5.58	6.92	73.59	31.25	C4	S2	C4S2	Poor
Site 9	65.8	7.73	13.2	-4.93	6.6	74.11	31.23	C4	S2	C4S2	Poor
Site 10	66.21	8.64	15.3	-6.57	7.37	72.91	31.26	C4	S3	C4S3	Very poor
Summer months											
Site 1	65.59	6.93	11.83	-3.44	5.8	75.84	28.57	C3	S2	C3S2	Acceptable
Site 2	65.32	6.59	10.57	-3.35	5.52	75.7	28.57	C3	S2	C3S2	Acceptable
Site 3	65.19	6.61	10.53	-3.44	5.53	75.5	28.54	C3	S2	C3S2	Acceptable
Site 4	64.84	6.03	9.68	-2.35	5.05	77.69	28.54	C3	S2	C3S2	Acceptable
Site 5	64.46	5.74	9.2	-1.86	4.8	78.95	28.57	C3	S2	C3S2	Acceptable
Site 6	65.02	6.39	10.4	-2.8	5.35	76.84	28.57	C3	S2	C3S2	Acceptable
Site 7	64.71	6.65	10.7	-3.46	5.56	75.5	28.54	C3	S2	C3S2	Acceptable
Site 8	64.75	6.59	10.4	-3.45	5.51	75.45	28.57	C3	S2	C3S2	Acceptable
Site 9	64.98	6.47	10.17	-3.37	5.42	75.45	28.54	C3	S2	C3S2	Acceptable
Site 10	65.08	6.77	10.97	-3.7	5.67	75.06	28.57	C3	S2	C3S2	Acceptable

CONCLUSION

Water is fundamental in agriculture, different source of pollution such as sewage and industrial wastewater which discharge onto the drains. So, the present study illustrates the water quality of El-Hoksa Drain. The water in this drain has very low quality, which in turn may cause hazards to soil and grown crops. It could be concluded that El-Hoks Drain may be used for irrigation purposes under controlled precautions with good soil management e.g, good tillage, deep plowing, land leveling, applying soil and water amendments, and finally suitable cropping system.

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تقييم جودة مياه الصرف لمصرف الهوكس شمال دلتا النيل، محافظة كفر الشيخ ، مصر

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نظرا للنقص الحاد في كميات مياه الري بمنطقة شمال الدلتا ، خاصة محافظة كفر الشيخ بمصر. لذا يجب تعويض النقص في كميات المياه العذبة للأراضي الزراعية الواقعة على في نهايات الترع. ولذلك، تم تقييم جودة مياه مصرف الهوكس للري فيما يتعلق بمحتوياته من الأمونيوم- N (NH₄-N) ، النترات- N (NO₃-N) ، الأملاح ، الأكسجين الذائب (DO) ، الأكسجين الحيوي الممتص (BOD) ، الأكسجين الكيميائي المستهلك (COD) ، والبورون (B) وبعض العناصر الثقيلة. حيث تم جمع ستين عينة مياه من عشرة مصارف فرعية لمصرف الهوكس خلال فصل الشتاء (2018/2017) وموسم الصيف (2018).- ووفقا لمنظمة الأغذية والزراعة (1985) والمعايير المصرية (قانون 1982/48) أظهرت النتائج أن الأكسجين الحيوي الممتص و الأكسجين الكيميائي المستهلك و النترات هي الملوثات الرئيسية ، حيث تم تصنيف قيم الأكسجين الحيوي الممتص و الأكسجين الكيميائي المستهلك بين سيئة إلى طفيفة ومتوسطة كما وجد أن قيم الامونيا و النترات تقع ضمن الحدود المسموح بها للري وفقاً لمنظمة الأغذية والزراعة (1985) ، لكنها تقع في خارج الحدود المسموح بها بناءً على المعايير المصرية (قانون 1982/48). إن قيم الأكسجين الذائب في النطاق الطبيعي للري وفقاً للمعايير المصرية (القانون 1982/48). أيضا ، تجاوزت العناصر الثقيلة (النحاس ، المنجنيز ، الرصاص والكاديوم) الحدود المسموح بها للري وفقاً لمنظمة الأغذية والزراعة (1985) والمعايير المصرية (قانون 1982/48) ، باستثناء النيكل الذي يقع ضمن المعدل الطبيعي (0.2 mg / لتر). كما وجد ان تركيز البورون في عينات المياه بين السيئ والجيد. بينما تراوحت قيم الأس الهيدروجيني من 7.37 إلى 7.95 ، وتراوحت قيم EC من 1.16 إلى 2.71 dSm⁻¹ ، بينما تراوحت قيم SAR من 5.74 إلى 8.64 وهي مناسبة للري وكانت قيمها أعلى في فصل الشتاء منها في الصيف.

الكلمات الدالة: الأكسجين الحيوي الممتص والاكسجين الكيميائي المستهلك، المعادن الثقيلة، النترات وجودة المياه.