Physico-chemical Study of Surface Seawater in The Northwestern Gulf of Suez

M.M. Emara, N.A. Farid^{*}, E. A. El-Sabagh^{**}, O.E. Ahmed^{**#} and E.M. Kamal^{**}

Chemistry Department, Faculty of Science (Boys), Al-Azhar University, *Egyptian Petroleum Research Institute and **Chemistry Department, Faculty of Science (Girls), Al-Azhar University, Cairo, Egypt.

S UEZ Canal is a narrow artificial water way in Egypt that joins the Mediterranean Sea with the Red Sea. The Suez Canal passes through an area with a considerable agricultual and industrial activities, so that the chemical and petrochemical pollution from oil transporting process can be effective. Eight surface seawater samples were collected from different sites extended from Suez Harbor up to Fertilizer factory, and both of Al-Osra and Al-Melaha beaches along Suez Gulf. The objective of this study is the monitoring and assessing the contamination of water by chemical and petrochemical pollutions. This study includes physical and chemical parameters of water such as PH values, conductivity, salinity, T.D.S, T.S.S, T.S, D.O., BOD, COD, Cl., SO₄, NO₃ and total alkalinity. Seasonal qualitative and quantitative studies have been carried out on the studied area. The concentration is relatively higher than recommended in the regulations of the Egyptian low environment No.4/1994. Results obtained from the present study should be considered when implementing a strategy for the protection and management of the Suez Gulf. Because the Suez Gulf is used in many areas as a recipient of waste water from different sources this study recommended that pollution control for wastewater disposing into it should be improved. More detailed studies are required to assess the impact of hydro- development projects and evaluate their effect on the studied area and monitoring should be conducted at least twice per year.

Keywords: Physico-chemical properties, Seawater analysis, Suez Gulf, Nutrient salts, Total alkalinity and Ion concentration.

The Suez Canal is located in the northeast of Egypt . Thee geographical position of the Suez Canal makes it the shortest route between east and west that leads to increase the passage of many tender convoys and ships through it (Fig. 1). It extends from Port Tewfik in South with overall length 162 kms and 22.5 m water depth, the Suez canal passes through an area of considerable agricultural, industrial and tourist activity. So the relation between human activities and environmental changes in Suez Canal require knowledge of trends in water quality and understanding of the management of aquatic habitats. The significance of observed contamination level must be measured and special

_

^{*}Corresponding author: dr.omaymamosa@yahoo.com

attention to a number of substance like synthetic biocides' radioactive materials, heavy metals and oil residues⁽¹⁾ must be paid. These substances can be sensitive for monitoring contaminants in aquatic environments. Therefore it is easy to dispose waste by dumping it into a river or lake and canals. In large or small amounts, dumped intentionally or accidentally, it may be carried away by the current, but will never disappear. It will reappear downstream, sometimes in changed form, or just diluted. Freshwater bodies have a great ability to break down some waste materials. This overload that results, called pollution, eventually puts the ecosystem out of balance. Sometimes nature itself can produce these imbalances. Most often our waterways are being polluted by municipal, agricultural and industrial wastes, including many toxic synthetic chemicals, which cannot be broken down at all by natural processes. Even in tiny amounts, some of these substances can cause a serious problem. Also, chemical pollution is an other source from factories and agricultural drainage which may be discharged into the canal, Chemicals can cause health problem to seafood present in seawater and their concentration can be increased by man's activities⁽²⁾. Water samples were collected from the Gulf of Suez during July, 1998. Nutrient salts (ammonia, nitrite, nitrate and phosphate) were measured spectrofluorometrically. The results show that nutrient salts were relatively high at the northern part of the Gulf of Suez (Suez Bay). Their concentrations were 4.14, 0.45, 4.6 and 1.04 ng at/L for NH/, NO₂", NO₃" and PO₄", respectively. The present study revealed that the Suez Bay area can be considered as eutrophic region. This is related to sewage and/or industrial waste discharges along the western coast of the bay. The rest of the Gulf area is considered as Oligotrophic region. Also, the results show that nitrogen is the limiting factor for phytoplankton growth in the Gulf of Suez (3). The wastewater of the Suez Gulf: As the Suez Canal becomes the passage to the international trade between east and west, several wastes are easily disposed and dumping into the boundaries of the studied area .The pollutants vary from raw sewage, oil spills, and industrial effluents to garbage which has tremendous impacts on the marine environment. The marine environment of the bays in Suez Gulf is subjected to mixed sources of pollution (industrial, agricultural and domestic sewage) through the direct discharge of El-Kabanon drain, which is considered as the main industrial and sanitary drain. Sewage of approximately 120,000 m3 /day is dumped through El-Kabanon drain into the bay in Suez Gulf. The sewage discharged contains 94 ton/year of ammonia, 0.31 ton/year of nitrite, 0.40 ton/year of nitrate, 53 ton/year of inorganic phosphate, 0.41 ton/year of copper, 3.7 ton/year of zinc and 0.12 ton/year of lead. Research was carried out on heavy metal pollution in the region, where the bay is subjected to industrial run-off from oil refineries, fertilizer plants, and power station in addition to sewage and garbage. The heavy metal concentrations ranged from 7.2 to 148 µg/l for Zn, 10 to 63 µg/l for Cu, 0.7 to 12 μg/l for Pb and 0.01 to 1.3 μg/l for Cd, respectively. Adabiya station showed the highest values, because of the various pollution sources discharged (i.e., harbors, sewage, and industrial drains), while in contrast the station of Moon Beach showed the lowest concentrations. The Moon Beach area in the Suez Gulf was also found to suffer from extensive chronic petroleum pollution inputs, as it is evident in the vicinity of the SUMED pipeline company terminals, which include Egypt. J. Chem. **56,** No. 5,6 (2013)

both floating and land-based receiving terminals⁽⁴⁾. Most of these pollutants occurred from the human activities. The Suez bay area is subjected to pollution from three main sources: (1) Industrial waste: which come from different factories such as petroleum companies (Misr Petroleum Company, Suez Petroleum Company and Al- Nasr Petroleum Company for refining the crude oil). Suez Oil Company produce the final products of fuel gases, propane, butane, gasoline, jet fuel, kerosene, gas oil, asphalt and coke, sulphur and lube oil . The amount of discharged waste water from Suez oil company to Suez bay is 360000-m3/day industrial water only. The concentration of heavy metals (pb, Cu ,Zn ,& Fe) from the companies effluent during the year 2003 and 2004 reached to 520, 69.5, 41.7 and 1186.2 mg/l, respectively (5). (2) Domestic draining of the Suez City: may contain sludge arising from the treatment of sewage. Municipal sewage sludge may have harmful effects such as oxygen depletion and eutrophication, while health risks may arise from the presence of pathogens if dump sites are not carefully selected. (3) Ship's oil: The waste from repairing ships in Port Tewfik also marine transportation; including tanker operations, other shipping activities and accidental spills from ships represents another input of oil to Suez Canal (6).

Materials and Methods

Area of study on Suez Gulf

Suez bay is the entrance of (Gulf sea) Red sea and is limited by latitudes 29° 54` and 29° 57` N, and longitudes 32° 28` and 32° 34` E, (Fig. 1). It is an important shipping route for oil tankers and other ships traveling through the Suez Canal $^{(7)}$.

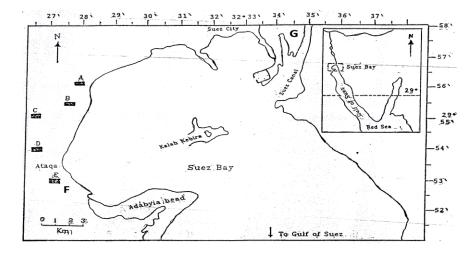


Fig. 1. Map of Suez Bay showing position of the sources of pollution. A: Al-Nasr Oil Company, B& C: inlet and outlet of Suez Oil Company & D: Electricity Station & E: Fertilizer Factory & F: AL-Osra &Al-Melaha beaches.

Samples collection and storage.

Surface water (0-2 cm) samples were collected from Suez Gulf, the studied areas including industrial zones and beaches (Table 1). It extended from Suez Harbor up to Fertilizer factory, and both of Al-Osra, Al-Melaha beaches a long Suez Gulf with a maximum distances nearly 21 km along Suez Gulf. Eight surface water samples with one liter for each other were taken from on-shore sites using precleaned glass bottles to minimize contamination with components of ambient air during storage of sampling and then the bottles were refrigerated for analysis (8).

TABLE 1. Studied sites, their location and distance from Suez Harbor.

Site No.	Location	Distance from Suez Harb. (Km)
1	Suez Harbor.	2(Km)
2	Al-Nasr Oil Company	5(Km)
3	Inlet of Suez Oil Company	7(Km)
4	Outlet of Suez Oil Company.	9(Km)
5	Outlet of Electrical Station Company.	10(Km)
6	Fertilizer Factory	18(Km)
7	Al-Osra beach	20(Km)
8	Al-Melaha beach	21(Km)

Harb.: Harbour Harb: Harbor.

General physico – chemical properties

The objective of this study is the monitoring and assessing the contamination of water by chemical and petrochemical pollutions. This study includes physical and chemical parameters of water such as PH values, electrical conductivity, salinity, T.D.S,T.S.S,T.S,D.O.,BOD,COD,Cl⁻,SO₄⁻,NO₃,PO₄⁻ and total alkalinity were determined for the seawater under study. All analyses had been achieved according to the following standard methods:

- 1. PH: The pH was determined using the electrometric method according to the ASTM D1293 $^{(9)}$ and APHA 4500H, $^{+\,(10)}$. The pH was measured at 25 ^{0}C using a digital pH meter, Hanna model pH 213, equipped with a combination pH electrode, HI 1230 B.
- 2. Biochemical and chemical oxygen demands for seawater were determined according to standard method described in details ⁽¹¹⁾.
- 3. *Electrical conductivity*: The electrical conductivity was determined using the instrumental method according to the ASTM D1125 ⁽⁹⁾ and APHA 2510, ^{+ (10)}. The conductivity of the seawater was determined at 25⁰C using a digital conductivity meter, WTW, model Cond 3301, equipped with a conductivity cell. The resistivity of the water samples was calculated from their obtained conductivity values.
- 4. Dissolved solids: Total dissolved solids were determined according to the ASTM D1888 $^{(9)}$ and APHA 2540 $^{(10)}$, methods at 105^{0} C by drying suitable aliquots of the water samples at 105^{0} C in an oven.

- 5. *Salinity*: The salinity of the water samples were estimated from the determined chloride concentration to give the salinity as sodium chloride.
- 6. *Alkalinity*: The bicarbonate alkalinity was determined by the titration method according to the ASTM D1067 and D3875, ⁽⁹⁾ and APHA 2320 ⁽¹⁰⁾. A suitable aliquot of the water sample was titrated to the pH 4.5, with a standard hydrochloric acid solution, calculations was done using Alkalinity Calculator Ver. 2.10. The alkalinity, as calcium carbonate, was calculated.
- 7. Determination of anions: Anions including fluoride, chloride, nitrate and sulfate were determined by the ion chromatography according to the ASTM D4327 ⁽⁹⁾ and APHA 4110, ⁽¹⁰⁾, using ion chromatography DX 600 gradient IC system (Dionex, Sunnyvale, CA, USA). The separated components were detected by auto suppressed conductivity using E D50 Electro chemical detector in the conductivity mode. An electrochemical self regenerated suppressor ASRS Ultra II 4-mm was used for suppression. Meanwhile, Ion Pac AS9-HC (4 x 250 mm). Separation column with Ion Pac AG9 (50 X 4 mm) guard column was used for separation by 9.0 mmol Na₂CO₃ solution at a flow rate of 1 ml/min. Integration was performed by Chrome-leon Ver. 6.30 software (Dionex).

Results and Discussion

Samples have been chosen for this study according to the location and activities taking place in each site. The main activities which may cause pollution are loading and unloading of ships in harbors, industrial and agricultural activities in addition to petroleum chemical and petrochemical wastes. Eight locations were chosen for this study starting from Suez Harbor site No. (1), to Al-Melaha beach site No. (8) (Table 1).

PH values ranged from 7.23 up to 8.20 in summer and 7.33 to 8.30 in winter with average values 7.62 and 7.65, respectively (Table 2 and Fig. 2). Normal pH value in seawater is between 8.0-8.2 at the surface, decreasing to 7.7-7.8 with increasing depths ⁽¹²⁾. In the Gulf of Suez, pH values were found to increase southwards, where the maximum pH of 8.20 & 8.30 was recorded at Inlet and outlet of Suez Oil Company for summer and winter, respectively while, a minimum pH of was recorded at Al-Osra beach and Suez Harbor for summer and winter, respectively which is affected by the disposal of mainly acidic sewage and industrial effluents, in addition to oil refineries effluents distributed in the Suez Bay. In general PH of Suez Gulf is slightly alkaline varing from region to another depending upon the location of sampling and independent on the season.

Electrical conductivity (EC): The Conductivity of an electrolytic solution depends on the ions present and their concentrations. The conductivity readings reflect the level of total dissolved solids (T.D.S) in wastewater ^(13, 4). Data showed that the conductivity values in the summer are greater than those recorded in the winter (Table 2 and Fig. 3). This may be related to mixing of salty water from the Suez Gulf with oil during the process of withdrawing petroleum considered as a source of salts in the petroleum wastewater. The

conductivity values ranged between 49.95-51.90 μ s/cm in summer and 49.80-51.80 μ s/cm in winter with average values of 50.92 and 50.75 μ s/cm, respectively, Low differences were noticed between values taken in the two seasons. Seasonal variation of salinity recorded high value 56.1 for inlet of Suez Oil Company in summer and low value 32.4 in outlet of Electrical Station Company site No. (5), in winter. Table 2 and Fig. 4 indicate that the salinity of the Suez Gulf was increasing from south to north, this accords with Morkos⁽¹⁴⁾ who stated that high salinity of Suez Gulf particularly at its northern part is attributed to the influence of high saline water of the Suez Canal.

TABLE 2. PH, EC and salinity values for the surface seawater samples from Suez Gulf.

	Season Summe	r				Winter	
S.No	No Location		E.C µs/cm	Salinity	PH	E.C μs/cm	Salinity
1	Suez Harbor.	7.45	51.90	33.9	7.33	51.60	33.4
2	Al-Nasr Oil Company	7.38	51.59	33.9	7.58	51.80	33.6
3	Inlet of Suez Oil Company	8.20	49.95	56.1	8.0	49.80	51.4
4	Outlet of Suez Oil Company.	8.03	50.80	54.3	8.30	50.50	51.6
5	Outlet of Electrical Station Company.	7.80	50.42	32.7	7.88	50.10	32.4
6	Fertilizer Factory	7.50	51.60	51.8	7.42	51.50	51.5
7	Al-Osra beach	7.23	50.70	32.9	7.35	50.60	32.8
8	Al-Melaha beach	7.40	50.40	50.6	7.35	50.10	50.1
Range		7.23- 8.20	49.95- 51.90	32.7- 56.1	7.33 -8.30	49.80- 51.80	32.4- 51.6
	Average	7.62	50.92	43.27	7.65	50.75	42.1

E.C: µs/cm

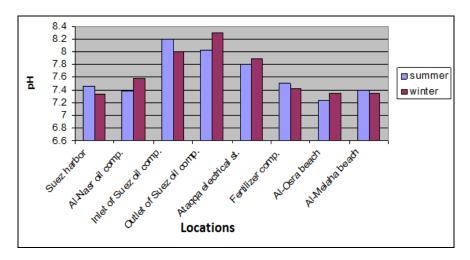


Fig. 2. The seasonal variation of pH value.

Egypt. J. Chem. 56, No. 5,6 (2013)

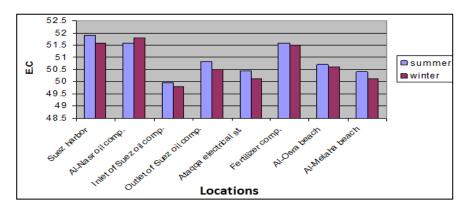


Fig. 3. The seasonal variation of E.C.

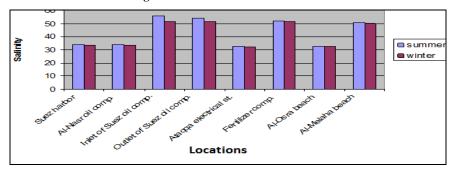


Fig. 4. The seasonal variation of salinity.

Total alkalinity: Alkalinity of water is a measure of its capacity to neutralize acid in other words to absorb hydrogen ions without significant PH change $^{(15)}$. Values of total alkalinity contents ranged between (80-180) mg CaHCO3 in summer and (115-200) mg CaHCO3 in winter with average values 128.12and 156.25 mg CaHCO3/L, respectively. This is due to evaporation during summer season .The highest values of alkalinity are in outlet of Suez Oil Company in winter, the lowest values of alkalinity are in Fertilizer Factory in summer (Table 3 and Fig. 5).

TABLE 3. The seasonal variation of total alkalinity.

	Season	Summer	Winter
S.No.	Location	Alkalinity	Alkalinity
		mg CaCO ₃ /L)	mg CaCO ₃ /L)
1	Suez Harbor.	110	135
2	Al-Nasr oil Company	120	145
3	Inlet of Suez Oil Company	140	150
4	Outlet of Suez Oil Company.	180	200
5	Outlet of Electrical Station Company.	90	150
6	Fertilizer Factory	80	115
7	Al-Osra beach	145	170
8	Al-Melaha beach	160	185
	Range	80-180	115-200
	Average	128.12	156.25

Alkalinity: mg (CaCO₃/L). Alkalinity, (mg (CaCO₃/L).

Egypt. J. Chem. 56, No. 5,6 (2013)

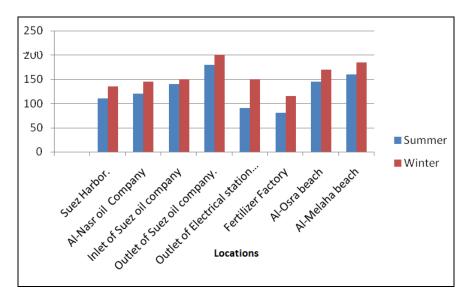


Fig. 5. The seasonal variation of total alkalinity, (mg (CaCO₃/L).

Total dissolved solids (TDS), total suspended solid (TSS) and total solids (TS)

Solid suspend: Due to the presence of oil processing companies in studied area which treated the oily- wastewater by flotation process in which solids contents or immiscible liquids that are lighter than water, they will float to the surface in the flotation tank, where they can be skimmed off. The gravity separator units for oil commonly are designed to remove free oil globules larger than 1.5 mm (16,15). So the outflow wastewater is not free from the oil and suspended solids, Table 4 and Fig. 6 show high values of suspended solids (SS) at the outflow effluent than those of the inflow water and also, the values in winter are higher 46851 mg\l than summer 36502 mg\l. This is due to the presence of the oil globules; such increase in the suspended solids (S.S) is most probably attributed to discharge of huge amounts of suspended solids from the dew-axing and coke distillation department as well as hydrodesulphurization unit directly into the drain. TDS comprise inorganic salts as calcium, magnesium, potassium, sodium, bicarbonates, chlorides and sulphates besides small amount of organic matter that are dissolved in water. The date obtained in Table 4 and Fig. 7 Show the highest values of TDS at Suez Harbor s. No. (1), TDS values ranged between 34965 mg\L to 36330 mg\L in summer and 34860 mg\L, to 36260 mg\L in winter with average values 356447.7 mg\L, 35525 mg\L respectively .In general TDS values increase than the maximum permissible limit of low 4\94 that TDS 2000 mg\L.

TABLE 4. Seasonal variation of TS, TDS and TSS.

	Season		Summer		Winter		
S.No.	Location	T.S mg\l	T.D.S mg\l	T.S.S mg/l	T.S mg\l	T.D.S mg\l	T.S.S mg\l
1	Suez Harbor.	38904	36330	11021	48058	36120	12038
2	Al-Nasr Oil Company	38300	36113	15219	48002	36260	17042
3	Inlet of Suez Oil Company	36432	34965	1466	46542	34860	16641
4	Outlet of Suez Oil Company.	38024	35566	15054	44675	35350	16036
5	Outlet of Electrical Station Company.	37040	35294	16640	47196	35070	16726
6	Fertilizer Factory	32709	36120	17728	46102	36050	20052
7	Al-Osra beach	34890	35490	11238	47830	35420	12018
8	Al-Melaha beach	35721	35280	17675	46404	35070	23334
	Range	32709- 38904	34965- 36330	1466- 17728	44675- 48058	34860- 36260	12018- 23334
	Average	36502	35644	13255	46851	35525	35610

T.S, T.D.S &T.S.S: mg\L.

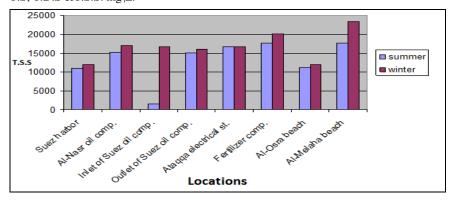


Fig. 6. The seasonal variation of T.S.S.

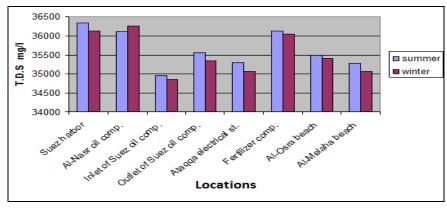


Fig. 7. The seasonal variation of T.D.S.

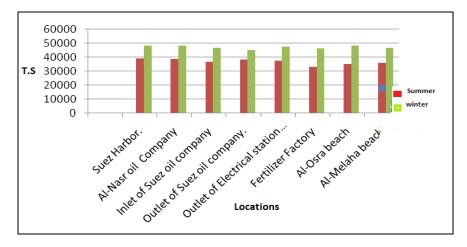


Fig. 8. The seasonal variation of T.S.

The content of dissolved oxygen (DO), chemical oxygen demand (COD) and biochemical oxygen demand (BOD)

The contents of dissolved oxygen ranged from 2.13 to 3.97 mgO₂ /L in summer and 1.49-2.24 mgO₂/L in winter (Table 5 and Fig. 9). This decrease in the content of dissolved oxygen may be due to the presence of mercaptans compounds in the oil participate in consuming the dissolved oxygen, or due to the presence of layer of oil on the surface of the water in the outlet effluent of Gulf prevent the oxygen to be dissolved. While, the content of BOD and COD in the outlet effluent exceeded than that recorded in the inflow water as shown in Table 5 and Fig. 10 & 11. Such increase in BOD and COD contents is mainly attributed to the raise in the content of total petroleum hydrocarbons and mercaptans compounds content in the drain, or may reflect the importance of active mixing and gas exchange in this area. Beside, the presence of fertilizer plant in the vicinity and also due to other activities in this area such as fishing farms, sewage stations and agriculture drainage^(4,15). Furthermore, the presence of high amount of organic compounds in the effluents enhances the growth of microorganisms in wastewater where the content of BOD in summer ranged between 172-362 mgO₂/L with average value 250 mgO₂/L also, BOD ranged between 86-212 mgO₂/L with average value 151 mgO₂/L in winter season. Generally, a BOD of 1 ppm is a characteristic of nearly pure water, as this value reaches 5ppm it is considered as doubtful pure water (17). Comparing this result of BOD presented in Table 5 with the tolerable level for pure water, higher level was particularly reported at the northern part of the Gulf, indicating the presence of perceptible load of domestic wastes discharging into this region.

TABLE 5. The content of dissolved oxygen, biochemical oxygen demand and chemical oxygen demand (mgO $_2\!/L)$ of surface water.

Season	1		Summer			Winter		
S.No.	Location	DO mgO ₂ /l	BOD mgO ₂ /l	COD mgO ₂ /l	DO mgO ₂ /l	BOD mgO ₂ /l	COD mgO ₂ /l	
1	Suez Harbor.	3.42	270	36.8	2.24	140	128	
2	Al-Nasr Oil Company	2.98	232	276	2.13	144	296	
3	Inlet of Suez Oil Company	2.46	172	202.4	1.92	86	264	
4	Outlet of Suez Oil Company.	3.97	272	231.2	2.13	146	240	
5	Outlet of Electrical Station Company.	2.13	362	55.2	2.02	180	160	
6	Fertilizer Factory	3.55	242	92	1.77	212	160	
7	Al-Osra beach	2.20	240	66	1.49	120	128	
8	Al-Melaha beach	2.23	210	94	1.81	180	150	
Range		2.13- 3.97	172- 362	36.8- 276	1.49- 2.24	86- 212	128- 296	
	Average	2.867	250	131.7	1.938	151	190.7	

BOD & COD $(mgO_2/1)$.

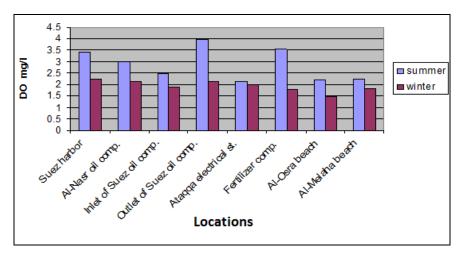


Fig. 9. The content of dissolved oxygen (mg O_2/L).

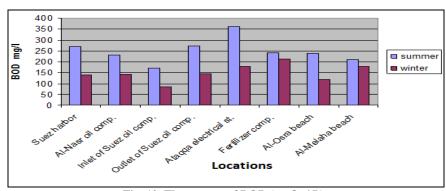


Fig. 10. The content of BOD (mgO_2/L) .

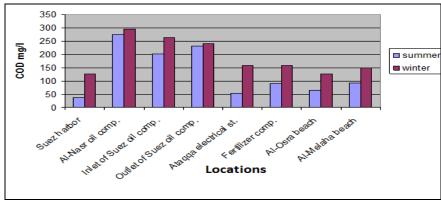


Fig. 11. The content of COD (mgO_2/L) .

Determination of nitrite, nitrate, phosphate, sulphate, chloride and fluoride

The nutrient salts are considered essential components for living organisms in natural water. Nitrite is found together with nitrate and ammonia in waste water but its concentration relatively low because of its instability; it is oxidized to nitrate or reduced to ammonia, by both chemical reaction and \or biological by bacteria (20, 13). Table 6 and Fig. 12 show that the nitrite values for the inflow water trend to decrease; this may be due to the reduction of it to ammonia by species of bacteria (19, 12), or may be due to the oxidation of it to nitrate also, the presence of petroleum compounds increases the values that confirmed from data recorded in Table 6, Al-Nasr Oil Company and Fertilizer Factory have the highest values 0.188, 0.180 ppm for summer and winter seasons, respectively. Nitrate considered the final stage in oxidation of nitrogen compounds and measure quantity of organic matter. Seasonal variation of nitrate recorded in Table 6 and Fig. 13 indicates that nitrate ranged between 0.01 ppm up to 8.02 ppm in summer with average value 2.39 ppm, while the concentration in winter ranged between zero ppm to 7.65ppm with average value 2.15 ppm. As shown from the recorded results, the nitrate content decreased during winter for both inflow and outflow effluent; this may be attributed to its uptake by microorganisms and phytoplankton while the increase during summer may be due to the nitrification of

ammonia to nitrite and nitrate produced by the biochemical decomposition of dead planktons⁽²¹⁾, the nitrate values were low in two seasons and within the permissible maximum limits 40 ppm for 4\94. Phosphate concentrations recorded in Table 6 and Fig. 14 show the average value in summer 0.034 mg\L and 0.020 mg\L in winter, the highest concentration of orthophosphate and total phosphate are found to be in summer, this may be attributed to decay of planktons which its body containing phosphate⁽²²⁾. Comparing the present data of nutrients with the previous records in the Suez Gulf and other regions indicates that they are comparable to other records (Table 7). The concentrations of nitrite, nitrate and phosphate were 0.0-0.188, 0.01-8.02 and 15-61ug/L, respectively at the area of study. Such high concentrations are indicators of the presence of pollutants of high activity viz. sewage discharge, industrial effluents and oil refineries concentrated at the northern part of the Gulf (Suez Bay). Also, it is obvious that the area of investigation can be divided into two regions, Suez Bay region as an eutrophic water (high productivity) and the part of the Gulf as going southward to Safaga, which is considered as oligotrophic water (low productivity). This conforms to Abd EL-Rahman⁽²³⁾ who stated that the Suez Bay is regarded among the eutrophic regions when considering the magnitude of the standing crop of zooplankton. This is particularly true at the in shore stations due to the high load of sewage and industrial wastes discharged into the area. Such pollution can obviously lead to a gradual deterioration of the water quality in the Bay. Accordingly, it is recommended that this wastewater should be treated to improve its quality before being discharged into the Bay. Generally, Thus nitrogen was the limiting factor for phytoplankton growth in the study area (24). The mean concentrations of sulphate, are shown in Table 6 and represented in Fig. 15, respectively, the increase of sulphate values may be due to the petroleum wastes containing sulphure which ranged from 0.1 to 3.4% (by weight) where sulphure compounds present as sulphate, sulphid and mercaptans. In addition, the reduction of sulphate in petroleum wastes depends on several factors; the most important of these are the concentration of sulphate, organic matter, and both of nutrient and dissolved oxygen necessary to sustain the life of bacteria (22, 4). The data recorded in Table 6 display the highest values 982.79 and 987.02 ppm of sulphate at Inlet of Suez Oil Company site no. (3) For summer and winter, respectively. Chlorides concentrations recorded in Table 6 and Fig. 16 showed that values in summer higher than winter, values in summer ranged between 100.57 -559.63 mg\L with average values 300.68 mg\L.while in winter values ranged between 98.68-386.79 mg\L with average 309.03mg\L. Flouride are present as fluorspar in rocks (limestone, sandstone) and as cryolite in igneous rocks (granite) (25). The seasonal variation of fluoride concentration is recorded in Table 6 and graphically represented in Fig. 17 for samples .The highest values 1.13, 1.11ppm are for Inlet of Suez Oil Company site no. (3) in summer and winter seasons, respectively due, to decay of living organism, and the presence of the crust earth, rocks, ground water containing fluoride during the drilling.

TABLE 6. Seasonal variation of nitrite, nitrate, phosphate, suphate, chloride and fluoride.

Season	uo			Sur	Summer					W	Winter		
S.No	Location	nitrite ppm	nitrate ppm	Phos. mg/l	Sulph ppm.	Chlor. mg/l	Flou. ppm	nitrite ppm	nitrate ppm	Phos. mg/l	Sulph ppm.	Chlor. mg/l	Flou. ppm
Т	Suez Harbor.	0.0	86'9	0.037	55.6	366.79	0.27	0'0	0.9	0.025	0.25	362.54	0.24
2	Al-Nasr Oil Company	0.188	1.31	0.057	127.03	288.21	0.075	0.165	1.00	0.047	129.00	386.79	0.007
3	Inlet of Suez Oil Company	0.183	8.02	0.020	982.79	559.63	1.13	0.180	7.65	0.015	20.786	554.01	1.11
4	Outlet of Suez Oil Company.	0.185	1.17	0.015	265.74	100.57	0.054	971.0	1.10	0.011	264.70	89'86	0.052
5	Outlet of Electrical Station Company.	0.0	0.19	0.028	638.002	289.72	0.045	0.0	0.10	0.020	00'8£9	280.71	0.040
9	Fertilizer Factory	0.184	1.45	0.027	148.2	270.43	090'0	0.180	1.40	0.013	145.02	265.24	0.050
7	Al-Osra beach	0.0	0.01	0.061	20.0	264.82	0.024	0.0	0.0	0.012	20.0	264.00	0.022
8	Al-Melaha beach	0.0	0.01	0.028	18.6	265.33	0.020	0.0	0.0	0.022	16.4	260.30	0.020
	Range	0.0-	0.01- 8.02	0.015-	18.6- 982.79	100.57- 559.63	0.020-	0.0-	7.65	0.011-	16.4- 987.02	98.68- 386.79	0.007-
	Average	0.0925	2.39	0.034	282.99	300.68	0.209	0.0875	2.15	0.020	281.89	309.03	0.192

nitrite (ppm), nitrate (ppm), phos: phosphate(mgU), Sulph: sulphate (ppm), Chlor: chloride (mgU) and Flou: fluoride (ppm).

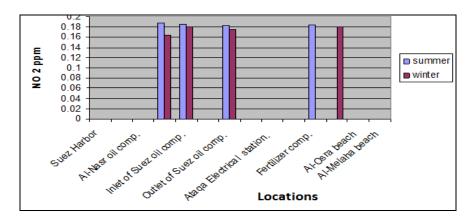


Fig. 12. The seasonal variation of nitrite content (NO₂) ppm.

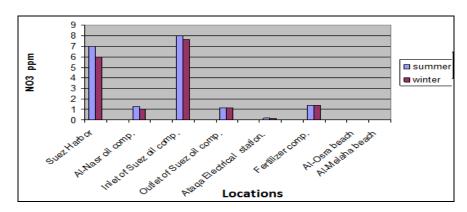


Fig. 13. The seasonal variation of nitrate (NO_3) ppm.

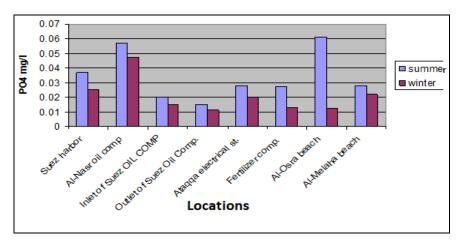


Fig. 14. The concentration of phosphate (PO_4) mg/L.

TABLE 7. Nutrient salts concentrations (ug/L) of seawater in the present study compared with other region.

Parameter					
Area	NH ₃ -N	NO ₂ -N	NO ₃ -N	PO ₄ -P	References
Gulf of Suez	0.18- 4.14	0.07- 0.45	0.48- 4.6	0.8-1.04	(3)
AL Khor, Suez	3.54	0.38	2.06	0.62	(26)
Gulf of Suez	1.54	0.20	0.52	0.47	(27)
Suez Canal	2.44	0.59	2.03	0.81	(27)
Suez Bay	3.25	0.49	1.48	0.85	(28)
Suez Bay	8.86	0.33	3.42	0.45	(29)
Jeddah		0.28	9.69		(3)
Dongonab Bay			4.9-7.2	0.3-1.5	(30)
Abu Qir Bay	1.23	0.07	0.88	0.58	(31)
Eastern harbor of Alexandria	3.16	0.95	6.79	0.56	(32)
EL Max Bay	20.38	0.48	1.87	1.62	(33)
Suez Gulf			0.17		(4)
Suez Gulf		0.0-0.188	0.01-8.02	15-61	This work
Oligotrophic level	0.5		0.5	0.05	(34,35)
Eutrophic level	2.02		4.0		(51,55)

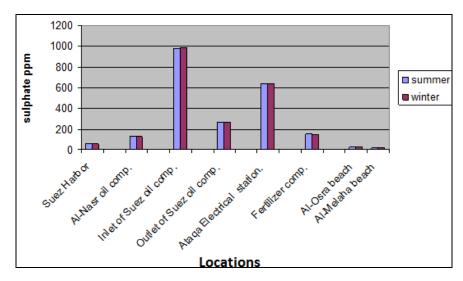


Fig. 15. The seasonal variation of sulphate (ppm).

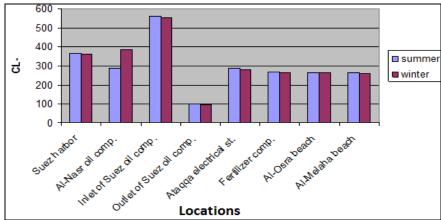


Fig. 16. The seasonal variation of Cl⁻mg/L.

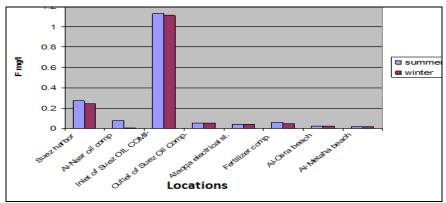


Fig. 17. The seasonal variation of (F) ppm.

Conclusion and Recommendations

The discussion revealed general physico-chemical properties of seawater such as PH values, alkalinity, T.S.S, conductivity, nitrates, phosphates, sulphates, chlorides, salinity, T.D.S, BOD and COD:

- 1. water samples were analyzed for PH values, alkalinity, salinity and electronic conductivity which indicated that PH values are slightly alkaline varing from region to another depending upon the location of sampling and independent on the season, alkalinity are higher in winter than summer this is due to evaporation during summer season, Conductivity value of seawater samples don't differ in all locations except conductivity in certain locations such as Suez Harbor and Al-Nasr Oil Company; this may be related to mixing of salty water from the Suez Gulf with oil during the process of withdrawing petroleum considered as a source of salts in the petroleum wastewater.
- 2. Results of BOD varied between $36.8~\rm up$ to $213~\rm mg$ O2/L in summer and $86~\rm up$ to $212~\rm mg$ O2/L in winter. BOD are higher in some samples due to activities

in these areas ⁽³⁶⁾, also COD varied from 36.8-276 mg O2/L in summer and 128-296 mg O2/L in winter, the highest values of COD are observed in site no.(2). Such increase in BOD and COD contents is mainly attributed to the raise in the content of total petroleum hydrocarbons and mercaptans compounds content in the drain beside, the presence of fertilizer plant in the vicinity and also due to other activities in this area such as fishing farms, sewage stations and agriculture drainage.

- 3. The distribution of some anions in seawater was studied, the higher values of anions are chlorides followed by sulphate, phosphate and the lowest one is nitrite and it was the limiting factor for phytoplankton growth in the Suez Gulf.
- 4. The Suez Bay area is regarded among the eutrophic region and highly polluted due to its relation with sewage and/or industrial activities which distributed along the western coast of the Bay.
- 5. The concentration is relatively higher than recommended in the regulations of the Egyptian low of environment of No.4/1994. Results obtained from the present study should be considered when implementing a strategy for the protection and management the Suez Gulf, Needless to say that such kind of pollution will obviously lead to a gradual deterioration of the water quality of the Bay.
- 6. More detailed studies are required to assess the impact of hydrodevelopment projects and evaluate its effect on the studied area and monitoring should be conducted at least twice per year.
- 7. Because of the Suez Gulf is used in many areas as a recipient for wastewater from different sources, this study recommended that pollution control for wastewater disposing into it should be improved.
- 8. Serious commitment should be applied for the environmental law for 1994 to protect water and beaches.

Acknowledgments: The authors gratefully thank the Chemistry Department, Faculty of Science (Boys) Al-Azhar University and Egyptian Petroleum Research Institute, Nasr city, Cairo, Egypt for helping to carry out the field work.

References

- Pekey. H., Karakas, D., Ayberk , S., Tolun, L. and Bakoglu, M., Ecological risk assessment using trace elements from surface sediments of Izmit bay (Northeastern Marmara Sea) Turkey. *Marine Pollution Bulletin*, 48, 964-953 (2004).
- 2. Samanidou V.F. and Papadoyanis I.N.J., Environ. Sci. Health, A27 (3), 587-601 (1992).
- 3. **Mohamed A. Hamed and Tarek, O.S.,** Effect of pollution on the water quality of the Gulf of Suez. *Egypt J. Aquat BioL* < *S Fish.*, **4** (d), 161-178 (2000)
- 4. **Mohamed I., Omayma A., Nazik.A F., Rabie S. and Moustafa M.,** Assessment of surface water quality of Suez Gulf. *Al-Azhar Bull.Sci.* **23** (1), 47-58 (June) (2012).

- Shahine, Eman A.A, Environmental pollution assessments for some human activities in light of chemical studies of water pollution, *M.Sc. Thesis*, Faculty of Science, (Girls), Al-Azhar University (2005).
- Van Vooren, L., Van De Steene, M., Ottoy, J-P. and Vanrolleghem, PA., Water Science, 43 & 7 (2003).
- 7. **REMIP WORKING GROUP 2 Remip, JICA and EEAA,** State oil pollution and management in Suez Gulf region, page 15 (2008).
- 8. **LAEA-TECDOC-**Collection and preparation of bottom sediment samples for analysis of radionuclide and trace elements International Atomic Energy Agency (2003b).
- 9. **Annual Book of ASTM Standards** Vols. 11.01 and 11.02, Am. Soc. *Test Mater.*, *Philadelphia* (1992).
- 10. Standerd Methods for the Examination of Water and Wastewater 20th ed. American Public Health. Association (APHA), American Work Water Association (AWWA), and Water Environment Federation 9WEF), Edited by Clescrei L.S., Greenberg A.E. and Eaton A.D (1998) .
- 11. Patnaik. Hand Book of Environmental Analysis, Lewis Publishers (1997).
- Capurro L.R. "Oceanography for Practicing Engineers". Barnea and Noble Inc., NewYork (1970).
- 13. **Barakat M.A.K., EL-Sayed and Ouf, E.A.,** Evaluation of drinking water quality in delta of Egypt. Egyptian Petroleum Research institutes (EPRI) Cairo, Egypt. (Accepted) October 22. *Egypt. J. Petrol.* **18** (1), 81-92 (2009).
- 14. **Morcos S.A.** The tidal currents in the southern part of Suez Canal. *Gen. Ass. of Helsinki, Finland*, **51**, 307-312 (1970).
- 15. Omayma, A. Mousa, Nazik. A Farid, Rabie Saad and Moustafa M. Emara, Assessment of surface water quality of Suez Gulf. Mohamed Ibrahim (a) Al-Azhar Bull. Sci. 23 (1), 47-58 (June) (2012).
- 16. **Paul L. Bishop**, *Pollution Prevention, Fundamental and Practice*, Th Mc Graw-Hill Companies (2000).
- 17. **Anon** "Environmental Central and Public Health, Water Analysis, Standards and Treatment, Eyre and Spottiswoode Ltd, ECPH, 131 (1975).
- World Health Organization (WHO). Nitrate and N- nitro compounds, Geneva, Environmental Health Criteria 5 (1975).
- Omayma E. Ahmed, Mageed M. Soliman, Nazik A. Farid and Abd EL- Rahman Mousa, Assessment of water quality of Alexandria sea water coast. (Accepted) Al-Azhar International Scientific Conference Environment, Development, and Nanotechn. 2010. ology (ISCAZ 2010) March 22-24 (2010) .

- 20. **Cecen. F. and Orak, E.,** Nitrification of fertilizer wastewaters in a biofilm reactor. *Journal of Chemical Technology and Biotechnology*, **65** (3), 229-238 (1996).
- 21. El-Sabah, M.I. and Beltagy, A.I., Bull. Institute, Oceanography. & Fish. 9, 78-82 (1982).
- 22. Fathy, K.A., Ph.D. Thesis, Faculty of Science, University of Suez Canal (2002).
- 23. **Abd EL-Rahman N.S.,** Ecological studies on the distribution of zooplankton communities in the northern part of the Suez Gulf 'Suez Bay. *M. Sc. Thesis*, Faculty of Science, Suez Canal University, 316 pp (1993)
- 24. **Redfield A.C., Katchum B. H. and Richard S. A.,** *The Influence of Organism on The Composition of Seawater in the Sea,* John Wiley and Sons, New York, Vol. 2, p26-77 (1963).
- Charless R. Cox., Operation and Control of Water Treatment Processes, World Health Organization, Monograph series No.49, Geneva (1969)
- 26. **Hamed M.A.,** Environmental studies on AL Khor area, Suez, Egypt. *J. Environmental Science*. Faculty of Science, Mansoura University. 75,181-201pp (1998).
- 27. **Hamed M.A.,** Determination of some microelements in the aquatic system and their relation to the efficiency of marine life. *Ph. D. Thesis*. Faculty of Science, Mansoura University, 120 pp (1996).
- 28. **Hamed M.A.,** Sea water quality at the northern part of the Gulf of Suez and the nearby area of the Suez Canal. *M. Sc. Thesis.* Faculty of Science, Mansoura University, 81pp (1992).
- 29. **Mohamed, A.A.,** Study of iodine cycle in Suez Bay. *M. Sc. Thesis*, Faculty of Science, Helwan University, Egypt, 90 pp (1994) .
- 30. Farah O.M., Symposium of Marine Chemistry in the Arab Region. Suez, Egypt. April, 21-23 (1991) .
- 31. **Fahmy, M. A.,** Hydrochemistry and nutrients of Abu-Qir Bay during summer, 1995. *Bull. Fac. Sci.*, Alex. Univ. **37** (2),171-186 (1997) .
- 32. **Aboul-Kissum, T.A.,** Cycles of carbon, nitrogen and phosphorus in the marine environment in Alexandria region. *M, Sc. Thesis*, Fac. Sci., Alex. University (1987).
- 33. **Tayel, F. F., Fahmy, M.A. and Shenadah, M.A.,** Studies on physicochemical characteristics of the EL Mex Bay and new Dekhaila Harbour waters of Alexandria. *Bull. Nat. Inst. Ocean. & Fisheries*, ARE, **22**,1-18 (1997).
- 34. Skrivanic, V. and Strin, J., Basic Physical, Chemical and Biological Data Reports. R.V. "A Mohorov ICIC" Adriatic cruises, 1974-76. Hydrographic Inst & Yugoslav Navy-split, 175 pp (1982).
- 35. **Franco**, **P.**, Fattori influent sulla productivity prim aria dell Adriatic© settentrionala. *Proc. Int. Conf.*, *Problems of the Adriatic Sea Trieste*, pp. 155-174 (1983).

36. Louis, M.A., Environmental Science and Technology .VI (10), p. 901 October (1974).

(Received 9/12/2013; accepted 9/12/2014)

دراسة فيزوكيميائيه على مياه البحر السطحية في منطقة شمال غرب خليج السويس

تهدف هذه الدراسه الى رصد و تقييم التلوث فى البيئه البحريه لقناة السويس حيث انها تتمتع بموقع جغرافى مميز يجعلها من أهم الممرات التجاريه العالميه والبحريه بين الشرق والغرب لذلك هى تعتبر من اهم مصادر الدخل القومى بجمهورية مصر العربيه وقد تتاثر البيئه البحريه فى قناة السويس بتفاعل الانشطه فى مناطق الدراسه و التي تشمل التلوث بالصرف الصحى والزراعى والصناعى حيث تتسرب هذه الملوثات الى داخل القناه وتهدف أيضا الى رصد تاثير هذه الملوثات على جودة مياه الشواطئ وتراكم هذه الملوثات بمواقع الدراسه فى المواسم المختلفه له نجميع عدد ثمانية عينات بين فصلى 7 (2000 صيفا 2010/2 شتاءا من المياه السطحيه لخليج السويس و طبقا لمواقع الدراسه.

التركيزات أعلى نسبيا من الموصى بها في لوائح قانون البيئة المصرية لسنة 1994/4 النتائج المتحصل عليها من هذه الدراسة يجب أخذها في الاعتبار عند تنفيذ استراتيجية لحماية وإدارة خليج السويس، وغني عن القول أن مثل هذا النوع من التلوث سيؤدي بالطبع إلى تدهور تدريجي في نوعية وجودة المياه في الخليج. هناك حاجة لدراسات أكثر تفصيلا لتقييم أثر المشاريع المائية والتنمية وتقييم تأثيرها على منطقة الدراسة والرصد ينبغي أن يجري على الأقل مرتين في السنة لأن خليج السويس يستخدم في العديد من المجالات مثل مستقبل لمياه الصرف من مصادر مختلفة، أوصت هذه الدراسة بأنه ينبغي مكافحة التلوث بمياه الصرف المحدي. وينبغي تطبيق الالتزام الجاد لقانون البيئة لعام 1994 الخاص بحماية المياه والشواطئ.