EFFECT OF EXOGENOUS APPLIED ANTIOXIDANTS ON ENDOGENOUS ENZYMATIC AND NON-ENZYMATIC ANTIOXIDANTS IN PEPPER PLANT GROWN UNDER SALINITY STRESS CONDITIONS Sakr, M. T. and Reda S.A. Metwally Bot.Dept.Faculty of Agric. Mansoura Uni. Egypt sakmoheb@yahoo.com

ABSTRACT

Two pot experiments were performed at the Experimental Station Farm , Faculty of Agriculture Mansoura university during two successive summer seasons of 2007 and 2008. This work was conducted to study the role of some exogenous applied antioxidant on endogenous enzymatic and non-enzymatic antioxidants under different salinity stress levels of sweet pepper cv. California wonder.

The data show that all salinity levels and each of applied antioxidants(presoaking+foliar spray) as well as their interactions slightly increased the contents of different non-enzymatic antioxidants (phenols, Ascorbic and Glutathione) in both shoot and fruits of pepper plant during the two growth seasons. Moreover SWE combined with highest salinity level treatment was the most effective in this respect.

Moreover each of salinity stress levels (2000, 4000 or 6000 mg/l) increased Super oxide dismutase (SOD) and Ascorbic peroxidase (APX) activity in both shoot and fruits of pepper plants. In addition The data show that any of applied antioxidants (HA, SA, ASA, Tochopherol, Put., SWE) alone or combined with different salinity stress levels also increased the activity of SOD, APX enzymes in both shoot and fruits of pepper plant, during the two growth seasons.

INTRODUCTION

Non-enzymatic antioxidants contents:

Regarding ascorbic acid (ASA), the sequence of events in the plant tissue subjected to drought stress are: increased production of ROS and of oxidized target molecules; increases in the expression of genes for antioxidant functions; increases in the levels of antioxidative systems and antioxidants; and increased scavenging capacity for ROS, resulting in tolerance against the drought stress (Mano, 2002).

The non-enzymatic plant antioxidants can be classified into two major types: (1) AA-like scavengers, and (2) pigments such as carotenoids. ASA is an important antioxidant, which reacts not only with H2O2 but also with O2, OH and lipid hydroperoxidases. On the other hand, ASA has been implicated in several types of biological activities in plants: (1) as an enzyme co-factor, (2) as an antioxidant, and (3) as a donor/ acceptor in electron transport at the plasma membrane or in the chloroplasts, all of which are related to oxidative stress resistance (Conklin, 2001). APX uses ASA and oxidizes it to monodehydroascorbate (MDA). MDA may give rise to dehydroascorbate (DHA). Both MDA and DHA will then be reduced to regenerate the ascorbate pool. This type of scavenging is thought to occur near PSI, thereby

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minimizing the risk of escape and reaction of ROS with each other (Foyer and Noctor, 2000). AA is water-soluble and also has an additional role in protecting or regenerating! oxidized carotenoids or tocopherols (Imai, *et al.* 1999). ASA is a major metabolite in chloroplasts of higher plants and represents about 10% of the soluble carbohydrate pool in leaves (Noctor and Foyer, 1998).

ASA regenerates tocopherol from tocopheroxyl radical providing membrane protection (Thomas, *et al.,* 1992).

As forPhenols,Navarro, *et al.*, (2005), found that pepper plants grown under three saline treatments (0, 15, and 30 mM NaCl) had some antioxidant activity fractions such as, lycopene, β -carotene, ascorbic acid, total phenolic compounds and reducing sugars were enhanced.

Alexieva, (2001), stated that in pea plants the increase in phenols was significantly induced by drought stress.

Hale, *et al.*, (2005), found that total phenolic glycoside concentration were increased due to drought stress.

Regarding,Glutathion,the antioxidant (GSH) defense must keep active oxygen understanding of the key roles of enzymes such as SOD, APX and catalase in antioxidant defense. While these individual enzymes have central roles in the antioxidant defense network, the exploration of the enzymes involved in the synthesis and metabolism of ascorbate and glutathione.

The enzymes ascorbate peroxidase, glutathione reductase, superoxide dismutase and monodehydroascrbate reductasem, among others, are involved in the regeneration of glutathione and ascorbate that are important in detoxification of ROS (Foyer and Mullineaux, 1994).

As for, Enzymatic antioxidants, plants posses antioxidant systems in the form of enzymes such as SOD, APX, GR, DIIAR, CAT and metabolites viz., ascorbic acid, glutathione, α -tocopherol, carotenoid, flavanoids etc. (Smirnoff ,1995). These antioxidant enzymes and metabolites are reported to increase under various environmental stresses (Yu and Rengel,1999) as well as comparatively higher activity has been reported in tolerant cultivars than the susceptible ones .It is possible that better salinity resistance maintain higher activity of antioxidant enzymes viz., SOD, GR and CAT resulting in lower H2O2 production, lipid peroxidation and higher membrane stability(Sairam, *et al.*,2002).

MATERIALS AND METHODS

Two pot experiments were performed at the Experimental Station Farm , Faculty of Agriculture Mansoura university during two successive summer seasons of 2007 and 2008. This work was conducted to study the role of some antioxidant materials in alleviation the harmful effect of different salinity stress levels on sweet pepper cv. California wonder.

Seed of sweet pepper (*Capsicum annum* L.) cv. California wonder were sown on 15 th February in both 2007 and 2008 seasons. Seedling were transplanted at the age 45days (6.7 leaves) on the 1 st of April in pots (50 cm inner diameter) containing 8 kg of air dried loamy soil at the rate of two plant/

pots. Each pot was supplied limit amounts of 20.5 % N in form ammonium sulphate at the rate (2.5 kg), 15.5 % P_2O_5 in the form calcium superphosphate at the rate (1.5 kg) and 48 % K2O in the form potassium sulphate at the rate (1 kg) before planting and added 1.5 kg ammonium sulphate 30, 60 and 120 days after transplanting.

Four levels of artificial sea water used by dissolving known weight of natural salt crust, in tap water. The natural salt crust was brought directly from the salterns of Rashid, El- Beheira Governorate, Egypt where the Mediterranean sea water is evaporated, air dried, thoroughly crushed using porcelain mortar and pestle. The four salinity levels used:

1)- Tap water (320 mg/l). ,2)- 2000 mg/l. ,3)- 4000 mg/l. ,4)- 6000 mg/l.

The amount of salt for each salinity level was calculated, dissolved in the proper amount of tap water and used for experimental investigation.

Applied antioxidants were: 1)- Tap water.,2)- Humic acid (1000 mg/L).,3)- salicylic acid (250 mg/L)., 4)- Ascorbic acid (250 mg/L).,5)- Putrescince (1 mg/L).,6)- Seaweed extract (1000 mg/L).

The seeds were presoaking in any of applied antioxidants for 8 hours before sowing and the plants were foliar sprayed with any of each applied antioxidants at 30,60,80,120 and 150 after trans planting under salinity stress levels.Plant samples were collected after 80 days to determine the contents of antioxidants as well as enzymatic antioxidant activity.

Non-enzymatic antioxidant determination

Total ascorbate determination:

0.5 g of leaf was ground in 50 ml of 2% (w/v) metaphosphoric acid using mortar and pastel and centrifuged for 30 min at 13 000 rpm at 4°C. The ascorbate content (μ .mol / g FW) was measured in the supernatant at 25°C, .The absorbance of red color was measured at 520 nm A blank value for each sample was obtained by adding a few crystals of ascorbic acid, completely reducing the dye and rendering it colorless according to Omaye et al (1979).

Total glutathione determination

The level of total glutathione(GSH) was determined with Ellman's reagent according to *De Vos et al (1992).* 300 ul of sample buffer were mixed with 630 ul of 0.5 M K₂HPO₄ and 25 ul of mM 5, 5 -dithiobis (2-nitrobenzoic acid) (final pH 7). The Absorbance at 412 nm was read after 2 min. GSH was used as a standard.

Total phenols determination:

1g of dry defeated ground leaves were macerated in 5-10 ml 80% ethanol for at least 24 hours at 0°C, the alcohol was clarified, the remained residue was re-extracted with 5-10 ml 80% ethanol 3 times. At the end, the clarified extract was completed to 50 ml using 80% ethanol. The colorimetric method of Folin-Denis as described by Daniel and George (1972) was employed for the chemical determination of phenolic compounds .quantities were determined by reading the developed blue color at 725 nm. Using 0.5 ml 80% ethanol and reagents only as a blank.(Daniel and Georg,1972 and A.O.A.C.,1967).

2-Enzymatic antioxidant activity determination: -Ascorbate peroxidase activity determination:

Ascorbate peroxidase (APX) was assayed spectrophotochemically according to Fielding (1978). The assay was carried out at 25°C in 1.0 cm light path cuvette and the reaction mixture was consisted of 1500 ul phosphate buffer, 20 μ l EDTA, 1000 μ l sodium ascorbate and enzyme extract (20 μ l). After mixing the reaction was initiated by adding the 480 μ l H₂O₂ and decreasing in optical density at 290 nm against blank (without extract) was continuously recorded every minute (for two minutes).

-Super oxide dismutase (SOD) activity determination:

Leaf samples were collected in a ice bucket and brought to the laboratory. Leaves were then washed with distilled water and surface moisture was wiped out. Leaf samples (0.5g) were homogenized in ice cold 0.1M phosphate buffer (pH 7.5) containing 0.5Mm EDTA with pre-chilled pestle and mortar. The homogenate was transferred to centrifuge tubes and was centrifuged at 4 $^{\circ}$ C in Beckman refrigerated centrifuge for 15 min at 15000 X g .The supernatant was transferred to 30 ml tubes and referred to enzyme extract .

SOD activity was estimated by recording the decrease in absorbance of superoxide-nitro blue tetrazolium complex by the enzyme . About 3ml of reaction mixture containing 0.1ml of 1.5M sodium carbonate, 0.2mL of 200mM methionine, 0.1ml of 2.25Mm Nitro-blue tetrazolium, 0.1ml of 3mM EDTA, 1.5ml of 100mM potassium phosphate buffer, 1ml distilled water and 0.05 ml of enzyme were taken in test tubes in duplicate from each enzyme sample. Two tubes without enzyme extract were taken as control. The reaction was started by adding 0.1ml riboflavin (60mM) and placing the tubes below a light source of two 15 w florescent lamps for 15 min. reaction was stopped by switching off the light and covering the tube with black cloth. Tubes without enzyme developed maximal color. A non- irradiated complete reaction mixture which did not develop color served as blank. Absorbance was recorded at 560nm and one unit of enzyme activity was taken as the quantity of enzyme which reduced the abaorbance reading of samples 50% of comparison with types lacking enzymes(Dhindsa *et al.*,1981)

The dates of all experiments were statistically analyzed as technique of the analysis of variance (ANOVA) according to Gomez and Gomez(1984). The treatment means were compared using the least significant differences (LSD).

RESULTS

Non-enzymatic antioxidants contents: Total phenols, Ascorbic and Glutathione content:

The data in table(1,2,3) show that all salinity levels and each of applied antioxidants as well as their interactions slightly increase the contents of different non-enzymatic antioxidants (phenols, Ascorbic and Glutathione) in both shoot and fruits of pepper plant during the two growth seasons. Moreover SWE combined with highest salinity level treatment was the most effective in this respect.

Table (1): Effect of salinity stress levels and antioxidants (presoaking and foliar spray) as well as their interactions on total phenols (mg/100g F.w) in shoot and fruits of pepper plant during the two growing seasons (2007 & 2008).

Trea	atment	Salinity levels (mg/L)											
		0.00	2000	4000	6000	Mean	0.00	2000	4000	6000	Mean		
		Season 2007				Weall	Season 2008				Weall		
Antioxidant (mg/l)			Pepper shoot										
Tap water		46	50	67	73	59	44	49	65	76	58		
HA	(1000)	43	51	58	71	55	45	54	64	73	59		
SA	(250)	40	45	54	63	50	43	47	57	67	53		
ASA	(250)	49	54	68	83	63	48	56	71	78	63		
Put.	(1)	41	45	68	78	58	40	48	60	79	56		
SW	'E (1000)	65	74	76	93	77	64	79	80	99	80		
Mean		45	51	63	74		45	52	63	74			
LSD at	t 5%	Antio	xidant:		Salinity								
		Interaction: 5.65 Interaction: 6.7											
-					Pepper					50			
Tap wa		26	31	38	45	35	28	34	40	50	38		
HA	(1000)	28	30	39	43	35	30	34	42	46	38		
SA	(250)	27	30	44	51	38	28	33	45	54	40		
ASA	(250)	34	39	45	43	42	35	38	46	52	42		
Put.	(1)	28	35	40	50	38	30	34	44	51	39		
SW	'E (1000)	36	42	56	63	49	38	45	57	64	51		
Mean		29.	34	43	49		31	36	45	51			
LSD at	t 5%		dant: 3		Salini	ity: 4.73 Antioxidant: 2.71 Salinity: Interaction: 5.7					/: 3.74		
		Interaction: 6.65 Interaction: 5.7											

HA : humic acid, SA : Salicylic acid, ASA : Ascorbic acid, Put : Putrescine, SWE : seaweeds extract

Table (2): Effect of salinity stress levels and antioxidants (presoaking and foliar spray) as well as their interactions on total Ascorbic acid (mg/100g F.w) in shoot and fruits of pepper plant during the two growing seasons (2007 & 2008).

	Plant	~~	gv .		<u></u>					•	
Treatment					linity lev						
	0.00	2000	4000	6000	Mean	0.00	2000	4000	6000	Mean	
	Season 2007				wean	Season 2008				wean	
Antioxidant (mg/l)	Pepper shoot										
Tap water	129	153	184	215	170	127	140	180	200	161	
HÁ (1000)	154	177	275	294	225	175	186	280	283	231	
SA (250)	164	203	280	290	234	160	189	277	280	226	
ASA (250)	175	245	300	364	271	167	200	288	294	237	
Put. (1)	180	200	300	329	252	175	200	288	315	244	
SWE (1000)	174	196	299	322	247	160	186	287	317	237	
Mean	163	194	261	289		162	183	260	275		
I CD at E%	Antioxidant:9.71 Salinit				y 8.73	73 Antioxidant: 8.71 Salinity: 9.					
LSD at 5%		Inter	action:	13.65	Interaction: 12.7						
				Pepper	fruits						
Tap water	79	86	123	150	109	80	99	104	143	106	
HÁ (1000)	107	164	171	174	154	110	152	155	168	146	
SA (250)	138	154	171	181	161	117	149	171	180	154	
ASA (250)	154	182	196	200	183	146	160	184	199	172	
Put. (1)	166	170	192	199	181	150	165	181	186	170	
SWE (1000)	141	165	183	189	169	158	170	192	199	179	
Mean	137	157	176	184		134	153	169	178	 I	
	Antioxidant: 7.71 Salinity 9.73 Antioxidant: 6.71 Salinity: 9.74										
LSD at 5%	Interac		2.65			Interac	tion: 13	3.7		y: 8.	

HA : humic acid, SA : Salicylic acid, ASA : Ascorbic acid, Put : Putrescine, SWE : seaweeds extract

Table (3): Effect of salinity stress levels and antioxidants(presoaking and foliar spray) as well as their interactions on glutathione (u mol/g.F.w in shoot and fruits of pepper plant during the two growing seasons (2007 & 2008).

	Salinity levels (mg/L)											
	0.00	2000	4000	6000	Mean	0.00	2000	4000 n 2008	6000	Mean		
		Seaso	n 2007		Wear			Wear				
Antioxidant (mg/l)	Pepper shoot											
Tap water	220	225	238	246	232	215	220	230	240	226		
HA (1000)	222	227	230	240	229	216	225	230	245	229		
SA (250)	227	232	235	242	234	220	230	230	245	231		
ASA (250)	245	260	280	240	256	235	240	275	280	257		
Put. (1)	225	231	250	257	240	215	225	240	250	232		
SWE (1000)	259	270	280	295	276	240	260	270	290	265		
Mean	230	237	251	254		222	230	243	255			
LSD at 5%	Antio	xidant:	8.71	Salinity	/: 5.73	Antiox	idant: 7	.71 4.8	8 Salinit	ty: 6.74		
LOD at J /0		Inter	action:	11.65	Interaction: 11.7							
				Pepper	fruits							
Tap water	226	231	244	252	238	221	220	236	246	230		
	228	233	236	246	235	222	225	236	251	233		
HA (1000)	233	238	241	240	040	226	000	236	246	236		
HA (1000) SA (250)				248	240	226	236					
HA (1000)	251	266	286	296	240	241	236	230	286	263		
HÁ (1000) SA (250) ASA (250) Put. (1)	251 231		286 256	296 263	274 246	241 221		281 246	286 256			
HA (1000) SA (250) ASA (250)	251 231	266	286	296 263 301	274	241	246	281	286 256 296	263		
HÁ (1000) SA (250) ASA (250) Put. (1)	251 231	266 237	286 256	296 263	274 246	241 221	246 231	281 246	286 256	263 238 271		
HA (1000)		230	241	240	240	220	230	230	240			

HA : humic acid, SA : Salicylic acid, ASA : Ascorbic acid, Put : Putrescine, SWE : seaweeds extract

Enzymatic antioxidants activity:

Data presented in tables (4,5) show that each of salinity stress levels (2000, 4000 or 6000 mg/l) increased Super oxide dismutase (SOD) and Ascorbic peroxidase (APX) activity in both shoot and fruits of pepper plants.

Table (4): Effect of salinity stress levels and applied antioxidants as well as their interaction on Superoxide dismutase(SOD) activity (units/g frech weight) in shoot and fruits of pepper plant during the two growing seasons (2007 & 2008).

Treatme	ent	Salinity levels (mg/L)										
Antioxidant (mg/l)		0.00	2000	4000	6000	Mean	0.00	2000	4000	6000	Mean	
			Seaso	n 2007		wear		Seaso		Weall		
			Pepper shoot									
Tap water		307	350	355	362	343	300	340	350	360	337	
HA (10)00)	310	335	347	358	337	325	340	350	360	343	
SA (25	50)	312	320	325	328	321	315	320	325	330	322	
ASA (25	0)	314	350	360	390	353	330	360	370	400	365	
Put. (1)		313	318	325	330	321	320	325	330	335	327	
SWE	(1000)	310	360	380	390	361	330	370	390	410	375	
Mean	· ·	310	332	341	351		316	336	345	356		
LSD at 5%	, D	Antio	oxidant: Inter	n.s raction:	Salinity 1.65	7: 0.73 Antioxidant:5.71 4.8 Salinity : 8.74 Interaction: 11.7						
					Pepper	fruits						
Tap water		315	358	363	370	351	308	348	358	368	345	
)00)	318	343	355	366	345	333	348	358	368	351	
SA (25	50)	320	328	333	336	329	323	328	333	338	330	
ASA (25	0)	322	358	368	398	361	383	368	378	408	384	
Put. (1)		321	326	332	338	329	328	333	338	343	335	
SWE	(1000)	323	368	388	398	414	338	378	398	418	3830	
Mean		318	340	349	359		329	344	353	364		
LSD at 5% Antioxidant: 7 Interaction: 12			.71 2.65	Salini	ty: 8.73	Antioxi Interac		.71 5.7	Salinity	y: 9.74		

HA : humic acid, SA : Salicylic acid, ASA : Ascorbic acid, Put : Putrescine, SWE : seaweeds extract

Table (5): Effect of salinity stress levels and applied antioxidants as well as their interaction on ascorbic peroxidase (APX) activity (units/g protein/min) in shoot and fruits of pepper plant during the two growing seasons (2007 & 2008).

Treatment	Salinity levels (mg/L)											
	0.00	2000	4000	6000	Mean	0.00	2000	4000	6000	Mean		
		Seaso	n 2007		Wear			Wear				
Antioxidant (mg/l)	Pepper shoot											
Tap water	125	133	137	141	134	120	125	130	140	128		
HA (1000)	130	135	140	146	137	135	139	142	148	141		
SA (250)	130	145	150	154	144	130	145	150	155	145		
ASA (250)	135	155	160	170	155	139	160	165	170	158		
Put. (1)	132	135	140	150	139	130	135	140	145	137		
SWE (1000)	140	148	155	160	150	150	155	162	170	159		
Mean	131	139	145	152		131	139	144	152			
LSD at 5%	Antic	xidant:	1.71	Salinity	/:2.73	Antioxidant:1.71 4.8 Salinity: 2.74						
LOD at 5%		Inter	raction:	3.65		Interaction: 3.7						
				Pepper	fruits							
Tap water	131	139	143	147	140	126	131	136	146	134		
HA (1000)	136	141	146	152	143	141	145	148	154	147		
SA (250)	136	151	156	160	150	136	151	156	161	151		
ASA (250)	141	161	166	176	161	145	166	171	176	164		
Put. (1)	138	141	148	156	145	136	141	146	151	143		
SWE (1000)	146	154	161	166	156	156	161	168	176	165		
Mean	137	145	151	158		137	145	150	158			
LSD at 5%	Antio	Antioxidant: 4.71 Salinity Interaction: 7.65				Antioxidant:4.71 Salinity: 5.74 Interaction: 7.7						
HA : humic acid, SA : Salicylic acid, ASA : Ascorbic acid, Put : Putrescine								rescine				

SWE : seaweeds extract

In addition The data show that any of applied antioxidants (HA, SA, ASA, Tochopherol, GSH, Citric , Put., SWE) alone or combined with different salinity stress levels also increased the activity of SOD, APX on enzymes in both shoot and fruits of pepper plant, during the two growth seasons.

DISCUSSION

Enzymatic antioxidant activity :

The metabolism of active oxygen, peroxidases in plants are also involved in the biosynthesis of cell wall including lignification and suberization. Considerable evidence shows that high peroxidase activity is correlated with the reduction of plant growth. This might be attributed to peroxidase catalysis of ferulic acid conversion to diferulic acid on polysaccharidcs, the feruloylation of hemicelluloses, or the insolubilization of hydroxyproline-rich glycoprotein causing cell wall stiffening. Morphologically, the most typical symptom of saline injury to a plant is retarded growth due to inhibition of cell elongation, resulting in a stunted plant.

To mitigate and repair damage initiated by AOS. Treated plants have developed a complex antioxidant system. The primary components of this

system include carotenoids, ascorbate, glutathione and tocopherols, and enzymes such as superoxidase dismutase(SOD), catalase(CAT), glutathione peroxidase(GPX), peroxidase(APX) and the enzymes involved in the ascorbate glutathione cycle. Ascorbate peroxidase, dehydroascorbate reductase, monodehydroascorbate reductase and glutathione reductase. Many components of these antioxidant defence system can be found in different subcelluler compartments.

Vaidyanathan, *et al.*, (2003). Plants posses antioxidant systems in the form of enzymes such as SOD, APX, GR, DIIAR, CAT and metabolites viz., ascorbic acid, glutathione, α -tocopherol, carotenoid, flavanoids etc. Smirnoff, (1995). These antioxidant enzymes and metabolites are reported to increase under various environmental stresses Yu and Rengel, (1999) as well as comparatively higher activity has been reported in tolerant cultivars than the susceptible ones , Sairam, *et al.*, (2002).

Non enzymatic antioxidants :

1- Ascorbic acid (AA) contents:

The non-enzymatic plant antioxidants can be classified into two major types: (1) AA-like scavengers, and (2) pigments such as carotenoids. ASA is an important antioxidant, which reacts not only with H₂O₂ but also with O₂, OH and lipid hydroperoxidases. On the other hand, ASA has been implicated in several types of biological activities in plants: (1) as an enzyme co-factor, (2) as an antioxidant, and (3) as a donor/ acceptor in electron transport at the plasma membrane or in the chloroplasts, all of which are related to oxidative stress resistance (Conklin, 2001). APX uses ASA and oxidizes it to monodehydroascorbate (MDA). MDA may give rise to dehydroascorbate (DHA). Both MDA and DHA will then be reduced to regenerate the ascorbate pool. This type of scavenging is thought to occur near PSI, thereby minimizing the risk of escape and reaction of ROS with each other (Foyer and Noctor, 2000). ASA is a major metabolite in chloroplasts of higher plants and represents about 10% of the soluble carbohydrate pool in leaves (Noctor and Foyer, 1998).

2- Total phenol contents:

Phenolics are diverse secondary metabolites (flavonoids, tannins, hydroxycinnamate esters and lignin) abundant in plant tissues (Grace and Logan, 2000). Polyphenols possess ideal structural chemistry for free radical scavenging activity, and they have been shown to be more effective antioxidants in vitro than tocopherols and ascorbate.

Another mechanism underlying the antioxidativc properties of phenolics is the ability of flavonoids to alter peroxidation kinetics by modification of the lipid packing order and to decrease fluidity of the membranes (Amor, *et al.*, 2000). These changes could sterically hinder diffusion of free radicals and restrict peroxidative reactions. Moreover, it has been shown recently that phenolic compounds can be involved in the hydrogen peroxide scavenging cascade in plant cells.

Phenolic compounds retard or inhibit lipid autoxidation by acting as radical scavengers (Namiki, 1990) and, consequently, are essential antioxidants that protect against propagation of the oxidative chain.

Total phenols in pepper tended increase gradually with increasing salinity levels in soil. This increase showed some tendency to adjust osmotically against salt stress. Moreover, stress condition leads to an increase in phenolic compounds (Amor, *et al.*, 2000). These phenolic compounds could be a cellular adaptive mechanism for scavenging oxygen free radicals during stress and this free radical scavenger and others such as ascorbate could be readily oxidized in the system of tissue representing subcelluler damages. In addition, Kurup, *et al.*, (1994), pointed out that stressed marigold plants contained higher concentrations of free phenolic compounds, due to enhanced biosynthesis and formation of lignin than controls.

Reduced Glutathion contents:

GSH can function as an antioxidant in many ways. (1)It can react chemically with singlet oxygen, superoxide and hydroxyl radicals and therefore function directly as a free radical scavenger Noctor and Foyer, 1998). (2) GSH may stabilise membrane structure by removing acyl peroxides formed by lipid peroxidation reactions (Price, *et al.*, 1990). (3)GSH is the reducing agent that recycles ascorbic acid from its oxidised to its reduced form by the enzyme dehydroascorbate reductase (Loewus, 1988). (4) GSH can also reduce dehydroascorbate by a non-enzymatic mechanism at pH > 7 and GSH concentrations greater than 1 mM. This may be a significant pathway in chloroplasts whose stromal pH in the light is about 8 and GSH concentrations may be as high as 5 mM (Foyer and Halliwell, 1976).

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تأثير المعاملة بمضادات الأكسدة على المحتوى الداخلى لمضادات الأكسدة الإنزيمية وغير الإنزيمية فى نبات الفلفل النامى تحت ظروف الاجهاد الملحى محب طه صقر و رضا سيد أحمد متولى قسم النبات الزراعى – كلية الزراعة – جامعة المنصورة

أجريت تجربتى أصص فى مزرعة محطة البحوث بكلية الزراعة حجامعة المنصورة فى الموسم الصيفى لعامى ٢٠٠٧ ٢٠٠٨ لدراسة دور بعض مضادات الأكسدة (معاملة خارجية) على المحتوى الداخلى لمضادات الأكسدة الإنزيمية و غي الإنزيمية فى نبات الفلفل صنف كاليفورنيا النامى تحت ظروف الإجهاد الملحى .

أظهرت النتائج أن كلاً من مستويات الاجهاد الملحى المختلفة وكذلك المعاملة نقعا ورشا بأى من مضادات الأكسدة وكذلك معاملات التداخل من مستويات الاجهاد الملحى و مضادات الأكسدة أدت الى زيادات واضحة فى المحتوى الداخلى من مضادات الأكسدة غير الانزيمية (الفينولات – الاسكوربيك – الجلوتاتيون) فى كل من المجموع الخضرى والثمار لنبات الفلفل خلال موسمى الزراعة.

ولوحظ أن المعاملة SWE (مستخلص أعشاب البحر) بالتداخل مع المستوى المرتفع من الاجهاد الملحي (٢٠٠٠ مج/لتر) كانت المعاملة الاكثر فاعلية فيهذا الشأن .

ولوحظ أن مستويات الأجهاد الملحى المختلفة (٢٠٠٠ & ٤٠٠٠ مج /لتر) أدت ولوحظ أن مستويات الأجهاد الملحى المختلفة (٩٠٠٠ م ٤٠٠٠ م م التر) أدت اللى زيادة النشاط الانزيمي لكلب من انزيمي SOD (سوبر أكسيد ديزميوتيز) & APX (أسكوربيك بيروكسيديز) في كل من المجموع الخضري وثمار نبات الفلفل كما لوحظ أن المعاملة بأي من مضات الاكسدة (هيوميك – سالسيليك – توكوفيرول –

حما لوحظ أن المعاملة بأى من مضات الأكسدة (هيوميك – سالسيليك – توكوفيرول – بتروسين - مستخلص أعشاب البحر) منفردة أو بالتداخل مع أى من مستويات الأجهاد الملحى المختلفة إلى زيادة النشاط الانزيميل لانزيمى SOD) & APX فى كل من المجموع الخضرى وثمار الفلفل صنف كاليفورنيا خلال موسمى ٢٠٠٧ & ٢٠٠٨.