

DESIGN, SYNTHESIS AND BIOLOGICAL EVALUATION OF NOVEL [1,2,4]TRIAZOLO[4,3-A] QUINOXALINONES

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ABSTRACT

A series of novel [1,2,4]triazolo[4,3-*a*]quinoxalinone derivatives have been synthesized. The structures of the newly synthesized compounds were confirmed by elemental analysis and different spectral techniques. The *in vitro* antimicrobial screening of some selected compounds was carried out using the diffusion agar technique. Compound **14_c** was found to be the most active antibacterial agent against *Escherichia coli* using gentamicin as reference drug. All the tested compounds showed no activity against *Candida albicans* and *Pseudomonas aeruginosa*.

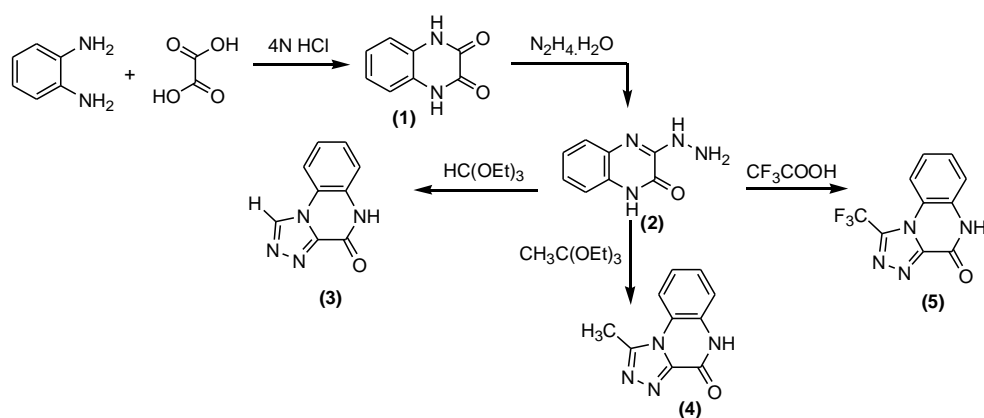
Introduction

Although antibacterial agents are life saving drugs, yet they suffer from serious drawbacks. The most important of which is anaphylaxis and developing resistance which implement a potential global health crisis. It is recommended to use new antibacterial agents with enhanced broad-spectrum potency. Therefore, recent efforts have been directed toward exploring novel antibacterial agents. Quinoxaline derivatives are important components of several pharmacologically active compounds [Tandon *et al.*, 2006; Sarges *et al.*, 1990; Sakata *et al.*, 1988; Arthur *et al.*, 2005; Seitz *et al.*, 2002; Szekelyhidi *et al.*, 2005]. Although rarely described in nature, synthetic quinoxaline ring is part of number of antibiotics such as echinomycin and actinomycin, which are known to inhibit the growth of Gram-positive bacteria and are also active against various transplantable tumors [Myers *et al.*, 2003; Iwashita *et al.*, 2005; Morokata *et al.*, 2004]. Substituted quinoxaline have received considerable attention during last decade as they are endowed with variety of biological activities and have wide range of therapeutic properties such as antibacterial, antiviral, anticancer, antifungal, anthelmintic, and insecticidal [Badran *et al.*, 2003; Catarzi *et al.*, 2004; Gali-Muhtasib *et al.*, 2005; Perez-Melero *et al.*, 2004; Kim *et al.*, 2004; Toshima *et al.*, 2004]. Relying on this evidence we became motivated to synthesize some novel [1,2,4]triazolo[4,3-*a*]quinoxalinone derivatives in an attempt to find an effective antimicrobial agent.

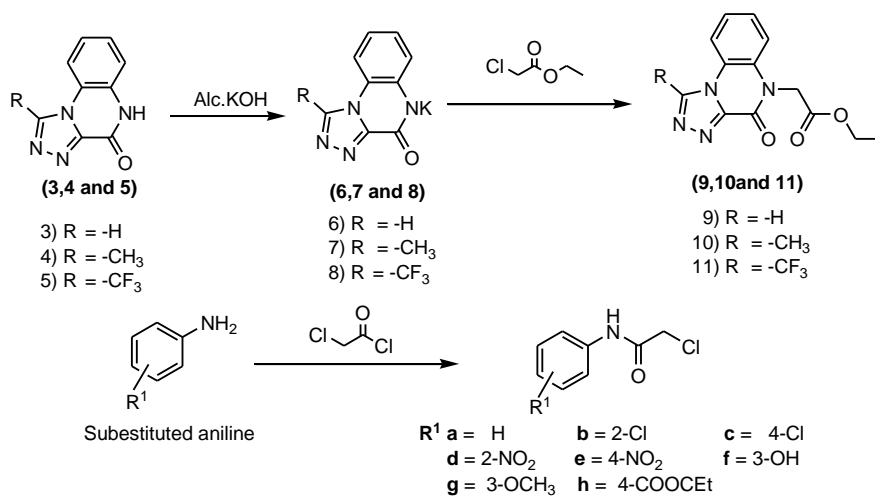
Results and Discussion

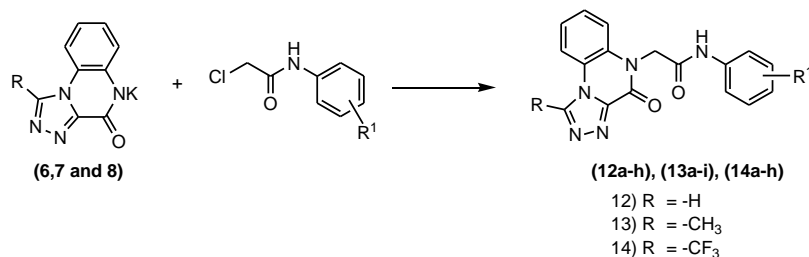
The present work involves preparation of some novel [1,2,4]triazolo[4,3-*a*]quinoxalinone derivatives. Achieving this goal was made by adopting the synthetic pathway depicted in **schemes I and II**. The main intermediate in this work is 3-hydrazinylquinoxalinone **2** which has been synthesized in two steps starting from *O*-phenylenediamine by treating it with oxalic acid to produce quinoxalinedione **1**. Reacting **1** with hydrazine hydrate afforded the hydrazinylquinoxalinone **2**. Triazoloquinoxalinone derivatives **3**, **4** and **5** have been prepared from compound **2** using established procedures **Scheme I**. Potassium salts **6**, **7** and **8** were obtained after treatment of compounds **3**, **4** and **5** with alcoholic KOH in quantitative yields. Reaction of potassium salts **6**, **7** and **8** with ethylchloroacetate gave the corresponding esters **9**, **10** and **11** respectively. Reaction of potassium salts **6**, **7** and **8** with chloroacetanilides produced the corresponding anilides **12a-h**, **13a-i** and **14a-h** **Scheme II**. The IR spectra of these compounds are characterized by appearance of amid carbonyl at 1681-1681 cm^{-1} and amidic -NH at 3245- 3448 cm^{-1} .

The structural assignments to the new compounds were based on their elemental analysis and spectral (IR, ^1H NMR and mass) data.



Scheme I





Scheme II

Experimental

All melting points were taken on Stuart SMP3 digital melting point apparatus and Gallen Kamp apparatus and were uncorrected. IR spectra were recorded on Pye Unicam SP 1000 IR spectrophotometer at Microanalytical Center, Cairo University. The ¹H NMR spectra were recorded in DMSO-d₆ at 300 MHz on a Varian Mercury VXR-300 NMR spectrometer at Research Services Unit, Faculty of Science, Cairo University. Chemical shifts were related to those of the solvent. TMS was used as a standard. Mass spectra were recorded on Hewlett Packard 5988 spectrometer at Regional Center for Mycology and Biotechnology, Al-Azhar University. Microanalysis was carried out at Microanalytical Center, Cairo University. Reactions progresses were monitored by TLC using TLC sheets precoated with UV fluorescent silica gel Merck 60 F254 plates and were visualized using UV lamp.

According to certain reported procedures, the following derivatives were prepared:

Quinoxaline-2,3(1*H*,4*H*)-dione(Romer, 2009), 3-hydrazinylquinoxalin-2(1*H*)-one(Newbold *et al.*, 1948), [1,2,4]triazolo[4,3-*a*]quinoxalin-4(5*H*)-one(Galal *et al.*, 2013), 1-methyl-[1,2,4]triazolo[4,3-*a*]quinoxalin-4(5*H*)-one(Rashed *et al.*, 1990) and 1-(trifluoromethyl)-[1,2,4]triazolo[4,3-*a*]quinoxalin-4(5*H*)-one (Bharat *et al.*, 1988), Chloroacetilides (El-Said *et al.*, 1991).

Potassium salt of [1,2,4]triazolo[4,3-*a*]quinoxalin-4(5*H*)-one (6), Potassium salt of 1-methyl-[1,2,4]triazolo[4,3-*a*]quinoxalin-4(5*H*)-one (7) and Potassium salt of 1-(trifluoromethyl)-[1,2,4]triazolo[4,3-*a*]quinoxalin-4(5*H*)-one (8):

To a solution of the appropriate (1,2,4)-triazolo[4,3-*a*]quinazolinone of type **3**, **4** and **5** (0.01mol each) in absolute ethanol, solution KOH (0.01mol) in absolute ethanol was added and the reaction mixture was heated to 78°C with stirring for 15minutes. The produced solid was collected by filtration, washed several times with absolute ethanol and dried producing the target compounds in quantitative yield.

Ethyl 2-(4-oxo-[1,2,4]triazolo[4,3-*a*]quinoxalin-5(4*H*)-yl)acetate (9):

Ethyl 2-(1-methyl-4-oxo-[1,2,4]triazolo[4,3-*a*]quinoxalin-5(4*H*)-yl)acetate (10):

Ethyl 2-(4-oxo-1-(trifluoromethyl)-[1,2,4]triazolo[4,3-*a*]quinoxalin-5(4*H*)-yl)acetate(11):

General procedure

In a conical flask equimolar quantities (0.01mol each) of the suitable salt **6** or **7** or **8** and ethyl chloroacetate were suspended in (20ml) DMF and heated on water bath for 3h. The reaction mixture was poured onto ice water (200 ml), and stirred for 30min. The obtained solid was filtered and crystallized from acetone: methanol (1:5) mixture.

2-(4-oxo-[1,2,4]triazolo[4,3-a]quinoxalin-5(4H)-yl)-N-substituted phenylacetamide(12a-h):

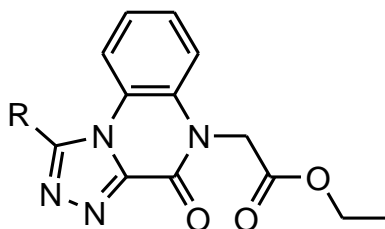
2-(1-methyl-4-oxo-[1,2,4]triazolo[4,3-a]quinoxalin-5(4H)-yl)-N-substituted phenylacetamide (13a-i):

2-(4-oxo-1-(trifluoromethyl)-[1,2,4]triazolo[4,3-a]quinoxalin-5(4H)-yl)-N-substituted phenylacetamide (14a-h):

General procedure

Equimolar quantities of the appropriate potassium salt **6** or **7** or **8** (0.01 mol) and the appropriate chloroacetatnilide (0.01 mol) in DMF (20 ml) were heated on water bath for 3h. The reaction mixture was poured onto ice water (200 ml), and stirred for 30 min. The obtained solid was filtered and crystallized from aqueous ethanol.

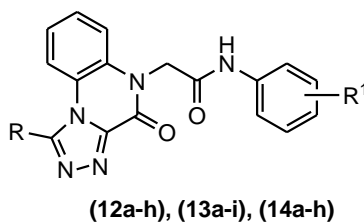
Table 1: Physical and analytical data of the synthesized compounds:



(9,10and 11)

No	R	Yield %	m.p. °C	M. Formula/ M. Wt	Analysis % Calcd /found		
					C	H	N
9	H	35	210-12	C ₁₃ H ₁₂ N ₄ O ₃ 272.2	57.35	4.44	20.58
					57.35	4.34	20.60
10	CH ₃	76	261-62	C ₁₄ H ₁₄ N ₄ O ₃ 286.2	58.73	4.93	19.57
					58.57	5.21	19.39
11	CF ₃	83	220-22	C ₁₄ H ₁₁ F ₃ N ₄ O ₃ 340.2	49.42	3.26	16.47
					49.22	3.15	16.50

Table 2: Physical and analytical data of the synthesized compounds:



No	R	R ¹	m.p. °C	Yield %	M. Formula/M.Wt	Analysis % Calcd/found		
						C	H	N
12a	H	H	316-317	70	C ₁₇ H ₁₃ N ₅ O ₂	63.94	4.10	21.93
					319.3	64.00	4.31	21.54
12b	H	2-Cl	325-326	65	C ₁₇ H ₁₂ ClN ₅ O ₂	57.72	3.42	19.80
					353.7	57.83	3.46	19.96
12c	H	4-Cl	314-315	82	C ₁₇ H ₁₂ ClN ₅ O ₂	57.72	3.42	19.80
					353.7	57.85	3.48	19.93
12d	H	2-NO ₂	284-386	63	C ₁₇ H ₁₂ N ₆ O ₄	56.05	3.32	23.07
					364.3	56.21	3.10	22.98
12e	H	4-NO ₂	345-346	57	C ₁₇ H ₁₂ N ₆ O ₄	56.05	3.32	23.07
					364.3	56.19	3.39	23.30
12f	H	3-OH	264-266	64	C ₁₇ H ₁₃ N ₅ O ₃	60.89	3.91	20.89
					335.3	61.07	3.94	21.04
12g	H	3-OCH ₃	292-295	63	C ₁₈ H ₁₅ N ₅ O ₃	61.89	4.33	20.05
					349.3	61.97	4.39	20.13
12h	H	4-COOEt	303-304	85	C ₂₀ H ₁₇ N ₅ O ₄	61.38	4.38	17.89
					391.3	61.47	4.46	18.02
13a	CH ₃	H	285-287	51	C ₁₈ H ₁₅ N ₅ O ₂	64.86	4.54	21.01
					333.3	64.97	4.59	21.13
13b	CH ₃	2-Cl	332-333	66	C ₁₈ H ₁₄ ClN ₅ O ₂	58.78	3.84	19.04
					367.7	58.63	3.89	19.17
13c	CH ₃	4-Cl	345-346	56	C ₁₈ H ₁₄ ClN ₅ O ₂	58.78	3.84	19.04
					367.7	58.86	3.86	19.17
13d	CH ₃	2-NO ₂	298-300	65	C ₁₈ H ₁₄ N ₆ O ₄	57.14	3.73	22.21
					378.3	57.25	3.81	22.36
13e	CH ₃	4-NO ₂	330-331	63	C ₁₈ H ₁₄ N ₆ O ₄	57.14	3.73	22.21
					378.3	57.05	3.65	22.09
13f	CH ₃	3-OH	327-328	55	C ₁₈ H ₁₅ N ₅ O ₃	61.89	4.33	20.05
					349.3	61.98	4.35	20.16
13g	CH ₃	3-OCH ₃	270-272	53	C ₁₉ H ₁₇ N ₅ O ₃	62.80	4.72	19.27
					363.3	62.87	4.78	19.35
13h	CH ₃	4-COOEt	300-302	70	C ₂₁ H ₁₉ N ₅ O ₄	62.22	4.72	17.27
					405.4	62.19	4.53	17.32
13i	CH ₃	4-COOMe	291-293	69	C ₂₀ H ₁₇ N ₅ O ₄	61.38	4.38	17.89
					391.3	61.09	4.42	17.63
14a	CF ₃	H	283-284	58	C ₁₈ H ₁₂ F ₃ N ₅ O ₂	55.82	3.12	18.08
					387.3	55.91	3.18	18.11
14b	CF ₃	2-Cl	237-238	56	C ₁₈ H ₁₁ ClF ₃ N ₅ O ₂	51.26	2.63	16.61
					421.7	51.38	2.97	16.34
14c	CF ₃	4-Cl	283-285	55	C ₁₈ H ₁₁ ClF ₃ N ₅ O ₂	51.26	2.63	16.61
					421.7	51.42	2.60	16.69
14d	CF ₃	2-NO ₂	279-280	60	C ₁₈ H ₁₁ F ₃ N ₆ O ₄	50.01	2.56	19.44
					432.3	50.21	2.45	19.31
14e	CF ₃	4-NO ₂	296-297	65	C ₁₈ H ₁₁ F ₃ N ₆ O ₄	50.01	2.56	19.44
					432.3	50.13	2.62	19.57

14f	CF ₃	3-OH	288-282	71	C ₁₈ H ₁₂ F ₃ N ₅ O ₃ 403.3	53.60 53.71	3.00 2.98	17.36 17.43
14g	CF ₃	3-OCH ₃	270-271	66	C ₁₉ H ₁₄ F ₃ N ₅ O ₃ 417.3	54.68 54.81	3.38 3.35	16.78 16.89
14h	CF ₃	4-COOEt	245-246	60	C ₂₁ H ₁₆ F ₃ N ₅ O ₄ 459.3	54.91 54.99	3.51 3.59	15.25 15.34

Table 3: Spectral data of the prepared compounds:

No		IR (KBr cm ⁻¹), ¹ H-NMR (DMSO-d ₆ , 300 MHz, δ ppm), MS M/Z (% relative abundance)
9	IR	1681(CO ring), 1732(CO ester), 2994(CH-aliphatic), 3118(CH-aromatic).
	¹ H-NMR	1.19(t, 3H, CH ₂ -CH ₃), 4.15(q, 2H, CH ₂ -CH ₃), 5.15(s, 2H, -N-CH ₂ -C=O), 7.52-8.26(m, 4H, ArH), 9.94(s, 1H, triazole ring).
	MS	272(M+1, 24.43%), 90(100 %)
10	¹ H-NMR	1.22(t, 3H, CH ₂ -CH ₃), 3.3(s, 3H, CH ₃ at position 1 of triazole ring), 4.2(q, 2H, CH ₂ -CH ₃), 5.15(s, 2H, -N-CH ₂ -C=O), 7.44-8.18(m, 4H, ArH).
12a	¹ H-NMR	5.15(s, 2H, -N-CH ₂ -C=O), 7.04-7.59(m, 3H, ArH), 8.23(d, 1H, ArH), 9.98(s, 1H, position 1 of triazole ring), 10.31(s, 1H, -CO -NH-Ph).
12b	IR	755(C-Cl), 1681(CO), 2957(CH-aliphatic), 3109(CH-aromatic), 3245(-NH-).
	¹ H-NMR	5.23(s, 2H, -N-CH ₂ -C=O), 7.19-7.68(m, 7H, ArH), 8.29(d, 1H, ArH), 9.98(s, 1H, position 1 of triazole ring), 10.00(s, 1H, -CO -NH-Ph).
	MS	353(M, 10.32%), 354(M+1, 10.77%), 227(100 % base), 199(92.63%).
12c	IR	752(C-Cl), 1688(CO), 3126(CH-aromatic), 3448(-NH-).
	¹ H-NMR	5.15(s, 2H, -N-CH ₂ -C=O), 7.26-7.63(m, 7H, ArH), 8.28(d, 1H, ArH), 9.98(s, 1H, position 1 of triazole ring), 10.46(s, 1H, -CO -NH-Ph).
	MS	353(M, 9.87%), 355(M+2, 6.14%), 227(100 % base), 199(96.58%).
12e	IR	1338,1566(sym. and asym. stretching of NO ₂), 1680(CO), 3127(CH-aromatic), 3460(-NH-).
	¹ H-NMR	5.23(s, 2H, -N-CH ₂ -C=O), 7.26-7.62(m, 3H, ArH), 7.79-7.87(m, 2H, ArH), 8.21-8.32(m, 3H, ArH), 9.99(s, 1H, position 1 of triazole ring), 10.98(s, 1H, -CO -NH-Ph).
12f	IR	1675(CO), 3100(CH-aromatic), 3213(m, broad-OH), 3592(w -NH-).
	¹ H-NMR	5.14(s, 2H, -N-CH ₂ -C=O), 6.43(d, 1H, ArH), 6.90-7.52(m, 6H, ArH), 8.27(d, 1H, ArH), 9.45(s, 1H, m-OH of the acetanilide group), 9.97(s, 1H, position 1 of triazole ring), 10.24(s, 1H, -CO -NH-Ph).
12g	IR	1682(CO), 2972(CH-aliphatic), 3108(CH-aromatic), 3450(-NH-).
	¹ H-NMR	3.71(s, 3H, -OCH ₃), 5.51(s, 2H, -N-CH ₂ -C=O), 6.66(d, 1H, ArH), 7.56-7.07(m, 6H, ArH), 8.29(d, 1H, ArH), 9.98(s, 1H, at position 1 of the triazole ring), 10.33(s, 1H, -CO -NH-Ph).
	MS	349(M ⁺ , 21.33%), 350(M+1, 5.01%), 227(38.48%), 199(64.0%), 123(100 % base).
12h	IR	1690(CO), 2991(CH-aliphatic), 3114(CH-aromatic), 3285(NH).
	¹ H-NMR	1.3(t, 3H, -OCH ₂ CH ₃), 4.28(q, 2H, -OCH ₂ CH ₃), 5.2(s, 2H, -N-CH ₂ -C=O), 7.42-7.61(m, 3H, ArH), 7.72(d, 2H, ArH), 7.93(d, 2H, ArH), 8.3(d, 1H, ArH), 9.98(s, 1H, position 1 of triazole ring), 10.7(s, 1H, -CO -NH-Ph).
	MS	391(M ⁺ , 2.95%), 199(0.16%), 119(20.75.0%), 92(34.52 %), 91(38.29 %), 54(100% base).
13a	IR	1681(CO), 2954(CH-aliphatic), 3084(CH-aromatic), 3526(-NH-).
	¹ H-NMR	3.04(s, 3H, CH ₃ at position 1 of triazole ring), 5.15(s, 2H, -N-CH ₂ -C=O), 7.07(t, 1H, ArH), 7.29-7.57(m, 7H, ArH), 8.17(d, 1H, ArH), 10.34(s, 1H, -CO -NH-Ph).
	MS	333(M, 2.3%), 241(100 % base), 213(87.95%).
13b	IR	754(C-Cl), 1675(CO), 3043, 3114(CH-aromatic), 3227(-NH-).

	¹ H-NMR	3.04(s, 3H, CH ₃ at position 1 of triazole ring), 5.23(s, 2H, -N-CH ₂ -C=O), 7.21-7.64(m, 7H, ArH _s), 8.17(d, 1H, ArH), 10.00(s, 1H, -CO-NH-Ph).
	MS	367(M, 10.4%), 368(M+2, 2.4%), 242(34.02%), 213(34.27%), 91(100% base).
13c	IR	749(C-Cl), 1662, 1689(CO), 2928(CH-aliphatic), 3118(CH-aromatic), 3323(-NH-).
	¹ H-NMR	3.04(s, 3H, CH ₃ at position 1 of triazole ring), 5.15(s, 2H, -N-CH ₂ -C=O), 7.35-7.6(m, 7H, ArH _s), 8.17(d, 1H, ArH), 10.48(s, 1H, -CO-NH-Ph).
	MS	367(M, 6.2%), 368(M+2, 2.4%), 241(100% base), 213(71.85%), 116(90.9%).
13d	IR	1340, 15139(sym. and asym. stretching of -NO ₂), 1688(CO), 2961(CH-aliphatic), 3111(CH-aromatic), 3339(-NH-).
	¹ H-NMR	3.04(s, 3H, CH ₃ at position 1 of triazole ring), 5.21(s, 2H, -N-CH ₂ -C=O), 7.37-7.79(m, 6H, ArH _s), 7.95(d, 1H, ArH), 8.17(d, 1H, ArH), 10.64(s, 1H, -CO-NH-Ph).
13f	IR	1670(CO), 3109(CH-aromatic), 3351(-OH), 3424(-NH-).
	¹ H-NMR	3.04(s, 3H, CH ₃ at position 1 of triazole ring), 5.12(s, 2H, -N-CH ₂ -C=O), 6.47(d, 1H, ArH), 6.92-7.1(m, 3H, ArH _s), 7.39-7.55(m, 3H, ArH _s), 8.17(d, 1H, ArH), 9.38(s, 1H, m-OH of the acetanilide group), 10.21(s, 1H, -CO-NH-Ph).
13g	IR	1683(CO), 2971(CH-aliphatic), 3144(CH-aromatic), 3311(-NH-).
	¹ H-NMR	3.04(s, 3H, CH ₃ at position 1 of triazole ring), 3.07(s, 3H, -OCH ₃), 5.15(s, 2H, -N-CH ₂ -C=O), 6.65(d, 1H, ArH), 7.07-7.55(m, 6H, ArH _s), 8.17(d, 1H, ArH), 10.35(s, 1H, -CO-NH-Ph).
13h	¹ H-NMR	1.3(t, 3H, -OCH ₂ CH ₃), 2.48(s, 3H, CH ₃ in position 1 of the triazole ring, overlapped with DMSO peak), 4.29(q, 2H, -OCH ₂ CH ₃), 5.24(s, 2H, -N-CH ₂ -C=O), 7.69-7.95(m, 8H, ArH), 10.7(s, 1H, -CO-NH-Ph).
13i	¹ H-NMR	3.3(s, 3H, CH ₃ in position 1 of the triazole ring, coincided with DMSO peak), 5.24(s, 2H, -N-CH ₂ -C=O), 7.69-7.95(m, 8H, ArH), 10.7(s, 1H, -CO-NH-Ph).
14a	¹ H-NMR	5.2(s, 2H, -N-CH ₂ -C=O), 7.06-7.69(m, 8H, ArH), 7.93(d, 1H, ArH), 10.32(s, 1H, -CO-NH-Ph).
14b	IR	1168(CO), 3263(CH-aromatic), 3470(-NH-).
	¹ H-NMR	5.28(s, 2H, -N-CH ₂ -C=O), 7.2-7.38(m, 2H, ArH), 7.5-7.38(m, 5H, ArH), 7.93(d, 1H, ArH), 9.97(s, 1H, -CO-NH-Ph).
	MS	421(M, 5.03%), 239(61.96%), 127(100% base).
14c	IR	1667(CO), 3194(CH-aromatic), 3320(-NH-).
	¹ H-NMR	5.2(s, 2H, -N-CH ₂ -C=O), 7.18-7.69(m, 7H, ArH), 7.93(d, 1H, ArH), 10.48(s, 1H, -CO-NH-Ph).
14d	IR	1352&1510(sym. and asym. stretching of -NO ₂), 1694(CO), 2990(CH-aliphatic), 3093(CH-aromatic), 3297(-NH-).
14e	IR	1342, 1509(sym. and asym. stretching of -NO ₂), 1692(CO), 2968(CH-aliphatic), 3157(CH-aromatic), 3480(-NH-).
	¹ H-NMR	5.28(s, 2H, -N-CH ₂ -C=O), 7.29-7.95(m, 6H, ArH), 8.23(d, 2H, ArH), 11.02(s, 1H, -CO-NH-Ph).
	MS	432(M, 0.01%), 295(100% base), 267(94.15%).
14f	IR	1667(CO), 3152(CH-aromatic), 3318(b, -OH and -NH overlapped).
	¹ H-NMR	5.18(s, 2H, -N-CH ₂ -C=O), 6.49(d, 1H, ArH), 6.5-7.68(m, 6H, ArH), 7.93(d, 1H, ArH), 9.39(s, 1H, m-OH of the acetanilide group), 10.18(s, 1H, -CO-NH-Ph).
14g	IR	1689(CO), 2847, 2966(CH-aliphatic), 3090, 3150(CH-aromatic), 3314(-NH-).
	¹ H-NMR	3.67(s, 3H, -OCH ₃), 5.16(s, 2H, -N-CH ₂ -C=O), 6.62-7.91(m, 8H, ArH), 10.36(s, 1H, -CO-NH-Ph).
	MS	417(M ⁺ , 3.45%), 289(100% base), 274(93.76%), 238(66.31%), 112(23.96%).
14h	IR	1602(C=C), 1694(CO), 2998(CH-aliphatic), 3196(CH-aromatic), 3458(-NH-).
	¹ H-NMR	1.3(t, 3H, -OCH ₂ CH ₃), 4.28(q, 2H, -OCH ₂ CH ₃), 5.12(s, 2H, -N-CH ₂ -C=O), 7.28-7.95(m, 8H, ArH), 10.74(s, 1H, -CO-NH-Ph).
	MS	459(M, 3.93%), 267(100% base), 239(40.15%), 165(46.18%), 119(61.75%).

Antimicrobial activity

Preliminary antibacterial and antifungal activity were performed for selected compounds against various types of bacteria and fungi, namely:

1. *Streptococcus pneumonia* (RCMB 010010) (Gram positive bacteria)
2. *Bacillus subtilis* (RCMB 010067) (Gram positive spore-forming bacteria)
3. *Escherichia coli* (RCMB 010052) (Gram negative bacteria)
4. *Pseudomonas aeruginosa* (RCMB 010043) (Gram negative bacteria)
5. *Aspergillus fumigatus* (RCMB 02568) (a representative of fungi).
6. *Syncephalastrum racemosum* (RCMB 05922) (a representative of fungi).
7. *Geotricum candidum* (RCMB 05097) (a representative of fungi).
8. *Candida albicans* (RCMB 05036) (a representative of fungi).

Materials

Culture media

Nutrient broth, Sabourauds broth and Nutrient agar were the products of Oxoid ltd., England.

Methodology: the agar plate disc-diffusion technique (Collins, 1964)

Sterilized filter paper discs (6 mm in diameter) were wetted each with 100 μ l of a solution of the tested compound containing 2mg/ml in DMF and the discs were allowed to air dry. The discs were then placed onto the surface of agar plates (nutrient agar for bacteria and sabourauds dextrose agar for fungi) seeded with the test organism. Each plate contained 15 ml of the agar medium, previously seeded with 0.2 ml of 18 hours broth culture of each organism. The inoculated plates were incubated at 37 °C for 48 hours and the mean zone of inhibition \pm standard deviation beyond well diameter were measured in mm. Discs impregnated with DMF were used as control. The antibacterial reference Ampicillin and Gentamicin and the antifungal reference Amphotericin B discs were tested concurrently as a standard.

Table 4: Results of antimicrobial activity

Comp.	Fungi				Gram Positive Bacteria		Gram Negative Bacteria	
	1	2	3	4	5	6	7	8
9	15.4 \pm 0.19	14.2 \pm 0.44	15.9 \pm 0.25	NA	14.4 \pm 0.44	16.8 \pm 0.25	NA	11.2 \pm 0.33
12a	20.1 \pm 1.5	20.9 \pm 2.1	21.4 \pm 0.58	NA	18.3 \pm 0.37	21.4 \pm 0.63	NA	20.3 \pm 1.5
12c	16.2 \pm 0.63	20.4 \pm 1.2	22.3 \pm 0.58	NA	19.4 \pm 1.2	21.3 \pm 0.58	NA	20.6 \pm 0.58
12h	18.7 \pm 0.36	16.9 \pm 0.44	20.8 \pm 0.65	NA	19.2 \pm 0.27	20.0 \pm 0.58	NA	16.4 \pm 0.19
13a	21.3 \pm 1.2	22.4 \pm 1.2	23.4 \pm 0.58	NA	21.6 \pm 1.5	25.4 \pm 0.37	NA	23.4 \pm 1.5
13b	14.6 \pm 0.58	12.8 \pm 0.19	15.9 \pm 0.44	NA	16.3 \pm 0.58	17.2 \pm 0.19	NA	12.7 \pm 0.53
13c	17.3 \pm 1.2	13.2 \pm 0.58	15.4 \pm 0.58	NA	18.2 \pm 1.2	20.3 \pm 0.58	NA	15.7 \pm 1.5
14a	18.3 \pm 0.58	20.2 \pm 1.5	20.9 \pm 0.58	NA	21.3 \pm 1.2	23.2 \pm 0.58	NA	22.1 \pm 0.63
14c	15.3 \pm 1.2	16.2 \pm 0.58	18.4 \pm 1.2	NA	22.4 \pm 0.44	25.3 \pm 1.5	NA	24.1 \pm 2.1
14g	20.7 \pm 0.63	17.0 \pm 0.37	24.2 \pm 0.35	NA	20.0 \pm 0.34	22.1 \pm 0.19	NA	18.7 \pm 0.19
Standard	<i>Amphotericin B</i>				<i>Ampicillin</i>		<i>Gentamicin</i>	
	23.7 \pm 1.2	19.7 \pm 1.5	28.7 \pm 2.1	25.4 \pm 0.58	23.8 \pm 1.2	32.4 \pm 2.1	17.3\pm 1.5	19.9 \pm 0.58

1= *Aspergillus fumigatus* (RCMB 02568) 2 = *Syncephalastrum racemosum* (RCMB 05922)

3= *Geotricum candidum* (RCMB 05097) 4 = *Candida albicans* (RCMB 05036)

5= *Streptococcus pneumonia* (RCMB 010010) 6 = *Bacillus subtilis* (RCMB 010067)

7= *Pseudomonas aeruginosa* (RCMB 010043) 8 = *Escherichia coli* (RCMB 010052)

- RCMB: Regional Center for Mycology and Biotechnology Antimicrobial unit test organism.
- NA: No activity.

Results of antimicrobial activity

From table 4, it was found that *in vitro* antimicrobial testing revealed that compounds **12_a**, **12_c**, **13_a** and **14_a** were found to be more active than the reference drugs amphotericin B and gentamicin on *Syncephalastrum racemosum* and *Escherichia coli*. compound 14c showed the most activity against *Escherichia coli*. Compound **14_g** which bearing trifluoromethyl group in position 1 and an electron donating (methoxy) group in position 3 of phenyl ring in the phenylacetamide residue has broad spectrum antimicrobial activity against the tested organisms. All the tested compounds showed no activity against *Candida albicans* and *Pseudomonas aeruginosa*. The presence of trifluoromethyl group in position 1 has profound effect on biological activity as noticed in compound **14_a**, **14_b** and **14_g**.

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تصميم وتشبيد بعض مشتقات [4,2,1] ترايازولو[3,4-*a*] كينوكسالينون الجديدة وتقييم
نشاطها كمضادات ميكروبيه

للدكتور

عادل حمدي غياتي

ممن

قسم الكيمياء العضوية- كلية الصيدلة- جامعة الأزهر- القاهرة

في هذا البحث تم تشبيد بعض المشتقات الجديدة من نواة الترايازولوكينوكسالينون وقد تم إثباتها بطرق التحاليل المختلفة كنقطة الانصهار والأشعة تحت الحمراء والرنين النووي المغناطيسي ومطياف الكتلة والتحليل الدقيق للعناصر (الكربون والهيدروجين والنتروجين) بهدف دراسة تأثيرها كمضادات ميكروبيه وذلك باستخدام الأمفوتريسين – ب والأمبسلين والجنتاميسين كعقاقير مرجعية (للفطريات – البكتريا موجبة الجرام – البكتريا سالبة الجرام) وقد وجد للبعض من هذه المواد الجديدة تأثير مماثل للعقار المرجعي في هذا الصدد .