



COMPARATIVE EFFECTS OF THERMAL TREATMENTS AND γ -IRRADIATION ON THE VOLATILE, NON-VOLATILE AND ANTIRADICAL ACTIVITY OF EGYPTIAN ANISE ESSENTIAL OIL

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ABSTRACT

The effect of various thermal treatments (electric oven, microwave) and γ -irradiation at three doses (6, 8 and 10 KGy) on the composition of volatile and non-volatile of anise essential oil and also their antioxidant properties were considered. The hydrodistilled oil (HD) of control and treated samples were subjected to gas chromatography–mass spectrometry (GC/MS) analysis.

The volatile profile of raw HD oil of anise consisted mainly of transe-anethole (79.68%) followed by hexahydrofarnesyl acetone (6.95%), para-anisaldehyde (5.49%); γ -himachalene (2.53%) and estragole (0.76%). Although the effect of roasting didn't cause significant changes in the total yield of major compounds of HD anise oil which are phenylpropanoid derivative (transe anethole, para-anisaldehyde, cis-anethole and estragole (=methylchavicol), it is found that gamma irradiation revealed the same behavior at the 10 KGy irradiated sample but decrease the total yield of these compounds in 6.8 KGy irradiated sample compared to control one. also the thermal and γ -irradiation caused drastic increase in the total yield of sesquiterpenes whereas decreased oxygenated

compounds in all samples under investigation compared to control one.

Such changes affected the antioxidant activity of the treated samples 1.1-diphenyl-2-picrylhydrazyl (DPPH) free radical scavenging as well as β -carotene bleaching test against butylated hydroxy toluene (BHT).

The strongest effect of reduction of DPPH radical as well as the highest inhibiting effect of the oxidation of linoleic acid and the subsequent bleaching of β -carotene was by 8 KGy irradiated sample which comprised (84.57% \pm 1.43); (85.21% \pm 0.12) respectively, in comparison to BHT (98% \pm 0.0) at the same concentration 30 μ g/mL besides all samples under investigation revealed high antioxidant activities due to their high content of phenylpropanoid and oxygenated compounds.

These confirmed by total phenolic content.

High performance liquid chromatography (HPLC) method was used for the analysis of phenolic compounds in the selected sample. Polyphenolic compounds were analysed on C₁₈ Reversed Phase (RP) HPLC. A total of 9 phenolic compounds were identified, the obtained results showed that the predominant compound was *P*-quumaric acid (43.36%) followed by ferulic acid (21.06%).

INTRODUCTION

Recently, spices have received attention also in their useful physiological functions and antimicrobial activity. There are a lot of reports about antimicrobial activity of spice extracts and its essential oils. However, the available information's are for a small group of microorganisms and they are tested at high concentrations which are no practical use. More research is required on antimicrobial effects to food-related bacteria such as food spoilage bacteria and food borne bacterial pathogens (Roby et al 2013).

In food processing, lipid oxidation not only causes a loss in nutritional and gustative quality of foods but also generates oxidized products such as free radicals which lead to various undesirable chemical reactions. To avoid or delay this autoxidation process, conventional artificial antioxidants such as butylated hydroxyanisole (BHA) and butylated hydroxy toluene (BHT), propyl gallate (PG) and tertiarybutylhydroquinone (TBHQ) have been used for more than five decades (Li and Yi., 2003). However, these synthetic antioxidants have been suspected to or promote negative health effects. For this reason, there is a growing interest in studies of natural additives as potential antioxidants. Many sources of antioxidants of plant origin have been studied in recent years. Among these, the antioxidant properties of many aromatic and medicinal plants have shown to be effective in retarding the process of lipid peroxidation in oils and fatty foods and have gained the interest of many research groups (Kulisic et al 2004). Therefore, the demands for these plants are increasing in industrialized and non-industrialized countries which lead to increasing their prices.

Anise (*Pimpinella anisum* L.) is one of the most widely used plants in the world; anise seed is used as a spice (Leela and Vipin, 2008). and Anise seed essential oil has a number of applications as an aromatic agent in the food and liquor industry (Leela & Vipin, 2008 and Tonutti & Liddle, 2010). Commercial production of anise seed and anise essential oil is concentrated in countries in south Asia, Europe, North Africa as Egypt, and also in Russia. *P. anisum* is primarily cultivated for its fruits, commercially called "seeds" that are currently used for flavouring. Anise fruits are widely used in the preparation of traditional bread and baking goods, in liquor, and for aromatization of traditional Bulgarian liquor annasonliika, whereas anise essential oil is used as an aromatic agent in the pharmaceutical, perfumery, cosmetics, and

candy production industries (Tonutti and Liddle, 2010). Many folk medicinal systems from Asia to the Mediterranean are still using anise seed as a medicinal plant (Leela and Vipin, 2008). For example, Bulgarian traditional medicine has been using anise seed or anise seed extract as an anesthetic, antispasmodic, carminative, for treating coughs, bronchitis, asthma, other inflammatory diseases, kidney stones, and for increasing lactation in nursing mothers (Stojanov, 1973).

With the increased use of herbs and spices for food preservation and their inclusion in nutritional recommendations (Tapsell et al 2006), the importance for adequate preservation methods for herbs has become more important. Treatment with ionizing radiation is an accredited preservation method. Although the safety of irradiated food is well documented (World Health organization "WHO", 1999), little is known about the effects of irradiation on the antioxidant activity of phytochemicals.

The Joint FAO/IAEA/WHO Expert Committee confirmed that irradiation up to 10 kGy does not produce toxicological hazards and nutritional or microbiological problems in foods (WHO, 1981). Dried foods, such as fennel powder, are less sensitive to irradiation than hydrated ones, and their irradiation has been authorized at a maximum dose of 10 kGy and 30 kGy in Korea and in the United States, respectively (Olson, 1998). An irradiation dose of 5 to 10 kGy is sufficient to reduce the population of microbes without changing essential quality attributes, and the spice flavour remained intact up to 7.5 kGy (Farkas, 1985). However, unsaturated hydrocarbons were created in gamma-irradiated ground chili and paprika at 5 and 10 kGy and increased with the dose of irradiation (Bendini et al 1998). Furthermore, spices are usually irradiated in prepackaged form to prevent postcontamination. In this case, possible abnormal compounds from packaging materials may be created by irradiation and may migrate into the foods. Consequently, this migration may be potentially harmful and impact food flavour (Krzymien et al 2001). In addition, microwave energy has been used in food processing applications mainly for its heating properties. Therefore, it seems natural to adapt microwave radiation to pasteurize or even sterilize foods at lower temperatures or for shorter times than conventional methods require (Fung and Cunningham, 1980).

Research during the past 30 years on spices and herbs proved that treatment with ionizing radiation is an effective process for destroying insects and

micro-organisms parkas, Ionizing radiation requires no additives and is a safe physical process with a controllable technology. A major advantage for ionizing radiation is that it is cold treatment and hence this treatment does not cause any loss of the volatile components of the spices. The use of microwave energy for heating has been around for 40 years (Basaran and Akhan, 2010). Its potential use in the food processing industry for pasteurizing, sterilizing, defrosting, blanching, dehydration and cooking has opened a new dimension in food preparation at home and in food service operation (Thostenson and Chou, 1999 and Abd-Elmageed et al 2011, 2012, 2014)).

Commercial and legal requirements regarding the safety, quality and storage of food products have still focused attention on the development and improvement of decontamination methods. Microwave processing and cooking of foods is a recent development, which is gaining momentum in household as well as large scale food applications (Basaran and Akhan, 2010). Microwave energy has been used in food applications mainly for its heating properties. It seems natural, therefore, to adapt microwave radiation to pasteurize or even sterilize foods at low temperatures in shorter times than required by conventional methods (Akgul et al 2008 and Abd-Elmageed et al 2011, 2012). Due to the absorption of electromagnetic energy, temperature of material of high dielectric capacity increases after microwave pasteurization. Therefore, microwave pasteurization offers similar benefits to conventional methods, but with an improved product quality and reduced time of exposure to energy (Hashem and Alamri, 2010). Processing of spices using microwaves is a newer dimension. This alternative methodology is preferred, due to the convenience and ease of handling (Abd-Elmageed, 2007).

Cooking or roasting alters the nature of many food constituents such as starches and proteins by changing their physical, chemical and nutritional characteristics (Belitz and Grosch, 1987). Heat processing also changes the bioavailability of proteins, carbohydrates, lipids and vitamins. Little information is available on the extent of destruction of bioactive principles of spices during food processing. Since the healthy, beneficial physiological effects of spices are attributable to their active principles (Srinivasan, 2005), there is a need to evaluate the availability of the spice active principles in their original form when spices are heat processed as in domestic cooking. Significant losses in the concentration of active principles

would raise a question as to whether the spices would retain beneficial health effects after conventional heat processing, as in domestic food preparation.

Generally, some spices are processed, for their microbial stability and removal of extraneous matter. Roasting is one of the important phases in the cooking process to release characteristic flavour volatiles (Susheela, 2000) and undesirable constituents. Hence, roasting of spices affects flavour quality.

The present work deal with the evaluation of some thermal treatments (conventionally roasting by electric oven and microwave heating) which were suggested to decontaminate the spices and to compare the results with a γ -irradiation doses (6,8 and 10 kGy) which (so-called cold sterilization) were recommended for this purpose as standard work. The evaluations included phenolic content, antiradical activities, volatile and non-volatile compounds to choose the best results in this comparative study for improving the quality of Egyptian anise.

MATERIALS AND METHODS

Plant materials

Seeds of Anise (*Pimpinella anisum* L.) were purchased from local market at Giza during 2014.

Chemicals

1,1-Diphenyl-2-picrylhydrazyl (DPPH), butylated hydroxy toluene (BHT), β -carotene, gallic acid and Folin-Ciocalteu reagents and standard hydrocarbons (C₈-C₂₂) were obtained from Sigma Chemical Co. Methanol and formic acid (HPL grade) were purchased from Aldrich Co. All other solvents and chemicals were of analytical grade.

Thermal processing and irradiation of anise samples

Fresh three dry anise seed samples (100 g each) were roasted in a separately conventional electric oven at 120°C and for 20 minutes. Similarly, three samples (100 g each) were subjected to microwave heating (Daewoo DE Microwave, Mod: KoG-181G, 200-240V 50Hz. Microwave input power was 1400 W, Korea) separately and roasted after 3 minutes.

The Anise-seed samples were packaged in a sanitized brown glass capped bottles (1L) and irra-

diated in Radiation Research Centre, Cairo, Egypt by γ -cell, cobalt-60 γ -irradiator at dose rate 1.29744 KGy/ hours. The applied doses in this study were 0, 6, 8 and 10 kGy. The actual doses were within 75.4% of the target dose (Choi et al 2010). The irradiation room temperature was 18°C. The non-irradiated control was placed outside the irradiation chamber to have the same environmental temperature effect with the irradiating sample. The irradiated anise samples were transferred and kept in dry place the raw and treated samples were separately ground in a spice mix grinder.

Isolation of essential oil

Portions (100 g) of each raw, heated and irradiated samples prepared plant material were hydrodistilled for 3 hours in a Clevenger type apparatus according to the method recommended in the European pharmacopeia to isolate the essential oil. The obtained essential oils were dried over anhydrous sodium sulphate. The collected essential oils of raw and treated samples were immediately analyzed using GC and GC-MS.

Gas chromatographic (GC) analysis

GC analysis was performed by using Hewlett-Packard model 5890, USA equipped with a flame ionization detector (FID). A fused silica capillary column DB-5 (60m x0.32 mm. id.) was used. The oven temperature was maintained initially at 50°C for 5 min., then programmed from 50 to 250°C at a rate of 4°C/min. Helium was used as the carrier gas, at flow rate of 1.1 ml/min. The injector and detector temperatures were 220 and 250°C, respectively. The retention indices (Kovats index) of the separated volatile components were calculated using hydrocarbons (C₆-C₂₂), Aldrich Co.) as references.

Gas chromatographic-Mass spectrometric analysis

The analysis was carried out by using a coupled gas chromatography Hewlett-Packard model (5890)/mass spectrometry Hewlett-Packard-MS (5970). The ionization voltage was 70 eV, mass range m/z 39-400 amu. The GC condition was carried out as mentioned above. The isolated peaks were identified by matching with data from the library of mass spectra (National Institute of Standard and Technology) and compared with those of authentic compounds and published data (Adams et al 2007). The quantitative determina-

tion was carried out based on peak area integration.

Antioxidant activity assay

Anise essential oils and their extracts used for the determination of the antioxidant activity assays, and total phenolic content (TPC) were prepared as follow: 1g of respective solid anise was mixed 100ml 80%(v/v) water/methanol or ethanol solution and the suspension was shaken for 1 hour using a laboratory shaker at 1000 rpm. The solid phase was separated using filtration and this step was carried out in triplicate and the final extracts were stored in closed vials in darkness at 4°C.

DPPH scavenging assay

Each extract (10, 20, 30 μ g/ml) in methanol was mixed with 3ml of methanolic solution containing DPPH radicals. The mixture was shaken vigorously and left to stand for 30 min in the dark, and the absorbance was then measured at 517nm using spectrophotometer (Shimadzu, UV-160-IPC, Japan) against a blank (Najja et al 2011).

β -Carotene scavenging activity assay

The antioxidant activity of ethanolic and methanolic extracts of raw and all treated samples of anise was performed using β -Carotene bleaching assay (BCBA) according to (Iqbal et al 2007). β -Carotene (0.1mg) in 0.2ml of chloroform, 10mg of linoleic acid and 100mg of tween 40 were mixed. The solvent was removed at 40°C under vacuum and the resulting mixture, 20ml of oxygenated water was added. four milliliter aliquots mixtures were pipetted into different test tubes containing 10 μ g of each extract (10, 20, 30 μ g/ml) in ethanol. All determinations were carried out in triplicate.

Determination of total phenolic content (TPC)

Total phenolic content (TPC) of raw and anise treated samples were spectrophotometrically determined by Folin Ciocalteu reagent assay according to Singleton (1998), using gallic acid in methanol (50-2500mg/L) served as an external standard. Samples, standards, and blanks were mad in triplicate. The sample absorbance (indicative for polyphenols) was determined photometrically at 760nm. Result are expressed as milligrams of gallic acid equivalent per 100g DW (mg GAE/100gDW).

HPLC analysis

Extraction procedure

According to the results obtained in the previous experiments, the extraction parameters finally selected were 25°C of temperature, and the use of a mixture of water: methanol (80:20). Two hundred and fifty milligram of powdered plant material were sonicated with 25ml of the solvent mixture in an ultrasonic bath for 20min. After centrifugation at 7600xg for 10min, the supernatant was adjusted to 25ml in a measuring flask. Samples were quantified immediately after extraction in order to avoid possible chemical alterations. Blanks and standards containing known concentrations were placed between the samples to monitor the quantification.

HPLC analysis instrumentation

Analyses were carried out in a Agilent 100 series 1050 chromatograph equipped with an automatic injector, vacuum degasser and a DAD system. A Discovery® HS C₁₈ (250 mm×4.8 mm, 5µm) column (Supelco, Bellefonte, PA, USA) was used for all the separations. The mobile phase was a gradient prepared from 0.1% formic acid in water (A) and 0.1% formic acid in acetonitrile (B). The composition ranged from 10% B to 26% B in 40min. The flow rate was 0.2ml min⁻¹, and the injection volume 50 µl. UV detection was performed at 280 nm.

Statistical analyses

Statistical analyses were performed using SPSS for Windows version 16.0. Student's t-test analyzed differences between samples and controls. One-way ANOVA tested differences within the irradiated and corresponding non-irradiated samples. Correlations were calculated using Pearson's correlation coefficient. All data are presented as mean ± SD. The significance level α was set at $p < 0.05$.

RESULTS AND DISCUSSION

Volatile components in hydrodistillation oil (HD) of raw, conventionally roasted, microwave heated and γ -irradiated at 6,8 and 10 KGy anise seeds.

Seeds of anise (*Pimpinella anisum L.*) subjected to conventionally roasting and microwave heat-

ing as well as exposed to gamma irradiation at 6,8 and 10 KGy in a ⁶⁰CO package irradiator. These samples were analyzed and a comparison was done between these treatments and raw sample. In the case of raw sample, the volatile oil recovered after 3h of hydrodistillation (HD) was 1.06%. Whereas HD volatile oil increased in all treated sample since in electric oven roasted sample yield 1.15%; microwave heated sample yield 1.5%; irradiated samples yield 1.08%; 1.16% and 1.31% (w/w) in 6,8 and 10 KGy, respectively. Eighteen volatile compounds were identified in HD oil of anise of all samples under investigation. All these compounds are listed with their area percentages in **Table (1)**. Identification of the volatile compounds was identified by KI values and MS spectra (**Adams, 2007**).

The typical gas chromatograms of the volatiles in HD oil of raw, roasted samples by electric oven and microwave heated; irradiated samples at different doses 6,8 and 10 KGy from anise seeds are shown in (**Figs. 1 and 2**). The total area percentages of the main chemical classes of all previous samples are shown in (**Fig. 3**).

The volatile profile of raw HD oil of anise consisted mainly of trans-anethole (79.68%) followed by hexahydrofarnesyl acetone (6.95%), para-anisaldehyde (5.49%); γ -himachalene (2.53%) and estragole (0.76%) **Table (1)**. These results are in accordance with many authors (**Ullah and Honermeier 2013; Yan et al 2011; Tepe et al 2006 and Arslan et al 2004**). It is well known that the major components of HD oil of anise are phenylpropanoid derivatives; trans-anethole, para-anisaldehyde, cis-anethole and estragole (=methylchavicol). Although the effect of roasted by electric oven and microwave heating did not cause significant change in the total area percentage of these compounds since it comprised 85.64% and 84.80% respectively, compared to control sample which comprised 85.93% (**Fig. 3**). It is found that γ -irradiation caused decrease in HD oil of irradiated samples at 6 and 8 KGy which comprised 80.69% and 80.64% whereas increased to reach 85.32% at 10 KGy compared to control sample (85.93%) **Fig. (3)**.

These compounds are the main constituents responsible for the antioxidant activities of oils which contain them (**Avlessi et al 2004**). In our study, the total percentage of these compounds ranged between 80.64% to 85.64% in the HD anise seed oil of 8 KGY irradiated sample and electric oven roasted sample, respectively. Then it is extremely expecting that all samples under investi-

gation must have high antioxidant capacity due to their high percentage of these compounds especially *trans*-anethol, *para*-anisaldehyde.

Tepe et al (2006) found that benzene derivatives are represent in high amount 97.6% in HD of anise oil.

Estragole, the flavouring agent is considered to have negative effects on animal and human health and was deleted from the list of flavourings in food stuffs (**Burt, 2004**). **The European Pharmacopeia (2005)** limit of estragole in essential oil of anise 0.5-6.0% was not exceeded in investigated samples. In our study estragole was found ranged between 0.66% to 1.49% in HD oil of 6 and 8 KGy irradiated samples, respectively. (**Table 1**).

Conventionally, roasted caused increase in the total yield of sesquiterpenes (4.47%) whereas microwave heating and γ -irradiation at 6,8 and 10 KGy caused drastic increase in the same compounds which comprised 7.43%, 7.18%, 10% and 8.18% respectively compared to control one 3.93% (**Fig. 3**). This is due to the increase of β -caryophellene, γ -elemene; γ -curcumene; α -muurolene; γ -cadinene and α -zingberene in the microwave and three irradiated samples whereas these compounds are not detected in control one (**Table 1**). Besides increase in the percentage of aromadendrene and remarkable increase in the major sesquiterpene γ -himachalene which comprised in microwave and γ -irradiated samples 3.90%, 4.40%, 5.83% and 4.84%, respectively in comparison to its concentration (2.53%) in control one (**Table 1**). These results are in accordance with **Zheljzakov (2013)**, who reported that, the yield of γ -himachalene increase steadily with increasing distillation time to reach maximum at 480 minute. Also, these results are in agreement with **Abd-Elmageed et al 2014, 2012, 2011, Leganani et al 2001; Chacko et al 1996 and Emam et al 1995**).

In contrary what takes place in our previous study (**Abd-Elmageed et al 2011**) the effect of roasted (electric oven or microwave heating) and γ -irradiation on the yield of oxygenated terpenoids of HD anise seeds oil, show a remarkable decrease in all samples under investigation in comparison with their percentage in control (unprocessed) sample (**Fig. 3**). Except irradiated sample at 6 KGy it appear a remarkable increase in the total yield of heavy oxygenated compounds which comprise 12.06% compared to their concentrations (10.12%) in control one (**Fig. 3**). This is due to a remarkable increase in the percentage of terpenylketone (hexahydrofarnesylacetone) in this sample (9.04%)

Table (1). Oxygenated terpenes exhibited a higher antioxidant power in comparison to other the identified classes (**Radonic and Milos, 2003**).

Higher percentages of phenolic content and oxygenated compounds was strongly contribute to the fragrance and antioxidant activity (**Reverchon and Senator, 1992**).

Antioxidant activity of the HD anise seeds essential oils

Radical scavenging ability tests aim to simulate basic mechanisms involved in lipid oxidation by measuring either the reduction of stable radicals or radicals generated by physical mechanisms or chemical reactions.

The profile of scavenging activity on DPPH radical as well as the evaluated antioxidant activity using B-carotene/linoleate assays are shown in **Fig. (4, 5)**, for raw, conventionally roasted, microwave and γ -irradiated 6,8 and 10 KGy anise seeds essential oil. The radical scavenging activity of essential oil on DPPH increased with increasing concentration of the oil. The strongest effect for reduction of DPPH radical was by 8 KGy irradiated sample (84.57%±0.08) followed by 6 KGy irradiated sample (83.33% ±0.07), respectively compared o BHT (98%±0.0) at the same concentration **Fig. (4)**, it can be seen that percent DPPH radical scavenging activities of the extracts were dose dependent. As shown in **Fig. (5)**, the highest inhibiting effect for oxidation of linoleic acid and the subsequent bleaching of B-carotene also was by 8 KGy irradiated sample followed by 6 KGy irradiated sample (85.21% and 82.16%) respectively compared to BHT (98%) scavenging radical. These results are confirmed in (**Fig. 6**) which showed the total phenolic content of irradiated samples besides all samples under investigation show high antioxidant activity due to higher percentage of phenolic propanioids and oxygenated compounds these results are in accordance with (**Reverchon and Senator, 1992; Tepe et al 2006 and Mansour et al 2010**). Polyphenols are thermally labile molecules; they get easily degraded upon heat treatment (**Podsdek et al 2008 and Roy et al 2007**). In case of roasted samples, occurrence of Maillard reactions at very high temperature (160 °C) might also contribute to the reduction of polyphenol levels. **Manzocco et al (2001)** reported that polyphenolic compounds take part in Maillard reaction which results in an increase in the Maillard reaction products and a decrease in the polyphenol level.

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Table 1. Volatile components isolated in the hydrodistillation anise seed oil of raw, electric oven, microwave and γ -irradiation at 6,8, and 10 KGy. (*Values expressed as relative area parentages to total identified compounds)

Peak No.	KI ^a	Compound	Control (Raw)	Roasted		γ -Irradiation (KGY)			Type	Identification method ^b
				Electric oven	Microwave	6	8	10		
1	1202	Estragole	0.76*	0.83	1.17	0.66	1.49	1.41	Ph.Pro.Der.	MS&KI
2	1264	(Z) Anethole	-	0.25	0.37	0.18	0.44	0.36	Ph.Pro.Der.	MS&KI
3	1271	Para-Anisaldehyde	5.49	6.04	5.60	4.79	6.32	4.72	Ph.Pro.Der.	MS&KI
4	1299	(E) Anethole	79.68	78.52	77.66	75.06	72.39	78.83	Ph.Pro.Der.	MS&KI
5	1389	β -Elemene	1.10	0.68	0.86	0.41	0.51	0.29	S	MS&KI
6	1457	β -Caryophyllene	-	0.34	0.31	0.27	0.30	0.27	S	MS&KI
7	1463	γ -Elemene	-	n.d	0.36	0.24	0.36	0.39	S	MS&KI
8	1480	γ -Curcumene	-	0.24	0.28	0.18	0.33	0.29	S	MS&KI
9	1486	Aromadrene	0.30	0.16	0.42	0.41	0.63	0.51	S	MS&KI
10	1491	γ -Himachalene	2.53	2.29	3.90	4.40	5.83	4.84	S	MS&KI
11	1498	α -Murolene	-	n.d	0.34	0.37	0.63	0.43	S	MS&KI
12	1512	γ -Cadiene	-	0.33	0.53	0.56	0.86	0.70	S	MS&KI
13	1532	α -Zingiberene	-	0.43	0.43	0.41	0.55	0.36	S	MS&KI
14	1594	β -Caryophyllene oxide	-	0.36	0.16	0.11	0.43	0.15	HOC	MS&KI
15	1687	α -Cadinol	-	0.34	0.40	0.07	0.57	0.30	HOC	MS&KI
16	1845	Hexahydrofarnesylacetone	6.95	7.42	5.73	9.04	6.09	4.88	HOC	MS&KI
17	1902	(E) Phytol	1.44	1.20	1.05	2.32	1.34	0.95	HOC	MS&KI
18	1963	(E) Phytol acetate	1.73	0.58	0.42	0.52	0.44	0.32	HOC	MS&KI

^a:KI: Linear Kovat indices:Compounds listed according to their elution on DB-5 column; ^b: compounds identified by GC-MS (MS) and / or Kovat index on DB5 (KI) and / or by comparison of MS and KI of standard compounds run under similar GCMS conditions. -: not detected.

Ph.Pro.Der.: Phenylpropanoids derivatives, S:Sesquiterpene hydrocarbons, HOC: Heavy oxygenated compounds;

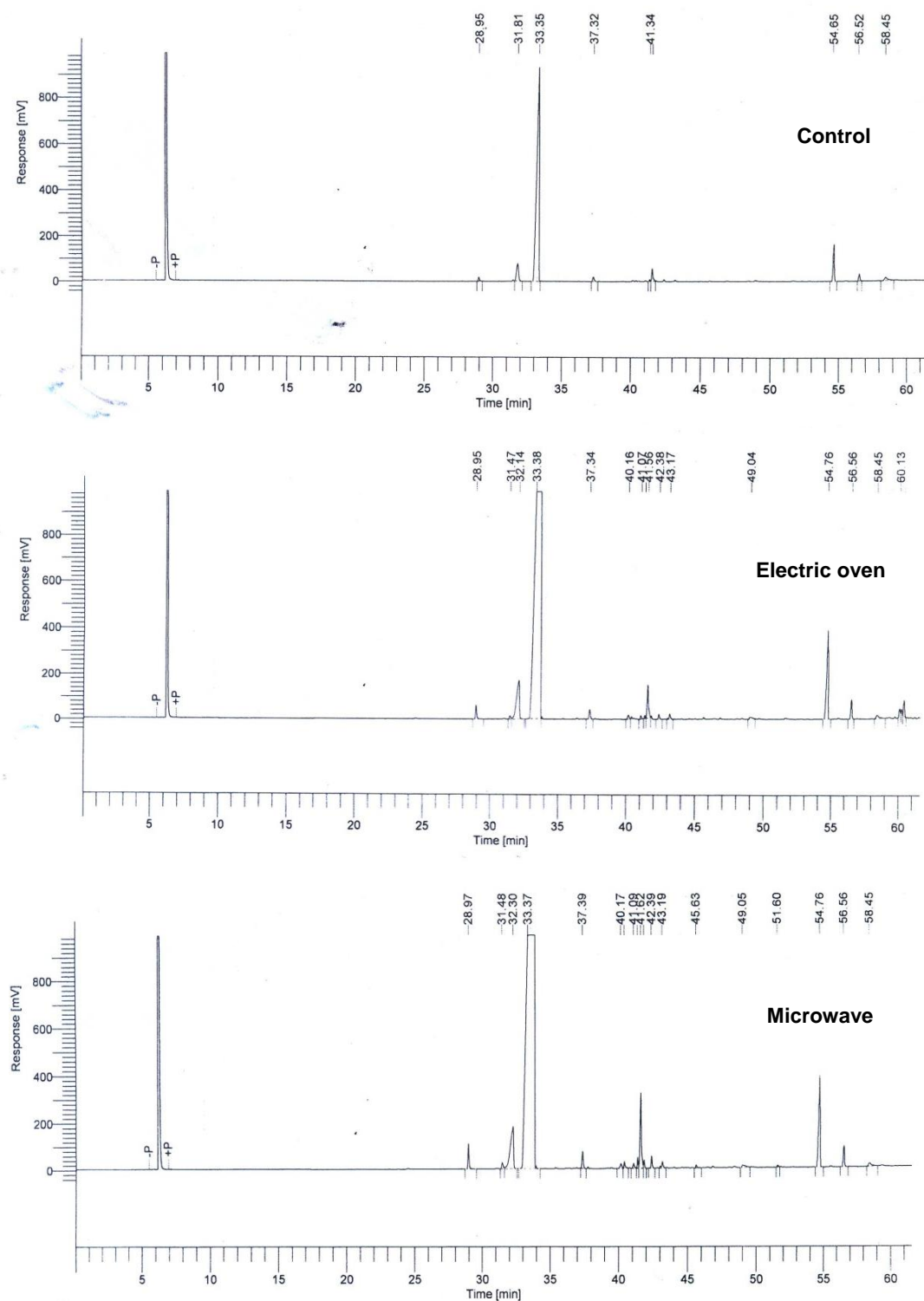


Fig. 1. Gas Chromatograms of volatiles in HD oil of raw (control), thermally roasted (electric oven) and microwave heated anise seeds

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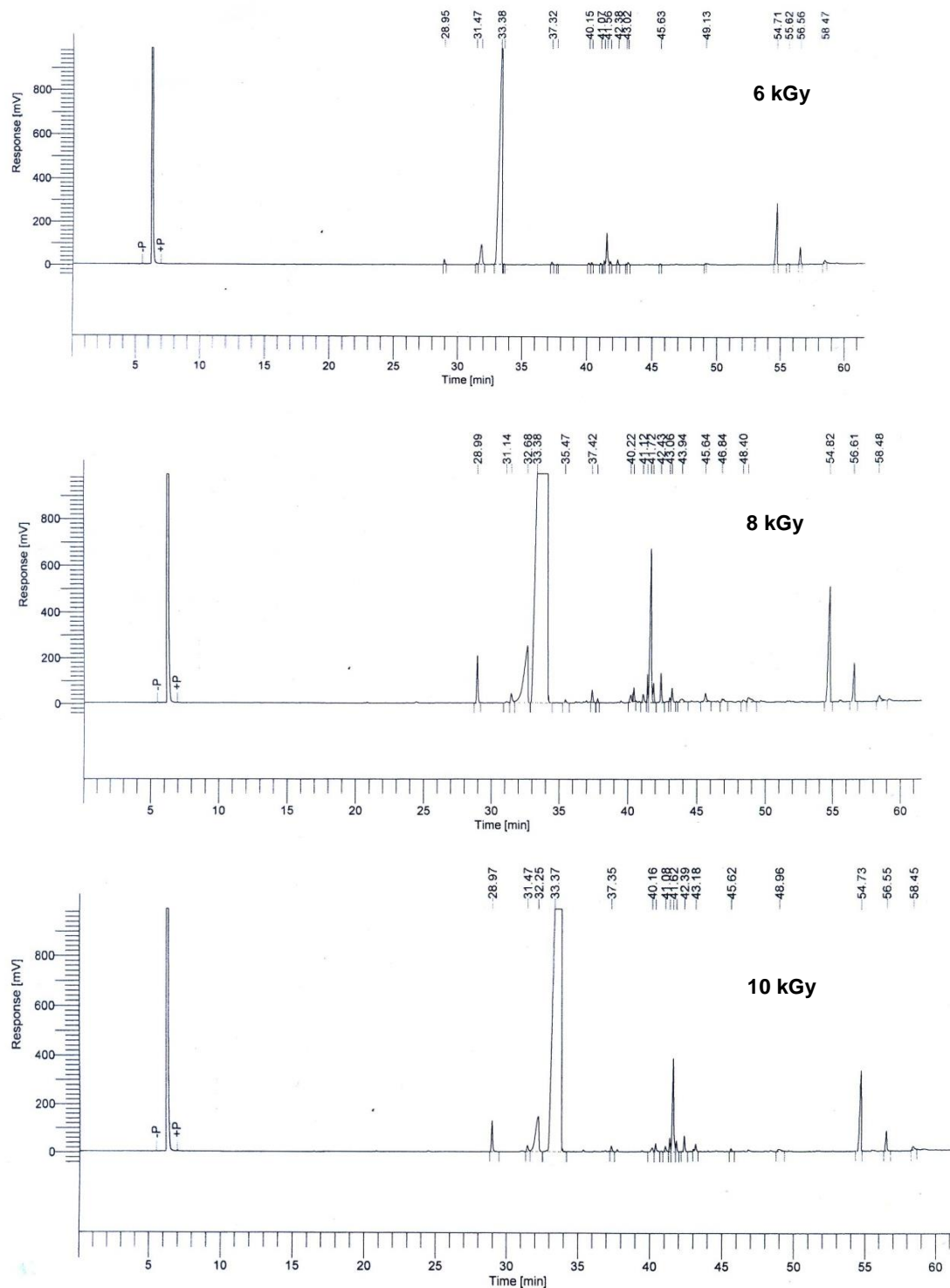


Fig. 2. Gas Chromatograms of anise essential oil treated with different γ - irradiation dosed (6,8 and 10 KGy)

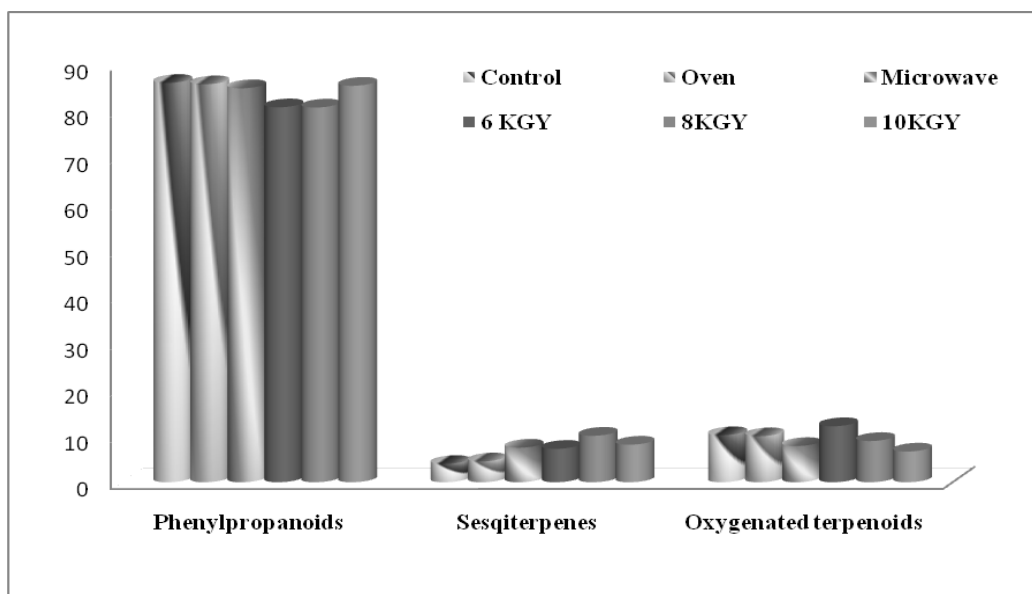


Fig. 3. The total area percentages of the main chemical classes in HD anise oil treated with thermal treatments (electric oven, microwave) as well as γ -irradiation at three doses

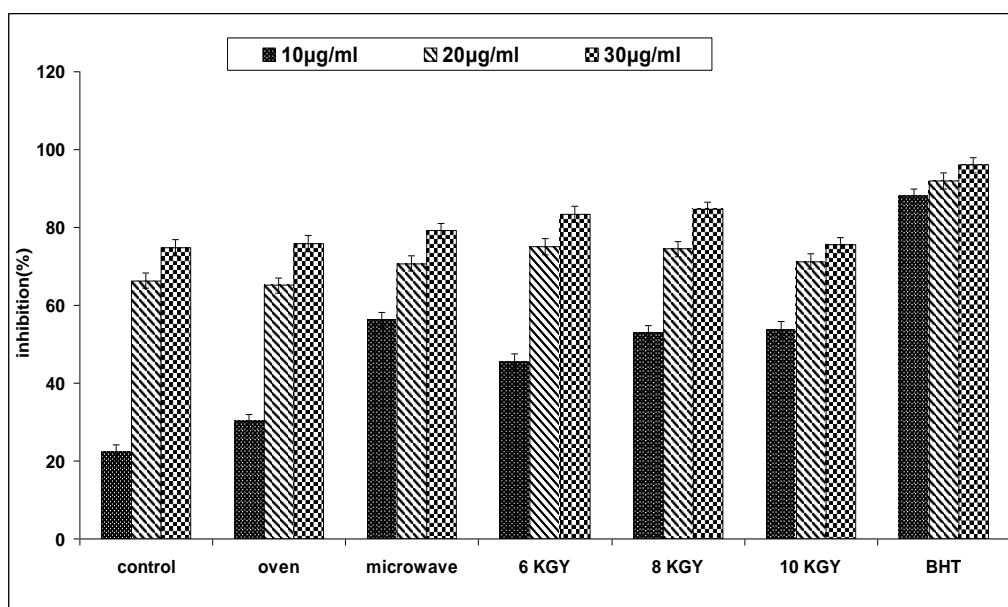


Fig. 4. Scavenging effect of Anise volatile oil on DPPH radicals with different concentrations compared to BHT \pm SD

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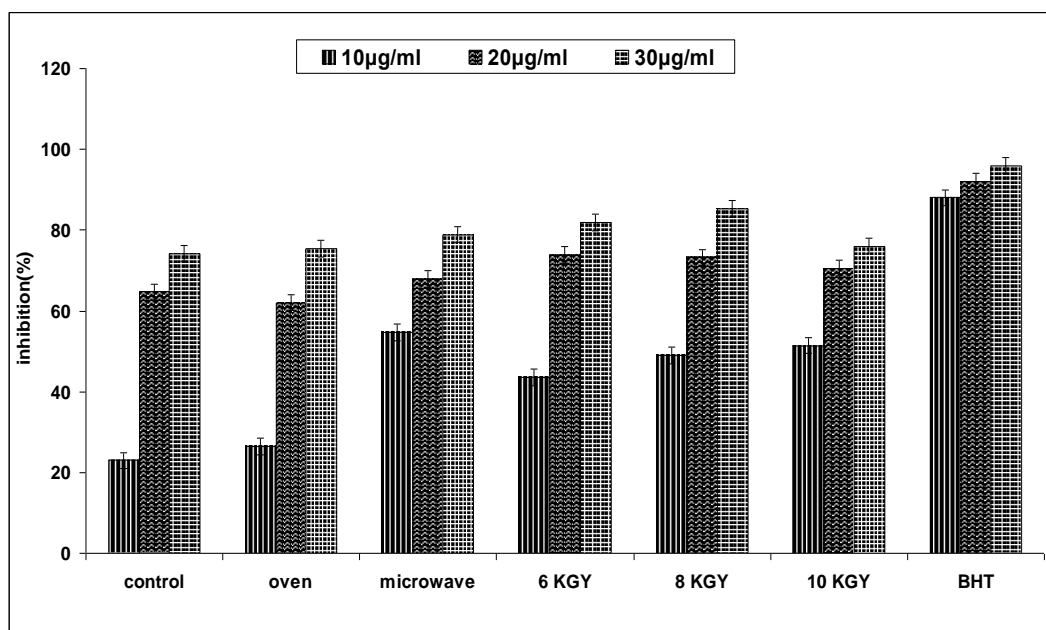


Fig. 5. Scavenging effect of Anise volatile oil on β -carotene linoleic acid bleaching with different concentrations compared to BHT \pm SD

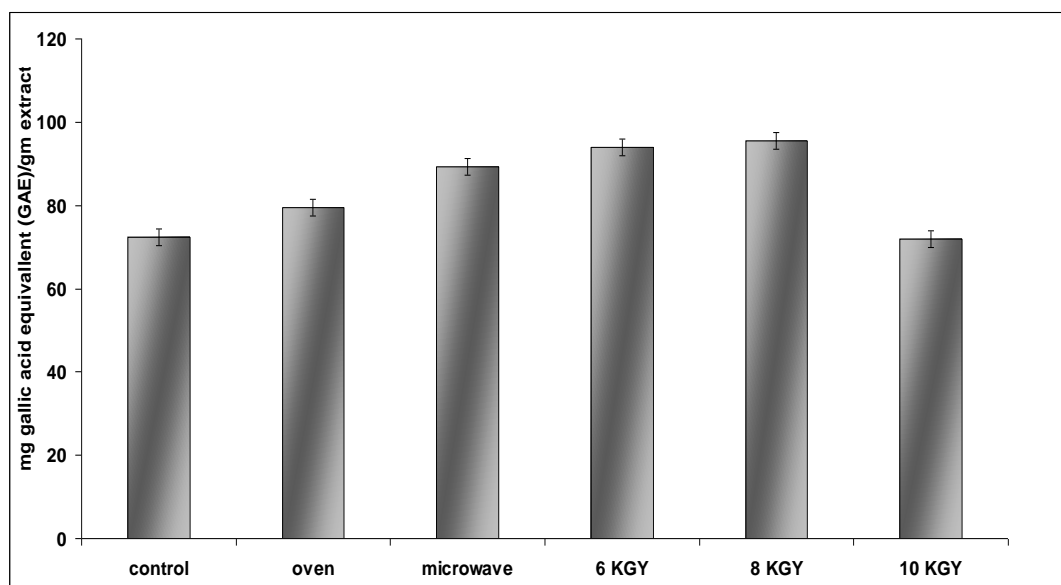


Fig. 6. Total Phenolic content of anise hydrodistilled oil in control, electric oven, microwave and irradiated doses at 6,8, and 10 KGy.

β -Carotene bleaching test is commonly used for determination of antioxidant activity of natural compounds because it is carried out in an emulsion, a condition frequently seen in foods. In this model system, the reduction in the orange colour of β -carotene caused due to the abstraction of a hydrogen atom from one of its methylene groups is assessed. From (Fig. 5), it can be seen that antioxidant activity coefficient (AAC) of the extracts was concentration dependent.

4. HPLC analysis of γ -irradiated selected sample

Identification of phenolic compounds in the selected treatment (8 KGy) –the most effective treatment-essential oil was preliminary performed by HPLC-UV, comparing the relative retention times and UV spectra with those of standard solutions. Concentration of phenolic compounds in the essential oil was comparing HPLC peak areas with those of external standard and the obtained data are reported in Table (2).

Table 2. Effect of γ -irradiation on phenolic compound of Anise essential oil (treated with 8 KGy).

Phenolic compound	Concentration
Catechin	4.72*
<i>P</i> -Qumaric acid	43.36
Ferulic acid	21.06
Benzoic acid	15.73
Cinnamic acid	6.07
Rurin	2.53
Quercetin	0.77
Gallic acid	2.33
Kampferol	0.86

*: Values are expressed ad relative area percentage

A total of 9 phenolic compounds (Table 2) were identified in anise essential oil after irradiation with γ -irradiation at 8 KGy which were gallic acid, catechin, ferulic acid, benzoic acid, Cinnamic acid, rutin, quercetin, *P*-qumaric acid and Kampferol. The obtained results showed that *P*-qumaric acid was the predominant phenolic compound (43.36%) followed by ferulic acid which represent (21.06%) and benzoic acid (15.73%). Several studies have determined the effect of γ -irradiation on total phenolics, but to our knowledge

no studies have investigated the effect of γ -irradiation on individual phenolics in anise essential oil. At irradiation doses of 2.5 to 10 kGy, various capsaicinoids in paprika increased or were unaffected with gamma irradiation (Topuz and Ozdemir 2004). Soy flour subjected to γ -irradiation ranging from 0.5 and 5 kGy resulted in reduced glycosidic conjugates with increased dosage (Variyar et al 2004).

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