SYNTHESIS OF SOME NEW 2,4,6-TRISUBSTITUTEDTHIAZOLO-[5,4-d] PYRIMIDINE-5,7(4H,6H)-DIONES.

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ABSTRACT: Reaction of 2,6-disubstitutedthiazolo[5,4-d]pyrimidin-7(6H)-ones 2 with ethyl chloroformate/ethanol mixture afforded 5-(ethoxycarbonylamino) -2-(substitutedthio) thiazole-4-(N-substituted)-carboxamides 4a-c. Thermal fusion of 4 followed by treatment with dimethyl sulfate gave the corresponding trisubstituted thiazolo[5,4-d]-pyrimidinediones 6. When the thiazole esters 7a,b were reacted with ethyl chloroformate, the corresponding ethyl 5-(ethoxycarbonyl)-aminothiazole-4-carboxylates 8 were obtained. Reaction of 8b with benzylamine gave the 2-benzylthio-5-(3-benzylureido)thiazole-4-(N-benzyl)carboxamide 9.

INTRODUCTION

Previously, ethyl chloroformate has been reported ⁽¹⁾ as a reagent for adding a carbonic acid-type carbon, fully oxidized, in the ring closure of some o-aminonicotinamides to their pyrido [2,3-d]-pyrimidine -2,4 (1H,3H)-diones. Also, the same compound was employed ⁽²⁾ in the presence of pyridine, for stepwise synthesis of some thiazolo [5,4-d]-pyrimidine - 5,7 (4H,6H)-diones.

Moreover, ethyl chloroformate/DMF mixture has been reported as a reagent for a facile ring closure of different o-aminocarboxamide heterocyclic derivatives to afford their condensed pyrimidines (3-5) and pyrimidotriazolone (6) derivatives. On the other hand, ethyl chloroformate/ethanol mixture has been reported as a ring fission reagent (7) for thiazolopyrimidinone derivatives.

In continuation of our studies (3-7), on the chemistry of ethyl chloroformate here we wish to report the behaviour of ethyl chloroformate/ethanol mixture towards 2,6-disubstituted thiazolo [5,4-d]pyrimidine -7 (6H)-ones.

EXPERIMENTAL

Melting points are uncorrected. Microanalyses were carried out by the Microanalytical Center, Cairo University. IR spectra were recorded (KBr-disc) by using Pye-Unicam SP-1100 spectrophotometer.

¹H-NMR spectra were measured in CDCl₃ (or DMSO-d₆ whenever reported) by using Hitachi Perkin-Elmer R-600 and Jeol GLM Ex Ft NMR systems with chemical shifts in δ (from Me₄Si). Mass spectra were recorded by using MS 5988 mass spectrometer.

Compounds 1a (8), 1b, 2c and 7 (9-11) were prepared according to reported procedures.

Reaction of products 1 and 5 with dimethyl sulfate

To a solution of 1 or 5 (1.0g) in sodium hydroxide solution (100 ml, 10%), dimethyl sulfate (5 ml) was added and the mixture was stirred for 1 hr. at room temperature (for products 5 the reaction mixture

was left over night). The solid product separated out (product 6a was obtained by extracting the reaction mixture with chloroform and then, the solvent was removed under reduced pressure) was filtered off, washed with water, dried and crystallized to give 2 and 6, respectively.

6-Methyl-2-methylthiothiazolo[5,4-d] pyrimidin-7 (6H)-one 2a: 71% yield (1hr), m.p. 229-30°C (n-butanol). IR ν /cm⁻¹: 1692 (C=O). ¹H-NMR (DMSO-d₆) δ 2.40 (s,3H, SCH₃); 3.50 (s, 3H, N-CH₃); 8.45 (s, 1H, H-5). MS m/e (M⁺): 213. Anal. Calcd. for C₇H₇N₃OS₂: C, 39.44; H, 3.29; N, 19.72. Found: C, 39.40; H, 3.20; N, 19.90%.

2-Benzylthio-6-methylthiazolo[5,4-d]pyrimidin-7(6H)-one 2b: 86% yield (1hr), m.p. 165-6°C (EtOH); IR v/cm⁻¹: 1685 (C=O), 1 H-NMR δ : 3.60 (s, 3H, CH₃); 4.55 (s, 2H, CH₂); 7.23 (m, 5H, C₆H₅); 7.85 (s, 1H, H-5). **Anal. Calcd.** for C₁₃H₁₁N₃OS₂: C, 53.98; H, 3.81; N, 14.53. **Found:** C, 53.80; H, 4.00; N, 14.40%.

4,6-Dimethyl-2-methylthiothiazolo[5,4-d]pyrimidine-5,7(4H,6H)-dione 6a: 85% yield (24hr), m.p. 186-7°C (EtOH). IR ν /cm⁻¹: 1719, 1678 (C=O). ¹H-NMR δ : 2.75 (s, 3H, SCH₃); 3.55 (d, 6H, 2NCH₃). MS m/e (M⁺): 243. Anal. Calcd. for C₈H₉N₃O₂S₂: C, 39.51; H, 3.70; N, 17.28. Found: C, 39.30; H, 3.60; N, 17.50%.

2-Benzylthio-4,6-dimethylthiazolo[5,4-d]pyrimidie-5,7(4H,6H)-dione 6b: 90% yield (24hr), m.p. 190-1°C (EtOH). IR v /cm⁻¹: 1717, 1663 (C=O). ¹H-NMR δ : 3.50 (d, 6H, 2NCH₃); 4.50 (s, 2H, CH₂); 7.35 (m, 5H, C₆H₅). MS m/c (M⁺): 319. Anal. Calcd. for C₁₄H₁₃N₃O₂S₂: C, 52.66; H, 4.08; N, 13.17. Found: C, 52.90; H, 4.10; N, 13.30%.

6-Benzyl-2-benzylthio-4-methylthiazolo[5,4-d]-pyrimidine-5,7(4H,6H)-dione 6c: 86% yield (24hr), m.p. 160-1°C (EtOH). IR v/cm⁻¹: 1706, 1662 (C=O).

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¹H-NMR δ: 3.44 (s, 3H, CH₃); 4.53 (s, 2H, SCH₂); 5.09 (s, 2H, NCH₂), 7.40 (m, 10H, 2C₆H₅). MS m/c (M⁺): 395. Anal. Calcd.: for C₂₀H₁₇N₃O₂S₂: C, 60.76; H, 4.30; N, 10.63. Found: C, 61.00; H, 4.50; N, 10.50%.

5 - (Ethoxycarbonylamino)-2-(substitutedthio)thiazole-4-(N-substituted)carboxamides 4a-c:

To a mixture of ethyl chloroformate and ethanol (30 ml, 1:5 ratio), compound 2a,b or c (0.01 mol) was added. The reaction mixture was heated under reflux for 3hr. and then concentrated. After cooling, the solid product obtained was filtered off, dried and crystallized from methanol to give 4a-c.

5-Ethoxycarbonylamino-2-methylthiothiazole-4-(N-methyl)-carboxamide 4a: 70% yield; m.p. 111-2°C. IR ν /cm⁻¹: 3408, 3157 (NH); 1720, 1635 (C=O). ¹H-NMR δ: 1.25 (t, 3H, CH₃-ethyl); 2.60 (s, 3H, SCH₃); 2.95 (d, 3H, NCH₃); 4.25 (q, 2H, CH₂); 7.10 (b, 1H, NH); 10.00 (b, 1H, NH-ester). MS m/e (M⁺): 275. Anal. Calcd. for C₉H₁₃N₃O₃S₂: C, 39.27; H, 4.73; N, 15.27. Found: C, 39.40; H, 4.50; N, 15.40%.

2-Benzylthio-5-(ethoxycarbonyl)aminothiazole-4-(N-methyl)-carboxamide 4b: 93% yield; m.p. 61-2°C. IR v /cm-1: 3392, 3230 (NH); 1714, 1638 (C=O). 1 H-NMR δ : 1.20 (t, 3H, CH₃-ethyl); 2.80 (d, 3H, NCH₃); 4.15 (m, 4H, 2CH₂); 6.85 (b, 1H, NH); 7.05 (s, 5H, C₆H₅); 10.20 (b, 1H, NH-ester). MS m/e (M⁺): 351. Anal. Calcd. for C₁₅H₁₇N₃O₃S₂: C, 51.28; H, 4.84; N, 11.97. Found: C, 51.20; H, 4.90; N, 12.10%.

2-Benzylthio-5-(ethoxycarbonyl)aminothiazole-4-(N-benzyl)-carboxamide 4c: 75% yield, m.p. 65-6°C. IR v/cm⁻¹: 3414, 3223 (NH); 1711, 1644 (C=O). ¹H-NMR δ: 1.30 (t, 3H, CH₃-ethyl); 4.20 (m, 4H, SCH₂, CH₂-ethyl); 4.60 (d, 2H, NCH₂); 7.30 (m, 6H, C₆H₅, NH); 10.50 (b,1H, NH-ester). MS m/e (M⁺): 427. Anal. Calcd. for C₂₁H₂₁N₃O₃S₂: C, 59.02; H, 4.92; N, 9.84. Found: C, 58.80; H, 4.90; N, 9.90%.

6-(Substituted)-2-(substitutedthio)thiazolo[5,4-d]- pyrimidine -5,7 (4H, 6H)-diones 5a-c:

When the thiazole-(N-substituted)carboxamide 4 (1g) was heated in a thermal decomposition tube at 200°C (oil-bath temperature) for 45 min., it melted and then solidified again. The solid was treated with methanol, filtered off, dried and crystallized to give 5a-c.

6-Methyl-2-(methylthio)thiazolo[5,4-d]pyrimidine-5,7 (4H, 6H)-dione 5a: 67% yield, m.p. 295-7°C (n-butanol). IR v/cm⁻¹: 3237 (NH); 1728, 1638 (C=O). ¹H-NMR (DMSO-d₆) δ: 2.55 (s, 3H, SCH₃); 3.5 (s, 3H, NCH₃); 11.80 (b, 1H, NH). MS m/e (M⁺): 229. Anal. Calcd. for C₇H₇N₃O₂S₂: C, 36.68; H, 3.06; N, 18.34 Found: C, 36.60; H, 3.10; N, 18.70%.

2-Benzylthio-6-methylthiazolo[5,4-d]pyrimidine-5,7(4H,6H)-dione 5b: 65% yield, m.p. 246-8°C (ethanol). IR ν /cm⁻¹: 3240 (NH); 1737, 1638 (C=0). (ethanol). IR ν /cm⁻¹: 3240 (NH); 1737, 1638 (C=0). (H-NMR (DMSO-d₆) δ : 3.20 (s, 3H, CH₃); 4.50 (s, 2H, 1H-NMR (DMSO-d₆)); 12.15 (b, 1H, NH). MS m/e (CH₂); 7.40 (m, 5H, C₆H₅); 12.15 (b, 1H, NH). MS m/e (M⁺): 305. Anal. Calcd. for C₁₃H₁₁N₃O₂S₂: C, 51.15; (M⁺): 305. Anal. Calcd. for C₁₃H₁₁N₃O₂S₂: C, 51.15; H, 3.61; N, 13.77. Found: C, 51.40; H, 3.60; N, 13.50%.

6-Benzyl-2-(benzylthio)thiazolo[5,4-d]pyrimid-ine-5,7(4H,6H)-dione 5c: 72% yield, m.p. 216-8°C (ethanol). IR v/cm⁻¹: 3276 (NH); 1734, 1658 (C=O). (ethanol). IR v/cm⁻¹: 3276 (NH); 1734, 1658 (C=O). (1H-NMR (DMSO-d₆) δ : 4.50 (s, 2H, SCH₂); 5.00 (s, 2H, NCH₂); 7.40 (m, 5H, C₆H₅); 12.25 (b, 1H, NH). MS m/e (M⁺): 381. **Anal. Calcd.** for C₁₉H₁₅N₃O₂S₂ C, 59.84; H, 3.94; N, 11.02. **Found:** C, 60.00; H, 3.90; N, 11.30%.

Ethyl 5- (ethoxycarbonylamino) -2- (substitutedthio)thiazole -4- carboxylates 8:

To a suspension of 7a or b (0.01 mol) in ethyl chloroformate (30 ml), pyridine (0.5 ml) was added and the mixture was heated under reflux for 1hr, concentrated and left to cool. The solid product obtained was filtered off, dried and crystallized from methanol to give 8.

Ethyl-5-(ethoxycarbonylamino)-2-(methylthio)-thiazole-4-carboxylate 8a: 84% yield, m.p. 88-9°C. IR ν /cm⁻¹: 3279 (NH); 1717, 1663 (C=O). 1 H-NMR δ: 1.35 (m, 6H, 2CH₃); 4.20 (m, 4H, 2CH₂); 9.73 (b, 1H, NH). MS m/e (M⁺): 290. **Anal. Calcd.** for $C_{10}H_{14}N_2O_4S_2$: C, 41.38; H, 4.83; N, 9.66. **Found:** C, 41.60; H, 4.60; N, 9.80%.

2-(Benzylthio)-5-(ethoxycarbonylamino)thiazole-4-carboxylate 8b: 85% yield, m.p. 79-80°C. IR v / cm-1: 3278 (NH); 1718, 1664 (C=O). 1 H-NMR 5 1.30 (m, 6H, 2CH₃); 4.20 (m, 6H, 3CH₂); 7.10 (m, 5H, C₆H₅); 9.75 (b, 1H, NH). MS m/e (M⁺): 366. Anal. Calcd. for C₁₆H₁₈N₂O₄S₂: C, 52.46; H, 4.92; N, 7.65. Found: C, 52.40; H, 4.70; N, 7.60%.

2-Benzylthio-5-(3-benzylureido)thiazole-4-(N-benzyl)-carboxamide 9:

A mixture of 8b (2g) and benzylamine (4 ml) was heated at 180°C (oil-bath temperature) for 2 hr. After cooling, the reaction mixture was treated with methanol. The solid product obtained was filtered off and

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organized from methanol to give 9 (85%), m.p. (NH); 1668, 1624 (85%), m.p. (85%), m.p. (85%), m.p. (85%), m.p. (85%), m.p. (85%), m.p. (85%), 1624 (C=O). 1024 (1H. NCONH-C); 7.00 (m, 16H, 3C₆H₅, 525 (h, 1H, NHCO-N). Me (m, 10H, 3C₆H₅, 10.30 (b, 1H, NHCO-N). MS m/e (M⁺):

(NH-C): 10.30 (b, 1H, NHCO-N). Calcd. For C₂₆H₂₄N₄O₄S₂. Calcd. Colod. For C₂₆H₂₄N₄O₂S₂: C, 63.93; H, 48 Found: C, 63.70; H. 500; M 48. Apa. C, 63.70; H, 5.00; N, 11.80%.

RESULTS AND DISCUSSION

Reaction of the 2-(substituted thio) thiazolo [54d] pyrimidin-7(6H)-ones 1 with dimethyl sulfate in [5.4-0] Production, afforded the corresponding 6 methyl-2 (substituted thio) thiazolo [5,4-d] pyrimidin-7-6H) ones 2a,b rather than the ether derivatives 3. (chi) Structure of products 2a and b was ducidated by careful studying of their IR spectra, which realed the presence of carbonyl absorptions at 1692 and 1685 cm⁻¹, respectively.

When products 2 were heated under reflux with a mixture of ethyl chloroformate/ethanol, the ring fission products,5-(ethoxycarbonyl-amino)-2- (substitutedthio)thiazole-4- (N-substituted) carboxamide derivatives 4a-c were smoothly obtained (Scheme 1).

Formation of products 4 from the 6- substitutedthiazolo[5,4-d]-pyrimidinones 2 can be explained according to the mechanism shown in scheme 2. Thus, the N-4 of the thiazolopyrimidinones 2 are attacked by the ethyl chloroformate and quarternized to the unstable intermediate (A), which in the presence of ethanol undergo ethanolysis to form (B). The latter intermediates undergo ring fission of the pyrimidine ring to the thiazole derivative intermediates (C), which in turn add EtO group to afford (D). The latter eliminates a formate residue to afford the thiazoles 4. (Scheme 2).

Structure of compounds 4a-c was elucidated by carefull studying of their spectral determinations. The IR spectra showed v NH at 3414-3392 cm⁻¹ and 3230-3157 cm⁻¹ regions, the spectra showed also v CO (ester) at 1720-1711 cm⁻¹ and v CO (amide) (12) at 1643-1635 cm⁻¹ regions. ¹H-NMR spectra of products 4a-e were characterized by the presence of ethyl proton signals at 8: 1.20-1.30 ppm (triplet) and 4.15-4.25 ppm (quartet) regions, this is beside the NH proton signals (D₂O exchangeable). Also, the structure of products 4a-c was accorded by mass spectra which revealed m/e (M⁺) at 275, 351 and 427 respectively.

On the other hand, when the thiazole derivatives 4a-c were heated above their melting points for a short time,the corresponding 6-substituted-2- (substitutedthio) thiazolo[5,4-d]-pyrimidine-5,7(4H,-6H)-diones5a-c were smoothly obtained in good yields.

IR spectra of products 5a-c showed v NH at 3276-3237 cm⁻¹ region and ν CO (two bands) at 1737-1728 cm-1 (NHCONH) and 1658-1635 cm-1 (-C=C-CON-R) regions (8) . The ¹H-NMR spectra of 5a-c revealed the stability of the amide substituent through the thermolysis.

Methylation of products 5a-c was also carried out by treating 5 with dimethyl sulfate in sodium hydroxide solution at room temperature to afford the corresponding 4-methyl-2,6-di(substituted)-thiazolo[5,4-d] - pyrimidine - 5,7 (4H,6H)-diones 6a-c (Scheme 3).

¹H-NMR spectra of products 6a-c were characterized by the presence of methyl proton signals at position-4 at δ: 3.44-3.55 ppm beside the other characteristic signals. Also, the mass spectra of 6a-c accorded their structures and revealed m/e (M+) at 243, 319 and 395, respectively.

Attempts to prepare products 5a-c from the ethyl 5-(ethoxy-carbonylamino)thiazole-4-carboxylates 8 (which obtained from reacting the ethyl 5aminothiazole-4-carboxylates 7a,b (9-11) with cthyl chloroformate in the presence of pyridine as catalyst) by reaction with the corresponding primary amine have been failed. Thus, in case of reacting of 8 with aqueous methylamine, the ethyl 5-aminothiazole-4-carboxylates 7 were obtained (13). Whereas, when 8b was reacted with benzylamine, the corresponding 2-benzylthio-5-(3benzylureido)thiazole- 4-(N-benzyl)carboxamide 9 was obtained.

Infrared absorption spectra of products 8a and b showed the presence of carbonyl ester moiety. Also, ¹H-NMR spectra revealed the presence of two ethyl proton signals at δ: 1.30-1.33 (CH₂) and 4.20 (CH₂) ppm, this is beside the NH proton signals at δ: 9.73-9.75 ppm (D2O-exchangeable) region.

RS
$$\stackrel{R_1}{\searrow}$$
 $\stackrel{R_1}{\searrow}$ $\stackrel{C_1CO_2E_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{E_1OH}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{E_1OH}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{R_1}{\searrow}$ $\stackrel{R_2}{\searrow}$ $\stackrel{$

(Scheme 2)

(Scheme 3)

Structure of product 9 was confirmed by IR spectrum which lack carbonyl ester absorption, instead it showed v (C=O (amide) at 1668 and 1624 cm⁻¹. Also, its 1H-NMR spectrum accorded the proposed structure which revealed the presence of three benzyl protons signals. Furthermore, the mass spectrum of 9 revealed an ion peak at m/e 488 which corresponded to M⁺.

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تشیید بعض من ۲، ٤، ٦- ثلاثی مشتق الثیازولو (٥، ٤ - د) بیر پیدین ٥، ٧ - (٤ هـ، ٦ هـ) ثنائی أونات الجدیدة

وحید محمد بسیونی ، هناء محمد حسنی

قسم مبيدات الآفات - المركز القومي للبحوث - الدقي - الجيزة -مصر

بتفاعل 7,7 – ثنائی مشتق الثیازولو (0 ، 3 –د) بیریمیدین -7 (1هـ) – اون 10 مع خلیط من الأیثیل کلورفورمات -کحول ایثیلی نتجت مشتقات -0 (ایثوکسی کربونیل أمینو -1 – مشتق ثیوثیازول -3 – (-3 – -4)

وقد أسفر الإنصهار الحرارى للمركبات (٤) متبعاً بالمعاملة مع كبريتات الميثيل عن أعطاء ثلاثي مشتق ثيازولو (٥، ٤-د) البيرميدينونات المقابلة (٦) وعند مفاعلة استر الثيازولات (١٧ ، ب) مع الايثيل كلورفورمات تم الحصول على ٥- (ايثوكسى كربونيل) أمينو . ثيازول -٤- كربوكسيلات (٨).