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Flavonoids from Albizia procera

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ARTICLE INFO	A B S T R A C T
<i>Article history:</i> Received 17 May 2011 Accepted 05 July 2011	he flavonoids luteolin, quercetin-3-O- α -L-rhamnopyranoside, quercetin-3-O- β -galactopyranoside, quercetin-3-O- β -D-glucopyranoside and quercetin-3-O-(2" β -D-xylopyranosyl)- β -D-galactopyranoside, were isolated from the methanolic
<i>Keywords: Albizia procera</i> ; Leguminosae; Flavonoids.	Identification of the flavonoid constituents was carried by analyzing their spectroscopic data and/or by comparing these data with the reported ones in the literature.

Introduction

The genus *Albizia* (Leguminosae) comprises about 150 species widely distributed in Africa and Central and South America. In Africa, several *Albizia* species are used in folk medicine for the treatment of rheumatism, stomach trouble, cough, diarrhea, wounds and as anthelmintic¹.

In traditional Chinese medicine, *Albizia* members are used for the treatment of insominia, irritability, wounds and as antidysentric, antiseptic and antitubercular². Phytochemical studies carried out on plants of genus *Albizia*, have revealed them as sources of different groups of natural products, triterpenoid saponins³⁻¹¹, diterpenoids¹², flavonoids¹³⁻²², lignans^{21,23}, phenolic glycosides^{24,25} and pyridineglycoside²⁶.

As a part of our continuous investigation on secondary metabolites from *Albizia* species^{9,11}, we describe in this report the isolation of five flavonoids from the methanolic extract of *A. procera*.

Materials and Methods

General experimental procedure

Column chromatography was run using silica gel 60 (Merck) and Sephadex LH-20 (Sigma). TLC was carried out using silica gel (Merck) as stationary phase. Chromatograms were first visualized by observation under UV and then spraying with FeCl₃ spray reagent. NMR spectra were recorded on a JEOL EX 500 MHz and a Varian GEMINT 200 MHz spectrometers. Chemical shifts were given on δ -scale with TMS as internal standard.

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Plant material

The leaves of *A. procera* were collected from the zoological garden in Giza, Egypt. Plant identification was confirmed by Mrs. T. Labib, head specialist for plant identification in El-Orman public garden, Giza, Egypt.

Extraction and Isolation

Air-dried and powdered leaves (750 g) of A. procera were extracted with chloroform (2L x 2) and then with MeOH (2L x 2) at room temperature. The total alcoholic extract was evaporated to dryness under reduced pressure. The residue (30 g) was suspended in distilled water (500 ml) and successively partitioned with CH₂Cl₂ (500 ml x 2), EtOAc (700 ml x 3) and BuOH (700 ml x 3). The BuOH fraction (10 g) was subjected to silica gel column chromatography. The column was eluted first with CH₂Cl₂ and then with CH₂Cl₂-MeOH mixture with increasing amount of MeOH to 20%. A total 40 fractions 100 ml each were collected. The fractions were monitored by silica gel TLC plates using solvent system CHCl₃-MeOH-H₂O (60: 30: 5) and examined under UV light followed by spraying with FeCl₃ reagent. The fractions which eluted with 10% MeOH were combined after TLC analysis contained compound 1 as a major component. The combined fraction was repeatedly chromatographed on Sephadex LH-20 column using methanol as eluent to yield pure 1 (40 mg). The fractions which eluted from silica gel column with 12-14% MeOH were combined contained compound 2 as a major component. The combined fraction was repeatedly chromatographed on Sephadex LH-20 column using methanol as eluent to yield pure 2

(50 mg). The fractions which eluted with 15% MeOH were combined. The combined fraction was subjected to preparative paper chromatography eluted with solvent system n-Butanol - Acetic acid – Water (4: 1: 5, upper layer) followed by repeated purification on Sephadex LH-20 column eluted with MeOH to yield inseparable mixture of **3** and **4** (30 mg). The fractions which eluted from the silica gel column with 17% methanol were combined after TLC analysis and found to contain compound **5** as a major constituent. The combined fraction was chromatographed on Sephadex LH-20 column chromatography using methanol as eluent to yield pure **5** (25 mg).

Luteolin (1)

Amorphous yellow powder. ¹HNMR (500 MHz, acetone-d₆) δ : 12.96 (1H, s, OH-5), 7.46 (1H, d, *J*=2.3 Hz, H-2'), 7.44 (1H, dd, *J*=8.4, 2.3 Hz, H-6'), 6.96 (1H, d, *J*= 8.4 Hz, H-5'), 6.55 (1H, s, H-3), 6.49 (1H, d, *J*= 2.3Hz, H-8), 6.21 (1H, d, *J*= 2.3 Hz, H-6), (OH-7 & OH-4' were not detected).

Quercetin-3-O-α-L-rhamnopyranoside (2)

Amorphous yellow powder. ¹HNMR (500 MHz, acetone-d₆) δ : 12.68 (1H, s, OH-5), 7.46 (1H, d, *J*=2.3 Hz, H-2'), 7.34 (1H, dd, *J*=8.4, 2.3Hz, H-6'), 6.95 (1H, d, *J*=8.4 Hz, H-5'), 6.43 (1H, d, *J*=2.3 Hz, H-8), 6.22 (1H, d, *J*= 2.3 Hz, H-6), 5.45 (1H, d, *J*= 1.5 Hz, Rha H-1"), 3.30-3.32 (Rha H-2" – H-5"), 0.88 (3H, d, *J*= 5.6 Hz, Rha Me-6"), (OH-7 & OH-4' were not detected). ¹³CNMR (500 MHz, acetone-d₆) δ : 178.5 (C-4), 164.5 (C-7), 162.3 (C-5), 157.6 (C-2), 157.2 (C-9), 148.5 (C-4'), 145.1 (C-3'), 135.0 (C-3), 121.9 (C-1'), 121.6 (C-6'), 115.9 (C-2'), 115.3 (C-5') 105.0 (C-10), 102.0 (C-1"), 98.7 (C-6), 93.7 (C-8), 71.3 (C-4"), 70.6 (C-2"), 70.5 (C-3"), 70.5 (C-5"), 16.9 (C-6").

Quercetin-3-O-β-D-galactopyranoside (3)

Amorphous yellow powder. ¹HNMR (500 MHz, acetone-d₆+D₂O) δ : 7.93 (1H, d, J=2.3 Hz, H-2'), 7.52 (1H, dd, J=8.6, 2.3Hz, H-6'), 6.88 (1H, d, J=8.6 Hz, H-5'), 6.45 (1H, d, J=2.5 Hz, H-8), 6.20 (1H, d, J=2.3 Hz, H-6), 5.12 (1H, d, J=7.7 Hz, Gal H-1"), 3.89 (1H, d, J=3.1 Hz, Gal H-5"), 3.82 (1H, t, J=7.7 Hz, Gal H-6"a), (OH-7, OH-4' & OH-5 were not detected). ¹³CNMR (500 MHz, acetone-d₆+D₂O) δ : 178.1 (C-4), 164.8 (C-7), 161.4 (C-5), 157.3 (C-2), 157.0 (C-9), 148.8 (C-4'), 144.6 (C-3'), 134.5 (C-3), 121.6 (C-1'), 121.4 (C-6'), 116.9 (C-5'), 115.2 (C-2') 104.2 (C-10), 104.0 (C-1"), 99.0 (C-6), 94.0 (C-8), 75.6 (C-5"), 73.6 (C-3"), 71.8 (C-2"), 68.1 (C-4"), 60.2 (C-6").

Quercetin-3-O-β-D-glucopyranoside (4)

Amorphous yellow powder. ¹HNMR (500 MHz, acetone- d_6+D_2O) δ : 7.79 (1H, d, J=2.3 Hz, H-2'), 7.52 (1H, dd, J=8.6, 2.3 Hz, H-6'), 6.89 (1H, d, J=8.6 Hz, H-5'), 6.44 (1H, d, J=2.5 Hz, H-8), 6.20 (1H, d, J=2.5 Hz, H-6), 5.21 (1H, d, J=7.6 Hz, Glc H-1"), 3.23 (1H, m, Glc H-5"), (OH-7, OH-4' & OH-5 were not detected). ¹³CNMR (500 MHz, acetone- d_6+D_2O) δ : 178.1 (C-4),

164.9 (C-7), 161.4 (C-5), 157.3 (C-2), 157.0 (C-9), 148.8 (C-4'), 144.6 (C-3'), 134.5 (C-3), 121.6 (C-1'), 121.4 (C-6'), 116.9 (C-5'), 115.2 (C-2') 104.2 (C-10), 103.0 (C-1"), 99.0 (C-6), 94.0 (C-8), 76.8 (C-5"), 76.6 (C-3"), 74.3 (C-2"), 69.5 (C-4"), 61.0 (C-6").

$Quercetin-3-O-(2''-O-\beta-D-xylopyranoxyl)-\beta-D-galactopyranoside \ (5)$

Amorphous yellow powder. ¹HNMR (500 MHz, DMSO-d₆) δ : 12.67 (1H, s, OH-5), 7.68 (1H, d, *J*=8.4 Hz, H-6'), 7.47 (1H, bs, H-2'), 6.78 (1H, d, *J*=8.4 Hz, H-5'), 6.35 (1H, bs, H-8), 6.14 (1H, bs, H-6), 5.66 (1H, d, *J*=7.7 Hz, Gal H-1"), 4.53 (1H, d, *J*=6.9 Hz, Xyl H-1"'), (OH-7 & OH-4' were not detected). ¹³CNMR (500 MHz, acetone-d₆+D₂O) δ : 178.3 (C-4), 164.4 (C-7), 161.5 (C-5), 156.9 (C-2, 9), 148.6 (C-4'), 144.7 (C-3'), 133.8 (C-3), 122.0 (C-1'), 122.6 (C-6'), 116.4 (C-5'), 115.4 (C-2') 104.4 (C-10), 104.2 (C-1"'), 100.0 (C-1"), 99.6 (C-6), 93.9 (C-8), 79.4 (C-2"), 75.4 (C-2"', 3"'), 73.7 (C-5"), 73.6 (C-3"), 69.5 (C-4"), 68.2 (C-4"'), 65.1 (C-5"'), 60.0 (C-6").

General method for acid hydrolysis:

Each glycoside (2 mg) in 3 ml 2N HCl and 3 ml methanol was heated at 100°C for 2h. The mixture was left to cool, diluted with H_2O and extracted twice with ethyl acetate. From the ethyl acetate layer, the aglycone was detected against reference sample by TLC (chloroform-methanol, 5: 1). The aqueous layer was repeatedly diluted with methanol and evaporated to dryness. The residue was investigated to detect the sugar by paper chromatography (PC) using solvent system n-BuOH-HOAc-H₂O (4: 1: 5, upper layer).

Results and Discussion

The methanolic leaf extract of A. procera was dissolved in water and the aqueous solution was successively extracted with chloroform, ethyl acetate then n-butanol. The material from the butanol extract was chromatographed on silica gel column chromatography. The collected fractions were examined by TLC and similar fractions were combined. The semi-purified compounds from the combined fractions were then purified using Sephadex-LH20 column chromatography and preparative paper chromatography to afford the flavonoids luteolin (1), quercetin 3-O-a-Lrhamnopyranoside (2)quercetin 3-O-β-Dgalactopyranoside 3-O-β-D-(3) quercetin glucopyranoside (4) and quercetin $3-O-(2"-O-\beta-D$ xyloxyl)- β -D-galactopyranoside (5). The structures of the isolates were determined by acid hydrolysis, ¹H and ¹³CNMR spectroscopic analysis and/or comparison with published data.

The flavonoid luteolin 1 was characterized by comparison of their spectral data with literature values²⁷.

In the ¹HNMR spectrum of Quercetin 3-O- α -L-rhamnopyranoside **2**, the presence of signals due to three *meta* – coupled protons at δ 6.22 (1H, d, *J*=2.3 Hz, H-6), δ 6.43 (1H, d, J= 2.3 Hz, H-8) and δ 7.46 (1H, d, J= 2.3 Hz, H-2') as well as two *ortho*-coupled protons at δ 6.95 (1H, d, J= 8.4 Hz, H-5') and δ 7.34 (1H, dd, J= 8.4, 2.3 Hz, H-6') along with carbon signal at δ 178.5 (C-4) in the ¹³CNMR spectrum, indicated a typical 3', 4', 5, 7-tetrahydroxyflavonol skeleton. The two doublets in the ¹HNMR spectrum of **2** at δ 5.45 (1H, J= 1.5 Hz) and δ 0.88 (3H, J= 5.6 Hz) were assigned to H-1" and H-6" of a rhamnose unit²⁷. The ¹³CNMR data of **2** and the result of acid hydrolysis indicated α -L-rhamnopyranoside. The attachment of the moiety at C-3 position of the aglycone was revealed from the δ value of C-3 (135.0 ppm) and that of C-2 (157.6 ppm).

Compound **2** was then assigned the structure of quercetin 3-O- α -L-rhamnopyranoside based on the above spectral evidences and comparison with literature values^{28,29}.

Quercetin-3-O-β-D-galactopyranoside 3 and Quercetin-3-O-β-D-glucopyranoside 4 were obtained as inseparable mixture with compound 3 as the major component. The ¹HNMR spectrum of **3** and **4** showed two sets of signals due to quercetin moiety. One set of signals was at δ 7.93 (1H, d, J=2.3 Hz, H-2'), 7.52 (1H, dd, J=8.6, 2.3Hz, H-6'), 6.88 (1H, d, J=8.6 Hz, H-5'), 6.45 (1H, d, J=2.5 Hz, H-8), 6.20 (1H, d, J= 2.3 Hz, H-6). The other set was at δ 7.79 (1H, d, J=2.3 Hz, H-2'), 7.52 (1H, dd, J=8.6, 2.3 Hz, H-6'), 6.89 (1H, d, J=8.6 Hz, H-5'), 6.44 (1H, d, J=2.5 Hz, H-8), 6.20 (1H, d, J= 2.5 Hz, H-6). The spectrum also exhibited two proton signals at δ 5.12 (1H, d, J=7.7 Hz) and δ 5.21 (1H, d, J=7.6 Hz) assigned to anomeric proton of β -galactose and that of β -glucose, respectively. The ¹³CNMR spectum of 3 and 4 showed two sets of signals due to the querctin moiety and two sets of signals attributable to a and glucose moieties, respectively galactose (experimental section). The signals of galactose moiety were at δ 104.0 (C-1"), 75.6 (C-5"), 73.6 (C-3"), 71.8 (C-2"), 68.1 (C-4") and 60.2 (C-6") while the signals due to the glucose moiety were at $\delta 103.0$ (C-1"), 76.8 (C-5"), 76.6 (C-3"), 74.3 (C-2"), 69.5 (C-4"), 61.0 (C-6"). Acid hydrolysis of 3 and 4 afforded quercetin identified by TLC comparison against authentic sample as well as the sugar component D-galactose and D- glucose recognized by paper chromatography against standard samples. The β -configuration of the anomeric centers was determined as β from the J_{1,2} values and the ¹³CNMR data indicated pyranose form for the two sugar units. The site attachment of the sugar moiety to C-3 position of quercetin, was established from the δ value of C-3 resonances at δ 134.5 and C-2 resonances at δ 157.3 for **3** and **4**. These values agreed well with the literature values^{28,30} and the corresponding ones in the ¹³CNMR data of compound **2** as well as other quercetin-3-O-glycosides^{28,29}.

Quercetin-3-O-(2"-O-\beta-D-xylopyranosyl)-\beta-D-galactopyranoside **5** is another quercetin 3-O-glycoside from the characteristic ¹H and ¹³CNMR resonances as well as acid hydrolysis which afforded quercetin moiety and the sugar components D-galactose and D-xylose. The ¹³CNMR spectrum of **5** showed in addition to the resonances of the quercetin moiety, eleven carbon resonances assigned to a 2-O-substituted ßgalactopyranose (C-2", 79.2 ppm; C-1", 100.0 ppm) and a terminal β -D-xylopyranose units. The β -configurations of the anomeric protons of D-galactose and D-xylose were deduced from the $J_{1,2}$ values for both sugar units. The pyranose form for the sugar units was determined from their ¹³CNMR data. The attachment of the terminal β -D-xylopyranosyl unit to the inner β -D-xylopyranosyl unit was established from the observed downfield position of C-2" and upfield position of C-1" of the inner sugar with reference to the corresponding values for compound **3** (δ C-2", 71.8; δ C-1", 104.0). The full structure of 5 was confirmed by comparing its spectral data with those reported in the literature 31 .

Luteolin²⁰, quercetin 3-O- α -L-rhamnopyranoside^{17,22} and quercetin-3-O- β -D-galactopyranoside²² were previously reported from *A. julibrissin*. This is the first reported occurrence of quercetin-3-O- β -Dglucopyranoside and quercetin-3-O-(2"-O- β -Dxylopyranosyl)- β -D-galactopyranoside in genus *albizia*.



	R	R 1
1	Н	OH
2	O-α-L-Rhamnopyranosyl	OH
3	O-β-D-Galactopyranosyl	OH
4	O-β-D-Glucopyranosyl	OH
5	$O-\beta-D-xylopyranosyl-(1\rightarrow 2)-\beta-D-galactopyranosyl$	OH

References

- Watt, J.M. Breyer-brandwijk, M.C. (1962). Medicinal and poisonous Plants of South and East Africa. Second ed. Livingstone, Edinburgh, p. 553.
- Chadha, Y.R. (ed.) (1985). Vol. 1-A, p. 126. CSIR, New Delhi.
- **3)** Pal, B.C., Achari, B., Yoshikawa, K., Arihara, S. (1995) *Phytochemistry* **38**, 1287.
- Ikeda, T., Fujiwara, S., Kinjo, J., Nohara, T., Ida, Y., Shoji, J., Shingu, T., Isobe, R. and Kajimoto, T. (1995) *Bull Chem Soc Jpn* 68, 3483.
- 5) Ma, L., Tu, G., Chen, S., Zhang, R., Lai, L., Xu, X., Tang, Y. (1996) *Carbohydrate Research* **281**, 35.
- Yoshikawa, K., Satou, Y., Tokunaga, Y., Tanaka, M., Arihara, S., Nigam, S. K. (1998) *J. Nat. Prod.* 61, 440.
- Liang, H., Tong, W., Zhao, Y., Cui, J., Tu, G. (2005) *Bioorganic & Medicinal Chemistry Letters* 15, 4493.
- 8) Zheng, L., Zheng, J., Zhao, Y., Wang, B., Wu, L., Liang, H. (2006) *Bioorganic & Medicinal Chemistry Letters* 16, 2765.
- Melek, F.R., Miyase, T., Ghaly, N.S., Nabil, M. (2007) *Phytochemistry* 68: 1261.
- 10) Liu, R., Ma, S., Yu, S., Pei, Y., Zhang, S., Chen, X., Zheng, J. (2009a) Journal of Natural products 72: 632.
- 11) Miyase, T., Melek, F.R., Ghaly, N.S., Warashina, T., El-Kady, M., Nabil, M. (2010) *Phytochemistry* 71: 1375.
- 12) Cheng, Z.Q., Yang D., Liu, Y.Q., Hu, J.M., Jiang, H.Z., Wang, P.C., Li, N., Zhou, J., Zhao, Y.X. (2010) Journal Brazilian Chemical Society 21, 1766.
- Rukunga, G.M., Waterman, P.G. (1996) Bulletin of the Chemical Society of Ethiopia 10(1), 47.
- 14) El-Mousallamy, A.M. (1998) *Phytochemistry* 48, 759.

- 15) Yadava, R. N., Tripathi, P. (1999) J. Inst. Chem. 71, 202.
- 16) Yadava, R. N., Tripathi, P. (2000) *Res. J. Chem. Environ.* 4, 57.
- 17) Li, Z.P., Gao, S., Hao, C.S., Fan, G.M. (2000) *Zhongguo Zhong Yao Za Zhi* 25, 103.
- 18) Rao, Y.K., Reddy, M.V.B., Rao, C.V., Gunasekar, D., Blond, A., Caux, C., Bodo, B. (2002) *Chem. Pharm. Bull.* 50, 1271.
- **19**) Pandey, M.K., Pandey, R., Singh, V.P., Pandey, V.B., Singh, U.P. (2002) *Mycobiology* **30**, 55.
- **20**) Jung, M.J., Chung, H.Y., Kang, S.S., Choi, J.H., Bae, K.S., Choi, J.S. (2003) *Arch. Pharm. Res.* **26**, 458.
- 21) Jung, M.J., Kang, S.S., Jung, H.A., Kim, G.J., Choi, J.S. (2004) Arch. Pharm. Res. 27, 593.
- 22) Lau, C.S., Carrier, D.J., Beitle, R.R., Bransby, D.I., Howard, L.R., Lay, J. J.O., Liyanage, R., Clausen, E.C. (2007). *Bioresource Technology* 98: 429.
- 23) Ito, A., Kasai, R., Yamasaki, K., Duc, N.M., Nham, N.T. (1994) *Phytochemistry* 37, 1455.
- 24) Ma, Y., Hsiao, S., Chen, H., Hsu, F. (1997) *Phytochemistry* 46, 1451.
- **25**) Liu, R., Yu, S., Pei, Y. (2009b) *Zhongguo Zhong Yao Za Zhi* **34**(16): 2063.
- **26**) Orsini, F., Pelizzoni, F., Pulici, M., Verotta, L. (1989) *Gazz. Chim. Ital* **119**, 63.
- 27) Harborne, J.B. (1993) p. 589-618. "The Flavonoids: Advances in Research since 1986". Chapman and Hall, London.
- **28**) Markham, K.R., Ternai, B., Stanley, R., Geiger, H., Mabry, T.J. (1978) *Tetrahedron* **34**: 1389.
- **29**) Asen, A., Horowitz, R.M. (1977) Phytochemistry **16**, 147.
- **30**) Yasukawa, K., Takido, M. (1987) Phytochemistry **26**, 1224.
- **31**) Larsen, L.M., Nielsen, J.K., Sorensen, H. (1982) Phytochemistry, **21**, 1029