Determination of available heavy metal concentrations in River Nile sediments, Damietta, Egypt

Ehab Assal¹, Magdy Khalil¹, Ahmed Rakha², Mohamed Elakad^{*1}

¹Geology Department, Faculty of Science, Damietta University, New Damietta, Egypt ²Egyptian Environmental Affairs Agency, Egypt

Received: 13 September 2020 /Accepted: 4 November 2020 * Corresponding author's E-mail: mohamed.a.elakad@gmail.com

Abstract

The present study investigates the distribution and available concentrations of As, Co, Cu, Fe, Sb, Zn, Cd, Se, Ni, Pb in the bottom sediment of the Damietta Nile branch. Available concentrations were determined using the DTPA extraction method. The mean available concentrations were arranged as follows As > Fe > Cu > Zn > Co > Se > Ni > Pb > Sb > Cd. Clustering analysis represents a clear difference between two groups of heavy metal concentrations before and after Damietta Dam. Textural classifications and Total organic matter have been studies. The matrix correlation coefficient, comparison between other studies has been evaluated.

Keywords: Heavy metals, River Nile, sediments, DTPA, Damietta, pollution.

Introduction

Nile River is the main source of freshwater in Egypt. Damietta branch is the main drinking and agricultural water source in Damietta, neighboring region, and connecting to North of Sinai through the Al-Salam canal.

Due to the construction of Aswan High Dam, major changes in the chemical, physical, and biological properties of Nile water were recorded (e.g., <u>Abd El-Hady, 2014</u>). As a result of the continuous population increase along the Nile River, many activities affect negatively on the environment, such as agricultural wastes discharge loaded with impurities of fertilizers and the industrial waste that may contain harmful elements. Fish farms that were located in parts of the study area are a major source of pollution. The above makes it necessary to study the possible pollutants in the area as heavy metal contamination in the bottom sediments.

Understanding heavy metals interaction in the sediment is essential to evaluate and predict their behavior in the sediment-water system. Sediments reflect the current quality of the water system and the chemical parameters history, they are considering as a carrier and probable source of contaminants (Haiyan et al., 2013). Heavy metals contamination in the Nile River and its tributaries can have destructive effects on the ecological system of the river resources and food diversity. The amount of heavy metals in water and even some regions of the Nile is well known to be higher than the limit set by the Egyptian General Authority for Standards and Quality (Anwar et al., 2001).

Most heavy metals in the aquatic environment can be indifferently deposited in the bottom sediments (Salomons et al., 1995). The

of sediments composed different are components, which include carbonates, quartz, clay minerals, feldspar, and organic matter (Tsirambides, 1992). In the river sediments; heavy metals have divided into three groups of components: Erosional and weathering of rocks and solid origin (Detrital solids), water column processes origin (Endogenic fractions), and sediments process (Diagenetic fractions) (Förstner et al., 1976; Jones and Bowser, 1978). The availability of heavy metals is depending on pH, Eh, salinity, organic matter concentrations, and inorganic complexation agents (Calmano and Förstner, 1983).

Two main sources that heavy metals may be derived from, it can be natural (lithogenic source) from the parent material and various anthropogenic (contamination) sources (Alloway, 2012; Ramos-Miras et al., 2011). The main cause of heavy metal contamination in the aquatic ecosystem are anthropogenic activities, such as mining, industrial waste, agricultural production, vehicle emission, and air precipitation (Keshavarzi et al., 2015; Liu et al., 2018). In the study area, fish farms also can be included as a possible source of pollution.

The study area is located between latitudes 31°22'24.45"N, 31°31'12.89"N and longitudes 31°42'46.07"E, 31°50'33.13"E (Figure. 1). It's about 32 Km long starting from Fareskour city and northward to the river mouth near to Ras El-Bar City. It passes through Kafr Albatikh, El-Horany, El-Bostan, Al-Adleiia, Damietta, and Ezbet El-Borg in front of Ras El-Bar city.

This study aims to investigate the presence and distribution of ten heavy metals in the bottom sediments of the study area.



Figure. 1 Study area and sampling stations

Materials and methods

Using a stainless-steel bottom sampling dredge,

thirty-three bottom sediment samples were collected from the main-stream of the Nile from eleven sites along the study area (Figure. 1). The depth of collected samples varies from 4 to 13 m with a mean of 7.3m (Table 1). Collection sites were determined using Geographic Position System (GPS), the study area map was drawn using adobe illustrator. At each site, three samples were collected to carry out the grain size analysis, measure the trace metals concentrations, and Collected samples that were preserved in polyethylene bags and carried to the laboratory. Storage and manipulation of sediments for chemical analysis were carried out according to the United States Environmental Protection Agency (USEPA, 2001). Sediment samples were air-dried, homogenized using an agate mortar.

Using a pH meter, pH values were measured in wet sediments at the faculty of science laboratories, Damietta university (Model: Walklab -TI9000). The determination of total organic matter (TOM) content was carried out on the oven, samples dried at 105°C and sieved (<2mm).

Five grams of samples were weighed and placed in an ignition oven at 550°C for 4 hours. After ignition, the samples were left for an hour. Total organic matter is the final loss of weight (Mucha et al., 2003; 2004). Sand-Silt-Clay content was determined according to Folk (1980) at the faculty of agricultural laboratory, Mansoura university (Table 1).

Heavy metal was extracted using a buffered diethylene triamine penta acetic acid (DTPA) solution method (Lindsay and Norvell 1978). This method is specified by the International Standardization Organization for (ISO 14870:2001) and provides a rapid procedure to assess metal solubilities and contamination in sediments. Heavy metals (As, Co, Cu, Fe, Se, Zn, Cd, Sb, Ni, Pb) were determined using a flame atomic absorption spectrometer (Model: PinAAcle 500) at faculty of science, Damietta University laboratories.

The statistical analyses were carried out using the Statistical Package for the Social Sciences (SPSS) version 26, calculating the range (MIN.-MAX.). average. Standard deviation (SDV). and the correlation coefficient was performed to potential relationships evaluate between available heavy metals concentration and sediment properties.

Site	Sample	Coordinates	Sediment type	Depth (m)	pН	TOM (%)
Site-1	D1				6.83	07.11
	D2	31°22'24.45"N – 31°42'46.07"E	Sandy siltstone	6.50		07.65
	D3					07.67
	D4			8.00	6.91	23.68
Site-2	D5	31°22'48.41"N – 31°43'2.56"E	Sandy siltstone			23.13
	D6					20.01
	D7				6.71	11.44
Site-3	D8	31°22'49.69"N – 31°44'9.51"E	Muddy sandstone	6.10		11.35
	D9					10.95
	D10				6.88	07.56
Site-4	D11	31°23'39.36"N – 31°45'0.48"E	Silty sandstone	7.50		07.27
	D12					08.68
	D13				6.64	07.98
Site-5	D14	31°24'26.67"N – 31°45'27.31"E	Sandy siltstone	10.00		08.26
	D15					08.67
	D16				6.59	09.35
Site-6	D17	31°23'49.78"N – 31°45'47.77"E	Sandy siltstone	11.50		09.88
	D18					10.28
	D19			13.00	6.65	04.56
Site-7	D20	31°23'52.33"N – 31°46'18.58"E	Silty sandstone			04.13
	D21					03.85
	D22	31°24'31.40"N – 31°47'5.94"E		4.00	6.44	12.65
Site-8	D23		Silty sandstone			12.74
	D24					11.93
	D25				6.53	06.75
Site-9	D26	31°25'26.32"N – 31°48'15.48"E	Sandy mudstone	7.50		06.85
	D27					05.80
	D28			6.10	6.71	14.23
Site-10	D29	31°27'32.38"N – 31°48'9.23"E	Sandy mudstone			14.69
	D30					13.08
Site-11	D31					13.87
	D32	31°31'12.98"N – 31°50'33.13"E	Muddy sandstone	4.00	6.78	13.23
	D33					13.53
Min.					6.44	3.85
Max.					6.91	23.68
Average					6.70	10.69
SDV					0.15	4.80

Table. 1 Samples location, sediment type, depth, pH. Total organic matter in the bottom sediment samples along the study area.



Figure. 2 Ternary plot displays the abundances of sand-silt-clay in collected sediments from the Damietta Nile river (Folk, 1980).

Results

Textural classification

The distributions of Sand-Silt-Clay along the Damietta branch are listed in (Table 1) and

illustrated in the triangular diagram after Folk (1980) (Figure 2). The textural classification of the Damietta Nile branch is described as sandy siltstone, silty sandstone, sandy mudstone, and muddy sandstone arranged in descending order of abundance. pH data ranges from 6.44 to 6.91 with an average of 6.70. The highest pH value is recorded at Damietta Power Plant (site-2), while the lowest was observed at Damietta Dam (site-8). Total organic matter (TOM) percentage varied between 3.85% to 23.68% with an average of 10.69%. Hight TOM was recorded at Damietta Power Plant (site-2) while the lowest percentage was after Al-Salam Canal (site-7) (Table 1).

Available heavy metals Distributions

Table 2 presents the concentrations of 33 available heavy metals in 11 sampling sites

along the study area. Heavy metals concentration can afford important information for the bottom sediment pollution levels and environmental risk assessment (Palma et al., 2015). Heavy metal levels average are in the following order As > Fe > Cu > Zn > Co > Se > Ni > Pb > Sb > Cd. The matrix correlation coefficient is illustrated in Table 3. The following is a brief description of each studied heavy metals.

Arsenic (As)

Concentrations of As varied from 80.19 µg/g at Damietta to 126.24 μ g/g with averaging 99.9 μ g/g. Distributions of As in the bottom sediments exhibit a general increase at sites Aland Al-Adliia-2 and Adliia-1 lower concentrations at Damietta City site. As has a highly significant positive correlation coefficient with Cu (r = +0.628) and there is a significant negative correlation coefficient with Pb (r = -0.667), Cd (r = -0.600) Zn (r = -0.548)

and Sb (r = -0.542).

Cobalt (Co)

Cobalt concentrations range from 02.42 μ g/g to 8.95 μ g/g with averaging 5.37 μ g/g. The maximum content of cobalt was at Al-Adliia-1 station and the lowest content at the Kafr Albatikh Water Station site. Co has a highly significant positive correlation coefficient with Fe (r = 0.781) and Ni (r = 0.916).

Copper (Cu)

Cupper ranges from 2.68 μ g/g to 16.96 μ g/g with an average of 10.71 µg/g. Cu has a maximum content at the Al-Adliia-2 site and the lowest at the Damietta City site. Cu has a highly significant positive correlation coefficient with Fe (r = 0.701) and Ni (r = 0.650) and a highly significant negative correlation coefficient with Pb (r = -0.741)

Table. 2 Available heavy metal concentrations in the bottom sediment samples along the study area.

Site	Sample	As	Со	Cu	Fe	Sb	Zn	Cd	Se	Ni	Pb
Site-1	D1	88.10	02.42	08.36	11.81	00.49	05.76	00.04	00.00	00.91	01.26
	D2	84.71	03.67	07.66	13.32	00.41	05.15	00.05	00.00	01.12	01.07
	D3	86.10	02.86	10.59	14.49	00.48	05.12	00.04	00.00	00.84	00.99
	D4	93.43	04.14	11.18	85.69	00.35	06.08	00.04	04.31	01.44	00.83
Site-2	D5	96.76	04.56	11.16	81.76	00.38	06.16	00.03	04.39	01.56	00.77
	D6	95.47	04.67	10.83	86.77	00.41	05.91	00.05	04.28	01.62	00.80
	D7	103.41	06.03	14.47	123.35	00.53	06.15	00.04	02.57	02.35	00.89
Site-3	D8	104.61	06.33	14.39	121.74	00.52	06.44	00.04	02.77	02.39	00.95
	D9	102.54	06.51	14.49	119.65	00.46	06.27	00.04	03.11	02.57	00.73
	D10	99.34	05.31	12.10	128.62	00.23	05.43	00.04	01.09	02.14	00.59
Site-4	D11	99.93	05.24	12.05	129.46	00.19	06.23	00.04	01.00	02.36	00.54
	D12	99.59	05.32	11.98	131.56	00.21	06.18	00.04	00.92	01.90	00.53
	D13	104.96	08.69	14.43	237.90	00.48	06.66	00.04	03.79	02.69	00.45
Site-5	D14	104.95	08.95	14.85	238.40	00.48	06.78	00.04	03.56	02.71	00.45
	D15	106.65	08.91	14.47	242.50	00.51	07.17	00.05	03.76	02.76	00.46
	D16	125.93	07.46	16.96	138.91	00.60	06.54	00.04	04.49	02.48	00.45
Site-6	D17	126.24	07.42	16.12	140.15	00.56	06.67	00.04	04.42	02.46	00.44
	D18	125.77	07.48	16.19	140.88	00.61	06.49	00.04	04.47	02.50	00.34
	D19	115.91	05.60	13.56	38.01	00.63	06.79	00.05	05.09	02.24	00.82
Site-7	D20	116.56	05.62	12.99	37.96	00.61	06.60	00.04	05.11	02.53	00.86
	D21	116.67	05.82	13.07	38.21	00.67	06.95	00.05	04.97	02.06	00.78
	D22	118.96	02.50	07.10	04.85	00.57	04.87	00.04	02.45	00.65	00.77
Site-8	D23	118.53	02.55	07.15	04.52	00.50	04.71	00.03	02.59	00.56	00.80
	D24	118.79	02.57	07.06	05.13	00.61	04.62	00.03	02.44	00.73	00.87
	D25	80.19	06.94	02.68	35.33	03.52	08.76	00.23	13.09	01.92	03.51
Site-9	D26	81.12	07.23	03.24	34.98	03.19	09.24	00.20	13.02	02.14	03.75
	D27	80.31	06.83	03.05	36.19	03.99	08.80	00.20	13.10	01.93	03.47
	D28	84.16	05.61	10.84	76.50	02.12	08.82	00.13	03.66	01.59	02.02
Site-10	D29	84.40	05.21	10.95	76.22	02.34	08.80	00.13	03.71	01.56	02.01
	D30	84.94	05.06	10.94	76.48	01.96	09.28	00.15	03.91	01.83	02.02
Site-11	D31	82.91	03.26	06.23	38.11	01.02	08.19	00.06	04.22	01.05	01.33
	D32	82.54	03.29	06.22	38.09	01.07	08.11	00.07	04.14	01.08	01.40
	D33	82.23	03.09	06.21	38.16	00.95	08.12	00.07	04.30	01.09	01.27
Min.		80.19	02.42	02.68	04.52	00.19	04.62	00.03	00.00	00.56	00.34
Max.		126.24	08.95	16.96	242.50	03.99	09.28	00.23	13.10	02.76	03.75
Average		99.90	05.37	10.71	83.81	00.96	06.78	00.07	04.08	01.81	01.16
SDV		15.24	01.93	03.97	67.63	00.98	01.35	00.05	03.25	00.68	00.89

Iron (Fe)

Fe ranges from 04.52 μ g/g to 242.5 μ g/g with an average of 83.81 μ g/g. The high content of Fe was recorded at the Al-Adliia-1 site, and the lowest values were at the Damietta Dam site. Fe has a highly significant positive correlation coefficient with Ni (r = 0.764) and a significant negative correlation coefficient with Pb (r = -0.428).



Figure. 3 Heavy metal concentrations in the bottom sediment samples along the study area.

Antimony (Sb)

Sb ranges from 0.19 μ g/g to 3.99 μ g/g with an average 0.96 μ g/g. The maximum concentration was at Damietta City site and the lowest at El-Bostan site. Sb has a highly significant positive correlation coefficient with Zn (r = 0.779), Cd (r = 0.976), Se (r = 0.827) and Pb (r = 0.956).

Zinc(Zn)

Zn ranges from 4.62 μ g/g to 9.28 μ g/g with an average of 6.78 μ g/g. The higher value was recorded at Damietta harbor canal station and the lowest was at Damietta Dam station. Zn has highly significant positive correlation a coefficient with Cd (r = 0.789), Se (r = 0.659) and Pb (r = 0.711).

Cadmium (Cd)

Cd ranges from 0.03 μ g/g to 0.23 μ g/g and averaging 0.067 μ g/g. Cd distribution has high content at Damietta Dam and low content at the Damietta Plant site. Cd has a highly significant positive correlation coefficient with Se (r =0.796) and Pb (r = 0.959).

Selenium (Se)

Se concentration ranges from zero to $13.10 \,\mu\text{g/g}$ an average of 4.1 μ g/g. The maximum concentration was at the Damietta City site and the lowest at Kafe Albatikh site. Se has a highly significant positive correlation coefficient with Pb (r = 0.775).

Nickel (Ni)

Ni ranges from 0.56 μ g/g to 2.76 μ g/g at and averaging 1.81 μ g/g. Al-salam canal site was recorded the highest value of Ni and, the lowest value was at Damietta Dam. Ni has a significant negative correlation coefficient with TOM (r =-0.356).

Lead (Pb)

Pb ranges from 0.34 μ g/g at Al-Adliia to 3.75 μ g/g at Damietta city site and averaging 1.16 µg/g. Pb recorded the maximum values at Damietta City site and the lowest at Al-Adliia-2.

Correlations											
	As	Со	Cu	Fe	Sb	Zn	Cd	Se	Ni	Pb	TOM
As	1.000										
Со	0.231	1.000									
Cu	0.628 ^a	0.497 ^a	1.000								
Fe	0.274	0.781 ^a	0.701 ^a	1.000							
Sb	-0.542 ^a	0.196	-0.611 ^a	-0.284	1.000						
Zn	-0.548 ^a	0.348 ^b	-0.288	0.026	0.779 ^a	1.000					
Cd	-0.600 ^a	0.204	-0.609 ^a	-0.249	0.976 ^a	0.789 ^a	1.000				
Se	-0.240	0.407 ^b	-0.454 ^a	-0.127	0.827 ^a	0.659 ^a	0.796 ^a	1.000			
Ni	0.315	0.916 ^a	0.650 ^a	0.764 ^a	-0.012	0.229	0.009	0.234	1.000		
Pb	-0.667 ^a	0.030	-0.737 ^a	-0.428 ^b	0.956 ^a	0.711 ^a	0.959 ^a	0.775 ^a	-0.144	1.000	
TOM	-0.183	-0.330	-0.052	-0.041	-0.128	-0.039	-0.117	-0.118	-0.356 ^b	-0.098	1.000

 Table. 3 Correlation coefficient in the study area.

Bold is a significant positive correlation between metal pairs

^a Correlation is significant at the 0.01 level (2-tailed).

^b Correlation is significant at the 0.05 level (2-tailed).



Figure. 4 Hierarchical clustering, based on the correlation coefficient of similarity using betweengroup linkage method

Discussion

The study area can be separated into two regions; the first starts about 16 Km before Damietta Dam, the second after to the end of the river mouth in front of Ras El-Bar city. In comparison between concentrations of heavy elements before and after Damietta Dam shows that there are abnormally high levels of (Co, As, Cu, Fe, Ni) than the average values in Al-Adliia 1 and Al-Adliia 2 area (sites-5 & site-6), this increase may be due to the random fish farms and agricultural discharge in the mainstream. The second abnormal increases of (Co, Cd, Sb, Zn, Cd, Se, Pb) concentrations than average, observed in Damietta City (site-9), It may be related to anthropogenic activities and domestic waste discharge.

Hierarchical cluster analysis was operated on the analyzed heavy metals to recognize different groups that mav represent the sources that controlling their distribution, using between-group linkage (Figure 4) these results are deduced by a person's correlation coefficient (Table 3). Two main groups (A and B) at the level of similarity of 0.94 appeared in figure 4. The first group (A) involved Sb, Cd, Pb, Se, and Zn additional separated into two sub-clusters (Sb, Cd, and Pb/Zn, Se) clustered at the level of (0.82). The second group (B) involved Co, Ni, Fe, Cu, and As further can be parted into two sub-clusters (Co, Ni, and As, Cu, Fe) clustered at the level of (0.65). The intense correlation between available (Sb, Cd) in group A and (Co, Ni) in group B leads to the statement that they are initiated from the same sources (Massas and Ehaliotis, 2010). metals assembled under this cluster can be considered to have an anthropogenic source (Imperato et al. 2003; Moller et al. 2005; Wang et al. 2005). Probable high levels of Pb and As may be related to industrial activity and pesticides used in agriculture activity (Sany et al., 2013). Fertilizers, petroleum, textile, and sewage sludge is the main source of Cd (El Nemr et al., 2007), in this area also observed an active ships industry, domestic and agricultural discharge. The obtained data of the available heavy metals concentrations in the present study are compared with other studies (Table 4). In the present study; the concentrations of Ni and Fe are more than El-Namla (2019), although low in Cd, Cu, and Co.

Locations	Heavy Metal concentrations (mg g ⁻¹)										Defenences
Locations	As	Co	Cu	Fe	Sb	Zn	Cd	Se	Ni	Pb	References
Nile River (Damietta), Egypt	80.19- 126.24	2.42- 8.95	2.68- 16.96	4.52- 242.5	0.19– 3.99	4.62- 9.28	0.03- 0.23	0- 13.1	0.56– 2.76	0.34- 3.75	Present study
Nile River (Mansoura), Egypt	_	0.00- 29.18	01.04- 21.99	07.69- 128.23		_	00.31- 8.89	-	0.02- 1.57	-	El-Namla (2019)

Table. 4 The available heavy metal concentrations in Nile River sediment samples from various studies.

.

Conclusion

The collected bottom sediments from form Damietta branch have an average metal levels in the following order As > Fe > Cu > Zn > Co> Se > Ni > Pb > Sb > Cd. Damietta Dam separated the concentrations into two clustering A, B according to their abundance in the same sites. there are abnormally high concentrations levels of (Co, As, Cu, Fe, Ni) than the average values in Al-Adliia-1,2 (sites no. 5 & 6), this increase may be due to the random fish farms and agricultural discharge in the mainstream. And other abnormal increase of (Co, Cd, Sb, Zn, Cd, Se, Pb) concentrations than the average observed in Damietta City (site no.9). It may be related to anthropogenic activities and domestic waste discharge.

References

- El-Hady, H. H. A. (2014). Alternations in biochemical structures of phytoplankton in Aswan Reservoir and River Nile, Egypt. J. Biodivers. Environ. Sci, 4, 68-80.
- Haiyan L., Anbang, S., Mingyi, L., Xiaoran, Z. (2013). Effect of pH, temperature, dissolved oxygen, and flow rate of overlying water on heavy metals release from storm sewer sediments. Journal of Chemistry, 2013, 104316.
- Anwar, S. M., Aziza, E. S., El Serafy, S. S., Ibrahim, I. I., Ali, E. A. (2001). Accumulation of trace elements in fish at Lake Quaroun as a biomarker of environmental pollution. Journal-Egyptian german society of zoology, 36(A), 443-462.
- Salomons, W. (**1995**). Long-term strategies for handling contaminated sites and large-scale areas. In Biogeodynamics of pollutants in soils and sediments (pp. 1-30). Springer, Berlin, Heidelberg.
- Tsirambides, A. (**1992**). Texture and mineralogical composition of Quaternary terrestrial and marine sediments from Macedonia and Thrace, Greece. Bulletin of the Geological Society of Greece, 55(1), 137-157.

- Förstner, U., Müller, G. (**1976**). Heavy metal pollution monitoring by river sediments. Fortschritte der Mineralogic, 53, 271-288.
- Jones, B. F., Bowser, C. J. (**1978**). The mineralogy and related chemistry of lake sediments. In Lakes (pp. 179-235). Springer, New York, NY.
- Calmano, W., Förstner, U. (**1983**). Chemical extraction of heavy metals in polluted river sediments in central Europe. Science of the Total Environment, 28(1-3), 77-88.
- Alloway, B. J. (Ed.). (2012). Heavy metals in soils: trace metals and metalloids in soils and their bioavailability (Vol. 22). Springer Science & Business Media.
- Ramos-Miras, J. J., Roca-Perez, L., Guzmán-Palomino, M., Boluda, R., Gil, C. (2011).
 Background levels and baseline values of available heavy metals in Mediterranean greenhouse soils (Spain). Journal of Geochemical Exploration, 110(2), 186-192.
- Keshavarzi, B., Mokhtarzadeh, Z., Moore, F., Mehr, M. R., Lahijanzadeh, A., Rostami, S., Kaabi, H. (2015). Heavy metals and polycyclic aromatic hydrocarbons in surface sediments of Karoon River, Khuzestan Province, Iran. Environmental Science and Pollution Research, 22(23), 19077-19092.
- Liu, J., Liu, Y. J., Liu, Y., Liu, Z., Zhang, A. N. (2018). Quantitative contributions of the major sources of heavy metals in soils to ecosystem and human health risks: A case study of Yulin, China. Ecotoxicology and environmental safety, 164, 261-269.
- USEPA (2001). Methods for Collection, Storage and Manipulation of Sediments for Chemical and Toxicological Analyses: Technical Manual, EPA-823-B-01-002. Washington, DC: U.S. Environmental Protection Agency, Office of Water.
- Mucha, A. P., Vasconcelos, M. T. S., Bordalo, A. A. (2003). Macrobenthic community in the Douro estuary: relations with trace metals and natural sediment characteristics. Environmental pollution, 121(2), 169-180.
- Mucha, A. P., Vasconcelos, M. T. S., Bordalo, A. A. (2004). Vertical distribution of the macrobenthic community and its relationships to trace metals and natural sediment characteristics in the lower

Douro estuary, Portugal. Estuarine, Coastal and Shelf Science, 59(4), 663-673.

- Folk, R. L. (1980). Petrology of sedimentary rocks. Hemphill publishing company.
- Lindsay, W. L., Norvell, W. A. (1978). Development of a DTPA soil test for zinc, iron, manganese, and copper. Soil science society of America Journal, 42(3), 421-428.
- International Organization for Standardization (ISO), (2001). ISO 14870:2001, Soil quality. Extraction of trace elements by buffered DTPA solution.
- Palma, P., Ledo, L., Alvarenga, P. (2015). Assessment of trace element pollution and its environmental risk to freshwater sediments influenced by anthropogenic contributions: the case study of Alqueva reservoir (Guadiana Basin). Catena, 128, 174-184.
- Massas, I., Ehaliotis, C., Kalivas, D., Panagopoulou, G. (2010). Concentrations and availability indicators of soil heavy metals; the case of children's playgrounds in the city of Athens (Greece). Water, Air, & Soil Pollution, 212(1-4), 51-63.
- Imperato, M., Adamo, P., Naimo, D., Arienzo, M.,

Stanzione, D., Violante, P. (2003). Spatial distribution of heavy metals in urban soils of Naples city (Italy). Environmental pollution, 124(2), 247-256.

- Wang, X. S., Qin, Y., Sang, S. X. (2005). Accumulation and sources of heavy metals in urban topsoils: a case study from the city of Xuzhou, China. Environmental Geology, 48(1), 101-107.
- Sany, S. B. T., Salleh, A., Rezayi, M., Saadati, N., Narimany, L., Tehrani, G. M. (2013). Distribution and contamination of heavy metal in the coastal sediments of Port Klang, Selangor, Malaysia. Water, Air, & Soil Pollution, 224(4), 1476.
- El Nemr, A., Said, T. O., Khaled, A., El-Sikaily, A., Abd-Allah, A. M. (2007). The distribution and sources of polycyclic aromatic hydrocarbons in surface sediments along the Egyptian Mediterranean coast. Environmental Monitoring and Assessment, 124(1-3), 343-359.
- El-Namla, S. K., (2019). Geochemical investigations and environmental assessment on the bottom sediments and water quality of Damietta Nile branch, Egypt. Unpublished MSc Thesis, Mansoura University, Egypt. 147 pp.

الملخص العربي

عنوان البحث : تعيين تركيزات العناصر الثقيلة في رسوبيات نهر النيل، دمياط، مصر

إيهاب عسل'، مجدى محمود خليل'، أحمد رخال، محمد العقاد

· كلية العلوم جامعة دمياط قسم الجيولوجيا · جهاز شئون البيئة مصر

هذه الدراسة تدرس توزيع تركيزات عناصر القصدير، الكوبلت، النحاس، الحديد، الأنتيمونيوم، الزنك، الكادميوم، السيلينيوم، النيكل والرصاص وذلك في روسوبيات قاع فرع النيل بدمياط. تم استخدام طريقة حمض دأى إيثيلين تراى أمين بنتا أسيتيك لاستخلاص العناصر الثقيلة. ترتيب متوسطات العناص مرتب کالتالی : قصدیر > حدید > نحاس > زنك > کوبلت > سیلینوم > نیکل > رصاص انتیمونیوم > کادمیوم. التحليل العنقودي قسم العناصر الي مجموعتين بينهما تباين واضح في تركيزات العناصر قبل وبعد سد دمياط. تم عمل تصنيف لعينات الرسوبيات، حساب نسبة المواد العضوية الكلية، معامل ارتباط المصفوفة بين تركيز إت العناصر ومقاربة بين البحث الحالي وأبحاث أخرى.