## APPLICATION OF ANTIBROWNING AND FIRMNESS SUPPORTING COMPOUNDS TO MAINTAIN THE QUALITY OF FRESH-CUT GUAVA

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#### ABSTRACT

The present study aimed to extend fresh-cut guava life and retain quality using safe compounds without affecting quality or palatability. Different concentrations of ascorbic acid, citric acid and calcium lactate were used before storage at 5°C for 12 days in two successive seasons. It has been found that citric acid at 2% was the most effective in the case of fresh-cut guava compared to untreated ones in different traits under study in terms of general appearance, total soluble solids/acid ratio, colour changes, browning index and decay percentage. Citric acid and ascorbic acid treatments showed significant impacts on fresh-cut guava content of total phenols and ascorbic acid. While, calcium lactate at 2% showed the least significant values of weight loss and the highest significant firmness compared with the control.

Key words: Antioxidants, ascorbic, citric, calcium lactate, cold storage, guava slices.

#### **1. INTRODUCTION**

Guava (*Psidium guajava* L.) is a popular fruit and is considered a source of abundant ascorbic acid and high nutritional value. Most guava varieties are usually classified as climacteric fruits (Brown and Wills, 1983), which are characterized by high respiration rate and rapid deterioration. There is growing consumers demand for fresh-cut products, and there are special sections of those products in most hypermarkets. Fresh-cut fruits are very desirable and have great market potential, as they are easy and fast to eat. Fresh-cut products should be in a good appearance and palatability, free of defects and have appropriate shelf life period.

Fresh-cut products are extremely perishable owing to the absence of the outer protective layer of their tissue. The cutting process increases the injured tissues, which are very sensitive to browning and pathogenic diseases, thus promoting rapid deterioration (Alley *et al.*, 1996). Watada and Ling (1999) reported that fresh-cut fruits are very perishable due to cut surfaces and exposed tissues and lack of protective skin.

Amiot *et al.* (1995) reported that browning of fresh-cut pieces is instigated by polyphenol oxidase effect on phenolic compounds released during the practice of cutting, disorders resultant from cutting can be reduced by the use of sharp cutting implements, enzymatic browning inhibitors, modified atmospheres and low temperatures. Gil *et al.* (2006) reported that the response of fresh cut fruits is different compared with full fruits in response to the same treatments.

Although there is a large amount of ascorbic acid in guava fruit compared to other fruits, it is not enough to stop the deterioration and browning especially after cutting and storage prolongation (Abd El-Moneim *et al.*, 2015).

Calcium as a constituent of the cell wall plays an important role in establishing cross-links, which affects strength and regarded as the last barrier before cell separation due to cutting practice (Fry, 2004). Calcium treatments used for tissue firmness have also been reported to reduce browning (Drake and Spayed, 1983). Varela *et al.* (2007) proposed the valuable use of calcium treatments in fresh-cut apples stored for 16 days at 8 - 10°C.

Irene and Diane (2000) reported that calcium lactate did not cause undesirable bitterness to fresh-cut fruits compared with calcium chloride that commonly used as a source of calcium in postharvest treatments. Also, Raheem *et al.* (2013) and Patchimaporn *et al.* (2014) reported that pretreatment of calcium lactate could be an alternative to calcium chloride in fruits as it improved quality.

Ascorbic acid is frequently used as enzymatic browning inhibitor as it chelates copper ions, reduces o-quinones and acts as a polyphenol oxidase inhibitor (Lozano-De-Gonzalez *et al.*, 1993). Citric acid and other acidifying compounds are also used as enzymatic browning inhibitors in vegetables and fruits in combination with ascorbic acid. Santerre *et al.* (1988) found that citric acid inhibited enzymatic browning to a similar degree in apple slices with ascorbic acid.

Ascorbic and citric acids are effective methods to reduce oxidation and deterioration in fruits. Dipping fruit slices in ascorbic and citric acids (0.2-10 g/l) solutions for 5 min. had an impact on polyphenol oxidase activity, colour changes and deterioration in fruits (Pizocaro *et al.*, 1993). Also, Gil *et al.* (2006) found that ascorbic acid treatment at 2% was effective in deferment the browning of apple slices.

The main objectives of the current study were to investigate the influence of ascorbic acid, citric acid and calcium lactate on fresh-cut guava to extend the shelf life and determine the acceptability and quality characteristics of treated fresh-cut guava fruits.

### 2. MATERIALS AND METHODS

Fruits of guava cv. Baladi were harvested manually at the commercial maturity stage (Kader, 1999 and Sing and Pal, 2008) from a private orchard in El- Beheira Governorate, Egypt in 2014 and 2015 seasons.

Fruits were almost similar in size and skin colour, and free of obvious pathological defects and mechanical injuries. Fruits were washed by tap water, sliced (1cm thickness) and the central seeds portion of the fruits were removed according to the procedure of Durigan et al. (2005), then dipped in one of the following solutions; 1 or 2% ascorbic acid, 1 or 2% citric acid, 1 or 2% calcium lactate, and distilled water (control) for 5 min., then the treated slices were left to dry at room temperature. Slices were packed in plastic packages (average 250 g/package) and stored at 5°C, 90-95% relative humidity for 12 days. Nine pieces from each replicate of each treatment were sampled for analysis after 3 days intervals.

The influence of cutting and cold storage on the quality indices and chemical characteristics of fresh-cut guava were evaluated.

Any signs of spoilage during storage were scored and discarded, decay percentage was calculated as the number of discarded slices / total number of slices  $\times$  100.

General appearance of slices was observed visually using the described procedure by Mitcham *et al.* (2003); on a scale from one to nine with 1= unacceptable, 3= poor, 5= fair, 7= good, and 9= excellent.

Instrumental slice colour was measured in the CIE  $L^* a^* b^*$  on two opposed sides of each slice objectively. Each measurement was taken at three locations for each slice, using a Minolta CR-400 chroma meter (Minolta, Osaka, Japan) according to McGuire (1992).

Browning index was measured using the formula mentioned by Ruangchakpet and Sajjaanantakul (2007) as following; browning index (BI) = (100 (x - 0.31)) / 0.17, where x =  $(a^* + 1.75L^*) / (5.645L^* + a^* - 0.3012b^*)$ .

Weight loss percentage was calculated using the following equation; (package initial weight – package weight at each sampling date) / package initial weight  $\times$  100.

Guava firmness was determined according to Mitcham *et al.* (2003) using a fruit pressure tester, data were presented as  $lb/inch^2$ .

TSS / Acid ratio was calculated via estimating total soluble solids and titratable acidity. Total soluble solid percentage was assessed by using a digital refractometer (model Palette, ATAGO), using drops of the fruit juice. Total acidity was measured using the titration method by 10 ml of the extracted juice against 0.1 N of NaOH, using Phenolphthalein indicator (A.O.A.C. 1980), and mentioned as percentage of citric acid.

Total phenols as mg gallic acid per 100 g fresh weight were determined colourmetrically using Folin Denis reaction method at 765 nm according to A.O.A.C. (2012).

Ascorbic acid was determined using titration method by 2,6 dicholorophenol-indophenol solution. Results were expressed as mg ascorbic acid per 100 g FW (Mazumdar and Majumder, 2003).

A completely randomized block design was followed, the treatment means were estimated using the method of LSD at the 5% level of significance (Snedecor and Cochran, 1989).

# **3. RESULTS AND DISCUSSION**

### **3.1. Decay percentage**

Data in Fig. (1) present the effect of ascorbic acid, citric acid and calcium lactate treatments on decay percentage in both seasons. Decay percentage increased gradually during storage especially after 9 days of storage at 5 $^{\circ}$ C. Citric acid at 2% recorded the lowest significant values compared with the control in 2014 and 2015 seasons.

Both citric acid treatments showed the lowest percentages of decay after 12 days of storage at  $5^{\circ}$ C, whereas 2% citric acid exhibited the least percentage of decay in the second season compared with the control that showed the highest rates as 66.67 and 61.11% in the first and the second season respectively.

The shelf or marketing life of a produce can be defined as how long the produce will be agreeable for the consumers (Labuza and Schmidl, 1988). Damage of external tissues during preparative stage of cutting process, leaks cytoplasm and provides a potentially richer source of nutrients for the microorganisms. In addition, high water contents induce microbial growth in fresh cut fruits and vegetables (Shah and Nath, 2006).

Citric acid is an anti-browning agent that prevents polyphenol oxidase activity by decreasing pH and binding the copper in an active site of this enzyme to form a defused complex (Martinez and Whitaker, 1995) which reduces deterioration and make them less susceptible to pathogens. In this way, GonzalezThis decrement appeared clearly in the ninth day of storage at  $5^{\circ}$ C. Different treatments maintained slices appearance compared with untreated ones. Citric acid at 2% was the most acceptable compared with the control that was fair after 6 days of cold storage in both seasons.

Fresh-cut fruits loss its attractive appearance rapidly because of colour changes and soft tissues (Alley *et al.*, 1996). The obtained findings declare that citric acid was effective in maintaining good appearance and freshness, which may be explained as a limiting factor of discolouration (Sams, 1999). Also, citric acid treatments did not exhibit any pitting on guava slices surface.

#### **3.3.** Colour measurement

L<sup>\*</sup> value and h<sup>°</sup> (hue angle) are the most precise values about colour changes in fresh-cut. The L<sup>\*</sup> value was used as an indicator of brightness, a<sup>\*</sup> point to chromaticity on a green (-) to red (+) axis, b<sup>\*</sup> indicate to yellow/blue rival colours with blue at negative b<sup>\*</sup> values and yellow at positive b<sup>\*</sup> values. Hue (h<sup>°</sup>) is the common correspondence between colours positioned around a colour wheel, chroma is the value of a colour's pureness, intensity or saturation (Nambi *et al.*, 2015).

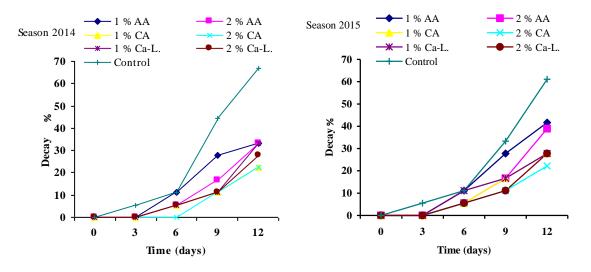


Fig. (1): Impact of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on guava decay (%) in 2014, 2015 seasons.

Agular *et al.* (2000) reported similar findings concerning anti-browning agents.

## **3.2. General appearance**

Fig. (2) presents the effect of citric acid, ascorbic acid and calcium lactate treatments on fresh cut guava general appearance. Guava slices freshness decreased sharply during cold storage.

The results presented in Fig. (3) indicate that 1 and 2% citric acid treatments exhibited the highest significant values in the first season, whereas 2% citric acid showed the highest significant value in the second season compared with untreated ones which showed the lowest values in both seasons.

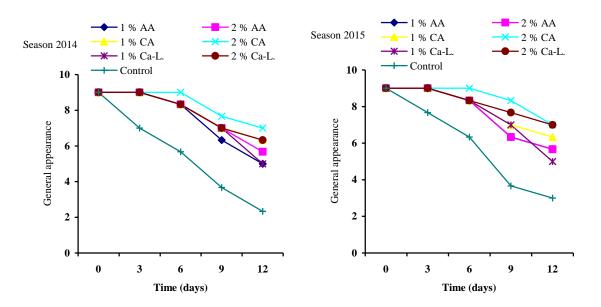


Fig. (2): Effect of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on guava general appearance (general appearance was evaluated visually on a scale, 1= unacceptable, 3 = poor, 5 = fair, 7 = good, and 9 = excellent ) in 2014, 2015 seasons.

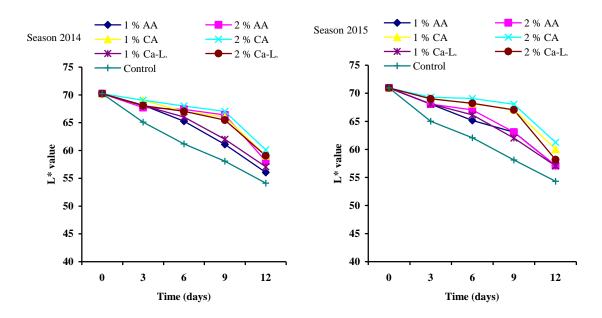


Fig. (3): Influence of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on L\* value in 2014, 2015 seasons.

 $L^*$  value decreased gradually during cold storage, where high concentration of citric acid showed the highest values after 12 days of storage at 5°C as it recorded 60.11 and 61.25 in 2014 and 2015 seasons, respectively, compared with untreated specimens that showed the least

brightness values.

Findings illustrated in Table (1) show that citric acid at 2% recorded the highest significant  $h^{\circ}$  values compared with the control in both seasons.

A gradual reduction in  $h^{\circ}$  values was observed during storage. Slices treated by

	Days of storage at 5°C (D)							<b>Days of storage at <math>5^{\circ}C(D)</math></b>						
Treatment (T)	0	3	6	9	12	Mean	0	3	6	9	12	Mean		
	First season							Second season						
1 % AA	97.32	95.03	94.60	93.11	92.53	94.52	97.93	97.13	95.31	92.84	91.80	95.00		
2 % AA	97.32	95.37	95.08	94.21	94.03	95.20	97.93	97.33	95.24	92.75	91.97	95.04		
1 % CA	97.32	96.07	95.49	95.10	94.52	95.70	97.93	97.83	96.53	93.53	92.63	95.69		
2 % CA	97.32	96.51	95.31	95.14	95.05	95.87	97.93	97.83	96.83	94.37	92.90	95.97		
1 % Ca-L.	97.32	95.50	95.24	94.24	93.38	95.14	97.93	97.03	95.30	92.70	91.53	94.90		
2 % Ca-L.	97.32	95.50	95.12	94.53	94.37	95.37	97.93	97.10	96.39	93.53	92.23	95.43		
Control	97.32	95.39	93.40	92.30	91.14	93.91	97.93	96.04	94.23	92.30	91.31	94.36		
Mean	97.32	95.62	94.89	94.09	93.58		97.93	97.19	95.69	93.14	92.05			
L.S.D <sup>0.05</sup>	(T) = 0.6	57, (D) = 0	.56, (T×D	) = 1.94			$(T) = 0.35, (D) = 0.30, (T \times D) = 0.79$							

Table (1): Effect of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on h° values of guava in 2014, 2015 seasons.

different concentrations of citric acid kept the highest values until the end of the storage period at  $5^{\circ}$ C compared with the control, which showed the lowest standards in both seasons.

Decrement in lightness and h<sup>°</sup> values were possibly caused by an increase in the respiration rate and the promotion of enzymatic processes that were responsible for a decrease in quality that involved browning and other reactions (McGuire, 1992).

Although citric acid was more effective on colour preservation under this study conditions, Pizocaro *et al.* (1993) found that dipping fresh cut apple in ascorbic acid for 5 min. decreased browning (maintaining lightness) effectively compared with citric acid by decreasing oxidation.

Moreover, citric acid has been reported widely for its inhibitory activity on polyphenol oxidase, and its antibrowning role in fresh-cut fruits and vegetables (Rosen and Kader, 1989).

## **3.4. Browning index**

Browning index is defined as brown colour purity, and it is considered one of the most common indicators of discolouration in sugar containing fruit products (Buera *et al.*, 1986).

The results presented in Fig. (4) reveal that browning index increased gradually under all conditions. Untreated ones showed the highest significant indices in 2014 and 2015 seasons, while 2% citric acid showed the lowest significant browning index in both seasons.

At the end of storage period in the first season, the control showed the highest

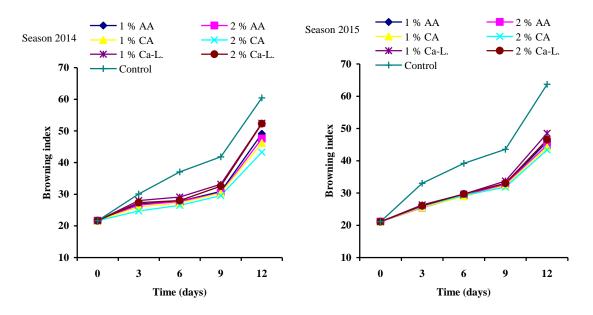


Fig. (4): Influence of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on BI values in 2014, 2015 seasons.

significant value of 60.46, while 2% citric acid showed the lowest significant browning index of 43.29. Also, at the end of the storage period in the second season, the control showed the highest significant value of 63.72, while citric at 2% acid showed the lowest significant value of 43.33. The antioxidant desired effect may be due to inhibition or reducing the enzymatic transformation of phenols to quinones, which slow down browning effectively (Lozano-De-Gonzalez *et al.*, 1993).

### **3.5.** Weight loss percentage

Product weights are very important values that express the final biological changes result in fresh commodity and they have a big impact on commercial standards. Decrement of weight in fresh-cut fruits is lower compared with whole fruit during cold storage (Abd El-Moneim *et al.*, 2015) because of airtight packages in the case of fresh-cut.

Table (2) indicates that, under all circumstances weight loss percentage increased gradually during storage periods. Calcium lactate at 2% showed the lowest significant weight loss percentages after 12 days of storage at 5 °C compared with the control in both seasons.

In agreement with our findings Lamikrana and Watson (2007) found that application of calcium lactate depressed respiration and improved cell wall turgidity as it establish strong links between cell surfaces that may result in reduction of water loss rates. control after 12 days of storage at  $5^{\circ}C$  compared with the control, which showed the lowest values in both seasons.

Also, Coci *et al.* (2006) showed a structural breakdown with subsequent softening in apple tissue. Firmness reduction of slices occurs due to dissolution of middle lamella and cell separation as a result of the activity of pectin esterase, even at low temperature (Roccui *et al.*, 2005).

Studied treatments with ascorbic and citric acid did not result in tissue softening, which were not agreeable to those reported by Gil *et al.* (1998).

Calcium lactate has been widely reported for its potential role in maintaining postharvest quality and decreasing softening by reducing the ripening processes along with reducing the deterioration (Pereira *et al.*, 2009), Postharvest application of calcium lactate lowered respiration and increased firmness (Lamikrana and Watson, 2007), treatment of calcium may led to strong cell wall structure through the bond between calcium and pectin (Gorny *et al.*, 2002).

Rodrigues *et al.* (2010) pointed out that the most obvious change in slices was a fast loss in firmness a few hours after processing, even at low temperatures, calcium lactate was effective in retaining firmness in most fresh-cut fruits.

The promising effect of calcium lactate could be due to maintaining the cell wall structure in fruit by interacting with pectins in the cell wall to form calcium pectate that supports molecular

Table (2): Effect of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on guava weight loss (%) in 2014, 2015 seasons.

		Days of storage at 5°C (D)							Days of storage at 5°C (D)						
Treatment (T)	0	3	6	9	12	Mean	0	3	6	9	12	Mean			
			First	season			Second season								
1 % AA	0.000	0.151	0.311	0.442	0.773	0.335	0.000	0.085	0.205	0.336	0.694	0.264			
2 % AA	0.000	0.187	0.329	0.478	0.800	0.359	0.000	0.176	0.346	0.528	0.727	0.355			
1 % CA	0.000	0.106	0.282	0.331	0.493	0.242	0.000	0.112	0.192	0.358	0.412	0.215			
2 % CA	0.000	0.177	0.216	0.346	0.463	0.241	0.000	0.225	0.291	0.425	0.491	0.286			
1 % Ca-L.	0.000	0.161	0.239	0.267	0.450	0.223	0.000	0.159	0.192	0.263	0.374	0.198			
2 % Ca-L.	0.000	0.060	0.157	0.235	0.440	0.178	0.000	0.083	0.157	0.196	0.372	0.162			
Control	0.000	0.443	0.505	0.516	0.965	0.486	0.000	0.172	0.313	0.528	0.939	0.391			
Mean	0.000	0.184	0.291	0.373	0.626		0.000	0.145	0.242	0.376	0.573				
L.S.D 0.05	(T) = 0.0	007, (D) =	= 0.006, (1	$\Gamma \times D$ ) = 0.0	16		$(T) = 0.007, (D) = 0.006, (T \times D) = 0.016$								

# **3.6.** Firmness (Ib/inch<sup>2</sup>)

The results presented in Table (3) indicate that, there was a negative proportion between firmness and periods of storage in all treatments in both seasons.

Calcium lactate treatments showed the highest significant values compared with the

bonding between constituents of cell wall (Degrave *et al.*, 2003). Calcium also increases cell wall turgor pressure and stabilizes the membrane turgidity (Luna-Guzman and Barret, 2000). Also, the free carboxylic acid liberated by pectin methyl esterase can interact with calcium to form a molecular complex which results in an

	Days of storage at $5^{\circ}C(D)$							Days of storage at $5^{\circ}C(D)$						
Treatment (T)	0	3	6	9	12	Mean	0	3	6	9	12	Mean		
	First season							Second season						
1 % AA	1.82	1.64	1.40	1.22	0.82	1.38	1.93	1.66	1.42	1.23	0.85	1.42		
2 % AA	1.82	1.54	1.22	1.00	0.73	1.26	1.93	1.59	1.29	1.02	0.76	1.32		
1 % CA	1.82	1.61	1.55	1.32	0.82	1.42	1.93	1.66	1.56	1.33	0.83	1.46		
2 % CA	1.82	1.60	1.32	1.10	0.81	1.33	1.93	1.72	1.33	1.13	0.83	1.39		
1 % Ca-L.	1.82	1.71	1.50	1.42	1.02	1.49	1.93	1.76	1.51	1.45	1.05	1.54		
2 % Ca-L.	1.82	1.71	1.60	1.41	1.11	1.53	1.93	1.79	1.61	1.41	1.13	1.58		
Control	1.82	1.20	0.90	0.80	0.55	1.05	1.93	1.25	0.92	0.86	0.65	1.12		
Mean	1.82	1.57	1.36	1.18	0.84		1.93	1.63	1.38	1.20	0.87			
L.S.D <sup>0.05</sup>	(T) = 0	).06, (D) =	0.05, (T×	D) = 0.14			$(T) = 0.07, (D) = 0.06, (T \times D) = 0.15$							

 Table (3): Impact of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on guava firmness (lb/ inch<sup>2</sup>) in 2014, 2015 seasons.

enhancement of firmness (Degrave *et al.*, 2003). Moreover, Robles-S'nchez *et al.* (2009) proved that the dipping fresh cut fruits in calcium lactate maintained the pectin methyl esterase and polygalacturonase activities during storage.

### 3.7. TSS/Acid ratio

Table (4) presents the influence of different concentrations of ascorbic acid, citric acid and calcium lactate on TSS/Acid ratio. Citric acid at 2% and 1% showed the highest significant ratios compared with the control in both seasons. Ratios of TSS/Acid increased regularly during cold storage but it decreased at the end of storage period. Both concentrations of citric acid maintained the highest ratios after 12 days of storage at 5°C compared with the control in both seasons.

In this regard, Gonzalez-Agular *et al.* (2000) reported considerable changes in fructose and glucose in fresh-cut mango treated with anti-browning means and stored at 10°C.

The increase in total soluble solids/acid ratio could be attributed mainly to the decrease of

acidity due to consumption of organic acids in fruits during respiration and biological changes of starch, soluble sugars, sucrose and glucose during ripening, but with storage prolongation total soluble solids decrease (Ahvenainen, 1996). **3.8. Total phenols (mg gallic acid/100g FW)** 

Table (5) presents the effect of different concentrations of calcium lactate, ascorbic acid and citric acid on total phenols content. Total phenols decreased continuously during cold storage prolongation. Citric acid at 2% showed the highest significant value compared with the control in the first season, whereas citric acid and ascorbic acid treatments showed the highest significant values compared with untreated ones in the second season.

Both citric acid concentrations and 1% ascorbic acid treatments showed the highest values after 12 days of storage at  $5^{\circ}C$  compared with the control that showed the lowest content in both seasons.

Decrement of the total phenolic content with the prolongation of storage period is probably

Days of storage at 5°C (D) Days of storage at 5°C (D) Treatment 0 3 6 9 12 Mean 0 3 6 9 12 Mean **(T)** First season Second season 1 % AA 8.83 9.89 10.14 11.51 12.13 10.50 8.53 9.59 10.04 11.37 11.82 10.27 2 % AA 8.83 9.93 10.24 11.65 12.24 10.58 8.53 9.74 10.09 11.36 11.82 10.31 1 % CA 8.83 10.04 10.25 12.26 13.04 10.88 8.53 9.85 10.18 11.59 12.44 10.52 2 % CA 8.83 10.15 10.70 12.63 13.39 11.14 8.53 9.89 10.25 11.92 12.71 10.66 1 % Ca-L. 8.83 9.93 10.14 11.52 12.16 10.51 8.53 9.81 10.03 11.32 11.66 10.27 2 % Ca-L. 8.83 9.97 10.44 12.00 12.79 10.81 8.53 9.52 9.78 11.39 11.84 10.21 8.83 9.85 9.95 11.40 11.97 10.40 8.53 9.22 9.63 10.48 10.91 9.75 Control 10.27 12.53 8.53 9.66 10.00 11.35 11.89 Mean 8.83 9.96 11.85 L.S.D 0.05  $(T) = 0.10, (D) = 0.09, (T \times D) = 0.23$  $(T) = 0.14, (D) = 0.12, (T \times D) = 0.32$ 

 Table (4): Effect of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on TSS/Acid ratio of guava in 2014, 2015 seasons.

		Days of storage at $5^{\circ}C$ (D)							Days of storage at 5°C (D)						
Treatment (T)	0	3	6	9	12	Mean	0	3	6	9	12	Mean			
	First season							Second season							
1 % AA	125.42	117.50	113.50	109.07	99.47	112.99	132.07	128.27	118.77	114.07	102.50	119.13			
2 % AA	125.42	119.13	114.07	109.13	96.63	112.88	132.07	130.20	119.80	114.74	102.40	119.84			
1 % CA	125.42	115.27	112.17	108.07	99.60	112.10	132.07	128.50	120.10	113.10	103.11	119.38			
2 % CA	125.42	120.07	115.07	110.07	100.10	114.14	132.07	129.13	119.07	116.03	102.50	119.76			
1 % Ca-L.	125.42	114.14	108.10	96.04	88.07	106.35	132.07	127.50	113.07	105.13	99.50	115.45			
2 % Ca-L.	125.42	116.17	108.43	96.13	88.07	106.84	132.07	128.03	110.07	103.03	99.80	114.60			
Control	125.42	110.13	94.04	89.40	80.17	99.83	132.07	115.03	100.03	90.07	82.07	103.85			
Mean	125.42	116.06	109.34	102.56	93.16		132.07	126.67	114.41	108.03	98.84				
L.S.D <sup>0.05</sup>	(T) = 0.9	9, (D) = 0.5	84, (T×D)	= 2.23			$(T) = 0.72, (D) = 0.61, (T \times D) = 1.61$								

Table (5): Effect of ascorbic acid (AA), citric acid (CA) and calcium lactate (Ca-L) treatments on total phenols (mg gallic acid /100g FW) of guava in 2014, 2015 seasons.

due to the oxidation by polyphenol oxidase (Yamaguchi et al., 2003). Also, quercetin is oxidized directly by polyphenol oxidase. Moreover, Robards et al. (1999) found that phenolic compounds have significant role in oxidation progressions as antioxidant substrates in browning reactions, leading to extend storage life. Their effects on the inhibition of oxidation of low density lipo proteins have been widely considered (Eberhardt et al., 2000). Rodrigues et al. (2010) illustrated that citric acid was more efficient than ascorbic acid and calcium lactate in maintaining or increasing the total phenol contents in the fresh-cut fruits. These findings are also in agreement with those obtained by Raheem et al. (2013), indicating to reduction of browning in treated fresh-cut slices.

### **3.9. Ascorbic acid (mg/100g FW)**

Table (6) reveals that ascorbic acid decreased dramatically during storage. Citric acid at 2% showed the highest significant value after 12 days in the first season, whereas citric acid at 2% and ascorbic acid at 2% treatments showed the highest significant values in the second season compared with the control that showed the lowest values.

The reason for ascorbic acid losses during storage is oxidation that occurs spontaneously when the ascorbic acid is combined with oxygen in the air (Owusu-Yaw *et al.*, 1988). Ascorbic acid is very unstable, this disintegrate is mainly due to the action of ascorbic acid oxidase and the reaction with oxygen, heavy metal ions and light (Bode *et al.*, 1990); therefore, ascorbic acid is taken as an indicator of fruit freshness and retention of other constituents.

The obtained results are in line with those obtained by Pizocaro *et al.* (1993), who found that dipping the fresh cut fruits in ascorbic and citric acids solutions for 5 min. had a varied impact on ascorbic acid content, polyphenol oxidase activity, discolouration and deterioration in fruits. In addition, Abd El-Moneim *et al.* (2015) declared that fresh-cut guava treated by citric acid at 2% exhibited significant less degradation of ascorbic acid content.

Table (6): Effect of ascorbic acid, citric acid and calcium lactate treatments on guava content of ascorbic acid (mg/100g FW) in 2014, 2015 seasons.

	Days of storage at 5°C (D)							Days of storage at 5°C (D)						
Treatment (T)	0	3	6	9	12	Mean	0	3	6	9	12	Mean		
			First	season		Second season								
1 % AA	94.68	94.13	86.84	84.67	79.66	88.00	95.39	95.41	89.11	82.87	79.87	88.53		
2 % AA	94.68	94.55	87.20	83.17	79.49	87.82	95.39	95.42	91.75	84.64	79.30	89.30		
1 % CA	94.68	94.11	85.02	81.27	79.35	86.89	95.39	95.56	88.43	84.52	79.56	88.69		
2 % CA	94.68	94.20	92.38	85.05	79.97	89.25	95.39	95.53	89.69	84.60	80.13	89.07		
1 % Ca-L.	94.68	93.93	86.58	79.46	77.46	86.42	95.39	94.88	87.80	82.21	79.21	87.90		
2 % Ca-L.	94.68	93.99	84.95	80.82	78.93	86.67	95.39	94.91	86.33	81.00	78.93	87.31		
Control	94.68	85.54	79.16	69.71	64.92	78.80	95.39	84.68	78.91	72.27	64.93	79.24		
Mean	94.68	92.92	86.02	80.59	77.11		95.39	93.77	87.43	81.73	77.42			
L.S.D <sup>0.05</sup>	(T) = 0.	.92, (D) =	0.78, (T×E	() = 2.06			$(T) = 1.01, (D) = 0.86, (T \times D) = 2.26$							

## Conclusion

The present work declared that citric acid, ascorbic acid and calcium lactate maintained the quality of guava pieces effectively for 9 days of storage at 5°C. Citric acid at 2% treatment maintained the general appearance, total soluble solids/acid ratio, ascorbic acid and total phenols of fresh-cut guava, it also recorded the lowest colour changes and decay percentage. Ascorbic acid treatments showed significant impacts on fresh-cut guava content of the total phenols and ascorbic acid. Calcium lactate at 2% showed the lowest significant values of weight loss and the highest significant firmness compared with untreated ones. The obtained results clarified that it is possible to prolong the life of fresh cut guava by using anti-browning and supporting firmness treatments. whereas citric acid maintained the characteristics of quality effectively for 9 days and calcium lactate contributed to maintain the rigidity and weight without causing bitterness or bad appearance.

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# استخدام بعض المركبات المضادة للتلون البنى والداعمة للصلابة للحفاظ على جودة قطع الجوافة الطازجة

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## ملخص

أجريت هذه الدراسة بهدف إطالة العمر التسويقى للجوافة الطازجة والحفاظ على جودتها وذلك باستخدام مركبات امنة دون التأثير على الجودة او الطعم. تم استخدام تركيزات مختلفة من حامض الاسكوربيك، حامض الستريك ولاكتات الكالسيوم قبل التخزين على درجة حرارة 5°م لمدة 12 يوم خلال موسمين متتالين. وجد أن المعاملة بحامض الستريك بتركيز 2% كانت هى الأكثر فعالية فى الحفاظ على الجودة المظهرية، نسبة السكريات الذائبة الكلية الى الحموضة، التغيرات اللونية، دليل التلون البنى ونسبة الفاقد من ثمار الجوافة الطازجة. إن استخدام حامض الستريك وحامض الاسكوربيك كان له تأثير معنوى على محتوى ثمار الجوافة الطازجة. إن استخدام حامض الستريك وحامض المعاملة باستخدام لاكتات الكالسيوم الى في محتوى ثمار الجوافة الطازجة. إن استخدام المن الكلية بينما أدت المعاملة باستخدام لاكتات الكالسيوم الى خفض نسبة الفاقد فى الوزن، والاحتفاظ بصلابة أعلى وذلك مقارنة بالمعاملة القياسية.

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