The seaweed (green macroalgae), *Ulva* sp. as bioindicator of metal pollution in the Mediterranean Coast, Alexandria region, Egypt

Samir M. Saeed and Yasser T. A. Moustafa* Central Lab. for Aquaculture Research, Agricultural Research Center. Limnology dept. E-mail: ymoustafaonline@yahoo.com

ABSTRACT

Changes in heavy metals (Fe, Mn, Zn, Cu, Cd and Pb) concentrations in marine seaweed (green alga) *Ulva* sp. and seawater collected from the the Mediterranean Coast, Alexandria region, Egypt were investigated during monthly samplings from January to December 2012 to assess the spatial and seasonal variation of these elements as well as the pollutional status at this area. Fe had the highest concentration and Cd found the lowest one. There were statistically positive significant correlations among the levels of some metals in the macroalgae. The mean metals concentration in the *Ulva* tissue decreased in the order: Fe>Mn>Zn>Cu>Pb>Cd. Only, concentrations, Fe, Zn and Pb, in water showed significant (P<0.05) seasonal changes. Fe, Mn, Zn and Cu showed the greater spatial and temporal differences (P<0.05) in macroalgae tissues. The results showed that the studied area not faced heavy metals pollution. It is also concluded that *Ulva* sp. play an important role as a bioindicator for heavy metals contamination in seawater.

Keywords: Seaweed, Ulva sp., heavy metals, Mediterranean Coast

INTRODUCTION

Macroalgae are major primary producers in the marine environment and play an important role in food chains. They also used as a natural source of food and medicines especially in Asian countries. The seaweeds (macroalgae) are widely distributed in the ocean and seas, ranging from tide level to considerable depths or attached to substrates such as sand, mud, rocks, shells, and coral (Apaydn *et al.*, 2010). Green macroalgae *Ulva sp.* is a widespread macroalgae occurring at all levels of the intertidal zone and grows along rocky or sandy coasts of oceans and seas (Apaydn *et al.*, 2010). It was the most abundant seaweed at the Egyptian marine environment throughout the year (Abdallah, 2010).

The bioaccumulation and biomagnifications of heavy metals along trophic chains increase their toxicity in the aquatic environment over time (Kamala-Kannan *et al.* 2007). Marine macroalgae accumulate such elements, which are further transferred along the trophic chain by herbivores and detritivores. So, they are important bioindicator of the marine environmental pollution by these elements (Villares *et al.*, 2001).

Ulva spp. have high accumulation capacity for metals, so they are used as bioindicators of metal pollution in different parts of the world (Villares *et al.*, 2001; Lozano *et al.*, 2003; Topcuoglu *et al.*, 2010; Kamala-Kannan *et al.*, 2007; Chaudhuri *et al.*, 2007; Apaydn *et al.*, 2010; Okuku and Peter, 2012; Rybak *et al.*, 2013) and in the Mediterranean coast of Egypt (El-Moselhy *et al.*, 2004; Abdallah, 2010; Abdallah and Abdallah, 2008).

Many factors may influence the bioavailability of metals in algae including pH, salinity, temperature, light, nutrient concentrations, oxygen, particulate matters and organic matters (Jothinayagi and Anbazhagan, 2009) and on the structural differences among the algae (Favero *et al.*, 1996). It is also worth noting that physiological changes and growth can affect concentrations of metals in the macroalgal tissue (Villares *et al.*, 2002).

The physical characteristics and locations of many coastal areas in Alexandria, Egypt make them vulnerable to contaminants. So, this study was conducted to assess the health of this aquatic ecosystem and contribute to the overall picture of metal inputs into this marine environment. Also, the seasonal variability of heavy metals concentration in water and macroalgal tissues of *Ulva* sp. was studied.

MATERIALS AND METHODS

Sampling

Water and seaweeds (green macroalgae, *Ulva* sp.) samples were collected monthly from January to December 2012 from two sites namely; Ras Al-Tin (A) and Al-Mountazah (B) along the north coast of Egypt, Alexandria region (Fig. 1).



Fig. 1: Map illustrate the study sites (A and B)

Laboratory analysis

The sampled seaweeds and seawater were subsequently analyzed for iron (Fe), manganese (Mn), zinc (Zn), copper (Cu), cadmium (Cd) and lead (Pb) as follow: **a) Seaweed**

All samples were dried, ground and homogenized in the laboratory prior to analysis. Samples were attentively washed in tap water to remove salt water, sand, and particulate matter, next rinsed in distilled water to remove any mineral particles and organisms adhering to algal surface and then dried in an oven at 60-70 °C. The dried samples pulverized in a mill and sieved to provide particle size homogeneity using a 100 mesh sieve. From each separate sample, 0.5 g was weighed into a 100 porcelain crucible; ignite at 500 °C in a muffle furnace overnight. After cooling, the ashed samples dissolved in 5 ml 20% HCl, filtered and completed to 50 ml with distilled water, as described by AOAC (1990). Atomic Absorption Spectrophotometer (Model Thermo Electron Corporation) instrument was used to detect metals concentrations which were expressed as $\mu g/g dry$ wt in algae tissues and $\mu g/l$ in water samples.

b) Water

Metals concentration in filtered water samples were measured after digestion with conc. HNO₃ and HCl according to EPA (1992).

Bioconcentration of metals in tissues of seaweed can be described by a bioconcentration factor (BCF), which is the ratio of the chemical concentration in an organism (C_B) (mg of chemical per kg of organism to the concentration in water (C_w) (mg of chemical per liter of water):

BCF = C_B / C_w (Gobas and Morrison, 2000).

Statistical analysis

Two-way ANOVA was employed to evaluate the variability of the concentration of each metal with respect to different seasons and sites. The interelemental relationships were performed through Pearson's correlation coefficient matrix. Significant differences are stated at P<0.05 (Bailey, 1981).

RESULTS AND DISCUSSION

Metals in seaweed

The seasonal variations in the concentrations of Fe, Mn, Zn, Cu, Cd and Pb in seaweed (*Ulva* sp.) collected from Ras Al-Tin and Al-Muntazah region on the shore of the Mediterranean coast are shown in Table (1) and Figure (2). Only, Mn, Zn and Cu showed significant (P<0.05) variations in concentration with respect to seasons. Among the metal concentrations, Fe showed the highest average (687.60 μ g g⁻¹ dw) and Cd was found to be the lowest level (0.31 μ g g⁻¹ dw). Accumulation levels of heavy metals have been detected in the order Fe>Mn>Zn>Cu>Pb>Cd. This order was also obtained in *Ulva* sp. by Alp *et al.* (2012). As for essential metals, the concentration order may correspond to their biological role in algal tissues (Munda and Hudnik, 1986; Dadolahi *et al.*, 2011).

In *Ulva* sp., the highest Fe concentrations were obtained in spring and autumn. The highest concentrations of Fe observed in autumn was similarly to that observed in Greece by Haritonidis and Malea (1999) and in Spain by Villares *et al.* (2002), when biomass is minimal. However, Misheer *et al.* (2006), in South Africa found the highest values in winter and the lowest during summer and autumn. Pérez *et al.* (2012) recorded high levels of Fe in *Ulva* sp. and declared that this algae is an excellent source of these metals. The possibility that seaweeds may have been absorbed Fe present in fine sediment particles (Giusti, 2001). Some species are able to absorb elements directly from sediment (through thallus and rhizoids, in which metal concentrations are much higher than in the water column) (Zbikowski *et al.*, 2006). In this study, the algae samples collected were attached to sand and rocks distributed along the coast.

Manganese concentrations exhibit higher values in summer, while the lowest one detected in winter and this may be related to anthropogenic discharges, including domestic, industrial and agricultural discharges (Rybak *et al.*, 2013).

The variation of the Zn concentration follows the pattern in which maximum concentrations occur in autumn and winter, followed by a decrease in spring and summer. A significant seasonal and spatial variation is observed all year long (p<0.05) (Table 3 and Fig. 2). Pérez *et al.* (2012) observed the highest concentrations of Zn in autumn and winter. Similar results were reported by Villares *et al.* in 2002

(18 μ g g⁻¹ dw), by Haritonidis and Malea, 1999 (16.9 μ g g⁻¹ dw) and Pérez *et al.*, 2012 (1.3-31.3 μ g g⁻¹ dw). Higher values than those were reported by Caliceti *et al.* (2002) in Italy with 64 μ g g⁻¹ dw; and lower values were reported by Chaudhuri *et al.* (2007) in USA, which fluctuated between 6 and 12 μ g g⁻¹ dw. Finally, it is worth noting that Zn concentrations for *Ulva* sp. were reported to vary seasonally (Brown *et al.*, 1999), with higher figures in winter than in spring or summer. It has been reported that the average zinc residues in plants collected from polluted waters are within the range of 100 to 500 μ g g⁻¹ dw (Moore and Ramamoorthy, 1984). Applying these guidelines to the present study, it could be inferred that the study area are uncontaminated sites with respect to Zn.

Cu concentrations showed significant differences among seasons and the sampling sites (p<0.05), for *Ulva* spp. Its levels decrease in spring and summer and increase in winter and autumn. Favero *et al.*, 1996) mentioned that essential metal (Cu) levels gradually decrease from spring to summer. This metal would be actively uptaken by algal tissues in the spring proliferation. Topcluogu *et al.* (2010) mentioned that, industrials effluents from the extensive oil production, high traffic shipping, loading and transport facilities off coasts could act as source of Cu pollution. According to Giusti (2001) and Caliceti *et al.* (2002) copper contamination is associated with algal levels of >20.0 μ g g⁻¹ dw. Similarly, (Sawidis *et al.*, 2001) have considered that a range of 20 to 70 μ g g⁻¹ dw in green macroalgae as a characteristic of contaminated sites. So, this area is uncontaminated with copper.

The competition between metals for algal sites could decrease the accumulation of Cd and Pb which exhibited lower concentration levels in water or which are present in less available forms (e.g. Cd) (Dadolahi et al., 2011). Foster (1976) reported that elevated Zn concentrations inhibited Cd uptake by seaweeds due to competition for binding sites. It has been reported that algal runs samples containing lower than 2 μ g g⁻¹ dw of cadmium can be considered not polluted (Lozano et al., 2003). In algae collected from the two sites, cadmium concentrations varied between 0.27 and 0.35 $\mu g g^{-1}$ dw. So, there is no cadmium pollution problem in the study area. The poor accumulation of cadmium seems to be a feature typical for all species of Ulva (Rybak et al., 2012). High levels of Pb in alga of study area could be attributed to combustion of fossil fuels and oil pollution (Dadolahi et al., 2011). Also, potential contamination from sediment particles appears theoretically possible for Pb (Giusti, 2001). A concentration of $<10 \ \mu g \ g^{-1}$ dw. has been considered as a border line between contaminated and uncontaminated species (Lozano et al., 2003). We can conclude that the lead contents of the algal specimen studied were within the expected limits of uncontaminated areas.

Increased Cu, Cd and Pb in winter and autumn may be as a result of inputs of materials via atmosphere in marine environment of the Alexandria shore which are probably significant due to different oil industry activities and high shipping traffic. There is evidence of seasonal variations of trace metal concentration in *Ulva* sp. (Favero *et al.*, 1996; Caliceti *et al.*, 2002; Villares *et al.*, 2002; Misheer, 2006; Pérez *et al.*, 2012), especially for Mn, Zn and Cu.

Fe, Mn, Zn and Cu showed significant (P < 0.05) variations between the two sites. In general, the highest concentrations of Fe, Mn and Cd being recorded at Al-Muntazah region (B), whereas Cu, Zn and Pb highly accumulated in algae from Ras Al-Tin area (A) (Table 1 and Fig. 2).

Season	Metals											
	Fe		Mn		Zn		Cu		Cd		Pb	
	А	В	А	В	А	В	А	В	А	В	А	В
Winter	516.2	550.7	9.46	9.60	15.12	12.55	17.20	7.39	0.62	0.48	6.03	4.09
	±96.0	±53.6	± 0.68	± 0.90	±1.76	± 1.02	±1.29	±0.58	±0.22	±0.19	±0.71	±0.26
Spring	513.3	891.0	21.41	28.27	9.34	9.89	12.48	10.20	0.085	0.36	4.04	4.86
	±39.7	±145.7	± 4.93	± 0.80	±1.52	± 0.01	±0.09	±0.72	±0.03	±0.19	±0.45	±0.28
G	662.9	696.3	21.65	51.47	9.30	9.73	11.12	7.14	0.207	0.21	5.34	4.22
Summer	± 88.5	±186.6	± 2.44	± 14.78	±0.71	±0.82	±0.16	±0.43	±0.02	±0.10	± 1.41	±0.39
	719.3	951.2	13.02	25.79	21.01	14.09	16.45	11.67	0.16	0.34	6.31	4.75
Autumn	±101.7	± 49.8	±0.35	±4.33	±0.36	± 0.08	±0.22	±0.25	± 0.01	± 0.07	± 0.58	±0.44
	602.9	772.3	16.39	28.78	13.69	11.57	14.31	9.10	0.27	0.35	5.43	4.48
Mean	±60.27	±105.9	±3.53	±6.45	±3.22	±1.23	± 1.71	±1.27	±0.13	± 0.06	± 0.58	±0.22
O. M."	687.60		22.59		12.63		11.71		0.31		4.96	
Гwo-way А	NOVA (P-v	alue)										

Table 1: Seasonal mean (±SE) of heavy metals concentration (µg g⁻¹ dry weight) in Ulva sp. during the four sampling seasons from Ras Al-Tin (A) and Al-Mountazah (B).

0.0000 *** 0.0000 *** 0.0863 0.4406 Season 0.0801 0.0023 ** 0.0081 ** 0.0075 ** 0.0000 *** 0.4665 0.0592 Site 0.0386 *. ** and *** are significant at (P<0.05), (P<0.01) and (P<0.001), respectively. O. M.": overall mean of the study area.

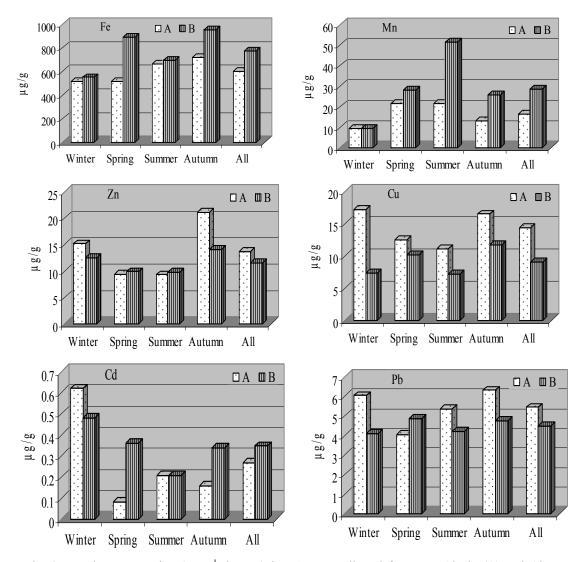


Fig. 2: Metals concentration ($\mu g g^{-1} dry wt$.) in Ulva sp. collected from Ras Al-Tin (A) and Al-Mountazah (B), Mediterranean coast, Alexandria region, during the four seasons.

Abdallah and Abdallah, (2008) showed that the metal variation in species from different sampling sites may be related to factors such as tidal range, temperature, salinity, dissolved nutrients, type of tissue, age of algae, its nutritional history and the geological structure of the study area. Forsberg *et al.* (1988) showed that the dynamic factors could affect the metals uptake in seaweed.

A comparison of the present results with data reported for similar algal species from other Mediterranean marine environment in Egypt and other regions (Table 2), suggest that the heavy metal concentrations showed higher and lower than the others studies. Studied metals are lower in the Egyptian Coast Mediterranean Sea algae than the Turkish Coast algae (Topçuoğlu *et al.*, 2010) for Fe, Mn and Zn; Italy Coast (Caliceti *et al.*, 2002) for all metals except Cd and Spain Coast (Villares *et al.*, 2001) for Mn and Zn. Abdallah and Abdallah (2008) recorded high concentrations of Mn, Zn, Cu and Cd in *Ulva* sp. in the coast of Mediterranean Sea, Egypt. In addition, Cu, Cd and Pb (except Pb in Abdallah, 2010) in this study are higher in *Ulva* sp. than that recorded for the same species by Abdallah, 2010; Topçuoğlu *et al.*, 2010). On the other hand, other references recorded lower metals concentrations than the present study, are also delineated.

Table 2: Comparison of heavy metals concentrations (ug g⁻¹ dry wt.) of *Ulva* sp. in the present study and other studies at the Mediterranean Sea coast.

Reference	Location	Fe	Mn	Zn	Cu	Cd	Pb
Abdallah and Abdallah (2008)	Egypt	514.55±2.1	73.95±4.8	63.10±3.2	14.52±4.7	1.84±0.93	-
Abdallah (2010)	Egypt	-	8.07-92.1	10.0-97.5	2.0-8.61	0.18-1.6	4.5-6.65
Topçuoğlu <i>et al.</i> (2010)	Turky	3195.1±12.3	68.02±0.51	76.37±1.68	7.93±0.11	< 0.02	< 0.10
Alp et al. (2012)	Turky	94.52-360.51	3.07-11.17	3.38-5.46	0.22-2.27	0.02-0.05	0.70-1.11
Villares (2001)	Spain	365.0	37.5	23.6	7.84	-	-
Caliceti <i>et al.</i> (2002)	Italy	1033 ± 564	-	64.0±55.0	13±7.0	0.2±0.0	7.3±6.4
Present study	Egypt	602.9-772.3	16.39-28.78	11.57-13.69	9.10-14.3	0.27-0.35	4.48-5.43

Heavy metal distributions in Water

The seasonal mean of different metals concentration in seawater are shown in Table (3) and Figure (3). Metal levels exhibited the same pattern across the two sites: Fe>Zn>Mn>Cu>Pb>Cd. Iron (Fe) distribution revealed high levels at site A (134 μ gl⁻¹) and decreased at site B (131 μ g l⁻¹). In general, Fe concentration revealed irregular variation among seasons. Manganese (Mn) varied between 4.5 and 7.5 μ g l⁻¹ at Ras Al-Tin to 5.9 and 7.7 μ g l⁻¹ at Al-Mountazah. The mean value tend to be increased slightly at site B (6.6 μ g l⁻¹) than in site A (6.0 μ g l⁻¹). The highest values of Fe and Mn were recorded during winter and autumn.

The results revealed wide variation in the distribution of zinc concentrations. Zn concentration varied between 4.5 and 7.0 μ g l⁻¹ at site A and 3.8 and 11.5 μ g l⁻¹ at site B, with the highest concentrations during winter. The total average Zn content ranged from 5.8 μ g l⁻¹ at site A and 7.1 μ g l⁻¹ at site B. Copper concentration is remarkable in surface water; the values tend to be higher at the site A (3.2–4.9 μ g l⁻¹) than site B values (3.4-4.3 μ g l⁻¹). In general, the total average values of Cu ranged from 4.0 μ g l⁻¹ at site B to 4.2 μ g l⁻¹ at site A (Table 3) with the maximum value in winter.

The lead (Pb) concentrations were generally increased at winter and remained relatively low at summer (Table 3 and Fig. 3). Pb concentrations in the study area ranged from 3.4 to 3.9 μ g l⁻¹ at site A and 3.4 to 3.9 μ g l⁻¹; 1.5 to 7.8 μ g l⁻¹ at site B.

The average values of Pb concentration ranged from 3.8 to 4.8 μ g l⁻¹ at site A and site B, respectively. Table (3) and Figure (3) shows the concentration of cadmium (Cd) is not detected at the two sites during the study period.

	Metals											
Season	Fe		Mn		Zn		Cu		Cd		Pb	
	Α	В	А	В	А	В	Α	В	Α	В	Α	В
Winter	157	170	6.9	7.7	7.0	11.5	4.9	4.3	ND	ND	4.0	7.8
winter	±11	±17	±1.4	±1.3	±1.5	±2.1	± 1.0	± 1.0	± 0.0	± 0.0	±0.9	±1.4
Sumin a	121	140	4.5	5.9	5.3	6.2	4.1	4.2	ND	ND	3.7	5.9
Spring	±7	±19	± 1.1	±1.3	±1.2	±1.3	± 1.0	± 1.0	± 0.0	± 0.0	±0.7	± 1.1
C	82	91	5.2	6.1	4.5	3.8	4.4	4.0	ND	ND	3.4	1.5
Summer	± 6	±11	±1.2	±1.5	± 1.1	± 0.8	± 1.0	±0.7	± 0.0	± 0.0	± 0.8	±0.3
A	176	122	7.5	6.6	6.2	6.8	3.2	3.4	ND	ND	3.9	3.9
Autumn	± 10	±13	±1.2	±1.5	±1.4	±1.2	±0.6	±0.7	± 0.0	± 0.0	± 0.8	±0.7
Mean	134	130.8	6.0	6.6	5.8	7.1	4.2	4.0	ND	ND	3.8	4.8
Mean	±24	±19	±1.3	± 1.4	±1.3	±1.5	± 1.0	± 0.8	± 0.0	± 0.0	±0.7	± 1.0
O. M."	1	32	6.3		6.5		4.1		ND		4.3	
				Two	way AN	IOVA (I	P-value)					
Season	0.00	0.0000 *** 0.5363		0.0216 *		0.5217		ND		0.0487 *		
Site	0.1	019	0.6	0.6513 0.227		278	0.7761		ND		0.1627	
1	•		· (D .0.)		(D 0 0	0.4.	. 1 0 1 (1) 11				0	.1

 Table 3. Seasonal mean (±SE) of heavy metals concentration in water of Ras-Al-Tin (A) and Al-Mountazah (B) on shore of the Mediterranean coast, Alexandria region.

* and *** are significant at (P<0.05) and (P<0.001), respectively. O. M.": overall mean of the study area.

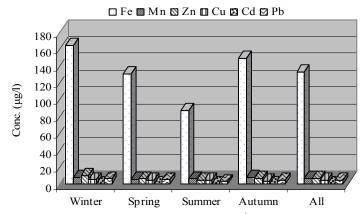


Fig. 3: Seasonal variations of metals concentration ($\mu g l^{-1}$) in water of the Mediterranean coast during the four seasons.

Metals concentrations varied significantly (P<0.05) through seasons (except Mn and Cu) (Table 3). On the other hand, no significant difference was recorded between the sites. This pattern may be as a result of similar conditions affecting these sites as local anthropogenic sources, sediments and/or the release of metals into the water as a freshwater and seawater combination (Perez *et al.*, 2003). The effect of wind as mentioned earlier not only increase the transfer of soil particles into the sea, but it also caused water agitation and mixing of sediment particles, both of which may increase the heavy metal loads in the water. Spring and summer showed the lowest values than other seasons. This may be due to the contribution of phytoplankton biomass in these seasons (El-Gohary *et al.*, 2012). Faragallah *et al.* (2009) suggested that the biological uptake may the main factor controlling the removal mechanisms of dissolved heavy metals. They found significant negative relationship between dissolved Cu, Fe and Zn content and the values of chl-a concentration and pH.

Table (4) shows that concentrations of heavy metals in seawater increased or decreased as compared to other studies on Mediterranean coastal waters, Egypt. Elevated values of Fe in water comparing with those previously reported may be explained as a result of the effect of the agitation of the bottom sediments which have relatively high metal concentrations (Faragallah *et al.*, 2009) as samples collected on shore. Generally, the levels of all studied metals in the two sites are still below the background values of the natural water recommended by WHO (2011) for surface or natural water.

Table 4: Comparison of heavy metals concentration ($\mu g l^{-1}$) in the present study and other references on the Mediterranean Sea water in Egypt.

Reference	Metals								
	Fe	Mn	Zn	Cu	Cd	Pb			
El-Nady (1996)	-	-	1.0-39.8	0.6-12.2	N.D-2.5	0.4-18.8			
Okbah and Nasr, 2006	11.92-30.45	5.79-17.36	0.87-7.80	0.40-1.87	-	1.53-10.31			
Abdallah, 2008	27.21-70.91	2.7-31.28	20.79-59.29	3.69-4.90	0.66-6.45	2.65-26.14			
Faragallah et al., 2009	28.66-43.73	-	25.94-38.22	2.0- 6.74	-	2.93- 4.33			
El-Gohary et al., 2012	-	-	4.67-31.39	3.04-6.08	0.57-1.26	15.96-22.64			
Present study	132	6.3	6.5	4.1	ND	4.3			
*WHO (2011)	0.5-50.0	0.1-0.4	0.01-0.05	1.0	0.003	0.01			

*Metals concentration in natural water (mg/l).

The bioconcentration factor (BCF) provides an index of the ability of the plant to accumulate metal with respect to the metal concentration in the substrate. The overall consistency in metal uptake by *Ulva* sp. at sampled sites was reconfirmed by its BCF values (Fig. 4). These values were varied in orders of magnitude and location and decreased as for Fe>Mn>Cu>Zn>Pb>Cd. *Ulva* sp. showed a consistent, high affinity for Fe, Mn and Cu and to less extent for Cd and Pb in the two sites. This phenomenon indicates that this seaweed has a bioavailability to accumulate metals from the surrounding medium and can suggest using it as a good bioindicator for the presence of essential and highly toxic metals. Also, it is clear that Fe, Mn and Cd more accumulated in algae collected from site II, whereas Zn, Cu and Pb highly concentrated in algae tissues at site I.

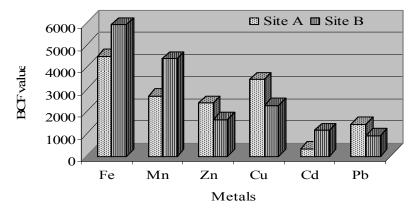


Fig. 4: Bioconcentration factors (BCF) of metals in studied sites, Ras-Al-Tin (A) and Al-Mountazah (B).

The high accumulation of metals in *Ulva* sp. reflect the high bioavailability of these metals in the study area, and the capacity of the alga to take them up and

sequencer them. This coincides with Shanmugam *et al.* (2012) who stated that seaweeds are excellent agents of filtering the metals like iron, zinc, copper and cadmium from seawater and accumulate them in their body cells 4,000-20,000 times more than the surrounding water.

The correlation coefficient matrix between heavy metal concentrations in *Ulva* sp. tissues showed some positive significant correlations for some pairs of metals. There were positive relationships (P<0.05) between Fe and Mn; Mn and Cu; Zn and Cu; Zn and Pb; Cu and Pb and finally no significant correlations were found (P>0.05) between other paired of metals (Table 5). These results can show that the six elements measured in the present study having similar sources and this may be related to geographical structure of this area as well as due to anthropogenic activities caused by industrial effluents and domestic sewage. Statistically significant correlations were observed between Cu and Pb concentrations and Fe–Mn (Favero *et al.*, 1996). The synergistic interaction of Pb with Zn in *Ulva compressa* was observed (Dadolahi *et al.*, 2011). In our study, the significant positive correlation between Zn and Cu in seaweed rules out large competition effects between these two metals. Similar results were recorded by (Giusti, 2001).

Table 5: Correlation coefficient matrix (r) between concentrations of paired metals in green macroalgae (*Ulva* sp.)

Fe	Mn	Zn	()1		
1 0			Cu	Cd	Pb
1.0					
0.0261*	1.0				
0.4942	0.1192	1.0			
0.4830	0.0305*	0.0009***	1.0		
0.8958	0.5100	0.2629	0.4450	1.0	
0.6261	0.4429	0.0031**	0.0044**	0.0714	1.0
	0.0261* 0.4942 0.4830 0.8958	0.0261* 1.0 0.4942 0.1192 0.4830 0.0305* 0.8958 0.5100 0.6261 0.4429	0.0261* 1.0 0.4942 0.1192 1.0 0.4830 0.0305* 0.0009*** 0.8958 0.5100 0.2629 0.6261 0.4429 0.0031**	0.0261* 1.0 0.4942 0.1192 1.0 0.4830 0.0305* 0.0009*** 1.0 0.8958 0.5100 0.2629 0.4450 0.6261 0.4429 0.0031** 0.0044**	0.0261* 1.0 0.4942 0.1192 1.0 0.4830 0.0305* 0.0009*** 1.0 0.8958 0.5100 0.2629 0.4450 1.0 0.6261 0.4429 0.0031** 0.0044** 0.0714

*, ** and *** are significant correlation at (P<0.05), (P<0.01) and (P<0.001), respectively.

CONCLUSION

The average metal concentrations for seaweeds from the two shoreline areas (Ras Al-Tin and Al-Muntazah) were different. Fe, Mn, Cu and Zn showed the greater spatial (except Fe) and temporal differences (P<0.05). The results showed that the studied area in Mediterranean coast in Egypt not faced heavy metals pollution. *Ulva* sp. plays an important role as a bioindicator for heavy metals contamination in seawater. This conclusion was supported by the thousand times higher levels of metals concentrations observed in the *Ulva* than in water from the same sites

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ARABIC SUMMARY

الأعشاب البحرية الخضراء (Ulva sp.) كدليل بيولوجي للتلوث بالمعادن الثقيلة بساحل البحر الأبيض المتوسط، منطقة الإسكندرية، مصر

> سمير محمد سعيد موسى، ياسر ثابت عبد المجيد مصطفى مركز البحوث الزراعية، المعمل المركزي لبحوث الثروة السمكية بالعباسة، قسم الليمنولوجي

تم استخدام الأعشاب البحرية الخضراء (.Ulva sp) كدليل بيولوجى لتقييم التلوث بالمعادن الثقيلة بساحل البحر الأبيض المتوسط، منطقة الإسكندرية، مصر تم أخذ عينات المياه ونبات الأولفا شهريا من منطقة رأس التين والمنتزة فى الفترة من يناير -ديسمبر ٢٠١٢ لدراسة التغيرات الموسمية والمكانية لتراكم العناصر . أوضحت الدراسة أن تركيز العناصر فى المياه وأنسجة أعشاب البحر أقل من النسب المسموح بها عالميا للمناطق الملوثة. كما أوضحت الدراسة أن هذه العناصر سجلت فروق ذات دلالة معنوية بين المواسم (المنجنيز، الزنك والنحاس) وبين الأماكن (الحديد، المنجنيز، الزنك والنحاس) فى أنسجة الأعشاب. أظهرت الأعشاب البحرية قابلية كبيرة لاختزان العناصر وذلك من خلال قيم معامل التركيز البيولوجى وكان تركيز العناصر الثقيلة على النحو التالى: الحديد > المنجنيز > الزنك > النحاس > الرصاص > الكادميوم. وخاصت هذه الدراسة إلى أن الأعشاب البحرية الخضراء (.Uva sp) تلعب دورا هاما كدليل بيولوجى للتلوث بالمعادن الثقيلة على الأعشاب البحرية الحديرات العناصر وذلك من خلال قيم معامل التركيز البيولوجى وكان تركيز العناصر الثقيلة على النحو التالى: الحديد > المنجنيز > الزنك > النحاس > الرصاص > الكادميوم. وخاصت هذه الدراسة إلى أن الأعشاب البحرية الخضراء (.Ulva sp) تلعب دورا هاما كدليل بيولوجى للتلوث بالمعادن الثقيلة فى مياه البحر. كما أن منطقة الدراسة لاتواجه تلوث بالمعادن الثقيلة.