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**DESIGN AND PERFORMANCE OF WES/PEM FUEL CELLS
HYBRID ELECTRIC SYSTEM**

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Abstract:

This paper studies the design and performance of a WES / PFCS hybrid electrical power system. This system consists of Wind Energy System, WES, and Pem fuel cell system PFCS. The design of this system depends on the hourly energy balance technique. When the power output from WES, P_{ew} , exceeds the load demand, P_L , the exceed power used by the alkaline electrolyser to produce hydrogen. Then the H_2 is compressed and stored in pressurized storage tanks. Also, when $P_{ew} < P_L$ this means that there is deficit power and the stored H_2 is utilized for generation electrical power by the PFCS to overcome the deficit on the system. The methodology which developed here has been applied on EL-Zafarana site as a case study. EL-Zafarana is a remote site located at the coast of the red sea, Egypt.

Keywords:

Design, performance, WES, PFCS, Electrolyser

1-Introduction:

As energy demands around the world increase, the need for a renewable energy source that will not harm the environment has been increased. Some projections indicate that

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the global energy demand will almost triple by 2050 [1 , 2] .Wind is one of the renewable energy sources which can be harnessed in a commercial way.

On the other hand the fuel cell is an electrochemical device that converts the chemical energy of a fuel directly into electricity. The components of a fuel cell are anode, anodic catalyst layer, electrolyte, cathodic catalyst layer, cathode, bipolar plates/interconnects and sometimes gaskets for sealing/preventing leakage of gases between anode and cathode [3].In this hybrid system the PEM fuel cells are used with type of 100 kW fuel cell systems Nedstack PS100 and operational ambient temperature -20°C to +40°C fuel by hydrogen with purity of 99.95 % [4].

2. Description of the hybrid system:

Figure 1 shows the components of the WES/ PFCS hybrid system under study. These components are wind turbine generator, WTG, PEM, fuel cells, the alkaline electrolyser, DC/AC inverter, storage hydrogen storage tank and controlling the system operation.

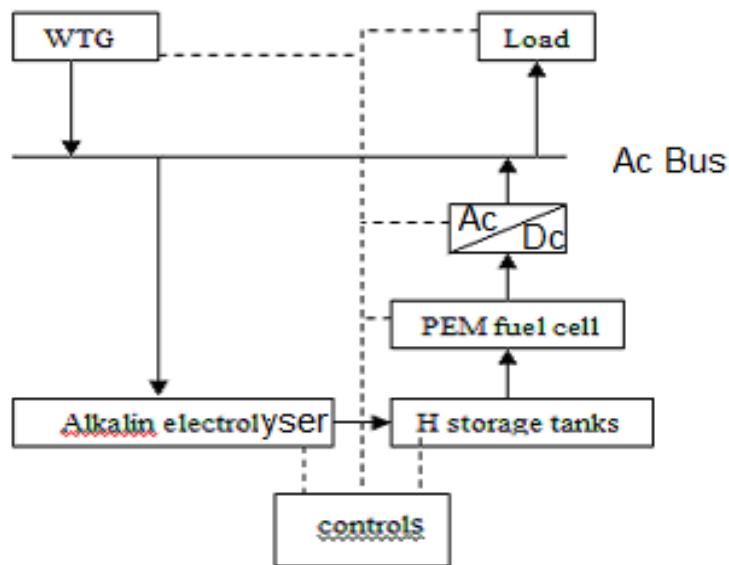


Figure (1): Components of the WES/FCS hybrid system

3. Methodology:

3-1 Design of WES

(a) Calculation of the output power from WES:

The wind speed is varying as the height from the ground and is varying with roughness of the ground. The boundary layers are formed near the surface of the earth. In this boundary layer wind velocity are smaller than the velocity of the free wind stream outside the boundary layer. The equation that relates all the previous elements are [5, 6]:

$$u_n = \left(\frac{h}{h_r}\right)^\alpha u_{n_0} \quad (1)$$

The average output power from ,WTG, can be calculated as follows [5, 6]

$$P_{ew,Ave} = P_{ew,r} CF \quad (2)$$

$$CF = \frac{\exp\left[-\left(\frac{u_c}{c}\right)^k\right] - \exp\left[-\left(\frac{u_r}{c}\right)^k\right]}{\left(\frac{u_r}{c}\right)^k - \left(\frac{u_c}{c}\right)^k} - \exp\left[-\left(\frac{u_f}{c}\right)^k\right] \quad (3)$$

(b) Calculation of the ANWTG:

The average number of WTG, ANWTG, can be estimated by knowing the average power production $P_{ew,Ave}$ and average load power $P_{L,Ave}$ which can be determined from eqn.(5). Then the ANWTG required is given by the following equation taking into consideration the efficiencies of PFCS:

$$ANWTG = \frac{\left(\frac{P_{L,Ave}}{\eta_{pfcs}}\right)}{P_{ew,Ave}} \quad (4)$$

$$P_{L,Ave} = \left(\frac{1}{8760}\right) * \sum_{i=1}^{8760} P_L(i) \quad (5)$$

Where; 8760 is the number of hours per year.

(c) Energy balance study:

To compute the optimum number of WTGs required an energy balance between the load and the output from WES must be taking into consideration η_{pfcs} . The output energy from WES must satisfy the load required energy. For periods with high wind speed (i.e. the output power from WES is greater than the load requirements) the excess power

used by the alkaline electrolyser to produce H₂. Then H₂ is compressed and stored in pressurized storage tanks. For the periods of low outputs from WES, which is less than the load demand, the stored H₂ is utilized for power generation in the PFCS to overcome the deficit power. The optimum number of WTGs required and the energy output for each case can be obtained as follows:

1-The hourly-generated power can be calculated according to the value of hourly wind speed from the following equation: [5, 6]

$$\begin{aligned}
 P_{ew}(t) &= 0 && \text{if } u(t) < u_c \\
 P_{ew}(t) &= (A+B*u^k) N && \text{if } u_c \leq u(t) < u_r \\
 P_{ew}(t) &= P_{wr} N && \text{if } u_r \leq u(t) < u_f \\
 P_{ew}(t) &= 0 && \text{if } u(t) \geq u_f
 \end{aligned} \tag{6}$$

Where:

$$A = \frac{P_{ewr} u_c^k}{u_c^k - u_r^k}$$

$$B = \frac{P_{ewr}}{u_r^k - u_c^k}$$

2-It is compared between the hourly generated power, P_{ew}(t), and hourly load power, P_{Lh}(t). If P_{ew}(t) is greater than P_{Lh}(t) then there is an hourly surplus power, but if P_{ew}(t) is lesser than P_{Lh}(t) then there is an hourly deficit power, at any value of N if the summation of hourly surplus power equal to the summation of hourly deficit power then this value of N represents the optimum number of WTG. The following process can be done to obtain the optimum number of WTG:

$$\text{IF } \sum_{i=1}^{8760} [P_{ew}(t) - P_{Lh}(t)] > 0 \tag{7}$$

Then, number of WTG must be decreased by one unit, and repeating the foregoing process.

$$\text{IF } \sum_{i=1}^{8760} [P_{ew}(t) - P_{Lh}(t)] < 0 \tag{8}$$

Then, number of WTG must be increased by one unit, and repeating the foregoing process.

$$\text{IF } \sum_{i=1}^{8760} [P_{ew}(t) - P_{Lh}(t)] = 0 \tag{9}$$

Then, number of WTG satisfies the energy balance condition and value of N is taken as the optimum number of WTG, ONWTG.

3.2 Design of PFCS:

In the case of a H₂/O₂ PEM fuel cell, H₂ and O₂ are the fuel and oxidant respectively. The product is pure water H₂O and electricity.

(a) Calculation of numbers of PEM fuel cells modules M:

Pem fuel cell must satisfy the deficit power load demand. Then the deficit load power should be used to compute the number of the Pem fuel cell modules needed, M as follows:

$$M = \frac{P_{\max, \text{def}}}{P_{\text{rat}}} \quad (10)$$

It is needed also to use inverter to convert the dc power output from the pem fuel cell module to AC power. [7]

(b) Calculation of fuel cell efficiency:

The most widely used efficiency value of the fuel cell is based on the change in the standard free energy of the fuel cell reaction. In H₂/O₂ PEM fuel cell, the chemical reaction produces water in the liquid form. At standard conditions of 25°C (298K) and 1 atm, the thermal input energy in the H₂/O₂ reaction is 285.84 kJ mol⁻¹ and free energy available for useful work is 237.2 kJ mol⁻¹. So, the maximum theoretical H₂/O₂ fuel cell efficiency is 0.83 [8]. The efficiency of actual fuel cell is often expressed in terms of the ratio of the operating cell voltage to the ideal cell voltage. The actual cell voltage is less than the ideal cell voltage due to the polarization losses, which include concentration polarization, activation polarization and ohmic polarization.

$$(11) \eta_{\text{pfcs}} = 0.83 \frac{V_{\text{act}}}{E_{\text{ideal}}}$$

The ideal voltage of a cell at 1 atm, and 25°C is 1.229 V, and the actual fuel cell operating at a voltage of V_c, then: [9]

$$\eta_{\text{pfcs}} = 0.83 \frac{V_c}{1.229} = 0.675 V_c \quad (12)$$

(c) Fuel cell usage Parameters:

1-Hydrogen usage

One mole of hydrogen would produce exactly 2 F of charge; the rate of usage of hydrogen is derived in a way similar to oxygen, except that there are two electrons from each mole of hydrogen, i.e,

charge $Q = 2F$ amount of H_2

Dividing this equation by time, and rearranging for M cells

$$(13) \quad H_2 \text{ usage} = \frac{IM}{2F}$$

This formula can be in terms of power, rather than current and also without needing to know the number of cells. If the voltage of each cell in the stack is V_c , Then,

$$P_{ef} = IMV_c \quad (14)$$

From the Equations (13) and (14) thus become

$$H_2 \text{ usage} = \frac{P_{ef}}{2V_c F} \quad \text{moles s}^{-1} \quad (15)$$

The molar mass of hydrogen is $2.02 \times 10^{-3} \text{ kg mole}^{-1}$, so this becomes

$$H_2 \text{ usage} = 1.05 \times 10^{-3} \frac{P_{ef}}{V_c} \text{ kg mole}^{-1} \quad (16)$$

The result of hydrogen usage can be transformed to a volume rate Nm^3 using the density of hydrogen, which is 0.084 kgm^{-3} at normal temperature and pressure (NTP).

2- Oxygen usage

Exactly two moles of hydrogen would be provided for each mole of oxygen. This would produce 4 F of charge, since two electrons are transferred for each mole of hydrogen from the above we get that four electrons are transferred for each mole of oxygen. So, on the same way get that [10]:

$$O_2 \text{ usage} = \frac{P_{ef}}{4V_c F} \quad \text{moles s}^{-1} \quad (17)$$

The molar mass of oxygen is $32 \times 10^{-3} \text{ kg mole}^{-1}$, so, this becomes:

$$O_2 \text{ usage} = 8.29 \times 10^{-3} \frac{P_{ef}}{V_c} \text{ kg mole}^{-1} \quad (18)$$

(d) Calculation of hydrogen production and water consumption:

When the power generated from WES is greater than the load power demand then there is surplus power can be used to generate the hydrogen and the oxygen fuel. To calculate the hydrogen and the oxygen productions fuel, considering the water alkaline

electrolyser operates at (NTP) normal temperature and pressure (20°C or 293.15 K and 1 atm, or 1.0132 bar), then:

$$H_{2\text{production}} = \frac{\text{surplus power}}{\text{conversion efficiency}} \quad (19)$$

Knowing the hydrogen production value, the oxygen production and the water Consumption can be calculated. [7]

The number of WTG satisfies the energy balance ONWTG that makes $H_{\text{usag}} = H_{\text{production}}$. It is needed to increase the optimum number of WTG, ONWTG, by N_{add} to overcome the difference between H_{usag} and $H_{\text{production}}$ and the losses of fuel cell and the electrolyser, the following equations can be used for this purpose:

$$H_{\text{need}} = H_{\text{usag}} - H_{\text{production}} \quad (20)$$

$$EH_{\text{need}} = H_{\text{need}} \cdot \eta_e \quad (21)$$

The yearly energy of one turbine YE_1 can be calculated from the following equation:

$$YE_1 = \frac{TYE}{\text{ONWTG}} \quad (22)$$

$$N_{\text{add}} = \frac{EH_{\text{need}}}{YE_1} \quad (23)$$

$$\text{ONWTGM} = \text{ONWTG} + N_{\text{add}} \quad (24)$$

3.3 Electrolyser Subsystem Design:

The decomposition of water into hydrogen and oxygen can be achieved by passing an electric current (DC) between two electrodes separated by an aqueous electrolyte with good ionic conductivity. [11]

To design the alkaline electrolyser subsystem the maximum hourly surplus power ($P_{\text{max.sur}}$) and also the characteristics of alkaline electrolyser (nominal flow rate of hydrogen in Nm^3/h and the Conversion efficiency for the electrolyser in kWh/Nm^3) must be known.

$$\text{The power consumed in alkaline electrolyser (P)} = \text{nominal flow rate of hydrogen} \cdot \text{Conversion efficiency of the electrolyser} \quad (25)$$

$$\text{Number of electrolyser modules} = \frac{P_{\text{max.sur}}}{P} \quad (26)$$

3.4 Hydrogen Tank Design:

Hydrogen can be stored in three types gas, liquid and solid hydrogen .There are certain metals like magnesium, titanium and iron absorb hydrogen when cooled and release it when it heated.[7]

The yearly hydrogen production when electrolyser operates as a gas according to the ideal gas law: [12]

$$P_i V_i = P_f V_f \tag{27}$$

$$\text{Number of storage tanks} = \frac{V_f}{\text{Size of the storage tank}} \tag{28}$$

4. Case Study:

El- Zafarâna site has been selected for this case study is at the coast of the red sea it’s in the south east of Cairo by 220 km Zafarana site features Flat area, Low turbulence and High capacity factor it has an economical wind speed for the WTG. [13]

4.1The Energy That Generated From WTG:

The previous methodology applied to determine the optimum number of WTG that required feed the demand load. The WTG that used in this system is **wes30 250kW** which characteristics from data sheet are:

Table (1): wes30 250kW characteristics

GENERAL SPECIFICATIONS	
Cut in wind speed	< 3 m/s – 6.7 mph
Cut out wind speed	25 m/s - 56 mph
Nominal wind speed	12 m/s - 27 mph
ELECTRICAL SPECIFICATIONS	
Power	250 kW
Voltage	400V/50 or (60) Hz 3 phase

By applying the load demands as the following:

Table (2): The load demand

Time	Load KW			
	Street lighting	residential	commercial	total

1A.M	100	200	200	500
2	100	200	200	500
3	100	200	200	500
4	100	200	200	500
5	100	200	200	500
6	100	200	200	500
7	100	300	200	500
8	-	400	300	700
9	-	500	500	1000
10	-	500	1000	1500
11	-	500	1000	1500
12 noon	-	500	1000	1500
1	-	500	1000	1500
2	-	500	1200	1700
3	-	500	1200	1700
4	-	500	1200	1700
5	100	600	1200	1800
6	100	700	800	1500
7	100	800	400	1200
8	100	1000	400	1400
9	100	1000	400	1400
10	100	800	200	1000
11	100	600	200	800
12 P.M	100	300	200	500

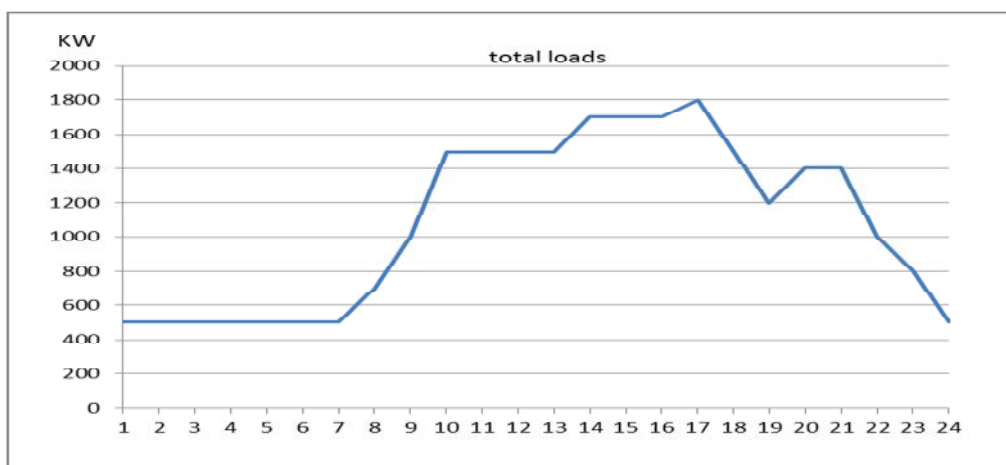


Figure (1): the total load demand curve

From the previous data and the wind speed of El- Zafarâna site can calculate the Weibull parameters and also:

shape Parameter	K=1.23
Scale Parameter C	7.069
capacity factor CF	0.278
Average wind power $P_{we,ave}$	69.5 kW
ONWTG	12 turbines
maximum surplus power $P_{max,sur}$	1765.261 kW
maximum deficit power $P_{max,def}$	1700 Kw
TYE	10221136.03 kWh

4.2 Electrolyser Subsystem:

In this system the alkaline electrolyser module type HySTAT®-30-10 is used to design the alkaline electrolyser subsystem must be known same characteristics as:

Module type	HySTAT®-30-10	
Hydrogen production	Nominal flow rate	30 Nm ³ /h
Oxygen production	Nominal flow rate	50% of H ₂
Conversion efficiency		5.2 kWh/Nm ³
power supply	Voltage	400 VAC ± 3%
water supply	Consumption	1.5 - 2 liters/Nm ³ H ₂
Operating conditions	Ambient temperature	-20°C to +40°C

By substitution in equations (25) (26) get that:

power consumed (P)	156 kW
Number of electrolyser	12 modules
By substitution in equation (19) get that	
Yearly H ₂ production	642543.7 Nm ³

4.3 PEM Fuel Cells Modules:

In this system the PEM fuel cell that used is **Nedstack Ps 100** module type which characteristics from data sheet:

Table (2): Nedstack Ps 100 characteristics:

Type	fuel cell Nedstack PS100	
Performance	Rated Power	100 kW
	Efficiency	55 – 57 %
Fuel	Hydrogen	

Applying equations (10) and (12) gets that:

number of fuel cell modules	17 modules
cell voltage V_c	0.8V
Applying equation (16) can get that:	
yearly hydrogen usage	1473148.1 Nm ³

Since the inverter to convert the output voltage DC of the fuel cell to be in case of AC voltage.

4.4 Calculate the Optimum Number of WTG's :

It is needed to increase the optimum number of WTG, ONWTG, by N_{add} to overcome the difference between H_{usag} and $H_{production}$.

Apply equations (20), (21), (22), (23) and (24) can calculate the following:

H_{need}	830604.4 Nm ³
EH_{need}	4319.14 MWh
YE_1	851.8 MWh
N_{add}	6 turbines
ONWTGM	18 turbines

4.5 Numbers of Hydrogen Tanks:

The yearly hydrogen production when electrolyser operates in Nm^3 will storage as a gas form by applying law equation (27) and considering that will storage in tanks at 9 bar, ($P_f=9$ bar.) and at NTP the initial pressure value ($P_i = 1.01325bar$) so $V_f= 72339.7 Nm^3$ Hydrogen storage at 9 bars could be in a xerxes fiberglass aboveground storage tank. The size of the xerxes storage tank at horizontal tank is $163.2 m^3$ [15] so can calculate the number of the xerxes storage tanks from equation (28)

Number of tanks = 444 tanks

4.6 The Optimal Operation of WES/PEM Fuel Cell Hybrid Electric System:

From the computer program get the relation between the load demand and the power generated from the WTG for all hours for the day for all month by taken four months which represent the all session January, April, July and October. when the load demands power is larger than the power generated from WTG this deficits power cannot confused and when the load demand power is leaser than the power generated from WTG there are surplus power not used this is the problem as shown in fig.2 (a,b,c,d) But when the WTG operates with the hybrid system using the PEM fuel cell and the alkaline electrolyser the output will be as shown in fig.3 (a, b, c, d) deficits power confused by fuel cell and surplus power used to produce hydrogen that feed the fuel cell.

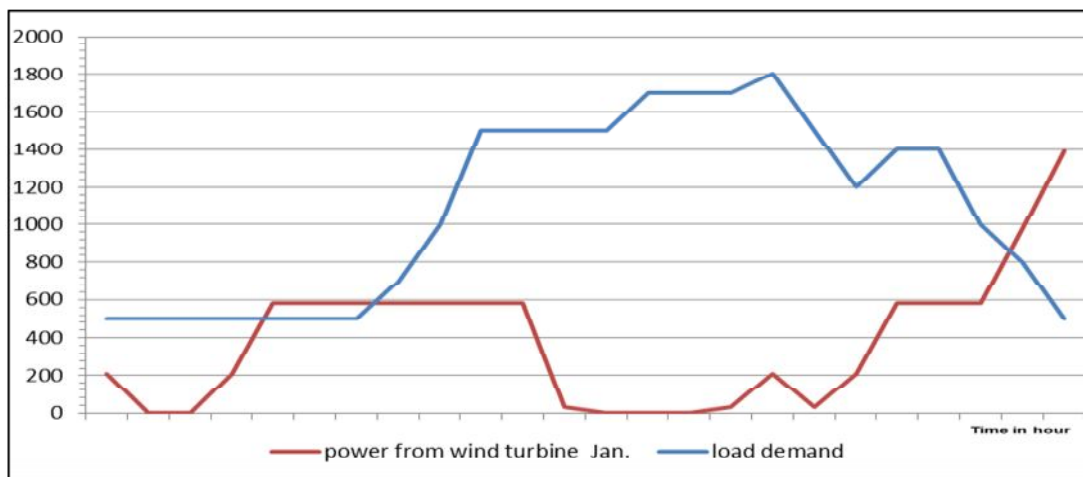


Figure (2.a): load demand with the power of WTG for Jan.

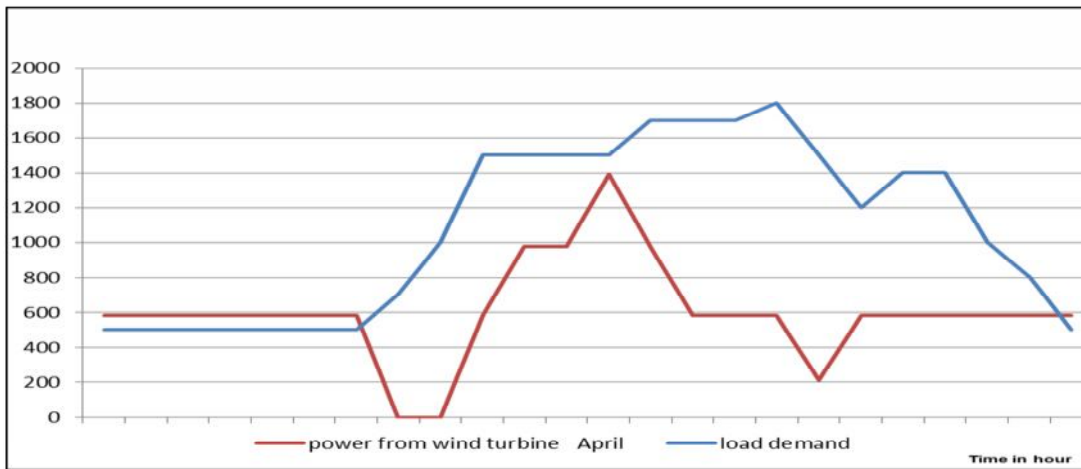


Figure (2.b): load demand with the power of WTG for April.

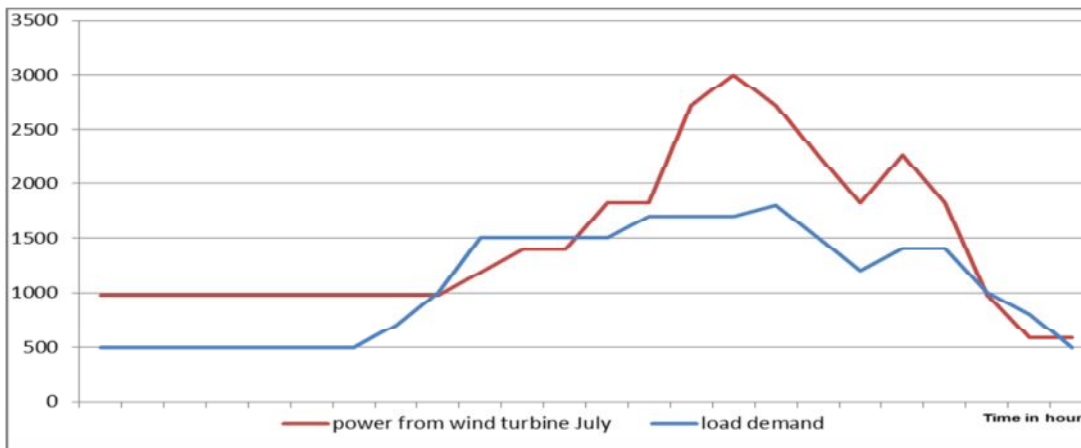


Figure (2.c): load demand with the power of WTG for July.

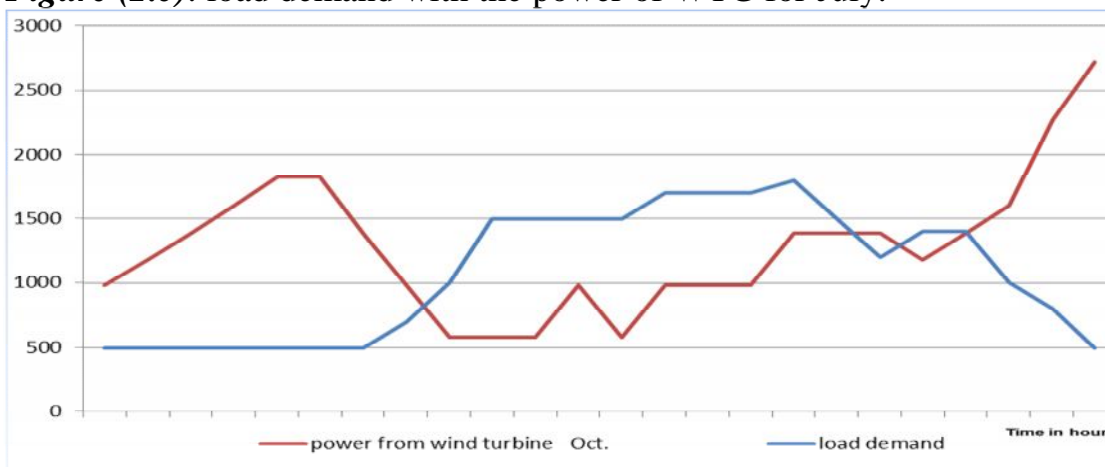


Figure (2.d): load demand with the power of WTG for Oct.

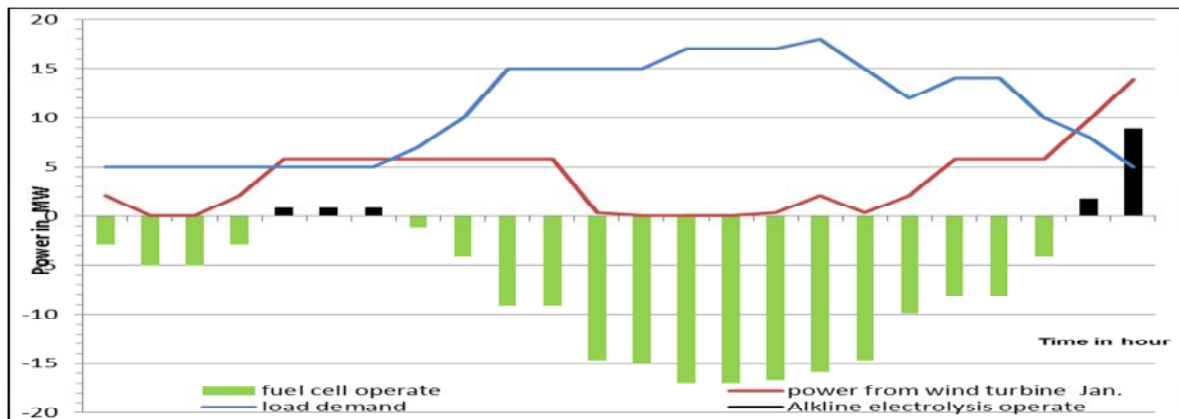


Figure (3.a): optimal operation of WES/PFCS hybrid electric System for Jan.

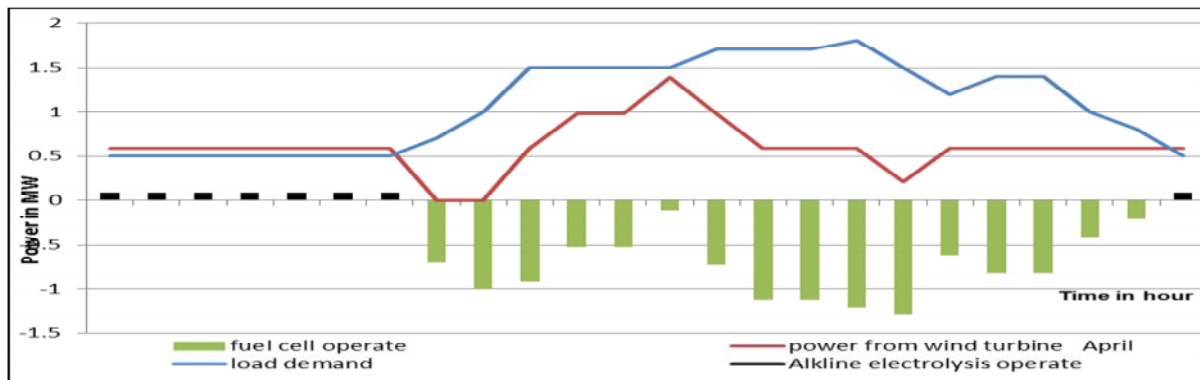


Figure (3.b): optimal operation of WES/ PFCS hybrid electric System for April.

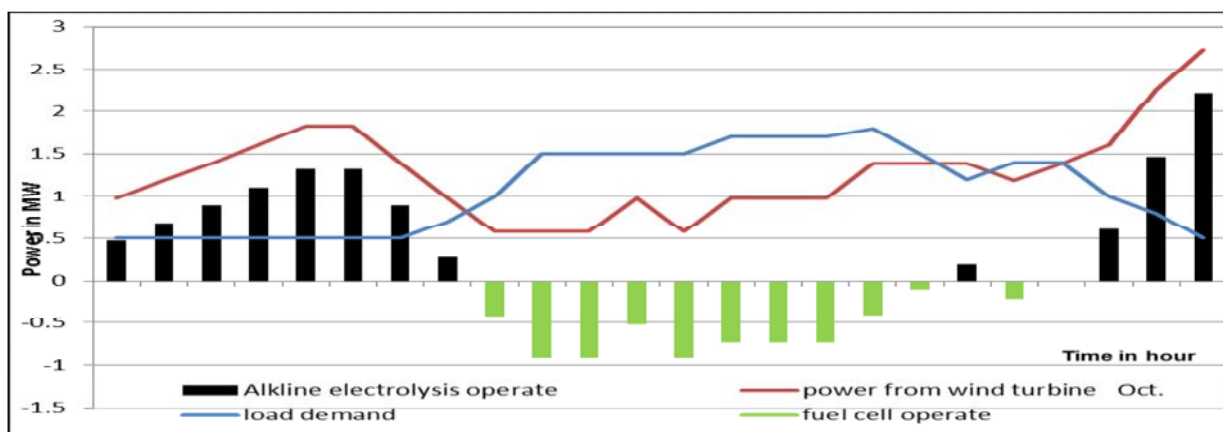


Figure (3.c): optimal operation of WES/PFCS hybrid electric System for July.

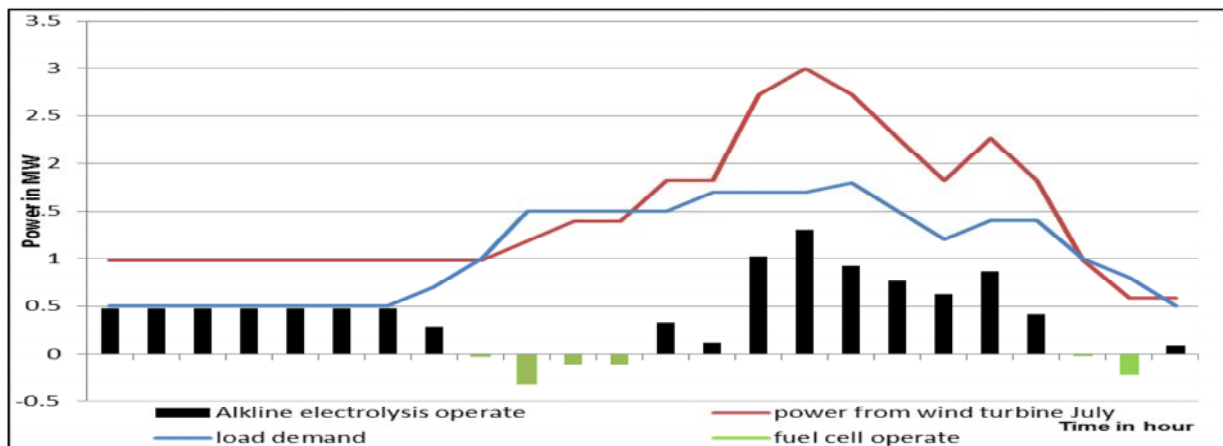


Figure (3.d): optimal operation of WES/ PFCS hybrid electric System for Oct.

5. Conclusion

In this paper using the fuel cell with the wind turbine generator give moor reliability for the system and able to make a quickly response enough to combined the change in power which given from the wind turbine generator due to the change in the wind speed. In this system the output power from the WTG is acts to loads throw the transmission line and use the AC bus to operate the alkaline electrolyser at time which have a surplus power and hence produced amount of hydrogen that storage in the xerxes storage tanks. This amount of hydrogen is used to be input of the fuel cell to produce electricity at time which faced a deficit power to confuse it. This operation of used the alkaline electrolizer or the PEM fuel cell is by the controller which depending on is the power generated from WTG deficit or surplus according to the wind speed.

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List of abbreviation

WEs	Wind Energy System
WTG	Wind Turbine Generator
FCS	Fuel Cells System
ONWTG	The optimum Number of WTGs
ONWTGM	The optimum Number of WTGs modified
CF	The capacity Factor
AVWTG	The Average Number of WTGs.
PFCS	Pem Fuel Cells System
Pem	Proton Exchange Membrane fuel cells.
M	Numbers of PEM fuel cells modules.
N _{add}	Number of turbines should add to make energy balance.
NTP	Normal temperature and pressure
AC	Alternating current
dc	Direct current.
H ₂	Hydrogen gas.

O_2	Oxygen gas.
TYE	The total yearly energy produced from WTG, kWh.
YE_1	The yearly energy of one WTG, kWh.

List of symbols

C	The Weibull Scale Parameter
k	The Weibull Shape Parameter.
u_r	The rated wind speed of WTGs, m/s.
u_c	The cut-in wind speed of WTGs, m/s.
u_f	The cut-off wind speed of WTGs, m/s.
P_{ew}	Electrical Power Produced from WES, kW.
P_L	Load demand power, kW
u_{h_0}	The wind speed at height of h_0 m.
u_h	The wind speed at height h m.
h_0	The height of the measurement, approximately 10 m.
h	The height at which the wind speed estimate is desired.
α	The exponent and is taken as usually 1/7.
$P_{ew,Ave}$	Average electric power production from WTG, kW.
$P_{L,Ave}$	Average load power. kW.
$P_{w,r}$	The rated electrical power of WTG.
η_{pfcs}	Efficiency of Pem Fuel Cells System
$P_{ew}(t)$	The hourly electrical generated power ,kW.
$u(t)$	The hourly wind speed, m/s.
$P_{ew}(t)$	The hourly generated power.
$P_{Lh}(t)$	The hourly load power.
$P_{max.def}$	The maximum load power deficit, kW.
P_{rat}	The rated power of pem fuel cell, kW.
V_c	The actual fuel cell operating voltage equal 0.8 v.
E_{ideal}	The ideal voltage of a cell at 1 atm, and 25 ^o c and equal 1.229 v.
F	Faraday constant = 96485.309.
EH_{need}	Amount of energy can get from the hydrogen need.
$P_{max.sur}$	The maximum hourly surplus power, kW.
P_i and P_f	The initial and finial pressure values in bar.
V_i and V_f	The initial and final volume values in Nm ³ .
P	The power consumed in the alkaline electrolyser Kw.
P_{ef}	Electrical Power Produced from fuel cells, kW.
η_e	The electrolyser Conversion efficiency kWh/Nm ³