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Characterization and Assessment of Fast Neutron Attenuation Properties of Developed Concrete Mixes at Different Temperatures

Moamen G. El-Samrah¹ and Mohamed A.E. Abdel-Rahman¹

Abstract

Concrete is regarded as one of the most proper materials that can be used in many applications concerning with shielding against gamma radiation and neutrons since they are the most penetrating radiation. In this study three different concrete mixes were prepared with three different types of coarse aggregates: barite, goethite and dolomite. Fine aggregates are selected to be local sand and limonite. Additives used are silica fume (SF) and fly ash (FA) by replacement of the total cement weight. To examine the performance of these concrete mixes to be used as radiation shielding concretes, a set of physical, mechanical and radiation attenuation measurements were carried out. Such investigations include compressive strength, slump test, bulk density as well as gamma ray and fast neutron attenuation properties for the prepared mixes at room temperature and after heating up to 450°C. The experimental results revealed that all concrete mixes; barite concrete (labeled Ba.C), goethite concrete (labeled Go.C) and dolomite concrete (labeled Do.C) show adequate physical and mechanical properties that successfully classifying them as high performance concretes. In addition Ba.C has the higher γ-ray attenuation coefficients. For fast neutron attenuation measurements, it was found that Go.C mix has the highest integrated fast neutron removal macroscopic cross section (Σ) emitted from ²⁵²Cf, at room temperature and thus the corresponding minimum required HVL. The Ba.C mix came in the second place following the Go.C mix. Heating up to 450°C, fast neutron attenuation properties of Go.C mix were fully deteriorated because of certain reasons that are illustrated in this study.

Keywords:

Radiation shielding concrete, TGA, Heating effect, Half value layer (HVL), Mass attenuation coefficient (μ_m) and integrated fast removal cross section (Σ)

N. I. E. . . . D. . . . M. . . T. I. . I. G. II. . G. .

¹Nuclear Engineering Department, Military Technical College, Cairo, Egypt



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1. Introduction

In nuclear shielding applications, the most important to be regarded is the indirect ionizing radiation (y-rays and neutrons) since they have high penetrating power and considerable ranges. On the other hand, the direct ionizing radiation (α, β) has small external hazard due to their small ranges in the transferring media and their weak penetration abilities so, they almost lose their full energies in quite small ranges [1]. Hence any shield that can attenuate indirect ionizing radiation to small level will automatically attenuate the others to tiny or ignored values. In attenuating gamma rays, elements with high atomic number are preferred thus dense materials are required like steel and lead for example. However, in attenuating neutrons, especially fast neutrons, light elements like hydrogen and oxygen are required and as a conclusion from the former, the shielding barrier should have high density and in the same time high hydrogen or light elements content so, considering these different demands, the optimum shielding barrier that can be used is concrete especially heavy weight concrete which achieves a good compromise between the high density and high hydrogen content [2]. The selection of local suitable aggregates and additives is very important for preparing radiation shielding concrete to achieve the optimum properties [3]. Some of the widely used aggregates in preparation of radiation shielding concrete are galena, magnetite, barite, limonite, goethite, serpentine, colemanite and some of artificial synthetic aggregates like steel punching, iron fibers, iron shots, steal slag and boron frits [4, 5]. It is essential that aggregates used in radiation shielding concrete must be inert with respect to alkalis and have good physic-mechanical properties to obtain adequate mix [2].

The main objective of this study was to investigate the physical, mechanical and radiation attenuation properties of the different prepared concrete mixes and the effect of heating on their attenuation properties against fast neutrons in order to assess durability and effectiveness of these shielding barriers even if at higher temperature in case of any accident that may occur. For this purpose, three different concrete mixes were prepared from naturally occurring ores using dolomite, barite and goethite, while local sand and limonite ore were selected to be used as fine aggregates. The specimens used in this study to cover the physic-mechanical, gamma rays and fast neutrons attenuation properties have two shapes, standard cubes with dimensions ($10 \text{ cm} \times 10 \text{ cm} \times 10 \text{ cm}$), standard cylinders (with dimensions 10 cm diameter and 20 cm length) that were cut to different thicknesses.



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2. Methodology of research

2.1 Materials

In this study, the aggregates selected were chosen mainly to enhance and develop the attenuation properties against gamma rays and fast neutrons and in the same time satisfy the basic requirements of construction applications. For this purpose, three different concrete mixes were prepared from naturally occurring ores, goethite [α -FeO(OH)], barite [BaSO₄] and limonite [Fe₂O₃·nH₂O] all obtained from El-Bahariya Oasis, Western Desert, Egypt, dolomite [CaMg(CO₃)₂] and local sand, obtained from Helwan, Egypt. The cement used in this study was Portland blast furnace slag cement CEM/B-S 42.5 N (labeled PBFSC) which is compatible with ASTM C-150 [6], from El-Aamryah Cement Company, Egypt. The additives used are fly ash (FA) class F, silica fume (SF) and super plasticizer Sikament-NN (type G) all from Sika Company, El-Obour, Egypt. All aggregates were sieved in order to get coarse aggregates in the range 5-20 mm and fine aggregates with particle size in the range 0.3-5 mm[7]. Some important physic-mechanical properties of aggregates are given in Table 1. Elemental analysis was performed for cement, additives and aggregates using XRF spectrometer as given in Table 2

2.2 Thermal gravimetric analysis (TGA)

Thermal gravimetric analysis (TGA) for limonite and goethite ores was performed using Instrument TGA Q500 V20.10 Build 36. Samples were heated from 30°C to 800°C at heating rate 10°C/min. Sample weight used in each test ranged from 5.9 mg to 6.66 mg.

2.3 Mix proportions

Three different concrete mixes were prepared using goethite, barite and dolomite as coarse aggregates. Fine aggregates used in all mixes (except Do.C mix) were (50% local sand – 50% limonite) in addition to 10% silica fume (SF) and 10% fly ash (FA) as a partial replacement from the total cement content. The three concrete mixes were prepared according to the American Concrete Institute method (ACI) of absolute volumes [8]. Mix proportions per 1m³ for the prepared concrete mixes are given in Table 3.All aggregates used in this study were used in saturated surface dry form to eliminate the effect of water absorption during mixing in order to evaluate the real effect of aggregates on concrete mixes properties [2]. The water to cement ratio for all mixes were selected to be 0.43.



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2.4 Mixing, curing and investigated specimens

The strength of concrete can be affected by several factors, in this study the main concern is with the effects of the change in concrete constituents such as using different aggregates and their effects on concrete strength and other mechanical and radiation attenuation properties of the studied concrete mixes [9]. All samples were placed at room temperature and were cured for 28 days in curing tanks under water until testing date. The curing water was changed every 7 days. The specimens prepared in this study have two geometrical shapes as given before.

2.5 Slump test

The slump test is an indication on the behavior of a compacted inverted cone of fresh concrete under the action of gravity. It measures the consistency or the wetness of concrete. It is also a simple method to evaluate the workability of the prepared concrete. The test was performed according to ASTM 143[10].

2.6 Bulk density

For concrete, expressing the density in kilograms per cubic meter is widely common when aggregates are to be actually batched by volume and here the density to be calculated is called bulk density. The bulk density for hardened concrete mixes was performed according to ECCS 203-2001code [11].

2.7 Compressive strength

The compressive strength of concrete is usually determined by applying a uniformly distributed increasing compression load on a cubic specimen using suitable testing equipment until failure. The testing equipment used for this test is 2000 KN universal machine. The test was performed using a set of three cubic specimens $[10\times10\times10~\text{cm}^3]$ for each concrete mix at curing age (28 days) and the formula used for calculating compressive strength of concrete is:

$$\sigma_c = P/A \quad Kg/cm^2$$
 (1)

Where, P is the maximum load and A is the surface area of the cubic specimen (= 100 cm²).

The compressive strength at 90 days was estimated using the following general equation:

$$(\sigma_c)_t = \frac{t}{a+bt} \times (\sigma_c)_{28}$$
 (2)



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Where: $(\sigma_c)_t$ and $(\sigma_c)_{28}$ are the compressive strength of concrete at any age t days and at 28-days, respectively. Both a and b are the coefficients which are varied for different cements and curing conditions. According to ACI 209/71 standard, this equation was obtained as a result of a study of concrete strength versus time with different types of concrete [12]. The constants a (days) and b both are related to the characteristics of the used mix design and curing conditions and they could be determined by solving two simultaneous equations at two different ages using previous experimental work which was suitable and agreeable with the mixes used in this study [2, 13].

2.8 Gamma rays attenuation measurements

In this study, cylindrical samples of dimensions 20 cm in length and 10 cm in diameter were prepared for all concrete mixes and then cut to different thicknesses. The gamma ray sources used in these tests are Ba-133, Cs-137 and Co-60 with four gamma lines and their specifications are given in Table 4.

The gamma rays of the sources were detected and measured using NaI(Tl), 3"x3", scintillation detector with multichannel analyzer using software (UCS-30) version 1.1.06 USB, Spectrum Technique 2010. To achieve a good geometry condition, the gamma source was placed inside 3 cm lead holder (source collimator) with an aperture of 3 mm in diameter, while the scintillation detector had been surrounded by blocks of lead (detector collimator) to prevent scattered gamma rays from entering the detector to get a more precise measurements [14].

Mass attenuation coefficient and half value layer (HVL) for each concrete mix were obtained from the following equations [2].

$$I_x = I_o e^{-\mu_m(\rho x)}, \ HVL(x_{1/2}) = \frac{\ln 2}{\mu}$$
 (3)

Where: I_x is the intensity after the shielding, I_o is the intensity without shielding, μ_m is the mass attenuation coefficient (cm²/g), μ is the linear attenuation coefficient and ρx is the density thickness (g/cm²).

2.9 Fast neutrons attenuation measurements

The study of the neutron attenuation properties is complicated because of the neutron wide energy spectra and their different modes of interaction with the investigated barriers (shielding materials) as well as the production of secondary gamma rays with different energies depending on the nuclear reactions occur between the incident neutrons and the different constituent atoms in the investigated barriers. This is in addition to the



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attenuation modes like inelastic scattering and radiative capture. In this study, fast neutrons and total gamma rays attenuation properties of the prepared concrete mixes were performed at normal temperature and after heating up to 450°C both with the same way by measuring the transmitted spectra of neutrons and total gamma rays produced primary from the source and during the interaction of neutrons with the different components of the samples.

The integrated fast neutrons removal macroscopic cross section (Σ) for each mix, the linear attenuation coefficient (μ) for the total gamma rays (primary in addition to secondary) and the Half value layer (HVL) for each concrete mix for both fast neutrons and total gamma rays were obtained from the following equations [1].

$$I_{x} = I_{0}e^{-\Sigma x}$$
 , $I_{x} = I_{0}e^{-\mu x}$, $HVL_{n \text{ or } \gamma} = \frac{\ln 2}{(\Sigma) \text{ or}(\mu)}$ (4)

For carrying out neutron and total gamma rays measurements, a collimated beam of ²⁵²Cf neutron source with an initial activity 100 mCi, Table 5, and organic scintillation counter with stilbene crystal with dimensions 4 cm for the diameter and 4cm for the crystal thickness and photomultiplier tube were used. Pulse shape discriminating technique based zero cross over method was used to discriminate between neutron and gamma pulses.

The distance between the front of the detector and the source is 21 cm. Source collimator is made from lead and cylindrical in shape with an outer radius of 3.5 cm and an inner one of 1.5 cm. Detector collimator is also made from lead used as slabs of thickness 10 cm surrounded the detector from all sides. These lead collimators were used to eliminate the side scattered radiation especially the highly energetic gamma rays that produced via the interaction of escaped neutrons with the high (Z) atoms in the walls of the lab thus enhancing the discrimination capability of the discriminating techniques. The geometry was fixed during all measurements. The setup and the associated electronics are displayed in Figure 1.

3. Results and Discussion

3.1 Thermal gravimetric analysis

3.1.1 TGA for limonite ore

Limonite or as called hydrated hematite Fe₂O₃.nH₂O in general is an intermediate phase during goethite transformation into hematite [15].



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Figure 2 shows two dehydration steps with average temperatures at 99°C and 276°C were observed. The first step was with temperature range (49°C - 150°C) and weight loss about 2.5%. The second step was the main with temperature range (200°C - 352°C) and weight loss about 5.5%. This curve or pattern of TGA corresponded probably to the bulk composition of Fe₂O₃.2H₂O [15]. The total weight loss was 8% of the total sample weight.

3.1.2 TGA for goethite ore

From the TGA pattern as shown in Figure 3, one main step of weight loss about 8.5 % in the temperature range (221°C - 341°C) with average temperature 281°C was observed for the ore. Generally, this TGA pattern represents goethite – hematite transformation:

$$2 \alpha \text{FeOOH} \rightarrow \text{Fe}_2\text{O}_3 + \text{H}_2\text{O}$$
 (5)

The average temperature is usually at 385°C for highly crystalline goethite and for amorphous goethite the transformation range and average temperature are always located at lower values and That indicated the amorphous nature of the used ore [16].

3.2 Slump test results

The results showed that the slump values of the prepared concrete mixes ranged from 6 cm to 10 cm as shown in Table 5, which means that all mixes were accepted from the workability point of view [10]. The highest slump value obtained with Do.C and the lowest value obtained with Go.C mix. These results can be related mainly to the differences in the water absorption ratios of the different aggregates in addition to the high specific gravity values for the used coarse aggregates except for dolomite used in Do.C mix (see Table 6).

3.3 Bulk density

The bulk densities for different concrete mixes are given in Table 7. The results obtained were predictable as the bulk density of the prepared concrete is directly proportional to the specific gravity of the aggregates used in the mix design (see Table 2); therefore, Ba.Cmix were found to have the highest value of bulk density. Ba.C and Go.C can be classified as heavy density concretes because both were found to have bulk density values more than 2600 kg/m³[17]. On the other hand Do.C approached to be heavy weight concrete as its density was 2570 kg/m³.

3.4 Compressive strength results



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3.4.1 Compressive strength values at room temperature

After curing for 28 days, the concrete mixes gained the most of their strength because of the formation of the hydration products and the domination of hydrated calcium silicate (C-S-H gel) among these hydration products. The compressive strength of all concrete mixes in this study was higher than that of traditional concrete. Using Portland blast furnace slag cement with high content was one of the reasons that lead to the concrete mixes high strength also the addition of fly ash and silica fume participated in the development of the strength of the mixes due to their good filling effect. The compressive strength of the different concrete mixes at 28 days and 90 days (estimated) is plotted in Figure 4. The results obtained revealed that Do.C was significantly higher than the other two mixes Go.C and Ba.C and the differences could be attributed to the physicmechanical properties of the coarse and fine aggregates used (see Table 2). The use of limonite as a portion of the fine aggregates was due to its good shielding properties but it had a bad effect on the strength of the mixes because its high water absorption value. The relative small crushing value and high specific gravity (4.04) had a good effect on gaining strength in Go.C mix but on the other hand its high water absorption value (13.5%) minimize this good effect. The high crushing value of barite (63.3%) had a significant bad effect on the Ba.C strength even if its low water absorption value (1.5%) and high specific gravity (4.4). Using sand only as fine aggregate beside dolomite as coarse aggregate in Do.C mix had a good effect on the strength due to their low water absorption values (0.4%, 0.7%) respectively, and the convergence in the specific gravity of both of them which had a great effect on strengthening the physical bonds between them and also with the binder.

3.4.2 Compressive strength values at elevated temperatures

The results of compressive strength at different temperatures for the prepared concretes are shown in Figure 5.

For the three prepared mixes, regular decrease was observed in the compressive strength after heating up to 100°C and that is because of vaporization of the residual mixing water. Then, in the range of (100°C - 200°C) the decrease continued for Do.C mix but the decrease was minimized for Go.C concrete mix. However, a large drop was observed in the compressive strength for Ba.C mix.

Finally, in the range of $(200^{\circ}\text{C} - 450^{\circ}\text{C})$ the rate of the decrease in compressive strength was minimized for Do.C and Ba.C mixes and at the final point the compressive strength for Ba.C concrete mix became lower than 20 MPa which is refused from the construction point of view. For Do.C mix the compressive strength value was higher than 25 MPa and that is accepted. On the other hand, large drop was noticed in the compressive strength for Go.C mix and the reason could be due to the dehydroxylation of goethite (α FeOOH)



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and transformation into ferric oxide Fe_2O_3 and H_2O . At the final point, compressive strength for Go.C mix became lower than 20 MPa (like Ba.C mix) and that is also refused from the construction point of view.

3.5 Gamma rays attenuation measurements

The main target of this part was to study gamma rays attenuation properties for the prepared mixes used in this study and to determine the best mix among the three mixes. The obtained values of mass attenuation coefficient (μ_m) and HVL for all mixes are shown in Table 8.

From the obtained results we found that Ba.C mix has the greatest linear attenuation coefficient against gamma rays and thus the corresponding minimum required HVL for these energies. The reason could be due to the high atomic number of barium (56) which is the effective element in barite that was used as coarse aggregate in Ba.C mix in addition to the high specific gravity for Ba.C mix (2.963). The priority of Ba.C is clear and obvious especially at low and intermediate energies because of the significant contribution of photoelectric effect while at higher energies Go.C gamma rays attenuation properties become so close to those of Ba.C mix because in this range of energies there is no notable contribution of photoelectric effect and the attenuation occurs mainly due to Compton scattering and pair production knowing that specific gravity of Ga.C mix (2.906) is very close to that of Ba.C mix. On the other hand, Do.C mix is in the last order because of its low density and the absence of the effective elements like barium or iron.

3.6 Fast neutrons attenuation properties at normal temperature

All values of integrated fast removal macroscopic cross section (Σ), linear attenuation coefficient (for total gamma rays) and half value layer (HVL) for the prepared mixes for fast neutrons and total gamma rays attenuation measurements are shown in Table 9.

Based on the results obtained, Go.C mix was found to have the greatest integrated fast neutrons macroscopic removal cross section (Σ) against neutrons emitted from ²⁵²Cf radioactive source at room temperature and thus the corresponding minimum required HVL while Ba.C mix came in the second order. The reason could be due to effective elastic scattering because of the high hydrogen content that contained in goethite (α FeO.OH) which was used as coarse aggregate and limonite (Fe₂O₃.nH₂O) which used as fine aggregate. In case of Ba.C mix, the high atomic number of barium (56) which is the effective element in barite that was used as coarse aggregate in Ba.C mix could be the main reason in having good attenuating properties against fast neutrons emitted from ²⁵²Cf neutron source because of the inelastic scattering of fast neutrons with barium atoms.



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On the other hand, for the total gamma rays include primary gamma photons emitted from 252 Cf neutron source and secondary gamma rays produced during interaction of neutrons with the samples, it was found that the Ba.C concrete mix has the greatest total linear attenuation coefficient (μ) at room temperature and thus the corresponding minimum required HVL.

3.7 Fast neutrons attenuation properties at elevated temperatures

After heating all mixes at 100°C, 200°C and 450°C, the results shown in Figure 6 revealed significant changes and differences between studied mixes in the priority of usage in attenuating fast neutrons at higher temperatures.

At first, after heating up to 100° C, all concrete mixes were found to be in the same order with observable decrease in the integrated fast neutrons removal macroscopic cross section (Σ) values that could be attributed to the evaporization of the residual mixing or additive water. Secondly, at 200° C, the above mentioned order was changed and the Do.C mix became in the second level after Go.C mix because this point can be considered as the start edge of the transformation of limonite (Fe₂O₃.nH₂O) to ferric oxide (Fe₂O₃) during a dehydration process and as mentioned before Do.C mix is the only concrete mix that had no limonite in its mix design. Finally, after raising heating temperature up to 450° C, the Go.C concrete mix attenuation properties against fast neutrons were deteriorated and the reason could be related to the dehydroxylation of goethite (α FeOOH) with a great weight loss and the same with limonite as mentioned earlier and that put Do.C mix in the first level at this point.

On the other hand, for total gamma rays, after heating at 100° C, 200° C and 450° C, the results shown in Figure 7 also revealed notable results. At the begining, after heating up to 100° C, there was no observable decrease in the total linear attenuation coefficient (μ) values for all concrete mixes that the evaporization of the residual mixing or additive water not make a difference in attenuating gamma rays. Then, at 200° C there was observable decrease in the total linear attenuation coefficient (μ) values for all concrete mixes and it was observed that the Ba.C concrete mix still have the highest attenuation coefficient. This observable decrease could be related to the notable weight loss for all mixes at this point. Finally, Go.C mix attenuation properties against total gamma rays deteriorated and became in the last order and the reason could be due to the transformation of goethite (α FeOOH) and limonite (Fe₂O₃.nH₂O) into ferric oxide (Fe₂O₃) and H₂O with a great weight loss and great production of voids.



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4. Conclusion

Slump values for all mixes were obtained fulfilling the requirements (higher than 5 cm) which means that all mixes were accepted from the workability point of view.

The results gained by the bulk density test showed that Ba.C mix has the highest bulk density among the investigated concrete mixes. Furthermore it should be noted that Ba.C and Go.C can be classified as heavy weight concrete because both mixes showed bulk density values more than 2600 kg/m³. On the other hand Do.C approached to be heavy weight concrete as its density was 2570 kg/m³.

The compressive strength results of such concrete mixes was higher than that of traditional concrete, such results revealed that Do.C had significantly higher compressive strength than the other two mixes (Go.C and Ba.C mixes) and the differences could be attributed to the physic-mechanical properties of the coarse and fine aggregates used.

From the chemical XRF analysis, it was obviously clear that the used aggregates (except dolomite) had high content of the desired effective elements like Iron (Fe) or Barium (Ba) or both of them, as in the case of barite ore, which was the main reason for the high density of the aggregates and thus the concrete mixes. Also using goethite and limonite with their high light element content had notable effect on fast neutron attenuation properties.

Due to the results obtained from gamma rays attenuation measurements, it was observed that Ba.C concrete mix have the greatest linear attenuation coefficient against gamma rays.

Concerning the fast neutrons and total gamma attenuation measurements and based on the results obtained, it was found that Go.C mix have the greatest integrated fast neutrons removal macroscopic cross section (Σ) emitted from ²⁵²Cf at room temperature and thus the corresponding minimum required HVL. The Ba.C concrete mix came in the second order. Ba.C mix was found to have the highest total linear attenuation coefficient (μ) at room temperature and thus the corresponding minimum required HVL.

After heating concrete mixes up to 450° C, a significant changes and differences between the investigated mixes in the priority of usage for attenuating fast neutrons were observed. At first, after heating up to 100° C, all concrete mixes was found to have the same order of observable decrease in the integrated fast neutrons removal macroscopic cross section (Σ) values. Secondly, at 200° C, the above mentioned order was changed and the Do.C mix became in the second level after the Go.C mix. Finally, after raising heating temperature up to 450° C, the Go.C mix attenuation properties against fast neutrons and total gamma were fully deteriorated.



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Table 1: Selected physical and mechanical properties of coarse and fine aggregates.

Property	Barite	Goethite	Dolomite	Limonite	Sand
Specific gravity	4.4	4.04	2.68	2.22	2.65
Water absorption, %	1.5	13.5	0.7	30.8	0.4
Crushing value, %	63.3	20	-	-	-

Table 2: Chemical analysis for cement, additives and aggregates using XRF

0-:1	PBFSC	CE.	EA	Co	Coarse aggregates			gregates
Oxides	PDFSC	PBFSC SF	FA	Barite	Goethite	Dolomite	Limonite	Sand
SiO ₂	23.33	96.81	61.13	1.16	11.2	2.24	16.3	94.84
Al ₂ O ₃	5.91	0.25	27.68	0.64	3.39	0.95	2.97	2.12
Fe ₂ O ₃	3.29	0.45	4.15	20.84	67.0	0.61	68.1	0.82
CaO	57.07	0.16	1.32	1.59	6.49	37.9	4.16	0.52
MgO	3.10	0.26	0.44	1.63	0.992	15.03	0.643	0.1
SO ₃	2.9	0.14	0.28	4.42	1.9	0.39	2.9	0.11
Cl	0.03	0.03	0.07	0.41	0.923	0.13	0.62	0.06
Na ₂ O	0.24	0.14	0.15	-	1.46	0.25	0.985	0.27
K ₂ O	0.25	0.28	0.85	0.34	1.8	0.07	0.74	0.69
TiO ₂	0.08	-	2.07	-	1.49	0.13	1.28	0.12
BaO	-	-	0.04	66.77	-	-	-	-
P_2O_5	-	0.03	0.61	0.28	0.91	0.03	0.83	0.05
Cr ₂ O ₃	-	-	-	0.14	0.416	-	-	-
MnO	-	0.05	-	1.1	0.292	-	-	-
CeO ₂	-	-	_	-	0.278	_	-	-
Sm ₂ O ₅	-	-	_	-	0.314	-	0.168	0.06
L.O.I	2.97	0.95	0.85	0.2	0.3	42.25	0.15	0.1
Total	99.17	99.55	99.64	99.52	99.15	99.94	99.84	99.86



Table 3: Mix proportions for the prepared concrete mixes

M:	Concrete ingredients, kg/m ³								
Mixe s	PBFS	Fine a	iggregates	Coarse aggregates			Additives		S.P
3	C	Sand	Limonite	Limonite Barite Goethite Dolomite				FA	3.P
Ba.C	400	270.77	226.83	1798.3	-	-	50	50	12.5
Go.C	400	270.77	226.83	-	1651.3	-	50	50	12.5
Do.C	500	554.8	-	-	-	1126.3	-	-	-

Table 4: Energies and activities of gamma radiation sources used in the measurements

Property/Source	Ba-133	Cs-137	Co-60
Gamma lines (keV)	356	662	1173 - 1332
Production Date	2014	2014	2013
Initial Activity (µCi)	10	10	6.95

Table 5: Some of the physical radiation properties of ²⁵²Cf neutron source

Physical radiation properties of ²⁵² Cf neutron source	Values	
Alpha particle energy	6.12 MeV	
Effective half life	2.65 years	
Alpha decay half life	2.73 years	
Spontaneous fission half life	85.5 years	
Gamma emission rate	$1.37 x 10^7$ photons/µgm	
Initial activity	100 mCi	
Production date	2001	
Average neutron energy	2.35 MeV	
Neutron emission rate	$1.721 \times 10^7 \text{ n/sec}$	
Fission rate	$6.2 x 10^{5}/\text{sec. } \mu\text{gm}$	



Table 6: Slump values for the concrete mixes.

Concrete mix	Ba.C	Go.C	Do.C
Slump value,	8.0	6.0	10.0

Table 7: Bulk densities of the prepared concrete mixes

Concrete mix	Ba.C	Go.C	Do.C
Bulk density, kg/m ³	2963	2906	2570

Table 8: The obtained values of mass attenuation coefficient (μ_m) and half value thickness (HVL) for the three concrete mixes.

Property	$(\mu_{\rm m})$, cm ² /g				(HVL), cm			
Gamma Energy, keV	356	662	1173	1332	356	662	1173	1332
Ba.C mix	0.112± 0.006	0.088± 0.004	0.059± 0.003	0.057 ± 0.003	2.11± 0.083	2.67± 0.078	4.01± 0.152	4.16± 0.171
Go.C mix	0.008 0.088± 0.005	0.004 0.076± 0.003	0.003 0.058± 0.004	0.003 0.056± 0.004	0.083 2.73± 0.096	3.17± 0.093	4.18± 0.193	4.34± 0.201
Do.C mix	0.096± 0.004	0.084 ± 0.004	0.060± 0.003	0.057± 0.003	2.81± 0.085	3.21± 0.097	4.50± 0.149	4.75± 0.165

Table 9: Obtained values of (Σ) , (μ) and (HVL) for the prepared mixes.

Concrete	(Σ), cm ⁻¹	(μ), cm ⁻¹	For fast neutrons	For total gamma
mix	(2), CIII	(μ), επ	HVL, cm	HVL, cm
Ba.C	0.1051±0.005	0.0918±0.002	6.61±0.205	7.53±0.127
Go.C	0.1119±0.004	0.0823±0.005	6.24±0.147	8.42±0.343
Do.C	0.1030±0.004	0.0797±0.002	6.73±0.184	8.69±0.165







Figure 1: Photo image for fast neutrons attenuation measurements including electronics associated with the stilbene detector.

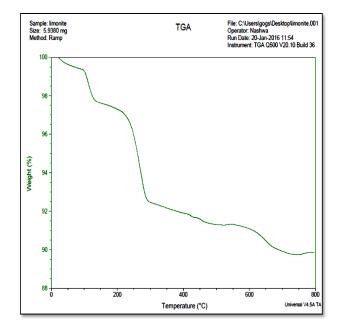


Figure 2: TGA pattern for limonite ore.



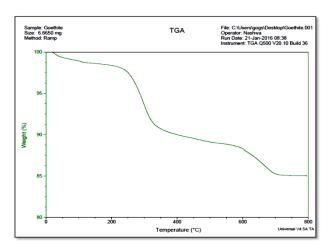


Figure 3: TGA pattern for goethite ore.

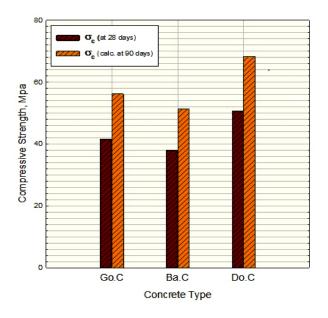


Figure 4: Compressive strength at room temperature for the prepared mixes.



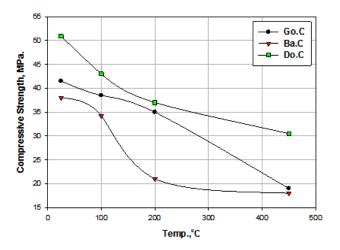


Figure 5: Compressive strength values for prepared concrete mixes at different temperatures.

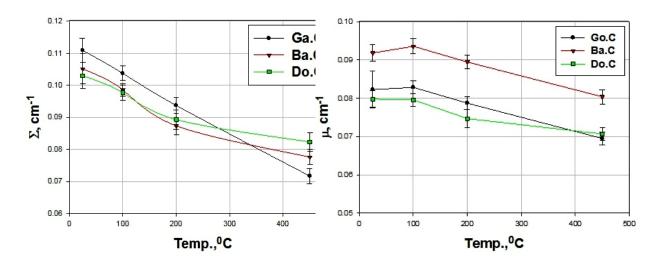


Figure 6: Fast removal cross section of neutrons for concrete mixes at different temperatures.

Figure 7: Linear attenuation coefficient of total gamma rays for the mixes at different temperatures.