SYNTHESIS OF SOME NOVEL 1, 3, 5-TRISUBSTITUTED [1,2,4]TRIAZOLE DERIVATIVES AS POTENTIAL ANTIBACTERIAL AGENTS

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> تم فى هذا العمل تحضير بعض مشتقات - الجديده لحلقة [التريازول وبعض قواعد شيف المحضره منهم تم التأكيد من التراكيب البنائيه للمركبات المستهدفة باستخدام الرنين النووى المغناطيسي التحليل الطيفى للكتلة والتحاليل الدقيقة للعناصر هذا وقد تم اختبار المركبات المستهدفة كمضادات للبكتريا (موجبة وسالبة الجرام) وقد ثبت أن هناك فاعلية لبعض المركبات المختبرة وأعطت نتائج مساوية أو متفوقة على المضادات الحيوية المرجعية المستعملة مثل الأمبسيللين وحمض الناليديكسيك

> In the present work, some new 1,3,5-trisubstituted[1,2,4]triazole derivatives and their Schiff's bases were synthesized. The chemical structure of the target compounds was confirmed by IR, ¹H-NMR, ¹³C-NMR, FAB-MS, EI-HRMS spectra and elemental analyses. The title compounds were tested for their in vitro antibacterial activity against Gram-positive and Gram-negative ones using ampicillin and nalidixic acid as reference drugs. Some of them showed antibacterial activity more significant than the reference drugs.

INTRODUCTION

The treatment of infectious diseases still remains an important and challenging problem due to the emergence of resistant strain of bacteria to major classes of antibacterial agents. The emergence of multidrug resistance of Grampositive pathogens is a problem of ever increasing significance¹. Organisms including methicillinresistant *Staphylococcus aureus* (MRSA) and *Staphylococcus*

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epidermids (MRSE), vancomycinresitant enterococci (VRE), penicillin and cephalosporin-resistant are continuously streptococci challenging chemists, physicians and patients^S. Consequently, the search of chemotherapeutic new agents constitutes real challenge facing microbiologists, Pharmacologists as well as medicinal chemists.

Currently, 1,2,4-triazole nucleus has been incorporated into a wide variety of the pharmacological activities. The following [1.2,4]triazole derivatives are applied in medicine: alprazolam (tranquilizer), estazolam (hypnotic, sedative. tranquilizer), rilmazafon (hypnotic, anxiolytic, used in the case of neurotic insomnia), benatradin trapidil (hypotensive), (diuretic). trazodon (antidepressant, anxiolytic), etoperidone (antidepressant), nefazodone (antidepressant, 5-HT2 Aantagonist), anastrozole, letrozole, vorozole (antineoplastics, nonsteroidal competitive aromatase inhibitors), ribavirin (the potent antiviral N-nucleoside), fluconazole, terconazole itraconazole. (the powerful azole antifungal agents)².

The literature survey revealed that depending on the type of substituent, the derivatives of [1,2,4]triazole have a high potential for biological activity, possessing antibacterial³⁻⁷ and antitumor⁸⁻¹¹ properties. The other ones show also anti-inflammatory¹², antihypertensive¹³, anticonvulsant¹⁴, and antiviral¹⁵.

Moreover, through the various molecules designed and synthesized

as new antibacterial agents, it was demonstrated that 1,2,4-triazoles and their derivatives could be considered as possible antimicrobial agents¹⁶⁻²².

The above mentioned facts motivate our interest to synthesize new 1,3,5-trisubstituteded [1,2,4]-triazoles and their Schiff bases as potential antibacterial agents.

EXPERIMENTAL

Chemistry

Materials and equipment

Melting points were determined an electrical melting point on apparatus (stuart Scientific, UK), and are uncorrected. Silica gel 60 F₂₅₄ (Merck) aluminum plates were used thin layer chromatography. for Dichloromethane/methanol (9:1 v/v)and Cyclohexane/EtOAc (3:2 v/v) were used as developing systems and visualized the spots were bv ultraviolet light. IR spectra were recorded on a Shimadzu spectrophotometer (IR-470) as potassium bromide discs, at the Faculty of Pharmacy, Assiut University, Egypt. ¹H-NMR spectra were performed on Varian EM-360, 60 MHz spectrometer at the Faculty of Pharmacy, Assiut University, Egypt. DMSO-d₆ was used as a solvent and the chemical shifts are (ppm) values downfield given in from TMS as internal standard. ¹H-NMR and ¹³C-NMR spectra of some compounds were taken using Bruker DPX 300 MHz instrument at Organic Chemistry institute, Bonn, Germany. DMSO-d₆ and CDCl₃ were used as solvent and the chemical shifts are

given in (ppm) values. The chemical shifts of the remaining protons of the deuterated solvents served as internal standard: 1 H: 2.49 ppm, ${}^{13}C:$ 39.7 ppm and ${}^{1}H:$ 7.26 ppm, ${}^{13}C:$ 77.36 ppm for DMSO-d₆ and CDCl₃ respectively. The EI-HRMS was obtained using El-Finnigan MAT 95XL (Thermo Organic Finnigan, Bremen) at Chemistry institute, Bonn, Germany. FAB-MS were carried out using JOEL JMS600 mass spectrometer at analyses center, Assiut University, Egypt. Elemental microanalyses were performed on a Perkin-Elmer 240 elemental analyzer at the Department of Chemistry, Faculty of Science, Assiut University, Egypt.

Compounds 9, 11, 12, 13, 15, 16, 17, 19 and 20 were prepared as reported²³.

Synthesis of methyl (4-methoxybenzoyl)thiocarbamate (10)

To a solution of 4-methoxybenzoyl chloride 2 (6.43 g, 37,84 mmol) in acetone 60 mL, potassium thiocyanate (3.68 g, 37.84 mmol, 1.0 eq.) was added under efficient stirring. The reaction mixture was refluxed at 60°C for 90 min and monitored by TLC until the start disappeared and the intermediate 4methoxybenzoyl isothiocyanate 6 was formed. Methanol (3.8 ml, 94.60 mmol, 2.5 eq.) was added dropwise to the reaction mixture. The mixture was refluxed for 8 h then cooled, filtered, washed with acetone (15 ml), and the filtrate was evaporated. The residue was purified by silica gel chromatography to afford the product as yellowish white oil.

¹H-NMR (300 MHz, CDCl₃): 9.13 (br, 1H, NH), 7.74 (d, J = 8.5 Hz, 2H), 7.88 (d, J = 8.5 Hz, 2H), 4.19 (s, 3H, OCH₃), 3.80 (s, 3H, OCH₃).

¹³C-NMR (75 MHz, CDCl₃): 190.65, 163.67, 161.88, 129.96, 124.79, 114.29, 59.52, 55.60.

EI-HRMS (m/z): calcd. for C₁₀H₁₁NO₃S 225.0460, found 225.0453 [M]⁺.

Synthesis of 3-methoxy-5-(4methoxyphenyl)-1-(4-nitrophenyl)-1*H*-[1,2,4] triazole (14)

Methyl (4-methoxybenzoyl)thiocarbamate 10 (5.41 g, 24 mmol, 1 eq.) and *p*-nitro-phenylhydrazine (3.66 g; 24 mmol, 1 eq.) were suspended in ethanol (45.0 ml). The reaction mixture was refluxed for 15 h. During the course of the reaction H₂S evolution was observed. The reaction mixture was cooled and the precipitated product was filtered washed with cold ethanol. The residue was dried in vacuum to afford the triazole derivative 14 as vellow solid.

IR (cm⁻¹): 3080 (Ar. C-H), 2980 (aliph. C-H), 1601 (C=N), 1528 (NO₂, asym.), 1340 (NO₂, sym.), 1180, 1070 (C-O), 827 (*p*-disubstituted benzene).

¹H-NMR (300 MHz, DMSO):

8.30 (d, *J* = 8.7 Hz, 2H), 7.62 (d, *J* = 8.7 Hz, 2H), 7.39 (d, *J* = 8.5 Hz, 2H), 6.98 (d, *J* = 8.5 Hz, 2H), 3.97 (s, 3H, OCH₃), 3.78 (s, 3H, OCH₃).

¹³C-NMR (75 MHz, DMSO): 168.55, 161.32, 154.18, 146.81, 143.11, 130.90, 126.19, 125.30, 119.71, 114.72, 57.12, 55.76.

EI-HRMS (m/z): calcd. for C₁₆H₁₄N₄O₄ 326.1015, found 326.1014 [M]⁺.

Synthesis of 1-(4-aminophenyl)-3methoxy-5-(4-methoxyphenyl)-1-*H*-[1,2,4] triazole (18)

3-Methoxy-5-(4-methoxyphenyl)-1-(4-nitrophenyl)-1H-[1,2,4] triazole 14 (4.50 g, 13.8 mmol) was dissolved in ethanol (100 mL) and the solution was washed with argon 3 times. added Pd/C10% (5.1)g) was portionwise and the reaction mixture was stirred in autoclave under H₂ atmosphere under 20 bar pressure for 20 h. The reaction mixture was filtered through celite pad and washed with ethanol (80 ml) and hot methanol (5x100 ml). The filtrate was evaporated to afford the desired compound 18. The compound was purified by recystallization from ethanol.

IR (cm⁻¹): 3340 (NH₂), 3060 (Ar. C-H), 2930 (aliph. C-H), 1634, 1599 (C=C, C=N), 1178, 1040 (C-O), 828 (*p*-disubstituted benzene).

¹H-NMR (300 MHz, DMSO): 7.37 (d, J = 8.6 Hz, 2H), 6.99 (d, J = 8.6 Hz, 2H), 6.92 (d, J = 8.5 Hz, 2H), 6.58 (d, J = 8.5 Hz, 2H), 5.49 (br s, 2H, NH₂), 3.90 (s, 3H, OCH₃), 3.74 (s, 3H, OCH₃).

¹³C-NMR (75 MHz, DMSO): 167.68, 160.68, 152.71, 149.91, 130.18, 127.48, 126.75, 120.44, 114.31, 114.09, 56.68, 55.67. EI-HRMS (m/z): calcd. for C₁₆H₁₆N₄O₂ 296.1273, found 296.1275 [M]⁺.

General procedure for synthesis of Schiff bases (21-44)

To a suspension of triazole derivative 17, 18, 19 or 20 (1.18 mmol) in ethanol (15 mL), an equimolar amount of the appropriate aldehyde (1.18 mmol) was added. The suspension was gently heated until a clear solution was obtained. Then few drops of glacial acetic acid were added and the solution was refluxed for 7-8 h on a water-bath. The reaction mixture was cooled to the ambient temperature and the formed precipitate was filtered and recrystallized from a mixture of ethanol and water (9:1) to yield the compounds 21-44.

1-[4-(Benzylidene-amino)phenyl]-3methoxy-5-(phenyl)-1-*H*-[1,2,4]triazole (21)

¹H-NMR (60 MHz, DMSO): 9.1 (s, 1H, N=<u>CH</u>), 7.9-8.6 (m, 14H, Ar H), 4.4 (s, 3H, OCH₃).

1-{4-[(4-Bromobenzylidene)amino]phenyl}-3-methoxy-5-(phenyl)-1-*H*-[1,2,4] triazole (22)

IR (cm⁻¹): 3010 (Ar. C-H), 2940 (aliph. C-H), 1612, 1591 (C=C, C=N), 1173 (C-O), 832 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.2 (s, 1H, N=<u>CH</u>), 7.9-8.6 (m, 13H, Ar H), 4.45 (s, 3H, OCH₃).

1-{4-[(4-Chlorobenzylidene)amino]phenyl}-3-methoxy-5-(phenyl)-1-*H*-[1,2,4] triazole (23)

IR (cm⁻¹): 3090 (Ar. C-H), 2940 (aliph. C-H), 1612, 1583 (C=C, C=N), 1078 (C-O), 832 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.2 (s, 1H, N=<u>CH</u>), 8.5 (d, J = 8.6 Hz, 2H, Ar H), 7.8-8.3 (m, 11H, Ar H), 4.4 (s, 3H, OCH₃).

3-Methoxy-1-{4-[(4-methoxybenzylidene)amino]phenyl}-**5-**(phenyl)-**1-***H*-**[1,2,4]** triazole (24)

IR (cm⁻¹): 3090 (Ar. C-H), 2935 (aliph. C-H), 1621, 1593 (C=C, C=N), 1156, 1101 (C-O), 825 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.0 (s, 1H, N=<u>CH</u>), 8.5 (d, J = 8.6 Hz, 2H, Ar H), 7.8-8.2 (m, 9H, Ar H), 7.5 (d, J = 8.5 Hz, 2H, Ar H), 4.4 (s, 3H, OCH₃), 4.2 (s, 3H, OCH₃).

3-Methoxy-1-{4-[(4-N,N-dimethylaminobenzylidene)amino]phenyl}-5-(phenyl)-1-*H*-[1,2,4] triazole (25)

IR (cm⁻¹): 3065 (Ar. C-H), 2905 (aliph. C-H), 1598, 1570 (C=C, C=N), 1178 (C-O), 830 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.1 (s, 1H, N=<u>CH</u>), 8.6 (d, J = 8.5 Hz, 2H, Ar H), 7.8-8.3 (m, 9H, Ar H), 7.3 (d, J = 8.6 Hz, 2H, Ar H), 4.4 (s, 3H, OCH₃), 3.3 (s, 6H, N(<u>CH₃)</u>₂.

FAB-MS m/z, (%): 398 [M+1]⁺ (5%), 420 [M+ Na] ⁺ (4%), 93.08 (100%).

3-Methoxy-1-{4-[(4-nitrobenzylidene)amino]phenyl}-5-(phenyl)-1-*H*-[1,2,4] triazole (26)

IR (cm⁻¹): 3030 (Ar. C-H), 2985 (aliph. C-H), 1616, 1586 (C=C, C=N), 1527 (NO₂, asym.), 1333 (NO₂, sym.), 1179, 1099 (C-O), 835 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.6 (s, 1H, N=<u>CH</u>), 9.2 (d, J = 8.6 Hz, 2H, Ar H), 8.4 (d, J = 8.5 Hz, 2H, Ar H), 7.9-8.2 (m, 9H, Ar H), 4.4 (s, 3H, OCH₃).

FAB-MS m/z, (%): 400.17 [M+1]⁺ (13%), 57.05 (100%).

1-{4-[(4-Hydroxybenzylidene)amino]phenyl}-3-methoxy-5-(phenyl)-1-*H*-[1,2,4] triazole (27)

IR (cm⁻¹): 3420 (OH), 3080 (Ar. C-H), 2970 (aliph. C-H), 1611, 1580 (C=C, C=N), 1170 (C-O), 823 (*p*disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 11.2 (br s, 1H, OH), 9.3 (s, 1H, N=<u>CH</u>), 8.5 (d, J = 8.5 Hz, 2H, Ar H), 7.8-8.2 (m, 9H, Ar H), 7.5 (d, J = 8.5 Hz, 2H, Ar H), 4.3 (s, 3H, OCH₃).

FAB-MS m/z, (%): 371.20 [M+1]⁺ (6%), 50.99 (100%).

1-{4-[(4-Carboxybenzylidene)amino]phenyl}-3-methoxy-5-(phenyl)-1-H-[1,2,4] triazole (28)

IR (cm⁻¹): 3470 (OH), 3030 (Ar. C-H), 2930 (aliph. C-H), 1679 (C=O), 1610 (C=N), 1162 (C-O), 835 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 11.1 (br s, 1H, COOH), 9.4 (s, 1H, N=<u>CH</u>), 8.5-8.7 (m, 5H, Ar H), 7.8-8.1 (m, 8H, Ar H), 4.3 (s, 3H, OCH₃).

1-{4-[(4-Carboxymethyloxybenzylidene)amino]phenyl}-3-methoxy-5-(phenyl)-1-*H*-[1,2,4] triazole (29)

IR (cm⁻¹): 3440 (OH), 3035 (Ar. C-H), 2920 (aliph. C-H), 1730 (C=O), 1614, 1502 (C=C, C=N), 1185, 1085 (C-O), 837 (*p*disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 10.8 (br s, 1H, COOH), 9.3 (s, 1H, N=<u>CH</u>), 8.5 (d, J = 8.6 Hz, 2H, Ar H), 7.5-8.1 (m, 11H, Ar H), 5.2 (s, 2H, O-<u>CH₂</u>-), 4.3 (s, 3H, OCH₃).

FAB-MS m/z, (%): 429.02 [M+1]⁺ (3%), 50.99 (100%).

1-{4-[(4-Bromobenzylidene)amino]phenyl}-3-methoxy-5-(4-methoxyphenyl)-1-*H*-[1,2,4] triazole (30)

IR (cm⁻¹): 3060 (Ar. C-H), 2945 (aliph. C-H), 1601, 1529 (C=C, C=N), 1179 (C-O), 830 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.4 (s, 1H, N=<u>CH</u>), 8.3-8.7 (m, 4H, Ar H), 8.0-8.2 (m, 4H, Ar H), 7.5-7.8 (m, 4H, Ar H), 4.4 (s, 3H, OCH₃), 4.0 (s, 3H, OCH₃).

1-{4-[(4-Carboxybenzylidene)amino]phenyl}-3-methoxy-5-(4methoxyphenyl)-1-*H*-[1,2,4] triazole (31)

IR (cm⁻¹): 3435(OH), 3035 (Ar. C-H), 2920 (aliph. C-H), 1675 (C=O), 1605, 1564 (C=C, C=N), 1180, 1105 (C-O), 835 (*p*disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.4 (s, 1H, N=<u>CH</u>), 8.6-8.8 (m, 4H, Ar H), 7.8-8.0 (m, 4H, Ar H), 7.1-7.7 (m, 4H, Ar H), 4.3 (s, 3H, OCH₃), 1.5 (s, 3H, OCH₃).

FAB-MS m/z, (%): 428.18[M]⁺ (12%), 93.10 (100%).

1-[4-(Benzylidene)amino)phenyl]-5-(4-*tert*.Butylphenyl)-3-methoxy-1-*H*-[1,2,4] triazole (32)

¹H-NMR (60 MHz, DMSO): 9.3 (s, 1H, N=<u>CH</u>), 8.3-8.6 (m, 4H, Ar H), 7.8-8.2 (m, 9H, Ar H), 4.4 (s, 3H, OCH₃), 1.5 (s, 9H, C(CH₃)₃).

1-{4-[(4-Bromobenzylidene)amino]phenyl}-5-(4-*tert*.butylphenyl)-3methoxy-1-*H*-[1,2,4] triazole (33)

IR (cm⁻¹): 3055 (Ar. C-H), 2930 (aliph. C-H), 1612, 1571 (C=C, C=N), 1160, 1047 (C-O), 830 (*p*disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.2 (s, 1H, N=<u>CH</u>), 8.3-8.6 (m, 4H, Ar H), 7.8-8.2 (m, 8H, Ar H), 4.4 (s, 3H, OCH₃), 1.5 (s, 9H, C(CH₃)₃).

5-(4-*tert*.Butylphenyl)-1-{4-[(4chlorobenzylidene)amino]phenyl}-3-methoxy-1-*H*-[1,2,4] triazole (34)

IR (cm⁻¹): 3040 (Ar. C-H), 2930 (aliph. C-H), 1615, 1582 (C=C, C=N), 1184, 1081 (C-O), 830 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.2 (s, 1H, N=<u>CH</u>), 8.5 (d, J = 8.5 Hz, 2H, Ar H), 7.7-8.2 (m, 10H, Ar H), 4.4 (s, 3H, OCH₃), 1.4 (s, 9H, C(CH₃)₃).

FAB-MS m/z, (%): 445.44 [M+1]⁺ (5%), 93.04 (100%).

5-(4-*tert*.Butylphenyl)-3-methoxy-1-{4-[(4-methoxybenzylidene)amino]phenyl}-1-*H*-[1,2,4] triazole (35)

IR (cm⁻¹): 3060 (Ar. C-H), 2925 (aliph. C-H), 1615, 1581 (C=C,

C=N), 1160 (C-O), 835 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.1 (s, 1H, N=<u>CH</u>), 8.4-8.6 (m, 2H, Ar H), 7.6-8.0 (m, 6H, Ar H), 7.1-7.5 (m, 4H, Ar H), 4.4 (s, 3H, OCH₃), 4.2 (s, 3H, OCH₃), 1.4 (s, 9H, C(CH₃)₃).

5-(4-*tert*.Butylphenyl)-3-methoxy-1-{4-[(4-N,N-dimethylaminobenzylidene)-amino]phenyl}-1-*H*-[1,2,4] triazole (36)

IR (cm⁻¹): 3080 (Ar. C-H), 2925 (aliph. C-H), 1583 (C=N), 1161 (C-O), 815 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.1 (s, 1H, N=<u>CH</u>), 8.5 (d, J = 8.5 Hz, 2H, Ar H), 7.9-8.2 (m, 8H, Ar H), 7.5 (d, J = 8.5 Hz, 2H, Ar H), 4.4 (s, 3H, OCH₃), 3.3 (s, 6H, N<u>(CH₃)</u>₂), 1.4 (s, 9H, C(CH₃)₃).

FAB-MS m/z, (%): 453.38 [M]⁺ (13%), 454.43 [M+1]⁺ (80%), 455.21 [M+2]⁺ (18%), 323.24 (100%).

5-(4-*tert*.Butylphenyl)-3-methoxy-1-{4-[(4-nitrobenzylidene)amino]phenyl}-1-*H*-[1,2,4] triazole (37)

IR (cm⁻¹): 3075 (Ar. C-H), 2930 (aliph. C-H), 1616, 1589 (C=C, C=N), 1530 (NO₂, asym.), 1330 (NO₂, sym.), 1162, 1048 (C-O), 825 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.5 (s, 1H, N=<u>CH</u>), 8.7-9.1 (m, 4H, Ar H), 7.8-8.2 (m, 8H, Ar H), 4.3 (s, 3H, OCH₃), 1.4 (s, 9H, C(CH₃)₃).

5-(4-*tert*.Butylphenyl)-1-{4-[(4hydroxybenzylidene)amino]phenyl} -3-methoxy-1-*H*-[1,2,4] triazole (38)

IR (cm⁻¹): 3395 (OH), 3070 (Ar. C-H), 2930 (aliph. C-H), 1601, 1562

(C=C, C=N), 1154 (C-O), 830 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 10.9 (s, 1H, OH), 9.2 (s, 1H, N=<u>CH</u>), 8.5 (d, J = 8.6 Hz, 2H, Ar H), 7.8-8.2 (m, 8H, Ar H), 7.5 (d, J = 8.6 Hz, 2H, Ar H),4.3 (s, 3H, OCH₃), 1.4 (s, 9H, C(CH₃)₃).

FAB-MS m/z, (%): 427.21 $[M+1]^+$ (18%), 323.09 (100%).

5-(4-*tert*.Butylphenyl)-1-{4-[(2hydroxybenzylidene)amino]phenyl}

-3-methoxy-1-*H*-[1,2,4] triazole (39) IR (cm⁻¹): 3390 (OH), 3035 (Ar. C-H) 2930 (aliph C-H) 1609

C-H), 2930 (aliph. C-H), 1609 (C=N), 1181, 1045 (C-O), 830 (*p*disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 12.7 (s, 1H, OH), 9.6 (s, 1H, N=<u>CH</u>), 7.9-8.4 (m, 10H, Ar H), 7.4-7.7 (m, 2H, Ar H), 4.3 (s, 3H, OCH₃), 1.4 (s, 9H, C(CH₃)₃).

5-(4-*tert*.Butylphenyl)-1-{4-[(4-

carboxybenzylidene)amino]phenyl} -3-methoxy-1-*H*-[1,2,4] triazole (40)

IR (cm⁻¹): 3445 (OH), 3090 (Ar. C-H), 2935 (aliph. C-H), 1679 (C=O), 1613, 1570 (C=C, C=N), 1185 (C-O), 835 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 10.9 (s, 1H, COOH), 9.5 (s, 1H, N=<u>CH</u>), 8.4-8.9 (m, 4H, Ar H), 7.8-8.3 (m, 8H, Ar H), 4.3 (s, 3H, OCH₃), 1.4 (s, 9H, C(CH₃)₃).

FAB-MS m/z, (%): 455.42 [M+1]⁺ (21%), 456.47 [M+2]⁺ (7%), 323.30 (100%).

5-(4-*tert*.Butylphenyl)-1-{4-[(4carboxymethyloxybenzylidene)amino]phenyl}-3-methoxy-1-*H*-[1,2,4] triazole (41)

¹H-NMR (60 MHz, DMSO): 11.0 (s, 1H, COOH), 9.3 (s, 1H, N=<u>CH</u>), 8.4-8.6 (m, 2H, Ar H), 7.8-8.3 (m, 6H, Ar H), 7.5-7.7 (m, 2H, Ar H), 7.1-7.3 (m, 2H, Ar H), 5.2 (s, 2H, O-<u>CH₂</u>-), 4.3 (s, 3H, OCH₃), 1.4 (s, 9H, C(CH₃)₃).

1-{4-[(4-Bromobenzylidene)amino]phenyl}-3-ethoxy-5-(2,5-diflourophenyl)-1-*H*-[1,2,4] triazole (42)

IR (cm⁻¹): 3080 (Ar. C-H), 2925 (aliph. C-H), 1579, 1540 (C=C, C=N), 1149 (C-O), 825 (*p*disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.3 (s, 1H, N=<u>CH</u>), 8.3-8.6 (m, 4H, Ar H), 7.7-8.2 (m, 7H, Ar H), 4.6 (t, J =7.3 Hz, 2H, -O<u>CH₂CH₃</u>), 1.5 (t, J =7.3 Hz, 3H, -OCH₂<u>CH₃</u>).

FAB-MS m/z, (%): 483.28 $[M+1]^+$ (7.6%), 484.32 $[M+2]^+$ (3%), 317.23 (100%).

1-{4-[(4-Carboxybenzylidene)amino]phenyl}-3-ethoxy-5-(2,5diflourophenyl)-1-*H*-[1,2,4] triazole (43)

IR (cm⁻¹): 3440 (OH), 3085 (Ar. C-H), 2960 (aliph. C-H), 1683 (C=O), 1615, 1571 (C=C, C=N), 1179 (C-O), 835 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.3 (s, 1H, N=<u>CH</u>), 8.4-8.7 (m, 5H, Ar H), 7.6-8.1 (m, 6H, Ar H), 4.6 (t, J =7.3 Hz, 2H, -O<u>CH₂CH₃</u>), 1.5 (t, J =7.3 Hz, 3H, -OCH₂<u>CH₃</u>).

3-Ethoxy-5-(2,5-diflourophenyl)-1-{4-[(4-nitrobenzylidene)amino]phenyl}-1-*H*-[1,2,4] triazole (44)

IR (cm⁻¹): 3065 (Ar. C-H), 2960 (aliph. C-H), 1619, 1570 (C=C, C=N), 1530 (NO₂, asym.), 1341 (NO₂, sym.), 1185 (C-O), 830 (*p*-disubstituted benzene).

¹H-NMR (60 MHz, DMSO): 9.6 (s, 1H, N=<u>CH</u>), 9.1 (d, J = 8.5 Hz, 2H, Ar H), 8.0-8.6 (m, 9H, Ar H), 7.6-8.1 (m, 6H, Ar H), 4.6 (t, J = 7.4Hz, 2H, -O<u>CH₂CH₃</u>), 1.5 (t, J = 7.4Hz, 3H, -OCH₂<u>CH₃</u>).

Antibacetrial Activity

The antibacterial activity of all the target compounds 17-44 was investigated in vitro at the department of microbiology and immunology, facultv of medicine. Assiut University. The title compounds were tested against methicillin resistant Staphylococcus aureus (MRSA), Bacillus cereus, Escherichia coli, and Klebsiella pneumoniae (clinical isolates obtained from Infection Control Unit, Assiut University Hospital, Faculty of Medicine, Assiut University) using agar cup diffusion method24&25 for susceptibility screening, and two-fold dilution method²⁵ for MIC determination. Ampicillin was used as a reference antibiotic, and DMSO was used as a solvent control.

Agar cup diffusion method

38 Gm of Mueller-Hinton agar medium (MH) (Hi-Media, M 001) were added to 1 L of distilled water, heated to boiling to dissolve the

ingredients completely, and sterilized by autoclaving at 121°C for 30 minutes. High density inocula were made by diluting 3-5 well isolated colonies grown overnight on selective media in 5 mL of distilled water to prepare a suspension equivalent in density to 0.5 McFarland Barium Sulfate standard unit with average turbidity 10⁸ CFU mL²⁵. The sterile petri dishes were seeded with 100 µL of the microorganism; a specified amount of the molten MH agar medium (45-50°C) was poured into the seeded Petri dishes to give a depth of 3-4 mm and allowed to solidify. Cylindrical plugs were removed from the agar using sterile cork borer. 100 µL of the tested compounds (20 mg/mL in DMSO), the blank solvent, and ampicillin sodium (20 mg/mL in DMSO) were added to the wells in triplicate. The seeded plates were incubated at 37°C for 24 h then the average diameters of the inhibition zones were measured in millimeters.

Minimum inhibitory concentration

The MIC values were determined using two fold-dilution method²⁶ for compounds having moderate to strong antibacterial activity. The squares of inhibition zone diameters were plotted against log concentrations of the tested compounds, extrapolation of the resulting straight line to intersect with log concentration scale in the curve corresponded to log MIC, and MIC was obtained as antilog²⁷.

RESULTS AND DISCUSSION

Chemistry

Methyl (4-methoxybenzoyl)thiocarbamate 10 was prepared through the reaction of 4-methoxybenzoyl chloride 2 with potassium thiocyanate in acetone to give 4-methoxybenzoyl isothiocyanate intermediate 6. followed by direct addition of methanol to the reaction mixture The structure (Scheme 1). of compound 10 was verified by ¹H-NMR, ¹³C-NMR and EI-HRMS. The ¹H-NMR spectrum of this compound was characterized by the presence of amide NH at 9.13 ppm, two methoxy groups at 4.19 and 3.80 ppm and its ¹³C-NMR showed C=O and C=S at 163.67 and 190.65 ppm respectively and the methoxy groups appeared at 59.52 and 55.60 ppm. EI-HRMS 10 showed a molecular ion at (m/z)225.0460 (its calculated mass = 225.0453, see exp. part).

3-Methoxy-5-(4-methoxyphenyl)-1-(4-nitrophenyl)-1*H*-[1,2,4] triazole 14 was prepared in analogy to the reported procedure²⁸ by reacting methyl (4-methoxybenzoyl)thiocarbamate **10** with *p*-nitrophenylhydrazine in ethanol, the reaction was ended after H₂S evolution was ceased completely (Scheme 1). The ¹H-NMR and ¹³C-NMR spectra revealed the disappearance of the amide NH and both C=O and C=S signals and appearance of new aromatic moiety which introduced to the triazole ring. Its EI-HRMS revealed (m/z) at $326.1014 [M]^+$ and the calculated one



1,5 R = H, **2,6** R= 4-OCH₃, **3,7** R = 4-C(CH₃)₃, **4,8** R = 2,5-di(F), **9,13,17** R = H, R1 = CH₃, **10,14,18** R = 4-OCH₃, R1 = CH₃, **11,15,19** R = 4- C(CH₃)₃, R1 = CH₃, **12,16,20** R = 2,5-Di(F), R1 = CH₂CH₃.

Compd.	R	R1	R2	Compd.	R	R1	R2
NO				NO			
21	Н	CH_3	Н	33	$4-C(CH_3)_3$	CH_3	4-Br
22	Н	CH_3	4-Br	34	4-C(CH ₃) ₃	CH_3	4-Cl
23	Н	CH_3	4-Cl	35	4-C(CH ₃) ₃	CH_3	4-OCH ₃
24	Н	CH ₃	4-OCH ₃	36	4-C(CH ₃) ₃	CH ₃	4-N(CH ₃) ₂
25	Н	CH_3	4-N(CH ₃) ₂	37	4-C(CH ₃) ₃	CH_3	NO_2
26	Н	CH_3	NO_2	38	4-C(CH ₃) ₃	CH_3	4-OH
27	Н	CH ₃	4-OH	39	4-C(CH ₃) ₃	CH ₃	2-OH
28	Н	CH_3	4-COOH	40	4-C(CH ₃) ₃	CH ₃	4-COOH
29	Н	CH_3	4-O-CH ₂ COOH	41	4-C(CH ₃) ₃	CH_3	4-O-CH ₂ COOH
30	4-OCH ₃	CH_3	4-Br	42	2,5-di(F)	CH ₂ CH ₃	4-Br
31	4-OCH ₃	CH ₃	4-COOH	43	2,5-di(F)	CH ₂ CH ₃	4-COOH
32	4-C(CH ₃) ₃	CH_3	Н	44	2,5-di(F)	CH ₂ CH ₃	4-NO ₂

Scheme 1: Synthesis of 1,3,5-trisbstituted-1H-[1,2,4]-triazole derivatives (21-44).

is 326.1015 and this confirmed the structure of compound **14**.

1-(4-aminophenyl)-3-methoxy-5-(4-methoxyphenyl)-1-*H*-[1,2,4] triazole **18** was prepared by catalytic reduction of the nitro derivative **14** using Pd/C10% catalyst (Scheme 1). The ¹H-NMR of compound 18 showed broad signal at 5.49 ppm which is a strong evidence for the formed NH₂. Its structure was confirmed by ¹³C-NMR and EI-HRMS which showed (m/z) at 296.1275 [M]⁺ and the calculated one is 296.1273.

Schiff bases of 1,3,5-trisubstituted [1,2,4] triazole **21-44** were prepared by the reaction of triazole derivatives

17-20 with equimolar amounts of the appropriate (un)substituted benzaldehydes in ethanol in presence of few drops of glacial acetic acid (Scheme 1). Comparative study of the ¹H-NMR of aminotriazole derivatives 17, 18. 19 or 20 and their Schiff bases 21-44. revealed easily the disappearance of the NH₂ group signal and appearance of N=CH signals at 9.1-9.6 ppm in addition to the introduced aromatic moiety. The structures of formed Schiff bases were also confirmed by elemental analyses (Table 1) and compounds 25-27, 29, 31, 34, 36, 38, 40 and 42 were further confirmed using FAB-MS (exp. part).

Elemental Analyses, Yield Compd. MW Formula Mp [°C] Calc./Found No. [%] С Η Ν 10 $C_{10}H_{11}NO_{3}S$ 225.26 HRMS* 91 Oil $C_{16}H_{14}N_4O_4$ 14 326.31 HRMS* 68 152-154 18 $C_{16}H_{16}N_4O_2$ 296.32 HRMS* 71 204-206 21 $C_{22}H_{18}N_4O$ 354.41 74.56 15.81 78 128-130 5.12 74.22 5.26 15.69 22 C22H17BrN4O 433.31 60.98 3.95 12.93 88 139-141 60.62 3.53 12.85 23 C₂₂H₁₇ClN₄O 67.95 4.41 14.41 85 388.86 134-136 67.62 4.56 14.44 5.24 14.57 384.44 71.86 24 $C_{23}H_{20}N_4O_2$ 86 118-120 71.90 5.39 14.39 25** C₂₄H₂₃N₅O. 406.49 70.91 5.96 17.23 83 178-180 70.57 5.92 17.09 0.5 H₂O 26** C₂₂H₁₇N₅O₃. 408.42 64.69 4.45 17.15 80 210-212 $0.5 H_2O$ 64.74 4.62 17.15 27** 79 C₂₂H₁₈N₄O₂. 379.42 69.64 5.06 14.77 195-197 69.28 5.26 14.90 $0.5 H_2O$

Table 1: Yields, melting points and analytical data of the new compounds.

Compd.	Formula	MW	Elemental Analyses,			Yield	Mp [°C]
28	$C_{22}H_{10}N_4O_2$	398.42	69 34 4 55 14 0e		14 06	72	233-235
20	02311181 (403	570.12	69.38	4.59	13.96	, 2	235 235
29**	$C_{24}H_{20}N_4O_4.$	437.46	65.89	4.85	12.81	71	200-202
	0.5 H ₂ O		65.70	4.90	12.75		
30	$C_{23}H_{19}BrN_4O_2$	463.34	59.62	4.13	12.09	68	154-156
			59.65	4.36	12.18		
31**	$C_{24}H_{20}N_4O_4$	428.45	67.28	4.71	13.08	67	220-222
			66.81	4.92	12.89		
32	$C_{26}H_{26}N_4O$	410.52	76.07	6.38	13.65	72	141-143
			75.89	6.59	13.34		
33	C ₂₆ H ₂₅ BrN ₄ O	489.42	63.81	5.15	11.45	81	179-181
			63.52	5.35	11.27		
34**	$\mathrm{C}_{26}\mathrm{H}_{25}\mathrm{ClN}_4\mathrm{O}.$	453.97	68.78	5.78	12.34	70	180-182
	0.5 H ₂ O		69.12	5.75	12.20		
35	$C_{27}H_{28}N_4O_2$	440.55	73.61	6.41	12.72	65	130-132
			73.23	6.37	12.51		
36**	$C_{28}H_{31}N_5O.$	462.60	72.70	6.99	15.14	66	148-150
	$0.5 H_2O$	155.50	72.86	6.73	14.94	70	200.210
37	$C_{26}H_{25}N_5O_3$	455.52	68.56	5.53	15.37	72	208-210
20**		425.52	67.90	5.48	15.26	74	220, 222
38**	$C_{26}H_{26}N_4O_2.$	435.53	/1./0	6.26	12.86	/4	220-222
20	$0.5 H_2 O$	126.52	72.03	0.30	12.89	70	157 150
39	$C_{26}H_{26}N_4O_2$	420.32	73.22 72.55	0.14	13.14	70	137-139
/0**	CHNO	162 51	60.06	5.99	12.00	60	220 222
40	$C_{27}H_{26}H_4O_3$.	403.34	69.90	5.80	12.09	09	230-232
41	CooHeeNLO	184 56	69.71	5.82	11.56	62	210-212
71	$C_{2811_{281}}$	404.50	69.11	5.82	11.50	02	210-212
42**	CaaHuzBrFaN(O	483 32	57.16	3 55	11.15	68	114-116
T#	C2311/D112140	105.52	56.75	3.66	11.53	00	117 110
43	$C_{24}H_{18}F_{2}N_{4}O_{3}$	457.44	63.01	4.20	12.25	80	191-193
	0.5 H ₂ O		63.35	4.37	12.39	20	
44	$C_{23}H_{17}F_{2}N_{5}O_{3}$	449.42	61.47	3.81	15.58	79	149-151
			61.59	4.33	15.39		

Table 1: Continued.

Compounds 14, 18, 20 were confirmed by HRMS (see the experimental part).
Compounds 25-27, 29, 31, 34, 36, 38, 40 and 42 were further confirmed by FAB-MS (see the experimental part).

Antibacterial activity

The newly synthesized compounds 17-43 were tested for their in vitro antibacterial activity against methicillin resistant Staphylococcus aureus (MRSA) and Bacillus cereus as representatives of Gram-positive strains and Escherichia coli and Klebsiella pneumoniae as representatives of Gram-negative ones using ampicillin and nalidixic acid as reference drugs. The results revealed that most of the newlv synthesized compounds exhibited promising antibacterial activity against all the test organisms (Table 2). Thus compounds 17-20 which were substituted by 4aminophenyl at the 1-position of [1,2,4]triazoles showed antibacterial activity similar to nalidixic acid and higher than that of ampicillin against all tested strains. Among the synthesized Schiff bases, compound 40 was the most potent one against all strains. Its antibacterial potency was superior to that of ampicillin and to that of nalidixic acid against

methicillin resistant *Staphylococcus aureus* (MRSA), *Bacillus cereus* and *Escherichia coli*.

Compounds 22, 25, 28, 31, 33, 36, 37, and 43 exhibited pronounced activity against all strains.

On the other hand, screening results revealed that in some cases, conversion of the amino derivatives to Schiff bases led to loss of antibacterial activity for example compounds 23, 27, 34, 35, 38 and 39.

Moreover, compounds **21**, **24**, **29**, **32**, **41**, **42** and **44** did not exhibit enhanced activity compared to the parent amino derivatives.

Also, it has been observed that introduction of carboxyl group at 4position to the 1-phenyl substituent regardless of the substituents of the 5phenyl moiety produced higly active compounds **28**, **31**, **40** and **43**. Replacement of this carboxyl group by oxymethylcarboxyl (O-CH₂-COOH) decreased the activity.

In general anticipation of SAR can not be attained.

Compd. No.	MRSA		Bacillus cereus		Klebsiella pneumoniae		E.coli	
	Inhibition zone (mm)	MIC µM/ml	Inhibition zone (mm)	MIC µM/ml	Inhibition zone (mm)	MIC µM/ml	Inhibition zone (mm)	MIC µM/ml
17	32	25	32	23	32	25	32	25
18	27	30	28	25	23	60	24	53
19	31	30	27	30	26	39.1	25	50
20	28	25	29	25	25	50	26	39.1
21	16		16		15		14	
22	31	20	23	65	16	130	16	135
23	-ve		-ve		-ve		-ve	
24	17		16		15		15	
25	30	20	20	70	17	125	17	125
26	17		20		-ve		-ve	
27	-ve		-ve		-ve		-ve	
28	28	30	25	50	27	31.6	18	79
29	19		18		20		20	
30	18		-ve		25		18	
31	32	20	21	66	16	132	16	135
32	15		17		16		15	
33]	29	25	21	60	18	79	18	79
34	-ve		-ve		19		-ve	
35	-ve		16		-ve		-ve	
36	27	30	22	60	20	70	20	70
37	27	30	28	30	26	40	17	125
38	17		-ve		-ve		-ve	
39	-ve		-ve		-ve		-ve	
40	33	20	30	25	31	30	33	20
41	19		22		21		26	
42	-ve		20		21		17	
43	29	30	28	31	25	42	17	127
44	-ve		-ve		22		17	
Ampicillin	20	69	22	60	23	50	20	70
Nalidixic acid	27	30	20	67	30	20	28	25

Table 2: Antibacterial activity of the tested compounds (expressed as the inhibition zone diameter and as MIC $\mu M/mL).$

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