# MINOR CONSTITUENTS FROM RUBIA CORDIFOLIA L. ROOT

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بإعادة فحص جزء الكلوروفورم لخلاصة الميثانول كلوروفورم (:) للجذور الجافة لنبات روبيا كوردفوليا ل تم فصل مركب جديد من نوع ثنائى النافثو هيدروكينونات () بالإضافة إلى أربعة مركبات معروفة - بيتا فريديلينول ()، حمض الأتراريك ()، حمض الفانيليك () دى - أ ميثوكسى كيروانوزيتول () وقد تم التعرف على المركبات المفصولة باستخدام الطرق الفيزيائية و الكيميائية و الطيفية المختلفة.

Reinvestigation of the chloroform soluble fraction of the chloroform- methanol (1:1) extract of the dried roots of Rubia cordifolia L. using different chromatographic techniques led to isolation of one new naphthohydroquinone dimer (1) and four known compounds identified as  $3-\beta$ -friedelinol (2), atraric acid (3), vanillic acid (4) and D-3-O-methoxy-chiro-inositol (5). The identification of the isolated compounds was carried out using different physical, chemical and spectral methods of analysis.

#### **INTRODUCTION**

Rubia cordifolia L. (Rubiaceae), is an important medicinal plant used for treatment of such as tumors<sup>1-10</sup>. various diseases inflammations, urinary disorders, stress and as antimicrobial<sup>10-21</sup>, hepatoprotective<sup>7,22&23</sup>, radio-protective<sup>24</sup>, hypoglycemic<sup>7,25&26</sup> and antipsychotic<sup>27</sup>. It has been listed officially as a herbal medicine in Chinese Pharmacopeia for treatment of arthritis, dysmenorrhea, hematorrhea and hemostasis which are free radical related diseases<sup>7,23,24&28-30</sup>. It is used also as natural food colourant and as natural hair and colour dve<sup>7</sup>.

Many classes of compounds were isolated from the roots of *R. cordifolia* L.<sup>2,7,23&31-45</sup>, besides large numbers of cyclic hexapeptides<sup>1-4&23</sup>. In the current study, a new naphthohydroquinone dimer (1) and four known compounds (2-5) were isolated and identified.

## MATERIALS AND METHODS

## Experimental

Melting points (uncorrected) were determined by electrothermal digital model 550. IR spectrum was recorded in KBr using Unicam SP 1025 spectrophotometer. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra were recorded in CDCl<sub>3</sub> (or  $C_5D_5N$  or  $CD_3OD$ ) at 400 MHz and 100 MHz respectively, with TMS as an internal standard on Bruker AM-400 spectrometer. Mass spectra were carried out using Hitachi M-80 spectrometer. For CC, silica gel (E. Merck, Germany 70-230 mesh) and reversed phase (RP-18, ODS) were used. Precoated silica gel  $G_{60}$  F<sub>254</sub> and RP-C18 F<sub>254</sub> S (E-Merck) were used for TLC and 10% H<sub>2</sub>SO<sub>4</sub> was used as spraying reagent followed by heating for 10 minutes. Final purification was carried out using Moderate Pressure Liquid Chromatography (MPLC) (22 mm, i.d. x 30 cm, Kusano Scientific co., Tokyo, Japan).

## **Plant material**

The dried roots of *Rubia cordifolia* L. were purchased from India and identified by Dr. Sang Rae Lee (Institute of Oriental Botanical Resources of Korea). A voucher sample (No. 20103) was kept in the Herbarium of Pharmacognosy Department, Faculty of Pharmacy, Assiut University, Assiut, Egypt as a reference material. The powdered roots were extracted and kept in refrigerator till used.

#### **Extraction and isolation**

The air-dried powdered root of *Rubia* cordifolia L. (20 Kg) was extracted with chloroform-methanol (1:1) till exhaustion (3x30 L). The combined extract was

evaporated till dryness. The dried extract was suspended in distilled water and extracted with chloroform to give 1890 g of chloroformic extract.

Part of the dried chloroform fraction (60 g) was chromatographed over silica gel column (2 kg), eluted with hexane-EtOAc gradiently and finally the column was washed with EtOAc-MeOH (9:1). Fractions (300 ml, each) were collected and monitored using TLC and 10%  $H_2SO_4$  as spraying reagent; similar fractions were pooled.

Fractions eluted with hexane-EtOAc (8:2) were subjected to silica gel CC using hexane-EtOAc (85:15), where sub-fractions 5-9 (containing similar spots) were rechromatographed over silica gel CC using  $CH_2Cl_2$ acetone (95:5), then final purification was done using (MPLC) RP-18 CC and methanol-water (7:3) to afford compound **1** (7 mg) and compound **2** (8 mg).

Fractions eluted with hexane-EtOAc (7:3) were subjected to silica gel CC using hexane-EtOAc (75:25) followed by RP-18 CC (MPLC) using methanol-water (7:3). Upon repeated crystallization, sub-fractions 3-8 afforded compound **3** (6 mg, methanol) while sub-fractions 10-12 afforded compound **4** (12 mg, methanol).

Fractions eluted with EtOAc-MeOH (9:1) were rechromatographed over silica gel CC using CHCl<sub>3</sub>-MeOH in a gradient elution technique. Fractions eluted with CHCl<sub>3</sub>-MeOH (9:1) were chromatographed over silica gel CC, followed by RP-18 CC (MPLC) using CH<sub>3</sub>CN-H<sub>2</sub>O (25:75), where sub-fractions (16-20) were purified by re-crystallization from methanol to afford compound **5** (11 mg).

Compound 1: yellowish-brown amorphous powder. IR (KBr): 3420, 1670, 1642, 1602 cm<sup>-1</sup>. <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) and <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>) (Table 1).

Compound **2**: white fine needles. m.p. 281-283°C. EI-MS *m*/*z* (% rel. int.): 428  $[M]^+$  (20) 413 (29), 275 (36), 248 (19), 231 (30), 220 (34), 205 (24), 177 (20), 165 (52), 125 (83), 95 (100, base peak), 69 (58) and 55 (56). <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>):  $\delta_{\rm H}$  0.84 (3H, s, CH<sub>3</sub>-25), 0.91 (3H, s, CH<sub>3</sub>-29), 0.92 (3H, s, CH<sub>3</sub>-26), 0.93 (3H, d, *J*= 8 Hz, CH<sub>3</sub>-23), 0.97 (6H, s, CH<sub>3</sub>-27 & 30), 0.98 (3H, s, CH<sub>3</sub>-24), 1.15 (3H, s, CH<sub>3</sub>-28), 3.72 (1H, br.s, H-3). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>):  $\delta_{\rm C}$  16.10 (C-1, t), 36.30 (C- 2, t),

72.97 (C-3, d), 49.38 (C-4, d), 38.05 (C-5, s), 41.94 (C-6, t), 17.77 (C-7, t), 53.41 (C-8, d), 37.32 (C-9, s), 61.56 (C-10, d), 35.56 (C-11, t), 30.86 (C-12, t), 38.59 (C-13, s), 39.89 (C-14, s), 32.55 (C-15, t), 35.77 (C-16, t), 30.24 (C-17, s), 43.03 (C-18, d), 35.40 (C-19, t), 28.40 (C-20, s), 33.03 (C-21, t), 39.50 (C-22, t), 11.85 (C-23, q), 16.62 (C-24, q), 18.47 (C-25, q), 20.34 (C-26, q), 18.88 (C-27, q), 32.01 (C-28, q), 35.25 (C-29, q), 32.31 (C-30, q).

Compound **3**: pale yellow amorphous powder. EI-MS: m/z (% rel. int.): 196 [M]<sup>+</sup> (45), 164 (100, base peak) and 136 (60). <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>):  $\delta_{\rm H}$  12.00 (1H, s, OH), 6.19 (1H, s, H-5), 3.90 (3H, s, OCH<sub>3</sub>), 2.43 (3H, s, C-6 methyl), 2.08 (3H, s, C-3 methyl). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>):  $\delta_{\rm C}$  C-1-C-6: 105.37 (s), 163.36 (s), 108.75 (s), 158.37 (s), 110.77 (d), 140.32 (s); Me-C-3: 7.88 (q); Me-C-6: 24.31 (q); COOMe: 169.76 (s), 52.01 (q).

Compound **4:** faint brown needles, m.p. 211-213°C. CI-MS m/z (% rel. int.): 169  $[M+1]^+$  (100), 148 (30), 125 (26), 79 (23), 69 (41) and 51 (20). <sup>1</sup>H-NMR (400 MHz, C<sub>5</sub>D<sub>5</sub>N):  $\delta$  8.13 (1H, dd, J= 1.9, 8.2 Hz, H-6), 8.04 (1H, d, J= 1.9 Hz, H-2), 7.31 (1H, d, J= 8.2 Hz, H-5), 3.94 (3H, s, OCH<sub>3</sub>).

Compound **5:** white crystals, m.p. 180-182°C. EI-MS m/z (% rel. int.): 194 [M]<sup>+</sup> (26), 163 (11) and 54 (100, base peak). <sup>1</sup>H-NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  3.27 (1H, t, J= 9.2 Hz, H-3), 3.58 (1H, t, J= 9.2 Hz), 3.69 (1H, dd, J= 2.4, 10.0 Hz), 3.73 (1H, dd, J= 2.4, 10.0 Hz), 3.88 (2H, m). <sup>13</sup>C-NMR (100 MHz, CD<sub>3</sub>OD):  $\delta_{\rm C}$  C-1-C-6: 72.34 (d), 70.62 (d), 83.51 (d), 72.91 (d), 71.13 (d), 72.06 (d); OMe: 59.39 (q).

## **RESULTS AND DISCUSSION**

From the chloroform soluble fraction of the chloroform-methanol extract of *Rubia cordifolia*, five compounds were isolated (1-5).

**Compound 1:** The IR spectrum of compound **1** showed the presence aromatic CH stretching, hydroxyl and ester groups (experimental part). The molecular formula was established as  $C_{34}H_{30}O_9$  from <sup>1</sup>H- and <sup>13</sup>C- including DEPT <sup>13</sup>C-NMR spectral data (Table 1). The <sup>1</sup>H-NMR spectrum showed the presence of ten protons in the aromatic region including a pair of AA`BB` type aromatic system in the region at

| С   | $\delta_{C}(m)$ | δ <sub>H</sub> (m)        | С   | $\delta_{C}(m)$ | $\delta_{\rm H}(m)$      |
|-----|-----------------|---------------------------|-----|-----------------|--------------------------|
| 1   | 158.61 (s)      | -                         | 1`  | 158.78 (s)      | -                        |
| 2   | 98.47 (s)       | -                         | 2`  | 98.34 (s)       | -                        |
| 3   | 120.93 (s)      | -                         | 3`  | 120.80 (s)      | -                        |
| 4   | 142.86 (s)      | -                         | 4`  | 142.58 (s)      | -                        |
| 5   | 119.20 (d)      | 7.76 (1H, br. d, 8.1 Hz)  | 5`  | 119.14 (d)      | 7.84 (1H, br. d, 8.0 Hz) |
| 6   | 129.53 (d)      | 7. 35-7. 43 (2H, m)       | 6`  | 129.50 (d)      | 7. 35-7. 43 (2H, m)      |
| 7   | 125.06 (d)      |                           | 7`  | 125.06 (d)      |                          |
| 8   | 124.53 (d)      | 8.27 (1H, br. d., 8.1 Hz) | 8`  | 124.30 (d)      | 8.30 (1H, br. d, 8.0 Hz) |
| 9   | 122.07 (s)      | -                         | 9`  | 122.28 (s)      | -                        |
| 10  | 125.06 (s)      | -                         | 10` | 125.06 (s)      | -                        |
| 11  | 104.12 (d)      | 6.16 (1H, s)              | 11` | 104.05 (d)      | 6.35 (1H, s)             |
| 12  | 163.38 (s)      | -                         | 12` | 160.89 (s)      | -                        |
| 13  | 72.87 (s)       | -                         | 13` | 35.74 (s)       | -                        |
| 14  | 27.67 (q)       | 1.46 (3H, s)              | 14` | 31.16 (q)       | 1.64 (3H, s)             |
| 15  | 54.66 (t)       | 2.79 d (1H, d, 14.5 Hz)   | 15` | 30.54 (q)       | 1.68 (3H, s)             |
|     |                 | 2.44 d (1H, d, 14.5 Hz)   |     |                 |                          |
| 16  | 171.43 (s)      | -                         | 16` | 171.43 (s)      | -                        |
| 17  | 51.82 (q)       | 3.56 (3H, s)              | 17` | 51.72(q)        | 3.59 (3H, s)             |
| -OH |                 | 11.93 (1H, br. s)         |     |                 | 11.87 (1H, br. s)        |

Table 1: <sup>13</sup>C- (100 MHz) and <sup>1</sup>H-NMR (400 MHz) data of compound 1 (in CDCl<sub>3</sub>).

m = multiplicity

 $\delta_{\rm H}$  7.53-8.30 (Table 1) and two singlet signals at  $\delta_{\rm H}$  6.35 and 6.16, in addition to two singlet protons in the most shielded region at  $\delta_{\rm H}$  11.93 and 11.87 for two chelated hydroxyl aromatic groups. Seventeen protons in the aliphatic region were observed as three singlet methyl signals at  $\delta_H$  1.46, 1.64, 1.68; two methoxy singlet signals at  $\delta_{\rm H}$  3.56, 3.59 and two doublets (each equivalent to one proton) at  $\delta_{H}$ 2.79 and 2.44. The identification of each proton was confirmed by proton decoupling <sup>13</sup>C-NMR spectral data The experiment. including DEPT mode measurement, revealed the presence of 30 carbon signals for 34 carbon atoms which include 10 methine, 1 methylene, 3 methyl, 2 methoxy and 18 quaternary atoms. The spectral data (Table 1) indicated that the two methoxy groups were observed at  $\delta_c$  51.72

and 51.82; the doublet signals at  $\delta_{\rm H}$  2.79 and 2.44 in the <sup>1</sup>H-NMR (each, 1H, d, J = 14.5 Hz) with  $\delta_C$  54.66 indicated the presence of geminally coupled methylene proton. The aromatic protons with their <sup>13</sup>C-NMR data in addition to the quaternary carbon atoms at  $\delta_C$ 171.43 and the two methoxy groups demonstrate the presence of two naphthohydroquinone moieties most probably C-12 substituted furomollugin<sup>32</sup>. The chemical shifts of 28 carbon atoms in the <sup>13</sup>C-NMR spectrum confirmed the presence of the two furomollugin moiety which are similar to that reported in the literature<sup>32&33</sup> while the remaining six carbons including two quaternary at  $\delta_C$  72.87 and 35.74; three methyls at  $\delta_C$ 27.67, 30.54 and 31.16 and one methylene at  $\delta_{\rm C}$ 54.66 were interpreted as shown in figure 1.

The location of the methylene group must be between the two aliphatic quaternary carbons since its presence between any of the two furomollugin groups and any of aliphatic quaternary carbons will make it resonate in the upfield region in the <sup>13</sup>C-NMR spectra<sup>46</sup>. Many naphthohydroquinones and naphthoquinone dimers were isolated from the genus Rubia and their formation was done through condensation of furomollugin or its derivatives with hydroquinone derivatives prenvlated bv different ways of condensation<sup>5,32,33,45&47</sup>. The possible biosynthetic pathway of formation of this compound can be suggested by the oxidation with concomitant allylic transposition of prenylated naphthoquinone [a] a known natural product previously isolated from family Rubiaceae which afford the reactive vinyl quinone [b] as a hypothetical biosynthetic intermediate that by acid-catalyzed or photochemical isomerization would yield furomollugin derivatives [c]<sup>5</sup>. The condensation of two furomollugin derivatives with the removal of water molecule will give compound 1 as shown in figure 2. From all the abovementioned data, compound 1 was identified as a new naphthohydroquinone dimer with the structure as shown in figure 1.

Compound 2: colourless needle crystals that gave positive colour with Liebermann-Burchard's test suggested its triterpenoidal nature. The molecular formula was deduced to be C<sub>30</sub>H<sub>52</sub>O from <sup>1</sup>H- and <sup>13</sup>C- including DEPT <sup>13</sup>C-NMR and MS spectra. The <sup>13</sup>C-NMR spectrum showed the presence of signals for 30 carbon atoms which with the DEPT <sup>13</sup>C-NMR showed the presence of eight methyl, eleven methylene, four methine, one oxymethine and six quaternary carbon atoms confirming the triterpenoidal skeleton. Comparing the <sup>1</sup>H- and <sup>13</sup>C-NMR data of the compound with similar compounds<sup>48</sup>, the data can be interpreted as follow: the <sup>1</sup>H-NMR showed the presence of one doublet methyl signal at  $\delta_{\rm H}$  0.93 (*J*= 8 Hz) ( <sub>C</sub> 11.85) which assigned for C-23; seven singlet methyl signals at <sub>H</sub> 1.15 (<sub>C</sub> 32.01); 0.98 ( c 16.62); 0.97 (2CH<sub>3</sub>) ( c 18.88 and 32.31); 0.92 ( c 20.34); 0.91 ( c 35.25) and 0.84 ( <sub>C</sub> 18.47) which assigned for C-28, C-24, C-27, C-30, C-26, C-29 and C-25, in addition to a broad singlet at  $\delta_H$  3.72 (  $_C$  72.97) which assigned for C-3. The chemical shifts were found to be close to friedelane triterpene nucleus having an oxygenated carbon<sup>48</sup>. Comparing these data with the data reported for 3- $\beta$ -friedelinol confirmed its identity<sup>49</sup>. The assignment of the structure was supported by the EI-MS, which exhibited beside the molecular ion peak at *m*/*z* 428 [M]<sup>+</sup>, other peaks at *m*/*z* 413 (29), 275 (36), 248 (19), 231 (30) and 95 (100, base peak). Based on these evidences, the structure of compound **2** was deduced as 3- $\beta$ -friedelinol (Fig. 1). Friedelinol type triterpene was previously isolated from the family Rubiaceae (*Paederia foetida*)<sup>50</sup> but this is the first report for its isolation from the genus *Rubia*.

**Compound 3:** The <sup>13</sup>C-NMR including DEPT spectra showed the presence of ten carbon signals including six quaternary, one methine, two methyls and one methoxy groups. The <sup>1</sup>H-NMR spectrum showed the presence of four singlet signals; one in the aromatic region at  $\delta_{\rm H}$ 6.19 (1H) and three in the aliphatic region at  $\delta_{H}$ 2.08, 2.43 and 3.90 (3H, each), in addition to chelated hydroxyl group at  $\delta_{\rm H}$  12.00. The aromatic proton and six carbons at  $\delta_{\rm C}$  105.37 (C), 163.36 (C), 108.75 (C), 158.37 (C), 110.77 (CH), 140.32 (C) indicated the presence of penta-substituted benzene ring; the two signals at  $\delta_{\rm C}$  7.88 and 24.31 were assigned for the two aromatic methyls and the two signals at  $\delta_{\rm C}$ 52.01 and 169.76 were assigned for the carbomethoxy group. The full identification of this compound was deduced by 2D NMR spectroscopy including HSQC and HMBC. The HMBC spectrum (Fig. 1) showed correlations between the proton at  $\delta_{\rm H}$  6.19 with carbons at  $\delta_{\rm C}$  105.37, 108.75 and 24.31; protons at  $\delta_{\rm H}$  2.43 with carbons at  $\delta_C$  110.77 and 105.37; protons at  $\delta_{H}$  2.08 with carbons at  $\delta_{C}$  163.36 and 158.37 and finally the methoxy protons at  $\delta_{\rm H}$  3.90 with the carbonyl carbon at  $\delta_{C}$  172.82. Final identification was supported by MS which showed  $[M]^+$  at m/z 196 corresponding to the molecular formula  $C_{10}H_{12}O_4$ . Comparing these data with that reported for similar compounds indicated that this compound is atraric acid which isolated from many plants as Pygeum africanum. Newbouldia laevis. Ekebergia Evernia prunastri<sup>51-54</sup>. pterophylla and Although, many phenolic acid derivatives were isolated from the genus Rubia<sup>33&55</sup>, this is the first report about the isolation of atraric acid from the family Rubiaceae.







Fig. 1: Structures of compounds (1-5) and HMBC of 3.



R= furomollugin moiety



**Compound 4:** The <sup>1</sup>H-NMR spectrum of compound 4 showed the presence of three peaks in the aromatic region at  $\delta_{\rm H}$  8.04 (1H, d, J= 1.9 Hz), 8.13 (1H, dd, J= 1.9, 8.2 Hz) and 7.31 (1H, d, J = 8.2 Hz) indicated the presence of trisubstituted benzene ring with ABX system while the singlet signal at  $\delta_{\rm H}$  3.94 (3H) suggested the presence of methoxy group. This data with the aid of MS at m/z 169  $[M+1]^+$ corresponding to the molecular formula C<sub>8</sub>H<sub>8</sub>O<sub>4</sub> and physical properties (experimental part) in addition to the reported data indicated that compound 4 is vanillic acid which was previously reported from Rubia yunnanensis<sup>55</sup> but this is the first report about its isolation from this species.

**Compound 5:** The <sup>13</sup>C-NMR data of the compound showed the presence of seven carbon signals for seven carbons including one methoxy at  $\delta_C$  59.39 and six oxygenated methine atoms resonating at  $\delta_{\rm C}$  70.62, 71,13, 72.06, 72.34, 72.91 and 83.51. The <sup>1</sup>H-NMR data revealed the absence of aromatic and presence of aliphatic groups as follow:  $\delta_{\rm H}$  3.27 (1H, t, J= 9.2 Hz), 3.58 (1H, t, J= 9.2 Hz), 3.69 (1H, dd, J= 2.4, 10.0 Hz), 3.73 (1H, dd, J= 2.4, 10.0 Hz), 3.88 (2H, m). The chemical shifts of the oxymethine groups indicated the presence of inositol moiety56 substituted with one methoxy group. Comparing the chemical shift with the reported inositol isomers<sup>46</sup> indicated that the compound is chiro-inositol substituted at C-3 with the methoxy group, thus the compound was identified as D-3-O-methoxychiro-inositol and this is the first report about its isolation from the genus Rubia.

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