SESQUITERPENE LACTONES AND FLAVONOIDS FROM INULA BRITANNICA (L.)

Omar M. A. El-Towesy

Deptartment of Pharmacognosy, Faculty of Pharmacy, Assiut University, Assiut, Egypt

هذا بالإضافة إلى ثلاثة مواد فلافونيدية هي: أبيجنين وابيجنين-٧-أ-جلوكوزيد وكورستين ٣-٤" داى ميثوكسي-٧-أ-جلوكوزيد.

From Inula britannica (L.) five known sesquiterpene lactones were isolated and identified as $4\alpha,5\beta$ epoxyeupatolide (1); $4\alpha,5\beta$ epoxydesacetylovatifolin (2); 5α -hydroxydehydroleucodin (3); 14-hydroxy-2-oxoguaia-1(10),3-dien- $5\alpha,11BH$ - $12,6\alpha$ -olide (4) and 2-oxo- $8\alpha,10\beta$ -dihydroxyguai-3-en-1- $\alpha,5\alpha,6\beta,11\beta H$ -12,6-olide (5) in addition to three known flavonoids viz. apigenin (6), apigenin-7-O- β -glucopyranoside (7) and quercetin 3',4'-dimethoxy 7-O- β -glucopyranoside (8). The structures of these compounds were depicted through spectral studies. All the isolated compounds were reported for the first time from the mentioned plant.

INTRODUCTION

Inula britannica (L.) is a perennial plant belonging to family Asteraceae. This plant is prescribed as a remedy for cough, vomiting and treatment of sore throat. Previous investigation of inula species gave in addition to some common compounds, several sesquiterpene lactones. As a part of our research program dealing with sesquiterpene lactones of compositae. We examined Inula britannica (L.) from which five sesquiterpene lactones and three flavonoids were isolated and identified.

EXPERIMENTAL

All mps are uncorrected. EIMS were recorded at 70 ev. NMR spectra were recorded at 200 and 50.3 MHz for ¹H and ¹³C, respectively. Chemical shifts are in δ value with

TMS as internal standard. TLC carried out on silica gel using solvent systems:

System I: Benzene-Acetone (8:2)

System II: Chloroform: Methanol (9.5:0.5)

System III: Chloroform:methanol (7:3)

Spots were visualized by their fluorescence at 254 nm in U.V. or by spraying with 5% H_2SO_4 (for sesquiterpenes), $AlCl_3$ (for flavonoids) and aniline hydrogen phthalate (for sugars).

Plant material

The plant material collected during the flowering stage in April 1995 from the Bontanical Garden of Faculty of Agriculture, Assiut University, identity was confirmed by Prof. Dr. A. Fayed, Professor of Taxonomy, Faculty of Science, Assiut University. Voucher specimens are deposited in the Herbarium of the Department of Pharmacognosy.

Extraction and isolation

The air-dried powdered aerial parts of the plant (2 kg) was defatted with petrol (40-60°C) followed by extraction with methanol by percolation. The concentrated methanolic extract (60 g) was fractionated with chloroform to give fraction A (20 g) and the mother liquor was concentrated under reduced pressure to give fraction B (40 g).

Fraction A was subjected to column chromatography (4x50 cm) packed with silica gel using CHCl₃ and CHCl₃-MeOH gradient for elution. The obtained subfractions were subjected to preparative TLC using silica gel plates and system II as a solvent system to obtain compounds 1 (50 mg) (R_f = 0.63), 2 (30 mg) (R_f = 0.31), 3 (10 mg) (R_f = 0.54), 4 (60 mg) (R_f = 0.38) 5 (55 mg) (R_f = 0.22) and system III to obtain compound 6 (70 mg) (R_f = 0.68).

Fraction B (40 g) was directly applied on the top of silica gel column (5x50 cm) and elution was done with CHCl₃, CHCl₃-MeOH graident to obtain compound 7 (60 mg) and compound 8 (50 mg).

Acid hydrolysis

Each of the isolated glucosides (5 mg) was dissolved in 5 ml N/2 H₂SO₄ mixed with an equal volume of water and refluxed for 2 hr. The aglycone was extracted with ether and subjected to TLC and spectral analysis. The acidic mother liquor of the hydrolysate containing the sugar moiety was neutralized, filtered and the filtrate was concentrated and spoted alongside with authentic sugar on whatmann No 1 filter paper using n-butanol-acetic acid-water (4:1:5) as solvent system.

Compound 1 (4α ,5ß epoxyeupatolide): was obtained as needles, m.p. 135-137°C (Lit. 136-138°C),⁴ IR ν (CHCl₃) cm⁻¹, 3450 (OH), 1760 (γ -lactone), 1625, 1430. HRMS m/z (rel. int.) 264.136 [M⁺] (1.3) [Calc. for C₁₅H₂₀O₄ 264.136], 246 [M-H₂O]⁺, (2.4), 161 (100).

Compound 2 (4α ,5 β -epoxydesacetylovatifolin): was occured as colourless crystals, m.p. 158-

160°C (Lit. 158-160°C),⁴ IR ν (CHCl₃) cm⁻¹, 3440 (OH), 1750 (γ -lactone), 1625, 1620. HRMS m/z (rel. int.) 280.172 [M⁺] (6) [Calc. for C₁₅H₂₀O₅ 280.175], 262 [M-H₂O]⁺, (44), 245 [M⁺-2H₂O] (64), 163 (100).

Compound 3 (5 α -hydroxy dehydroleucodin): Colorless crystals, m.p. 171-3°C (LIt. 173°C). ¹⁰ IR ν (CHCl₃) cm⁻¹, 3580 (OH), 1775 (γ -lactone), 1695, 1645. HRMS m/z (rel. int.) 260.105 [M⁺] (100) [Calc. for C₁₅H₁₆O₄ 260.104], 242 [M-H₂O]⁺ (36), 227 [242-Me]⁺. [α]²⁰_D-8 (CHCl₃ C 0.1).

Compound 4 (14-hydroxy-2-oxyguaia-1 (10), 3-dien-5 α , 11 β H-12,6 α -olide): Gum IR ν (CHCl₃) cm⁻¹, 3450 (OH), 3050, 1780 (γ -lactone), 1685, 1620. HRMS m/z (rel. int.) 262 [M⁺] (3) [Calc. for C₁₅H₁₈O₄ 262], 234 [M-CO]⁺ (62), 219 [M-CO-Me]⁺ (100). [α]²⁰_D +20 (CHCl₃ C 2.0).

Compound 52-oxo-8 α , 10 β -dihydroxyguai-3-en-1 α , 5 α , 6 β , 11 β H-12, 6 olide: Oil (Lit. oil), ¹² IR ν (CHCl₃) cm⁻¹, 3400 (OH), 1775 (γ -lactone), 1685 (C=O), 1625. HRMS m/z (rel. int.) 280 [M⁺] (3) [Calc. for C₁₅H₂₀O₅ 280], 262 [M-H₂O]⁺ (100). [α]²⁰_{β} +32 (CHCl₃ C 3).

Compound 6: Yellow amorphous powder m.p. 347-349°C (Lit. 349°C), UV (MeOH) λ_{max} nm 265, 338; + NaOMe 272, 292 sh, 390; + AlCl₃ 270, 298, 382; + AlCl₃/HCl 270, 381; + NaOAc 272, 375; + NaOAc/H₃BO₃ 268, 338. MS m/z (rel. int.) 270.121 [Calc. $C_{15}H_{10}O_5$) 270.120] (100).

Compound 7: Yellow crystals m.p. 226°C (Lit. 226°C), UV (MeOH) λ_{max} nm 270, 336; + NaOMe 270, 385; + AlCl₃ 275, 300, 385; + AlCl₃/HCl 280, 301, 345, 385; + NaOAc 270, 340; + NaOAc/H₃BO₃ 270, 340. EIMS m/z (rel. int.) 270 [M⁺] (100).

Compound 8: Yellow amorphous powder m.p. 210-212°C (Lit. 212°C), ¹⁶ UV (MeOH) λ_{max} nm 369, 325, 270 (sh), 255; + AlCl₃ 427, 357, 264; + AlCl₃/HCl 427, 357, 263; + NaOMe

413, 263; + NaOAc 409, 265; NaOAc/H₃BO₃ 365, 255. EIMS m/z (rel. int.) 330 (100), 315 (13), 301 (6); 287 (14), 259 (4), 244 (4), 165 (11), 153 (4).

All the ¹H and ¹³C-NMR spectral data for compounds 1-8 are compiled in Tables 1-4.

RESULTS AND DISCUSSION

The aerial parts of *Inula britannica* (L.) afforded five known sesquiterpene lactones (two germacranoides 1 and 2 and three guaianolides 3, 4, 5) in addition to three known flavonoids.

Compounds 1 & 2: The structures of these germacranolides were easily deduced from IR and 1H -NMR data. In each case H-5 was shifted upfield in the region typical of epoxide signals. In the HRMS, the molecular ion peak (M⁺) at m/z 264 in agreement with the molecular formula $C_{15}H_{20}O_4$ for compound 1 and at m/z 280 in agreement with molecular formula $C_{15}H_{20}O_5$ for compound 2. The spectral data of compounds 1 & 2 are in good agreement with those reported. 8.9

& 5 are in good agreement with those reported. 9-13

Compound 6: It showed physical, chemical, chromatographic characters and spectral data (UV, IR, ¹H-NMR, ¹³C-NMR and MS) which compared favorably with those of apigenin. 14-17

Compound 7: It is dihydroxy flavone glucoside

Guaianolides 3, 4 and 5: Exhibited IR

characteristic for γ lactones. The mass spectrum

showed the molecular ion peak (M⁺) at m/z 260

in agreement with the molecular formula

C₁₅H₁₆O₄ for compound 3. Peak at m/z 262 in

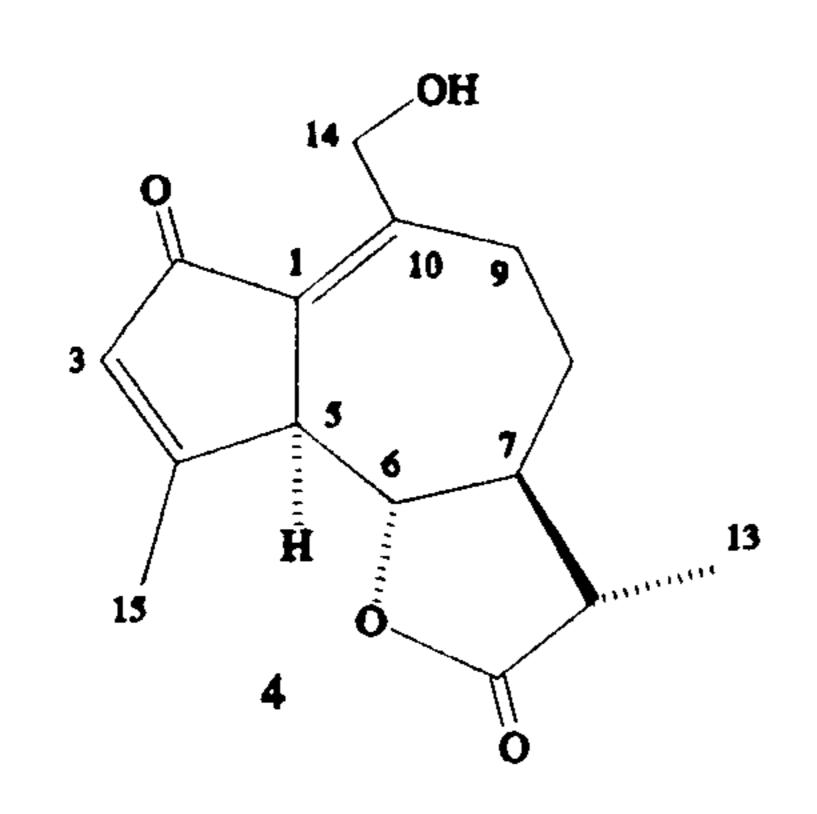
agreement with molecular formual C₁₅H₁₈O₄ for

compound 4 and at m/z 280 in agreement with

molecular formula $C_{15}H_{20}O_5$ for compound 5.

The above mentioned data for compounds 3, 4

Compound 7: It is dihydroxy flavone glucoside as indicated from UV, ¹H-NMR and ¹³C-NMR spectra (see Tables 3 & 4). Acid hydrolysis yielded aglycone which was identified as apigenin by direct comparison with authentic sample while the sugar moiety was characterized as glucose by co-chromatography with reference glucose. The position of the sugar was



	(1)	(2)	3	(4)	(5)
,	1 11 /10	7 27			2.68. d (6.7)
-	5,15, pr. ad (15,2)	7.47, UI. uu (13,4)			
2α	2.16, m	2.22, br. ddd (13,6,2)	•	•	
28		2.49, dddd	ł		
3α	1.21, ddd (13,6,1.5)	.32	6.13, q	6.19, brs	6.01, brs
38	ddd (13.5.5.	.16, ddd			
\ \	277 d (8 5)	.82, d (8	•	46,	3.2, br dd (10,6.7)
٠ <u>٧</u>	457 44 (8 5 8 5)	, dd (3.91, d (10)	96	4.32, dd (10,10)
) r	ישלים ליטיים	02, dddd	3.68, dddd	1.97, dddd	1.97, ddd (11.5,10,10)
\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	7.0, udus	46, m		8	
700	111 ()	•	1 49 m	38	4.1, ddd (10.5,10,3.5)
G o	* * / * *	76 37		7	1.72, ddd (14,10.5)
β 	ad (14,	./o, du			/
98	2.29. dd (14.5)	2.52, dd (14,5)	2.24, m	58	2.09, dd (14,5.3)
<u> </u>			1	.26	2.6, dq (11.5,7)
	6.33. d (3.5)	6.38, d (3.5)	6.17, d (3)	.26	1.45, d (7)
13,	(S) P (3)	.69. d			
\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \	1 84 hr c	, <u>†</u>		4.60, brd (13)	1.58, s
- ÷		86 4 (17)		50,	
+	-	, oo.	•) L	7 7 7
15	1.32 s	1.35, s	2.3, d (1.5)	.55,	2.3, UIS

J (paranthesis) in Hz.

Table 2: ¹³C-NMR data of compounds 1, 2, 4, 5, (50.3 MHz, CDCl₃).

Carbon	1	2	4	5
1	126.9 d	131.3 d	134.1 s	59.4 d
2	24.1 t	24.1 t	196.7 s	207.0 s
3	35.9 t	36.2 t	134.9 d	132.1 d
4	62.4 s	62.1 s	173.2 s	179.1 s
5	66.5 d	66.1 d	52.5 d	53.2 d
6	75.1 d	74.9 d	83.2 d	79.5 d
7	50.5 d	50.8 d	56.2 d	58.3 d
8	73.2 d	71.1 d	26.4 t	69.4 d
9	47.1 t	44.3 t	33.4 t	51.6 t
10	134.5 s	134.5 s	155.8 s	71.5 s
11	138.0 s	137.9 s	41.1 d	41.3 d
12	169.7 s	169.7 s	177.2 s	178.2 s
13	122.2 t	122.0 t	12.4 q	15.9 q
14	20.3 q	60.1 t	65.7 t	32.4 q
15	17.1 q	16.5 q	20.1 q	20.8 q

Table 3: ¹H-NMR spectral data of flavonoids 6-8 in DMSO-d₆.

Proton	6	7	8
H-3	7.51, s	7.5, s	
H-6	6.32, d (2)	6.35, d (2)	6.44, d (2)
H-8	6.71, d (2)	6.70, d (2)	6.86, d (2)
H-2'	7.7, dd (7,2)	7.75, dd (7,2)	7.77, d (2.2)
H-6'			7.85, dd (8.5, 2.2)
H-3'	6.8, dd (7,2)	76.82, dd (7,2)	-
H-5'			7.15, d (J = 8.5)
C ₅ -OH	12.5 (s)	12.5 (s)	12.50 (s)
C ₃ -OH			9.7 (s)
H-1"		5.1, d (7)	5.02, d (7)
Sugar protons		3.4-4.1, m	3.4-4.1 m
-OCH ₃			3.85 s
-OCH ₃			3.84 s

determined from UV spectra of both glucoside and aglycone and was found to be at position No. 7. Accordingly compound 7 was identified as apigenin-7-O-B-glucopyranoside^{14,15}.

Compound 8: Its UV absorptions indicated that it was a flavonol derivative and the ¹H and ¹³C-NMR spectra suggested that it had a monosaccharide residue. The ¹H-NMR spectra showed the presence of a set of meta-coupled

protons [δ 6.44 and 6.86 (J= 2 Hz)]. Three protons which showed an ABX spin system [δ 7.15 (J= 8.5 Hz), 7.77 (J= 2.2 Hz and 7.85 (J= 8.5, 2.2 Hz), two aromatic methoxyls (δ 3.84 and 3.85), two hydroxyl groups (9.7, 11.2)]. The above data and the EIMS spectral fragment at m/z 330 indicated that compound 8 is quercetin-3', 4'-dimethyoxy-7-O- β -glucopyranoside. 14-17

Table 4: ¹³C-NMR of flavonoids 6-8 (50.3 MHz, DMSO-d₆).

Proton	6	7	8
1			
2	164.1	164.5	146.9
3	102.8	103.2	136.5
4	181.8	181.9	176.3
5	161.5	161.5	160.3
6	98.8	99.6	98.7
7	163.5	163.1	162.7
8	94.0	95.1	94.5
9	157.3	156.9	155.6
10	103.7	105.4	104.4
1'	121.3	121.0	123.1
2'	128.4	128.5	111.4°
3'	116.1	116.5	148.3
4'	161.1	161.2	150.4
5'	116.0	116.5	110.8°
6'	128.4	128.5	121.6
1 "		100.2	100.0
2"		73.3	73.0
3"		76.6 ^b	76.4 ^b
4"		69.8	69.4
5"		77.4 ^b	77.2 ^b
6"		60.9	60.5
MeO-3'			55.5
MeO-4'			55.5

a-b Interchangeable within the same compound.

Acknowledgement

The author is grateful to Prof. Dr. Chihiro Ito, Meijo University, Japan for the spectral analysis.

REFERENCES

- 1- K. Bremer, In Asteraceae cladistic and classification, p. 273. Timber press, Portland, Oregon (1994).
- 2- L. M. Perry, T. Metzger, "Medicinal plants of east and south East Asia", Cambridge, London, p. 95 (1980).

- 3- F. Jeske, S. Huneck and J. Jakupovic, Phytochemistry 34, 1647 (1993).
- 4- B. N. Zhou, N. S. Bai, L. Z. Lin and G. A. Cordell, Phytochemistry 36, 721 (1994).
- 5- A. A. Ali, O. M. Abdallah and W. Steglich, Die Pharmazie 44 (H-11) 800-1 (1989).
- 6- A. A. Khalifa, O. M. Abdallah and M. K. Mesbah, Bull. Fac. Pharm., Cairo Univ. 29 (3), 63-66 (1991).
- 7- O. M. Abdallah, A. A. Ali and H. Itokawa, Die Pharmazie 46, 472-473 (1991).
- 8- F. Jeske, S. Huneck and J. Jakupovic, Phyochemistry, 41, 1539-1542 (1996).
- 9- H. Yoshioka, T. J. Mabry and B. M. Timmer-Manne, "Sesquiterpene lactones chemistry, NMR and plant distribution", University of Tokyo Press, Tokyo (1973).
- 10- C. Zdero and F. Bohlmann, Phytochemistry 28, 3101-3104 (1989).
- 11- F. Bohlmann and C. Zdero, Phytochemistry 18, 336 (1979).
- 12- J. A. Marco, F. J. Sanz-Cervera, E. Manglano, F. Sancenon, A. Rustalyan and M. Kardar, Phytochemistry 34, 1561 (1993).
- 13- R. X. Tan, Z. J. Jia, J. Jakupovic, F. Bohlmann and S. Huneck, Phytochemistry 30, 3033 (1991).
- 14- T. J. Mabry, K. R. Markham and M. B. Thomas, "The systematic identification of flavonoids" Spring Verlag, New York, Heidelberg (1970).
- 15- T. B. Harborne and T. J. Mabry, "The flavonoids, Advances in research", 1st ed., Chapman and Hall London, New York (1982).
- 16- Q. Xiong, D. Shi and M. Mizuno, Phytochemistry 39, 3, 723-725 (1995).
- 17- H. Yosiuoko, T. J. Mabry, M. A. Irwin, T. A. Geisman and Z. Samek, Tetrahedron Lett. 27, 3317 (1971).