

Egyptian Journal of Chemistry

http://ejchem.journals.ekb.eg/

Evaluation of Rare Earth Elements in Black Sand and Phosphate ores, EGYPT



Mahmoud S.A. Shain¹, Ebtissam A. Saad², Ahmed Shazly³, Amir Ezzat⁴, Sayed A. El-Mongy⁵*

^{1,3}Central laboratories, Almaza, Cairo, Egypt ^{2,4} Chemistry Department, Faculty of Science, Ain Shams University, Cairo, Egypt ⁵ Nuclear and Radiological Regularity Authority (ENRRA), Cairo, Egypt

Abstract

The rare earth elements (REEs), uranium (U) and thorium (Th) isotopes found naturally in soil, rock, water and plant. About 20 samples of Black sand and phosphate were collected from different locations in the Rashid city, Egypt. All the collected samples were prepared and chemically treated using different acids and microwave, then analyzed using ICP-MS. The present work was conducted to identify and evaluate the compositional trends of REEs, U and Th in the samples in terms of distribution patterns and indicator ratios. The results revealed that the black sand and phosphate ores are rich as natural sources of U, Th and other rare earth elements. The elements' concentrations in black sands and phosphate ores were found higher than the world average crust soil.

Keywords: Type your keywords here, separated by semicolons; REEs, Lanthanides, Actinides, Uranium, Thorium, Treatment, ICP-MS

1. Introduction

Rare Earth Elements (REEs) are relatively abundant in the earth's crust, but discovered economic concentrations are less common than for most other ores. REEs consist of a coherent group of elements (lanthanide series) with similar chemical properties[1]. They can be divided into light rare earth elements (LREE) which include lanthanum (La), cerium (Ce), praseodymium (Pr), neodymium (Nd), plus samarium (Sm); and Heavy rare earth elements (HREE) which include europium (Eu), gadolinium (Gd), dysprosium (Dy), holmium (Ho), erbium (Er), thulium (Tm) ytterbium (Yb), lutetium (Lu), plus yttrium(Y) [2]. The demand for rare earth elements has increased substantially owing to their unique physical ,chemical and light emitting properties [3].Some of such elements (Y, Nb, Ta, Zr, Hf, and Sc) are used in wide range of new technologies with which they are closely associated in many deposits, while others such as Nd, ,Dy, Eu, Tb, and Y are considered critical rare earth in terms of their importance to clean energy economy and the risk of supply disruption [4]. The source of REEs deposition into the environment. local in its influence, is from industrial plants

extracting and processing phosphoric raw materials, which result in geological and chemical anomalies in aerosols, soils, crops, aqueous systems and living organisms[5]. The naturally existing radio nuclides like Uranium (²³⁸U) and Thorium (²³²Th) are present everywhere in the Earth's crust[6]. The abundance of thorium is about four times that of uranium in the earth crust. The U content in soils is usually about 1 ppm, it can be as high as 8 ppm, means while Th contents are usually about 5 ppm, but can be as high as 50 ppm.

In fact, the general distribution of U in soils is very similar to that in stream sediments[7]. In addition, the detection of U in soil and water is of great interest in biological and environmental science due to its radioactivity[8].

The Egyptian black sand along the Mediterranean coast comprise huge reserves of several common economic minerals. The Rosetta black sands have been the subject of many articles and dissertations, as well as technical reports by private firms[9].Beach sand deposits are mostly rich sources of industrial minerals [10].

Phosphate rocks of sedimentary origin contain ²³⁸U and ²³²Th and its decay products in addition to phosphate minerals .Considerable variations are found in the chemical composition of rock phosphate from different mining areas. In general, sedimentary

*Corresponding author e-mail: <u>mahmoud_44493@yahoo.com</u>.

Receive Date: 03 January 2020, Revise Date: 18 March 2020, Accept Date: 18 March 2020 DOI: 10.21608/EJCHEM.2020.21042.2305

^{©2020} National Information and Documentation Center (NIDOC)

phosphate rocks, or phosphorites, originated in a marine environment, are characterized by activity concentrations of U much higher than those of volcanic and biological rocks [11].Phosphorus fertilizers contain varying amounts of heavy metals and other REEs as contaminants from either phosphate rock ores or other ingredients used in the phosphate fertilizer industry[12]. As some heavy metals are potentially unsafe to human health, attention is being given to its way of entry into the human food chain. Uptake of such elements by plants consumed directly or indirectly by humans is one path of entry, so the effects of heavy metal contaminants in phosphate fertilizers are of concern[5, 13]. In this study destructive technique ICP-MS is selected to be used to determine the concentrations of U, Th and REEs. ICP-MS technique is an effective device for its high accuracy of the detection of the trace elements. The aim from this work can be given as follows: i) quantification of REE, Th, and U in black sands and phosphate ores using ICP-MS, and ii) obtain the baseline information about the content of REE, Th, and U in black sands and phosphate ores.

2. Experimental work

2.1. Materials and Methods

All reagents and used materials were of high analytical grade (Merck PA, Accustandard® and Sartorius®).All samples under investigation have been collected from different locations in Egypt as follows: Ten samples of black sand from different locations along the Rashid city at the Mediterranean seacoast, they represent the raw black sand in 1.5 km² stretch and the width is variable from few meters up to 70 meters and ten phosphate samples were collected from Abu El Mahameed area. These samples were air dried and sieved on a 2-mm mesh nylon sieve and then ground in an agate mortar and passed through a stainless steel 0.3-mm mesh sieve to homogenize them. one g of sample was digested in Teflon vessels with 4 ml of 65 % nitric acid (HNO_3) and 4 ml of 48 % hydrofluoric acid (HF) in a microwave oven[14]. The extracts were transferred to 25-ml certified flasks (NBR ISO/IEC) filled with ultrapure water (Millipore Direct-Q System) and filtered in filter paper (Sartorius®).

A Four-point calibration curve for the U, Th and REEs from a standard solution containing 10 μ g/ml multi-element calibration standard (Accustandard ®) was prepared. Then analysis with the coefficient of determination (r2) of the calibration curve more than 0.990 was performed. Analytical data quality and standard process measures such as curve recalibration, analyses of blanks for the acid digestion

Egypt. J. Chem. 63, No. 11 (2020)

procedure; were conducted. Based on this method of digestion the collected samples are verified and validated by using IAEA- 313 and IAEA -375 soil standard reference material (SRM).All of the analyses were carried out in duplicate using ICP-MS[15, 16].

2.2. ICP-MS setup:

In this study destructive technique ICP-MS (Agilent 7700) is calibrated to determine the concentrations of U, Th, Sc, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Yb, Lu, Dy, Er, Ho, Tb, and Tm in black sands and phosphate samples using ICP-MS. The most important operating standard tune parameters and plasma Radiofrequency ICP-MS is given in Table 1.

Table .1. Operating conditions and ICP-MS settings:

	-	1	-			
ICP-MS operati	ng conditions	Mass Spectrometer acquisition setting				
plasma		mass gain	139			
radiofrequency	1550 W					
power		22				
plasma		mass offset	127			
radiofrequency	1.8 V					
power matching						
carrier gas	0.99 L/min	Dwelltime	50ms			
Tourch avis	-0.3 H and -0.5	Sweeps per reading	20			
1 Outen axis	V					
Interface cones	Nickle	Replicates	3			
Lense setting	Auto Lense	Reading per peak	1			
electron		Scan mode	Peak			
multiplier (EM)	4.5 mV		hopping			
discriminator			rr8			
analog	2665V	Detector mode	dual			
pulse			1666 V			
plasma carrier gas o	ffset		-0.06L/min			
-						

3. RESULTS AND DISCUSSION:

3.1. Uranium and thorium in black sand samples:

Table (2) shows the concentration of uranium, thorium in different Black sand samples collected from different places in the Rashid area in Egypt .The average concentration of U and Th were 18.62ppm and 82.67 ppm, respectively. The average concentrations are higher than the average world crust soil especially for thorium.

The concentration ratios in black sand samples to the world average crustal soil varied as, U: 4.9 ppm and Th: 5.3ppm[17]. The black sand contains more Th (82.67 ppm) than U (18.62ppm). The concentration of Uranium and Thorium decreases as we approach the beach starting from sample (BS1) this may be due to the effect of water currents as shown in Figure.1.



Figure 1 .Variation of concentrations for Th and U of beach black sand samples.

3.2. REEs in Black sand samples:

The rare earth elements are significant tracers and used in modeling of assorted geochemical samples, processes. In our the average concentrations of Sc, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Th, Yb and Lu were found 80.84, 42.95, 37.95, 88.44, 7.84, 29.607.05, 1.69, 8.41, 1.41, 8.29, 1.73, 5.57, 6.44 and 1.1 ppm, respectively as shown in table (3). We also quantified the fractionation of LREEs and HREEs according to the La_N/Sm_N and Gd_N/Yb_N ratios, respectively, to interpret the REE patterns. Cerium [Cerium [Ce_N / $(La_N * Pr_N)^{0.5}$] Eu anomalies [Eu_N $/ (\text{Sm}_{\text{N}} * \text{Gd}_{\text{N}})^{0.5}$] were calculated [18], where N implies normalized values .

The average ΣREE concentrations in black sand were higher than the values reported for Brazil, Japan, China, Swedish, Europe, and the Earth's crust. As well, the average $\Sigma HREE$ and $\Sigma LREE$ concentrations were much higher in black sand than elsewhere except world earth crust. All of the REEs U and Th exhibited concentrations higher than those found in the average Earth's crust except for Nd, Pr and Eu. The experimental data for elemental content in our samples were compared with the similar black sand beach in other studies using instrumental neutron activation analysis (INAA) as shown in (Table 4).

Rare earth element' concentrations tend to decrease with increasing atomic number according to the Oddo Harkins rule [19]: Ce > Nd >La > Pr > Sm > Gd > Dy > Er > Yb > Eu > Tb > Ho > Tm > Lu. In general, the average REE concentrations in this study exhibited a comparable ordering: Ce > La > Nd > Gd > Dy >Pr >Sm > Yb > Er > Tb > Ho > Eu >Lu > Tm . The concentrations distribution of REE Table .2: Concentrations of uranium and thorium in Black sa elements in black sand samples are shown in Figure. 2(a,b).



Figure 2. Histogram illustrates the distribution of LREEs and HREEs in beach black sand samples

Figure. 3 shows the proportions, in percent, of the concentrations of REE, LREE, and HREE in the studied area. LREEs comprised approximately 81% and HREEs 19% of the total REE concentrations.



Figure 3. Histogram illustrates the proportions of REE in Black sand samples.

	•	1110	concent	actions a	bulloution	01				
Tabl	e .2:	Conc	entrations	of uranium	and thorium	in	Black sand	sample	in ppm:	
							0	4		

		Concentrations (ppm)										
element	BS1	BS2	BS3	BS4	BS5	BS6	BS7	BS8	BS9	BS10	Ave	
U	12.3	29.23	19.13	23.42	17.62	16.54	25.32	16.54	14.02	8.5	18.62	

Egypt. J. Chem. 63, No. 11 (2020)

M.S.A.Shahin.	et.al.
---------------	--------

Th	62.23	132.11	91.02	90.01	85.21	78.11	105.01	71.74	69.12	42.21	82.67
a BS black sand sample are ordered according to the latitude. b ppm : part per million											
Table 3: E	le me ntal co	mposition	determine	ed in beac	h black sa	nd sample	s in ppm				
concentrations (ppm)											
element	BS1	BS2	BS3	BS4	BS5	BS6	BS7	BS8	BS9	BS10	Ave
Sc	109	100	131	121	120	118	2.65	146	55.3	14.5	80.84
Y	19.3	61.7	49.2	57.4	45.4	47.7	3.21	53.3	55.3	56.3	42.95
La	15.4	46.6	27.5	43	32.3	39	33.77	32.6	41.9	75	37.16
Ce	36.1	69	55.8	90	89	94	82.27	64.4	176	164	88.44
Pr	4.17	10.4	6.83	10.7	7.89	9.59	7.08	7.8	9.86	8.31	7.84
Nd	17.1	41.6	28.3	41	29.5	35.2	28.35	31.2	38.9	22	29.60
Sm	3.7	8.57	6.4	8.71	6.2	7.37	4.85	6.6	8.19	13.7	7.05
Eu	1.27	2.16	1.82	1.94	1.45	1.74	1.2	1.83	1.75	3.02	1.69
Gd	3.99	9.09	7.02	11.6	9.4	11.4	8.82	7.42	8.92	10.39	8.41
Tb	0.59	1.49	1.2	1.61	1.23	1.42	1.13	1.23	1.41	1.45	1.41
Dy	3.78	10	8.22	9.65	7.27	8.07	6.83	8.36	9.08	15.5	8.29
Но	0.83	2.31	1.84	2.12	1.64	1.75	1.17	1.93	2.01	2.57	1.73
Er	2.62	7.74	6.08	6.91	5.39	5.64	4.49	6.35	6.45	6.75	5.57
Tm	0.42	1.33	1.02	1.15	0.9	0.91	0.87	1.11	1.05	0.85	0.91
Yb	3.21	10.51	7.86	9.01	7.11	7.06	0.5	8.68	8.13	5.77	6.44
Lu	0.57	1.87	1.32	1.56	1.24	1.19	0.08	1.56	1.34	0.88	1.10
∑REE	93.75	222.6	161.0	238.9	200.5	224.3	181.4	181.0	315.0	330.1	205.7
∑LREE	77.74	178.3	126.6	195.3	166.3	186.9	157.5	144.4	276.6	286.0	171.8
∑HREE	16.01	44.33	34.36	43.6	34.18	37.44	23.89	36.64	38.42	44.16	33.90
Ce/Ce*	4.50	3.13	4.07	4.20	5.58	4.86	5.32	4.04	8.66	6.57	6.57
Eu/Eu*	0.33	0.24	0.27	0.19	0.19	0.19	0.18	0.26	0.20	0.25	0.25
Eu/Sm	0.343	0.252	0.284	0.222	0.233	0.236	0.247	0.277	0.213	0.220	0.239
Ce/La	2.344	1.480	2.029	2.093	2.755	2.410	2.436	1.975	4.200	2.186	2.379
La/Yb	4.797	4.438	3.580	4.772	4.542	5.524	67.54	3.755	5.153	12.99	5.767

a To calculate Ce/Ce and Eu/Eu* (McDonough and Sun, 1995).* Table. 4: Descriptive statistics of the concentrations of REE, \sum REE, \sum LREE and \sum HREE of black sand (in ppm) the obtained results are compared with national and international soil data and Earth crust.

Element	mean	SD	median	min	max	P value	Egypt Black sand	Brazil	Japan	China	Swedish	Europe	Earth's crust
U	18.262	6.26	17.08	8.5	29.23	0.000	16.32	ND	ND	ND	ND	ND	3.3
Th	82.6771	24.61	81.66	42.21	132.11	0.000	78.84	ND	ND	ND	ND	ND	12.1
La	38.707	15.56	36.385	15.4	75	0.305	29.4	20.8	18.0	37.4	17.4	25.9	35.0
Ce	92.057	44.82	85.635	36.1	176	0.009	44.1	43.5	40.0	64.7	37.7	52.2	66.0
Pr	8.263	1.98	8.1	4.17	10.7	0.047	ND	9.61	4.53	6.67	4.10	6.02	9.10
Nd	31.31	8.04	30.35	17.1	41.6	0.083	35	17.7	18.0	25.1	15.1	22.4	40.0
Sm	7.429	2.72	6.985	3.7	13.7	0.030	31.32	3.37	3.74	4.94	2.98	4.28	7.00
Eu	1.818	0.52	1.785	1.2	3.02	0.051	7.52	0.600	1.03	0.980	0.650	0.850	2.10
Gd	8.81	2.26	9.03	3.99	11.6	0.124	8.54	2.32	3.72	4.38	3.07	4.20	6.10
Tb	1.275	0.28	1.315	0.59	1.61	0.593	3.625	0.500	1.22	0.580	0.490	0.640	1.20
Dy	8.676	2.97	8.29	3.78	15.5	0.253	ND	0.930	3.33	3.93	2.95	3.58	4.50
Ho	1.816	0.51	1.885	0.83	2.57	0.289	ND	0.17	0.71	0.830	0.610	0.720	1.30
Er	5.441	1.31	5.69	2.62	6.9	0.206	ND	0.59	2.02	2.42	1.88	2.10	3.50
Tm	0.959	0.24	0.955	0.42	1.33	0.036	ND	nd	0.29	0.24	0.36	0.33	0.33
Yb	6.765	2.95	7.395	0.5	10.5	0.124	31.12	0.670	2.01	2.32	2.01	2.09	3.10
Lu	1.161	0.53	1.28	0.08	1.87	0.174	3.3	0.050	0.34	0.35	0.300	0.300	0.800
∑LREE	179.589	63.35	172.335	77.74	286.03	0.273	147	95.6	85.2	140	78.0	112	159
∑HREE	34.903	8.91	36.04	16.01	44.16	0.0329	46	12.0	13.2	14.8	11.3	13.6	51.5
∑REE	214.492	70.09	210.59	93.75	330.19	0.014	193	108	98.4	155	89.3	125	211

4188

EVALUATION OF RARE EARTH ELEMENTS IN BLACK SAND.....

References	This study	A. Ali et Silva et Yoshida Wei et Sadeghi Sadeghi al. (2018) al. et al. al. et al. et al. Olsson (2016) (1998) (1991) (2013) (2013) (2022)
------------	------------	--

3.3. Uranium and thorium in phosphate samples:

Most Phosphate rocks contain between 30 and 200 ppm U and less than 10 ppm Th [20] .Table (5) shows the concentrations of U and Th in phosphate samples. The average concentrations of U and Th were 127.26 and 22.94 ppm, respectively. These concentrations are higher than the average world crust soil [17, 20] especially for U (10%).

In the present study, the average concentrations of Sc, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Th, Yb and Lu were found 4.24, 53.81, 39.80, 58.74, 7.43, 31.13,5.80, 1.691.55, 6.9.2, 1.14, 6.75, 1.7,1.46, 4.75, and 0.84 respectively as shown in table (6).

Figure 4. Shows the proportions, in percent, of the concentrations of REE, LREE, and HREE in the studied area. LREEs comprised approximately 84% and HREEs 16% of the total REE concentration. The average Σ REE, Σ LREE and Σ HREE concentrations in phosphate samples were lower than the values reported for the Earth's crust.

The average concentrations of U and Th were higher than the values reported for the Earth's crust Table (7).on the other hand, the average concentrations of ΣREE , $\Sigma HREE$ and $\Sigma LREE$ were lower than the world earth crust. Other international studies for estimation of such elements in phosphate ores are scarce and conducting using different techniques [20-22].The previous studies in Egypt evaluated the radioactivity of phosphate using instrumental neutron activation analysis[23-25].



Figure 4. Histogram illustrates the proportions of REE in phosphate samples.

4. CONCLUSION

In general, the Egyptian black sand contains several economic minerals. It is a promising source of REEs: Th and U. Based on this work, the baseline information concerning the content of Th, and U in the analyzed black sand and phosphate ores are obtained. It was observed that the average concentrations of U and Th in the samples were found to be higher than the average world level (4.9, 5.3%). The ratios of Eu / Sm and Ce/La values are almost the same as that presented in the literature, which confirms the precision of the ICP-MS analytical techniques applied in this work. The black sands in Rashid city are not recommended to be used in building materials due to the high content of Th and U. The measured REEs elements have many applications in medical, biological, industrial and nuclear fields.

5. Conflicts of interest

There is No conflict of interest.

					concer	ntrations(p	opm)				
Element	P1	P2	Р3	P4	P5	P6	P7	P8	P9	P10	Ave
U	139.92	144.01	145.2	136	136.89	141.24	148.34	144	137.24	139.75	127.26
Th	25.213	25.33	26.34	21.03	23.57	27.34	28.23	29.34	21.3	26.94	22.942
Table .6	Table .6: Concentrations of REEs in Phosphate samples.										
Elements concentrations(ppm)										01/0	
Elements	P1	P2	Р3	P4	P5	P6	P7	P8	P9	P10	ave
Sc	5.32	2.74	2.58	3.98	7.56	6.32	5.22	2.78	2.88	3.08	3.7148
Y	70.69	40.10	52.55	50.87	43.08	67.11	70.69	50.10	42.55	50.40	46.745
La	49.78	33.02	35.33	39.89	32.13	50.65	49.87	33.02	35.33	38.89	34.813
Ce	69.92	44.69	69.21	51.67	40.66	74.76	69.92	44.69	69.21	52.71	51.752
Pr	8.44	5.55	8.04	6.34	8.20	9.19	8.44	5.55	8.04	6.59	6.594
Nd	35.52	23.49	32.23	26.01	36.22	38.58	35.52	23.49	32.23	28.01	27.578
Sm	6.81	4.61	6.58	5.42	3.61	7.56	6.81	4.61	6.58	5.47	5.125
Eu	1.92	1.29	1.57	1.34	1.01	2.08	1.92	1.29	1.57	1.52	1.359
Gd	8.31	5.66	7.15	5.46	6.20	8.88	8.31	5.66	7.15	6.49	6.096
Tb	1.21	0.82	1.18	0.91	1.86	1.31	1.21	0.82	1.18	0.93	1.022
Dy	7.73	5.27	7.8	5.93	5.86	8.14	7.73	5.27	7.80	5.98	5.978
Но	1.74	1.21	1.71	1.31	0.87	1.80	1.74	1.21	1.71	1.34	1.29
Er	5.33	3.73	5.51	4.33	4.53	5.41	5.33	3.73	5.51	4.10	4.217
Tm	0.75	0.53	0.91	0.68	0.38	0.76	0.75	0.53	0.91	0.58	0.603
Yb	5.12	3.63	6.95	3.76	2.47	5.19	5.12	3.63	6.95	3.96	4.166
Lu	0.83	0.59	1.11	0.63	1.31	0.82	0.83	0.59	1.11	0.65	0.764
∑REE	203.5	134.09	185.28	153.67	145.31	215.13	203.5	134.09	185.28	157.2	151.3
∑LREE	172.4	112.6	152.9	130.6	121.83	182.8	172.4	112.65	152.96	133.1	127.2
$\overline{\Sigma}$ HREE	31.02	21.44	32.32	23	23.48	32.31	31.02	21.44	32.32	24.03	24.13
Eu/Sm	0.281	0.279	0.238	0.247	0.279	0.275	0.281	0.279	0.238	0.27	0.265
Ce/La	1.40	1.353	1.958	1.295	1.265	1.476	1.40	1.353	1.958	1.3	1.48
La/Yb	9.740	9.096	5.083	10.609	13.0081	9.759	9.740	9.096	5.083	9.820	8.3567
Table. 7	:Descript	ive statist	ics of the	concent	rations of	REE, ∑R	EE, ∑LRÌ	EE and \sum	HREE of	phospha	te samples
(in ppm)) the obta	ined resul	ts are cor	npared wi	th the Ear	th crust.					_
Eler	nent	mean		SD	mediar	n min	ma	ax	P value	Ear	th's crust
Ţ	J	73.26		4.05	80.58	76.00	88.	34	0.000		3.3
Т	ĥ	22.94		2.79		21.03	29.	29.34			12.1
T	я	34.81		7 54	37.11	32 13		88.34	0.305		35.0
	μ 	51 75		13.06	60.96	40.66	50	65	0.009		66 0
P	r Pr	6 59		1 31	8 04	5 55	74 74	50.05 74.76			9 10
N	I Id	27 57		5 53	32 23	23.49	,	9	0.047		40.0
S	m	5 12		1 26	6.03	3.61	38	58	0.030		7 00
E	'n	1.35		0.34	1.55	1.01	7.5	56	0.051		2.10
G	d	6.09		1.24	6.82	5.46	2.0)8	0.124		6.10
T	b	1.02		0.31	1.18	0.82	8.8	38	0.593		1.20
D	v	5.97		1.18	6.86	5.27	1.8	36	0.253		4.50
Н	0	1.29		0.32	1.53	0.87	8.1	4	0.289		1.30
F	r	4.21		0.75	4.93	3.73	1.8	30	0.206		3.50
T	m	0.60		0.17	0.72	0.38	5.5	51	0.036		0.33
Ŷ	Ъ	4.16		1.47	4.54	2.47	0.9	91	0.124		3.10
L	u	0.76		0.25	0.83	0.59	6.9	95	0.174		0.800
ΣR	EE	151.35		30.44	171.25	5 134.0	9 1.3	31	0.273		211
ΣL	REE	127.22		25.90	143.08	3 112.6	5 215	.13	0.0329		159
Σ́HI	REE	24.13		4.89	27.53	21.44	182	.82	0.014		51.5
Refer	ences				This	study				Tyle	r and Olsson
						-					(2002)

Table .5: Concentrations of Uranium and thorium in Phosphate samples

6. References

- 1. Henderson, C.R., Applications of linear models in animal breeding. Vol. 462. 1984: University of Guelph Guelph.
- Goodenough, K., et al., Europe's rare earth element resource potential: An overview of REE metallogenetic provinces and their geodynamic setting. Ore Geology Reviews, 2016. 72: p. 838-856.
- Eliseeva, S.V. and J.-C.G. Bünzli, Rare earths: jewels for functional materials of the future. New Journal of Chemistry, 2011. 35(6): p. 1165-1176.
- Goodenough, K.M., F. Wall, and D. Merriman, The rare earth elements: demand, global resources, and challenges for resourcing future generations. Natural Resources Research, 2018. 27(2): p. 201-216.
- Aydin, I., et al., Hazardous metal geochemistry of sedimentary phosphate rock used for fertilizer (Mazıdag, SE Anatolia, Turkey). Microchemical Journal, 2010. 96(2): p. 247-251.
- 6. Ojovan, M.I., W.E. Lee, and S.N. Kalmykov, An introduction to nuclear waste immobilisation. 2019: Elsevier.
- Boyle, R.W., Geochemical prospecting for thorium and uranium deposits. Vol. 16. 2013: Elsevier.
- 8. Clark, I.D. and P. Fritz, Environmental isotopes in hydrogeology. 2013: CRC press.
- 9. Mahmoud, H., A. Abdel-Lateef, and A. Attiah, Distribution of some elements in the Egyptian black sands from Abu Khashaba beach area. 2013.
- Laznicka, P., Giant metallic deposits: Future sources of industrial metals. 2006: Springer Science & Business Media.
- 11. Sahu, S., et al., Natural radioactivity assessment of a phosphate fertilizer plant area. Journal of Radiation Research and Applied Sciences, 2014. 7(1): p. 123-128.
- Mar, S.S. and M. Okazaki, Investigation of Cd contents in several phosphate rocks used for the production of fertilizer. Microchemical Journal, 2012. 104: p. 17-21.
- 13. Abdel-Haleem, A., et al., Heavy metals and rare earth elements in phosphate fertilizer

components using instrumental neutron activation analysis. Applied Radiation and Isotopes, 2001. 55(4): p. 569-573.

- Chen, M. and L.Q. Ma, Comparison of four USEPA digestion methods for trace metal analysis using certified and Florida soils. Journal of Environmental Quality, 1998. 27(6): p. 1294-1300.
- Mihaylova, V., B. Todorov, and R. Djingova, Determination of uranium and thorium in soils and plants by ICP-MS. Case study of Buhovo region. Comptes rendus de l'Académie bulgare des Sciences, 2013. 66(4).
- 16. Ivanova, J., et al., On the microwave digestion of soils and sediments for determination of lanthanides and some toxic and essential elements by inductively coupled plasma source mass spectrometry. Talanta, 2001. 54(4): p. 567-574.
- Viers, J., B. Dupré, and J. Gaillardet, Chemical composition of suspended sediments in World Rivers: New insights from a new database. Science of the total Environment, 2009. 407(2): p. 853-868.
- Banks, C.E. and R.G. Compton, Ultrasonically enhanced voltammetric analysis and applications: an overview. Electroanalysis: An International Journal Devoted to Fundamental and Practical Aspects of Electroanalysis, 2003. 15(5-6): p. 329-346.
- 19. Laveuf, C. and S. Cornu, A review on the potentiality of rare earth elements to trace pedogenetic processes. Geoderma, 2009. 154(1-2): p. 1-12.
- 20. Menzel, R.G., Uranium, radium, and thorium content in phosphate rocks and their possible radiation hazard. Journal of Agricultural and Food Chemistry, 1968. 16(2): p. 231-234.
- Binnemans, K., et al., Towards zero-waste valorisation of rare-earth-containing industrial process residues: a critical review. Journal of Cleaner Production, 2015. 99: p. 17-38.
- 22. Papastefanou, C., et al., The application of phosphogypsum in agriculture and the radiological impact. Journal of

Environmental Radioactivity, 2006. 89(2): p. 188-198.

- Allam, K.A., et al., Analysis and statistical treatment of 238U series isotopic ratios using gamma-ray spectrometry in phosphate samples. Radiation Protection and Environment, 2017. 40(3): p. 110.
- 24. Salman, A., et al., A comparative study for 235U radioactivity concentration calculation methods in phosphate samples. Radiation Protection and Environment, 2019. 42(1): p. 5.
- Hassan, N.M., et al., Ele-mental analysis of Egyptian phosphate fertilizer components samples by TGA, DTA and IR methods. IOSR J. Environ. Sci. T oxicol. Food Technol.(IOSR-JESTFT), 2013. 7: p. 98-107.